

# IDAHO NATIONAL ENGINEERING LABORATORY

GEOLOGICAL ASPECTS OF AN  
ASSESSMENT OF THE NATIONAL  
POTENTIAL FOR NON-ELECTRICAL  
UTILIZATION OF GEOTHERMAL RESOURCES

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NON-ELECTRICAL UTILIZATION OF GEOTHERMAL RESOURCES

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AEROJET NUCLEAR COMPANY  
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## PREFACE

The work included in this report was performed by the Geology Department of Boise State University and was coordinated by the Geothermal Projects Branch of Aerojet Nuclear Company, prime contractor to the Energy Research and Development Administration at the Idaho National Engineering Laboratory.

Financial support for this work was supplied through funds assigned by the Energy Research and Development Administration. These funds were made available to collect and assemble existing geological, geophysical and geochemical data relative to geothermal resources in the Western third of the United States.

The information presented in this report has been made an integral part of the "National Program Definition Study for the Non-Electrical Utilization of Geothermal Energy," ANCR-1214, prepared by the Geothermal Projects Branch of Aerojet Nuclear Company.

This report is being issued separately since it contains a wealth of vital information never before collected in one place. It is anticipated that the data presented herein will be of value to many government and non-government enterprises engaged in geothermal activities.

The staff of the Aerojet Nuclear Company Geothermal Projects Branch wishes to thank the contributors to this report, particularly, Dr. K. M. Hollenbaugh, Dr. C. R. Nichols, Dr. J. K. Applegate and their cohorts for their fine work and for this valuable contribution.

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## ABSTRACT

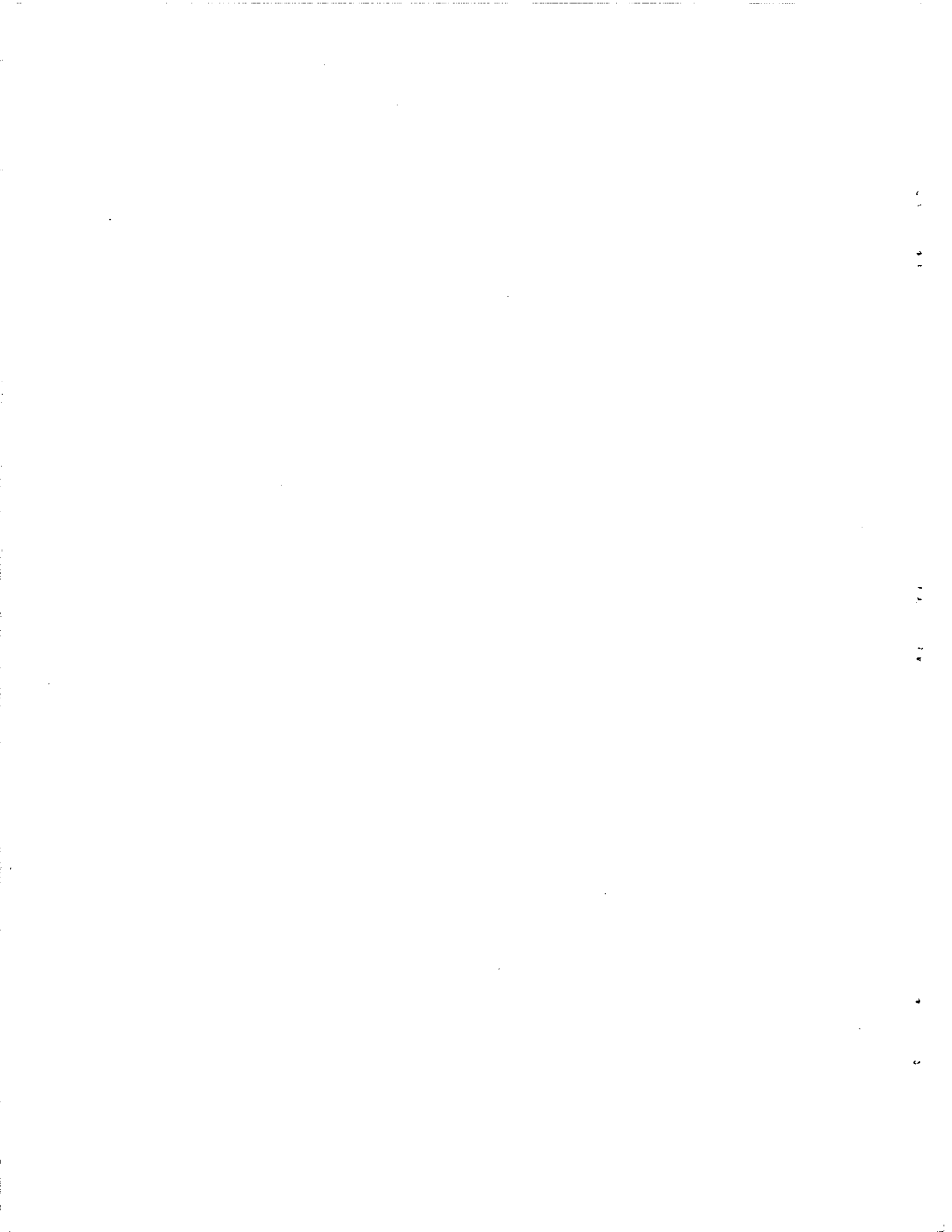
This report presents exhaustive data on the potential sources and uses of geothermal water particularly in the Western third of the United States. This data includes the currently known or expected location, quality, quantity, and temperatures of geothermal sources. Included are discussions of the more important sources. Recommendations are made relative to potential utilization of these sources. In addition, suggestions are presented for the scientific evaluation of these and other potential geothermal sources.

## ACKNOWLEDGEMENTS

The preparation of this report was a group effort. Dr. K. M. Hollenbaugh was the principal investigator and coordinated the effort at Boise State University. Dr. C. R. Nichols was instrumental in writing of the narrative and the presentation of background data and interpretations. Two students, Charles Bryan and Ann Cochrane, spent much time drafting maps and collecting references. The manuscript was typed by Kay Johnson. The principal investigator prepared the state-by-state summaries, coordinated the total effort, and assumes full responsibility for any inaccuracies or omissions that may have escaped detection. Finally, grateful appreciation is extended to the many authors and agencies who have cooperated so freely in providing data included in the report.

## TABLE OF CONTENTS

ACKNOWLEDGEMENTS	
INTRODUCTION	1
Previous Investigations	3
Objectives of the Study	4
ASSESSMENT OF GEOTHERMAL EVALUATION TECHNIQUES	5
Evaluation Criteria Applied by the Federal Gov't.	5
Geologic Environment	7
Geochemical Techniques	9
Geophysical Techniques	12
GEOGRAPHIC DISTRIBUTION OF THE GEOTHERMAL RESOURCE	13
Conclusion	13
Data Presentation	14
State Summaries	15
RECOMMENDATIONS FOR FURTHER STUDY	19
SELECTED REFERENCES	20
APPENDICES	
I. State Geothermal Discussions	21
II. Map Classification of Areas by Temperature Range, Water Quality and Boron Content	204
III. State Tectonic Maps (in envelope)	237





# GEOLOGIC ASPECTS OF AN ASSESSMENT OF THE NATIONAL POTENTIAL FOR NON-ELECTRICAL UTILIZATION OF THE GEOTHERMAL RESOURCE

## INTRODUCTION

The geologic assessment of the potential of the non-electrical utilization of the resource must consider the whole question of natural hot water origins, occurrences and characteristics as opposed to the rather narrow portion of the total resource which may be economically utilized at present for electrical power generation. As compared with the relatively rare coincidence of geologic factors necessary for economic geothermal power production utilizing presently available technology, the portion of the geothermal resource available for non-electrical utilization actually constitutes the major portion of the total geothermal picture. In a very real way, the non-electrical resource may be considered the bulk of the "iceberg" obscured beneath the surface, and the relatively rare high temperature brine or dry steam systems remain the very visible but volumetrically insignificant part of the resource.

There are two major types of geothermal systems as defined in terms of the nature of the fluid phase existing in the thermal reservoir (White, Muffler & Truesdell, 1971). In "vapor dominated" systems, a steam phase is present within the reservoir and may be produced directly for turbine generation of electrical power. "Hot water" systems may produce a steam-water mixture, due to boiling during fluid production, but water is the dominant reservoir fluid. For the purpose of this report, systems are defined in terms of their reservoir temperatures: low temperature,  $< 100^{\circ}\text{C}$ ; intermediate temperature,  $100\text{-}180^{\circ}\text{C}$ ; and high temperature,  $> 180^{\circ}\text{C}$ .

White (1970) has estimated that only one out of twenty geothermal systems are of the dry steam or vapor-dominated type. Both actual drilling of geothermal zones and chemical techniques for differentiating between wet and dry systems

are tending to confirm White's prediction. Likewise, the very high temperature brines, such as those produced in the geothermal fields of Cerro Prieto, Mexico, and Imperial Valley, California, are relatively rare in the wet type of geothermal systems as compared to the much more common systems which have base temperatures below 200°C. The majority of the geothermal systems under investigation in the western U.S. seem to have base temperatures clustered between 100 and 180°C. These low to intermediate temperature geothermal systems constitute the majority of the nation's known geothermal resource areas (KGRA's). When the number of these known geothermal resource areas with observed or predicted temperatures of 100°C or greater are compared with the number of lower temperature hot spring areas, the true preponderance of the lower temperature portions of the geothermal resource is clearly understood. The geologic assessment of the nation's non-electrical geothermal resource must then avoid the human tendency to concentrate on the higher temperature or even the intermediate temperature resources to the exclusion of the volumetrically more impressive and geographically more significant low temperature portions of the resource. Economic considerations already make it evident that the thermal water effluent produced as a by-product of either higher temperature geothermal or non-geothermal power production will ultimately be utilized. The utilization of the low temperature portion of the geothermal resource will, however, require some special attention to its unique geologic and hydrologic characteristics which will affect its potential for development.

The true assessment of the potential for non-electrical geothermal resource utilization must also take into consideration the thermal fluid resource which will be encountered routinely on a truly national scale as oil and gas exploration and production drilling reaches to greater and greater depths within sedimentary basins. Just as development of the Geysers area heralded the beginning of a geothermal rush in the western states, the geopressurized zones encountered at relatively shallow depths in the Gulf Coast states may be merely the forerunner

to discoveries and development of thermal fluids at greater depths in other sedimentary basins.

### Previous Investigations

Previous assessments of the geothermal resource relevant to its non-electrical utilization include the cataloging of thermal waters by Waring (revised, 1965). This study presents specific data as to individual hot spring locations, observed temperatures and flow rates, but lacks any comprehensive cataloging of geochemical data for the thermal water occurrences. The lists of thermal water occurrences by Waring are, however, relatively complete and very few major natural hot water occurrences were omitted.

Designations of known geothermal resource areas (Godwin, et al., 1971) and a growing number of statewide surveys sponsored by individual state governments and the U.S. Geological Survey have focused geologic interest on the higher temperature geothermal zones. The designation of known geothermal resource areas under the Geothermal Steam Act of 1970 provides a comprehensive guide to geographic areas in the western U.S. reasonably expected to have an economic potential for geothermal power exploitation as judged by 1972 standards. The listing of the areas, however, includes no quantitative data for the individual areas and much additional detailed geologic and geochemical data has been acquired since that time.

More recently, a number of comprehensive state surveys have begun to provide the quantitative geochemical and hydrologic data necessary for the initial pre-drilling evaluation of the total geothermal resource. Notable among these investigations are those for Utah (Mundorff, 1970), Idaho (Young and Mitchell, 1973), and New Mexico (Summers, 1965). A series of geothermal water quality investigations under way by geochemists at the Menlo Park office of the U.S. Geological Survey are presently available as open file reports. These include

studies of thermal waters of northern and central Nevada (Mariner, et al., May 1974) and hot springs in Oregon (Mariner, et al., March, 1974). Other studies in progress by this group include investigations in Montana, California, and New Mexico (J.B. Rapp, personal communications, September, 1974). California's Division of Oil and Gas investigators, particularly Marshall Reed, have collected samples according to established U.S. Geological Survey procedures. These samples from selected geothermal zones in California were analyzed in U.S. Geological Survey laboratories (Reed, 1974).

Geologic, geophysical and hydrological investigations of areas with a known geothermal potential are continuing under programs of the geologic and geophysical branches of the U.S. Geological Survey. Detailed geologic investigations have followed the initial phase of the resource investigations which have been primarily concerned with water chemistry and geothermometry.

#### Objectives of the Study

An important objective of the present investigation is the assessment of geologic techniques and evaluations especially applicable to the non-electrical utilization of the geothermal resource. The indicia and kinds of data generally considered to be crucial in the evaluation of the economic potential of the geothermal resource in a particular geographic area have been outlined by Godwin, et al. (1971). These criteria were intended specifically for the determination of areas with a power-producing potential. Obviously, the areas possessing these sorts of indications of the presence of a higher temperature geothermal resource will also have a potential for non-electrical utilization, at least as a by-product of power generation. The purpose of the present study is, however, to assess the techniques and evaluations applicable to the much broader geothermal resource with particular emphasis on the lower temperature portion of the spectrum.

A second objective of the present study is to assess the geographic distribution of the non-electrical resource on a national scale from presently available published and unpublished data. This involved a tabulation of data relevant to the distribution of the resource in terms of such variables as temperature, water quality and volume. A third objective is an assessment of additional data and types of investigations necessary for the completion of this resource evaluation phase.

### ASSESSMENT OF GEOTHERMAL EVALUATION TECHNIQUES

#### Evaluation Criteria Applied by the Federal Government

The indicia and techniques applied by agencies of the federal government to the evaluation of the economic potential of a geothermal zone have been outlined by Godwin, et al. (1971). Three indicia listed as necessary for the retention classification of lands in geothermal resource provinces are:

1. Volcanism of late Tertiary or Quaternary age--especially caldera structures, cones, and volcanic vents.
2. Geysers, fumaroles, mud volcanoes, or thermal springs at least 40°F higher than average ambient temperature.
3. Subsurface geothermal gradients generally in excess of two times normal, as reflected in deep water wells, oil well tests, and other test holes.

The authors further stated that the following kinds of data were to be considered in the geological, geophysical and geochemical evaluation of the economic potential of an area.

1. Siliceous sinter and natural geysers both imply high subsurface temperatures, generally 350°F or greater (D.E. White, "Geochemistry Applied to the Discovery, Evaluation and Exploitation of Geothermal Energy Resources": Rapporteur rept., Sec. V, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September, 1970) in hot-water systems, because of relationships generally existing between temperature and SiO<sub>2</sub> content of liquid water.
2. The temperatures of fumaroles, thermal springs, and mud volcanoes provide minimum subsurface temperatures.

3. The SiO<sub>2</sub> content of spring water is a very useful chemical geothermometer for indicating the reservoir temperatures of many hot-water systems (R.O. Fournier and A.H. Truesdell, "Chemical Indicators of Subsurface Temperature Applied to Hot Spring Waters of Yellowstone National Park, Wyoming, U.S.A": Paper V/2, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September, 1970; D.E. White, see above).
4. The Na/K ratio in spring waters of many hot-water systems is also a useful chemical geothermometer when there is adequate knowledge of competing influences (D.E. White, see above).
5. Most known potential geothermal systems occur in or near volcanoes and calderas of late Tertiary or Quaternary age.
6. Abnormally high conductive heat flow and the geothermal gradient are the best indicators of deep, concealed geothermal reservoirs. Although specific limits have not yet been established, two to 10 times the world-wide average (heat flow of 1.5 microcalories per cm<sup>2</sup> per second; temperature gradient of 1°F per 100 ft.) extended consistently over hundreds of feet of depth appears favorable.
7. The porosity and the permeability of a potential reservoir are important parameters, but can be established only by drilling and testing. Where stratigraphic control of the reservoir fluid or steam by a caprock is expected, near-surface characteristics of the rocks may provide preliminary evaluations.
8. Electrical resistivity surveys are probably the best geophysical means to geothermal evaluation available at this time, especially for the hot-water systems.
9. Magnetic, gravity, and airborne infrared geophysical surveys may provide useful supplemental data.
10. Other geophysical methods, such as microseismic, seismic ground noise, electromagnetic, and telluric surveys may have significant future use in evaluation.

These considerations constitute an excellent summary of the geological, geophysical and geochemical criteria applicable to the high temperature geothermal systems. However, as the non-electrical utilization of the geothermal resource also encompasses the lower end of the temperature spectrum, additional considerations are pertinent to the evaluation of the total resource. Specifically, the resource which may be economically utilized for non-electrical purposes may occur in a wider range of geologic environments. Surface indications

of the development at depth of a geothermal reservoir may be lacking in the lower temperature system which yet has a significant potential for non-electrical utilization.

Subtle differences exist in the techniques applicable to the evaluation of low and high temperature geothermal resources. These differences arise as a result of the differing origins of the majority of the lower temperature thermal water areas as opposed to the more unique geologic environment conducive to high temperature geothermal zone development.

#### Geologic Environment

The statement in the federal criteria that "Most known potential geothermal systems occur in or near volcanoes and calderas of late Tertiary or Quaternary age" is especially relevant to high temperature systems and may be expanded in terms of the petrologic type of volcanism. Rhyolitic volcanism is especially indicative of high, near-surface temperatures and the possibility of major, high-temperature geothermal zone development.

The geologic environments conducive to the high heat flow necessary for volcanism in general have been discussed by Muffler and White (1972). Regions with a high heat flow, roughly 1.5 to 5 times the world average of  $1.5 \times 10^{-6}$  calories per square centimeter per second occur along the margins of crustal plates. It is in these areas that molten rock is generated with resultant volcanism. In the absence of continental crustal material in the ocean basins, basaltic volcanism dominates. Rhyolite forms predominantly through crustal melting and is in itself prima facie evidence for an abnormally high heat flow situation in the crustal rocks. A late Tertiary-Quaternary age of the rhyolitic volcanism is necessary, as older volcanic rock may have lost its heat content and older thermal events in the crust may have terminated. The length of time required for the cooling of a shallow magma body within the crust will be dependent on a number of factors such as the size, initial temperature and depth

of emplacement of the molten body. Thermal history of an area intruded by a magma body, such as the study by Mundry (1971) indicates that a relatively rapid decrease in thermal gradients occurs after an initial heating up of the ground rock during magmatic intrusion. The rate of decrease of geothermal gradients in the natural situation will depend on the stability of regional temperature gradients subsequent to magmatic intrusion. The results of the modeling indicate, however, that a small magmatic intrusion will cease to be a viable geothermal heat source in a relatively short geologic time after intrusion (1-4 million years?). The maintenance of a localized high geothermal gradient associated with a shallow cooling magma body is a relatively short-term geologic phenomena. Regionally high heat flow may be a much longer term geologic event.

Types of volcanism other than "rhyolitic" may not provide the higher temperatures required for power generation but may be associated with higher than normal regional thermal gradients conducive to low temperature zone development. Basaltic volcanism, for example, originates through partial melting in the mantle at depths averaging 40-60 km. The relatively fluid, molten, rock flows through fissures to the surface where its heat content is rapidly lost. Andesite tends to form at even greater depths in the mantle but may have a greater tendency towards entrapment in relatively shallow magma chambers, thus forming a shallow, but still relatively short-lived heat source.

The net result of these considerations is the prediction of a relatively few intense high temperature geothermal zones directly associated with abnormal heat flow in the crust. These will be associated with crustal melting and will be evidenced by young rhyolitic volcanism if this flow occurs in continental areas with a silicic crust. Lower temperature geothermal occurrences related to broad areas of regionally high heat flow should be much more common. These "normal" low temperature occurrences constitute the portion of the geothermal



resource uniquely suited to non-electrical utilization. The rare high temperature zones and the thermal waste water produced by high temperature utilization will also be put to non-electrical uses, but the main portion of the geothermal resources, both volumetrically and geographically, appears to be of the low temperature type.

Most lower temperature geothermal occurrences are related genetically to faulting and deep water circulation in areas with a regionally high geothermal gradient. This seems to be the explanation for the numerous hot spring occurrences in granitic areas such as the "Idaho Batholith" of Idaho and Montana. The abnormally low regional heat flow throughout the Sierra Nevada is responsible for the relative absence of hot springs there. Porosity and permeability factors are equally important in either high or low temperature systems, although a premium may actually be placed on these factors in low temperature systems where high flow rates may be required in order to provide the total energy required from low enthalpy fluids.

#### Geochemical Techniques

Geochemical techniques referred to by Godwin, et al., are equally pertinent to the electrical and non-electrical resource assessment. At present, geochemical thermometry (particularly predictions based on  $\text{SiO}_2$  content assuming quartz equilibrium) provides the most useful single technique of the pre-drilling assessment of temperature characteristics of the geothermal resource. Basic assumptions pertinent to the application of geochemical thermometry as outlined by Fournier, White and Truesdell (1974) include:

1. A temperature-dependent reaction at depth.
2. An adequate supply of the solid phase involved in the reaction to permit saturation of the constituent used for geothermometry.
3. Water-rock equilibrium at depth.

4. Negligible re-equilibration as the water flows to the surface.
5. No dilution or mixing of hot and cold waters.

In cases where high flow rate springs discharge water rather directly from its high temperature reservoir without significant surface water dilution, the SiO<sub>2</sub>-predicted temperatures are generally valid. Normally, however, at least some dilution or mixing of thermal water with shallow, relatively cool ground water may occur, resulting in geochemically-predicted temperatures below the actual "base" of maximum temperature attained by the geothermal water during its convective circulation.

Fournier and Truesdell (1974) have suggested geochemical calculations for determining if mixing has occurred and present a technique for calculating the original, undiluted reservoir temperature. Preliminary applications of the technique to selected, relatively low temperature springs in Yellowstone National Park predict high reservoir temperatures. These temperatures appear reasonable in terms of reservoir temperatures encountered by test drilling in several of the areas. Studies by Mariner, et al. (1974) applying the mixing techniques of Fournier and Truesdell have indicated the probability of significant mixing in 11 of 32 hot springs in Oregon. The geochemical thermometers applied by Fournier and Truesdell (1974) and Willey, et al. (1974) are summarized in Table 1.

TABLE I

1. For quartz solubility with SiO<sub>2</sub> in mg/kg

$$T (^{\circ}\text{C}) = \frac{1309}{5.19 - \log (\text{SiO}_2)} - 273.15$$

- 1a. For quartz solubility with cooling by steam loss

$$T (^{\circ}\text{C}) = \frac{1522}{5.75 - \log (\text{SiO}_2)} - 273.15$$

2. For albite-orthoclase solubility, cations in mole/kg

$$T (^{\circ}\text{C}) = \frac{777}{.47 + \log (\text{Na/K})} - 273.15$$

3. For sodium, potassium, calcium in mole/kg

$$T (^{\circ}\text{C}) = \frac{1647}{2.24 + \log (\text{Na/K}) + \beta \log (\sqrt{\text{Ca}/\text{Na}})} - 273.15$$

Where  $\beta = 4/3$  is to be used, except if the calculated temperature is above 100°C; then  $\beta = 1/3$  is to be used.

from Reed, 1974

## Geophysical Techniques

Geophysical techniques provide two differing types of information for geothermal exploration. Most geophysical techniques provide varying degrees of structural information, depending on the geologic conditions. Some of the techniques used to provide structural information include: active seismic, gravity, magnetic, electrical, and passive seismic. Some geophysical techniques are also considered a direct detection tool. Among these techniques are: resistivity, thermal gradient measurements, and microseismic studies. However, the relatively limited experience in the actual drilling of geophysical anomalies thought to be associated with geothermal reserves is insufficient to label any tool or technique as a panacea.

In exploration for a low temperature geothermal resource, techniques such as active seismic, gravity and magnetics are utilized to define the structural relationships in the province. The direct detection techniques might not be as useful in directly delineating low temperature reserves. For example, normal ground water conditions may mask any thermal gradient associated with convective hot water systems driven by a regional anomaly. Also, a regional anomaly might not have the localized seismic activity which would be detected by a microearthquake program. Present knowledge in the area of seismic noise sources is probably not sufficient to be conclusive on the microearthquake activity and seismic noise level associated with low temperature geothermal activity.

The technique that seems to offer the most promise for low temperature geothermal resource exploration is electrical resistivity. Because the resistivity of rock is primarily a function of fluid content, fluid salinity and temperature, resistivity measurements may be extremely helpful in delineating the spatial distribution of the highly conductive geothermal reservoir.

The major factors controlling the applicability and the amount of information obtained from any of these geophysical techniques are the local geology and the grade of the geothermal deposit. In other words, there must be a variation in properties between the "country rock" and the resource. The larger the variation in properties, the greater the chance of detection. Thus, in any area the utilization of a number of geophysical techniques will probably be required. A premium will be placed on the application of all the geological and geophysical techniques designed to clarify the geology and hydrology of the low temperature geothermal occurrence.

#### GEOGRAPHIC DISTRIBUTION OF THE GEOTHERMAL RESOURCE

The cataloging of reported hot spring locations by Waring (1965) and the designation of KGRA's and AVP's (Areas Valuable Prospectively) by Godwin, et al. (1971) provide a starting point for consideration of the national geothermal potential. The assessment of the non-electrical utilization of the geothermal resource requires, however, an evaluation of the geographic distribution of the resource in terms of temperatures to be reasonably expected, water quality, and flow volumes. The present investigation involved a tabulation and graphic presentation of the resource distribution in terms of these variables which determine its potential for utilization.

#### Conclusion

Exploratory drilling and the application of geochemical thermometers indicate that the bulk of the major thermal occurrences in the western states have reservoir temperatures in the 100-180°C range. Waring listed 852 thermal springs and wells for the 10 western states, exclusive of Yellowstone National Park (which has more than 10,000 mapped thermal features). Analyses reported to date for the 852-plus known thermal occurrences indicate approximately 250 with silica predicted temperatures in the 100-140°C range, 67 in the 140-180°C

range, 15 in the 180-220°C range and two above 220°C. The remainder of the thermal occurrences (approximately 90%) have apparent temperatures below 100°C. Approximately 2% have known or predicted temperatures above 180°C, the minimum temperature presently required for economic geothermal power generation. The additional application of refined geochemical thermometers and the newly developed "mixing" techniques will undoubtedly increase the number of suspected high temperature systems. The present data, however, is weighted in favor of known high temperature occurrences, i.e., 10 reported analyses from Long Valley. Therefore, the inference of the overwhelming predominance of the low temperature portion of the resource is valid.

#### Data Presentation

Temperature data is shown graphically on a state by state basis for the 10 westernmost states. Predicted temperature ranges were subdivided into three categories: 15° above ambient - 100°C, 100-180°C, and > 180°C. Temperatures plotted in the intermediate and high temperature range were ordinarily those temperatures predicted by the SiO<sub>2</sub> geothermometer assuming quartz equilibrium. Temperature prediction for Idaho, Nevada, Oregon, New Mexico, Utah and California are based on relatively complete coverage by state or U.S. Geological Survey geothermal water quality reports. Data for Washington, Montana and Arizona is less complete, but work in progress will soon provide almost total geochemical coverage of the western states.

Data for the "geopressurized zone" of the Gulf Coast is extracted entirely from the work of Jones (1970). Temperatures shown are for an average depth of 8000 feet as calculated from observed gradients.

The water quality data is presented in both tabular form and graphically in terms of total dissolved solids and boron expressed in ppm. The total dissolved solids content was chosen for presentation as a general indication of overall water quality important to both industrial process and agricultural appli-

cation. The boron level is shown as it constitutes the limiting factor in many agricultural applications.

Flow data is included in tabular form and discussed on a local or regional basis where geothermal production is significant.

#### State Summaries

California contains two unique geothermal occurrences, Salton Sea-Imperial Valley high temperature brine reservoirs and the Geysers Area, Sonoma County, one of the world's prime examples of a developed dry steam geothermal system. The exploitation of the Geysers area for electrical development and attempts to develop the Imperial Valley area for desalination purposes have been well documented elsewhere (Armstrong, 1971). The 14 KGRA's and additional areas studied by Reed (1974) and Hannah (1974) possess an obvious potential for non-electrical geothermal development. Of the areas investigated by Reed, only the Long Valley area appeared to possess definite indications of reservoir temperatures above 180°C. In California, 86 analyses of geothermal waters with temperatures in the 100-180°C range have been reported in the literature, with the majority of these (74) between 100-140°C. Areas with a particular potential of non-electrical geothermal potential include: Alturas area, Big Valley, Wendel-Amedee area, Sierra Valley area, Long Valley area, and northern Napa Valley in Calistoga.

Idaho possesses at least 380 known hot springs and wells, 82 of which have silica predicted reservoir temperatures above 100°C. Of these, 73 are in the 100-140°C range, and nine are in the 140-180°C range. The state possesses eight major geothermal resource areas, including two areas, the Raft River in Cassia County and the Island Park caldera in Fremont County, which received the federal KGRA designation. In view of the abundant 100-140°C resource, the state possesses an exceptionally high potential for non-electrical geothermal development.

Numerous hot wells in northern Owyhee County in an area of ranch and farm land bordering the Snake River Plain and the Mt. Bennett Hills anomaly, an agricultural area in southern Elmore County, possess a major potential for agricultural uses of the geothermal resource. Water quality is excellent, and the hot water is used extensively for irrigation. The coincidence of high volume wells producing 178°F water within Boise, the state capital, make this a prime site for space heating applications. The Raft River KGRA has undergone extensive geological, geophysical and hydrological investigations in conjunction with its possible utilization as a low temperature power generation site; and there is a limited potential for non-electrical utilization of the effluent waste water.

Montana: The western third of the state is block faulted and composed in part of Tertiary granitic and volcanic rocks where the thermal springs occur. Approximately 40 thermal springs occur in the state, and their temperatures range up to 80°C. There is one KGRA near Yellowstone National Park in the southern part of the state. Flow rates range up to 50,000 gpm. The water quality is good in the areas underlain by granitic rocks of the Idaho Batholith, but water quality is variable in the region of Yellowstone Park.

The steepest geothermal gradient measured is in the area around Marysville, where conducted heat flow range is from 3.2 to 19.5 u. cal/cm<sup>2</sup> sec. At one locality, a temperature of 58°C was measured at a depth of only 220 meters. Drilling has indicated that the immediate source of the high heat flow is an unexposed reservoir of thermal fluids present within the fractured granite. The drilling confirms the convective circulation of water within the fractured granite to depths of 6000 feet, and thus casts doubts on the relative abundance of hot "dry" geothermal systems.



Nevada contains 152 thermal occurrences as reported by Waring (1965). Nineteen areas in Nevada have geochemically predicted temperatures in the 100-140°C range; sixteen areas have reported temperatures in excess of 140°C. Applications of refined geothermometry techniques by Mariner, et al. (1974) indicate that a relatively high number of the Nevada geothermal areas (five) have reservoir temperatures above 180°C. These higher temperature systems, including Beowawe, Sulfer Hot Springs and Steamboat Springs, have a potential for non-electrical utilization secondary to any attempted utilization of the wet steam resource for power generation. The principal deterrent to non-electrical utilization of the state's abundant intermediate temperature geothermal areas is the low population density.

New Mexico: The thermal occurrences of New Mexico are associated with high heat flow and recent volcanism along the Rio Grande trough. The presence of young silicic volcanism in the Jemez Volcanic Field and the Valles caldera west of Los Alamos indicate a marked potential for high temperature geothermal development and possible "non-electrical" utilization of the effluent. The presence of steam has been confirmed by exploratory drilling, and the area is also being explored for its hot, dry rock geothermal potential. The remainder of New Mexico's geothermal occurrences appear to be associated with the high regional heat flow and faulting rather than a localized shallow heat source.

Oregon: Waring reported 86 thermal occurrences in Oregon. Thirty-two analyses of the hotter springs were reported by Mariner, et al. (1974). Aquifer temperatures between 140-181°C were predicted for nine springs, based on both silica and Na-K or Na-K-Ca geothermometers. These results indicate that the principal-use of Oregon's geothermal potential will continue to be non-electrical. Individual areas with a definite non-electrical utilization potential include:

1. Lake County, where predicted temperatures range from 120-155°C, but boron contents will generally eliminate direct agricultural uses;
2. Vale area,

Malheur County, with the highest predicted temperature in the state, 181°C. Non-electrical utilization in Vale may be possible as an aftermath of exploration for higher temperature steam; 3. Klamath Falls - the established utilization of the low temperature fluids for space heating at Klamath Falls illustrates the economic value of the low temperature resource. The geographic distribution of the shallow hot water resource has been established by drilling, but studies of the hydrology of the system and its actual productive capacity are needed; 4. Breitenbush Hot Springs, Marion County and 5. Hot Lake, Union County, both possess high flow rates of hot or near-boiling water; 6. Alvord Hot Spring in southern Harney County and 7. Belknap Hot Spring in the Cascade Range are very saline, with total dissolved solids of 2,980 and 3,020 ppm respectively, and are better suited for industrial process use than for agricultural use.

Utah: The thermal springs of Utah are closely associated with the Wasatch-Hurricane fault zone and appear genetically related to deep circulation in this high-heat flow, tectonically active area. The known and predicted thermal occurrences are all low or intermediate temperature areas, and the major (if not exclusive) utilization of the geographically dispersed resource will be non-electrical.

## RECOMMENDATIONS FOR FURTHER STUDY

1. Detailed water quality analyses coverage should be extended to include all significant thermal water occurrences not presently reported in the literature or available on an open-file basis from state or federal agencies. Specifically, these analyses should be of a completeness and accuracy comparable with the standards met by the U.S.G.S. water quality program. These analyses are necessary both for the pre-drilling estimation of reservoir temperatures and an assessment of the best use of the thermal water as influenced by its total chemistry. The absence of boron analyses, for example, is a hindrance to the assessment of the potential of thermal waters in many areas.
2. Studies of the frequency distribution of geothermal systems as a function of observed and geochemically predicted base temperatures should be conducted as the geochemical data is obtained. Application of geothermometers to the almost complete data available at present indicates an overwhelming predominance of the low and intermediate temperature systems. The data is also susceptible to treatment in terms of the geologic environment (i.e., recent rhyolitic volcanism, basalt provinces, regional faulting, Basin and Range faulting, geopressurized zones, etc.).
3. In view of the apparent dominance of the low temperature resource, a major research effort should be directed towards the utilization of relatively low temperature, low enthalpy fluids.
4. The drilling of geothermal, oil and gas, and deep water wells should be closely monitored for situations where the "mixing function" may be applied and confirmed. The major unknown concerning the temperature distribution is the degree to which mixing has masked the presence of higher temperature wet systems.

## SELECTED REFERENCES

- Fournier, R.O. and Truesdell, A.H., 1974, Geochemical indicators of subsurface temperature Part II: Estimation of temperature and fraction of hot water mixed with cold water: In press: U.S. Geol. Survey Jour. of Res., v. 2, #3, May 1974.
- \_\_\_\_\_, White, D.E., and Truesdell, A.H., 1974, Geochemical indicators of subsurface temperature Part I: Basic Assumptions: In press: U.S. Geol. Survey Jour. of Res. v. 2, #3, May 1974.
- Godwin, L.H., et al., 1971, Classification of public lands valuable for geothermal steam and associated geothermal resources: U.S. Geol. Survey Circ. 647, 17 p.
- Mariner, R.H., et al., 1974, The chemical composition and estimated minimum thermal reservoir temperatures of selected hot springs in Oregon: U.S. Geol. Survey Open-File Rept., Menlo Park, CA, 27 p.
- \_\_\_\_\_, May 1974, The chemical composition and estimated minimum thermal reservoir temperatures of the principal hot springs of northern and central Nevada: U.S.G.S. Open-File Rept., Menlo Park, CA, 32 p.
- Muffler, L.J.P. and White, D.E., 1972, Geothermal energy: Sci. Teacher, v. 39, No. 3, pp.40-43.
- Mundry, E., 1970, Mathematical estimation concerning the cooling of a magmatic intrusion: Geothermics, v. 2, Part 1, pp. 662-668.
- Reed, Marshall J., 1974, Chemistry of thermal water in selected geothermal areas of California: California Div. of Oil and Gas Pub. (In Press October, 1974)
- Summers, W.K., 1965, A preliminary report on New Mexico's geothermal energy resources: New Mexico Bureau of Mines and Min. Res. Circ. 80, 41 p.
- Waring, G.A. (Rev. 1965), Thermal springs of the United States and other countries of the world - a summary: U.S. Geol. Prof. Paper 492, 833 p.
- White, D. E., 1970, Geochemistry applied to the discovery, evaluation, and exploitation of geothermal energy resources: Geothermics Special Issue No. 2.(In Press)
- \_\_\_\_\_, Muffler, L.J.P., and Truesdell, A.H., 1971, Vapor-dominated hydrothermal systems compared with hot-water systems: Econ. Geol., v. 66, No. 1, pp. 75-97.
- Willey, L.M., O'Neil, J.R., and Rapp, J.B., 1974, Chemistry of thermal waters in Long Valley, Mono Co., California: U.S. Geol. Survey Open-File Rept., Jan. 1974, Menlo Park, CA.
- Young, H.W. and Mitchell, J.C., 1973, Geothermal investigations in Idaho, Part I: geochemistry and geologic setting of selected thermal waters: Idaho Water Information Bull. 30, 43 p.

Appendix I.  
STATE GEOTHERMAL DISCUSSIONS

# A R I Z O N A

## INTRODUCTION

Intensive geological, geophysical, exploration surveys, and studies are actively underway; and one group has uncovered sufficient information to warrant the beginning of an exploration drilling program.

Hot springs are localized in the southeastern part of the state in the Basin and Range Province, plus a few in the southwest.

Heat flow in southern Arizona is 2 HFU.

Most recent volcanism occurred 900 years ago at Sunset Crater.

Most springs are related to fault structures and/or Cenozoic igneous rocks.

Hottest surface water - Gillard Hot Springs in Greenlee County 183°F (84°C)

Average temperature 100°F (38°C)

25 thermal areas (11 of which are wells)

Discharge average 200 gpm, up to 600 gpm

Dissolved solids range up to 100,000 ppm, and most have a high salt content.

### Gillard Hot Springs

Maximum temperature 184°F (84°C)

Discharge 100-400 gpm

### Clifton Hot Springs

Maximum temperature 160°F (71°C)

### Radium Hot Springs

Maximum temperature 140°F (60°C)

There are sparse data indicating a possibility that latent geothermal energy may be present at depth at several areas in southern Arizona. Data on springs and wells have been compiled on Table 1, which is a listing of localities where thermal waters of 100°F or greater are known. The locations of these occurrences, shown on Figure 1, are relatively close to population centers which could possibly utilize thermal energy. There are no known geothermal resources areas which produce geothermal steam in Arizona, and no known boreholes have been drilled to explore the potential of geothermal energy.

GEOLOGIC AND HYDROLOGIC RESUME OF CONDITIONS RELATED TO  
POTENTIAL GEOTHERMAL RESOURCES

Arizona contains parts of two distinct physiographic provinces, the Colorado Plateau and the Basin and Range, which are shown on Figures 1 and 2. In both provinces, rocks of nearly all the geological ages and types are known to be present.

The rocks of Arizona have been faulted and folded many times throughout geologic time. The faulting and folding occurred during short episodes, and concurrent periods of igneous activity were common. The entire area had the most intense deformation during the late Cretaceous and early Tertiary periods. In the Basin and Range Province these structural features were disrupted by later structural features which, in turn, have been followed by late Cenozoic volcanism.

In the middle Tertiary, about 30 million years ago, the first phase of block faulting began which developed the Basin and Range structural features; and many of these uplifted blocks cut across previous structures. Large amounts of volcanic material were emplaced in the basins which, in places, dammed the drainage systems. These structural troughs have been filled with heterogeneous mixtures of clastic materials ranging from fine-grained lake-bed deposits to boulder-cobble conglomerate interbedded with lava flows. During this phase, several intrusive bodies were emplaced in the southeastern part of the state. Some volcanic activity continued to about 1,000 years ago, such as in the Flagstaff area.

Arizona has been divided into three water provinces: Basin and Range Lowlands, the Central Highlands, and the Plateau Uplands. Large amounts of groundwater occur in the alluvial basins in the Lowlands which have been filled to depths of several thousands of feet of low permeable silt and clay bodies. Large amounts of groundwater are in storage in the sandstone and limestone aquifers in the Uplands area, but the lesser permeability of the sandstones constrain well yields. Both areas receive recharge from the Highlands area which occurs across the central part of the state.

The quality of water resources ranges widely throughout the state. The groundwater quality ranges from less than 200 to more than 100,000 ppm (parts per million). The streams and rivers carry variable amounts of dissolved salts. A large percentage of the salt in some drainages is from springs which contribute saline water to the streamflow.

An examination of the available thermal data in Table 1 in relation to the tectonic fabric and occurrence of Tertiary igneous rocks reveals several interesting suggestions. Nearly all the occurrences are in the Basin and Range Lowlands; most of the springs and some of the wells appear to be related to fault structures and/or Cenozoic igneous rocks; the highest temperatures occur in Greenlee County; and the geothermal gradient is about 30 feet per 1°F in southeastern Arizona. These data and geological features appear to indicate that most of southern Arizona comprises a target area to conduct studies and exploration for potential geothermal resources, particularly in the southeastern quadrant and western section. The Plateau Uplands area is not necessarily ruled out, as reservoir rocks may be deeply buried and the overlying extensive groundwater system could have a cooling effect on any leakage through the thick blanket of fine-grained sediments. The youngest volcanism activity occurs in this area, and a geothermal reservoir could be enclosed by fine-grained rock sequences.



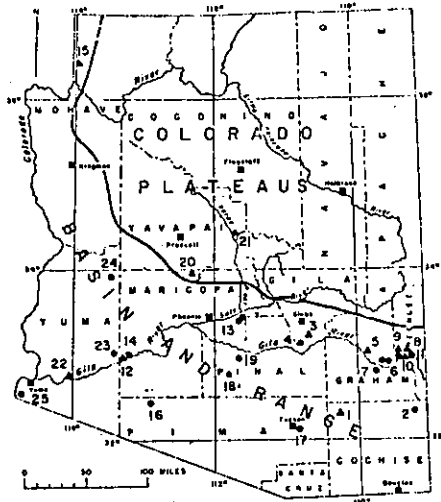
## CONCLUSIONS

1. There are no Known Geothermal Resources Areas in Arizona.
2. There are several investigations on geothermal resources being conducted by private enterprise which indicate a potential occurrence. Exploration borehole drilling is reported to start within the next 60 days.
3. In southern Arizona there are many geologic structural and rock characteristics similar to known geothermal reservoirs.
4. There are no surface indications of steam leakage, but the occurrence of thermal water nearby favorable geologic features suggests the potential occurrence of geothermal energy.
5. There are several favorable target areas in southern Arizona; the application of skillful study, creative imagination, and data from borehole drilling programs could lead to successful geothermal development.

TABLE 1 – SELECTED THERMAL SPRINGS AND WELLS IN ARIZONA

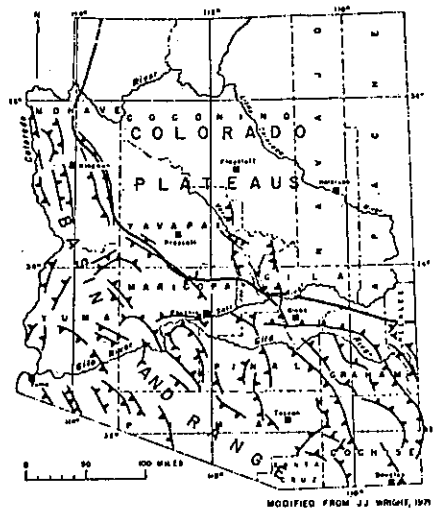
Locality No. on Fig. 1	County and Location	Maximum Temperature (°F) Reported	Discharge in Gallons Per Minute	Depth (feet)	Gradient (feet/ 1°F)
<b>Cochise</b>					
1	Hookers Hot Spring Sec 6, T 13 S, R 21 E	130	40	—	—
2	San Simon Valley-wells	103	—	920	26
<b>Gila</b>					
3	Spring Sec 1, T 3 S, R 16 E	95	90	—	—
4	Well, Sec 29, T 3 S, R 15 E	110	—	150	—
<b>Graham</b>					
5	Indian Hot Springs Sec 17, T 5 S, R 24 E	118	200-300	—	—
6	Wells in Safford Area	99+	—	1095	30
7	Safford Golf Course Well	100	—	2180-2420	30
<b>Greenlee</b>					
8	Clifton Hot Springs Sec 30, T 4 S, R 30 E	160	—	—	—
9	Spring, Eagle Creek Sec. 26, T 4 S, R 28 E	120	—	—	—
10	Gillard Hot Springs Sec 27, T 5 S, R 29 E	184	100-400	—	—
11	Well, Clifton	130	—	220-491	31
	Well, Clifton	143	—	140-500	33
<b>Maricopa</b>					
12	Agua Caliente Spring Sec 19, T 5 S, R 10 W	104	—	—	—
13	Buckhorn Area Wells	107	—	325	8
14	Well Sec 16, T 5 S, R 10 W	114	—	884-1268	—
<b>Mohave</b>					
15	Pakoon Spring Sec 24, T 35 N, R 16 W	100	200-400	—	—
<b>Pima</b>					
16	Wells near Ajo	99	—	1021	39
17	Wells, SE Tucson	126+	—	2500	42
<b>Pinal</b>					
18	Well, near Redrock	108	—	1950	49
19	Hana Well Sec 6, T 6 S, R 8 E	161	500-600	2700	—
<b>Yavapai</b>					
20	Castle Hot Springs Sec 3, T 7 N, R 1 W	122	280	—	—
21	Verde Hot Springs Sec 3, T 11 N, R 6 E	106	10	—	—
<b>Yuma</b>					
22	Radium Hot Springs Sec 12, T 8 S, R 18 W	40	—	—	—
23	Wells in Sections 2, 5, 11, T 4 S, R 11 W	105	—	280-585	—
24	Wells in Palomas Plain	108	—	443	13
25	Oil Test Well	125	—	6015	120

Data tabulated from: Haigler (1969) and Wright (1971) in Selected References



20 SPRING 100°F OR GREATER  
 16 WELL 100°F OR GREATER  
 (NUMBERS REFER TO LOCALITIES IN TABLE I)

FIGURE 1.--THERMAL SPRINGS AND WELLS IN ARIZONA.



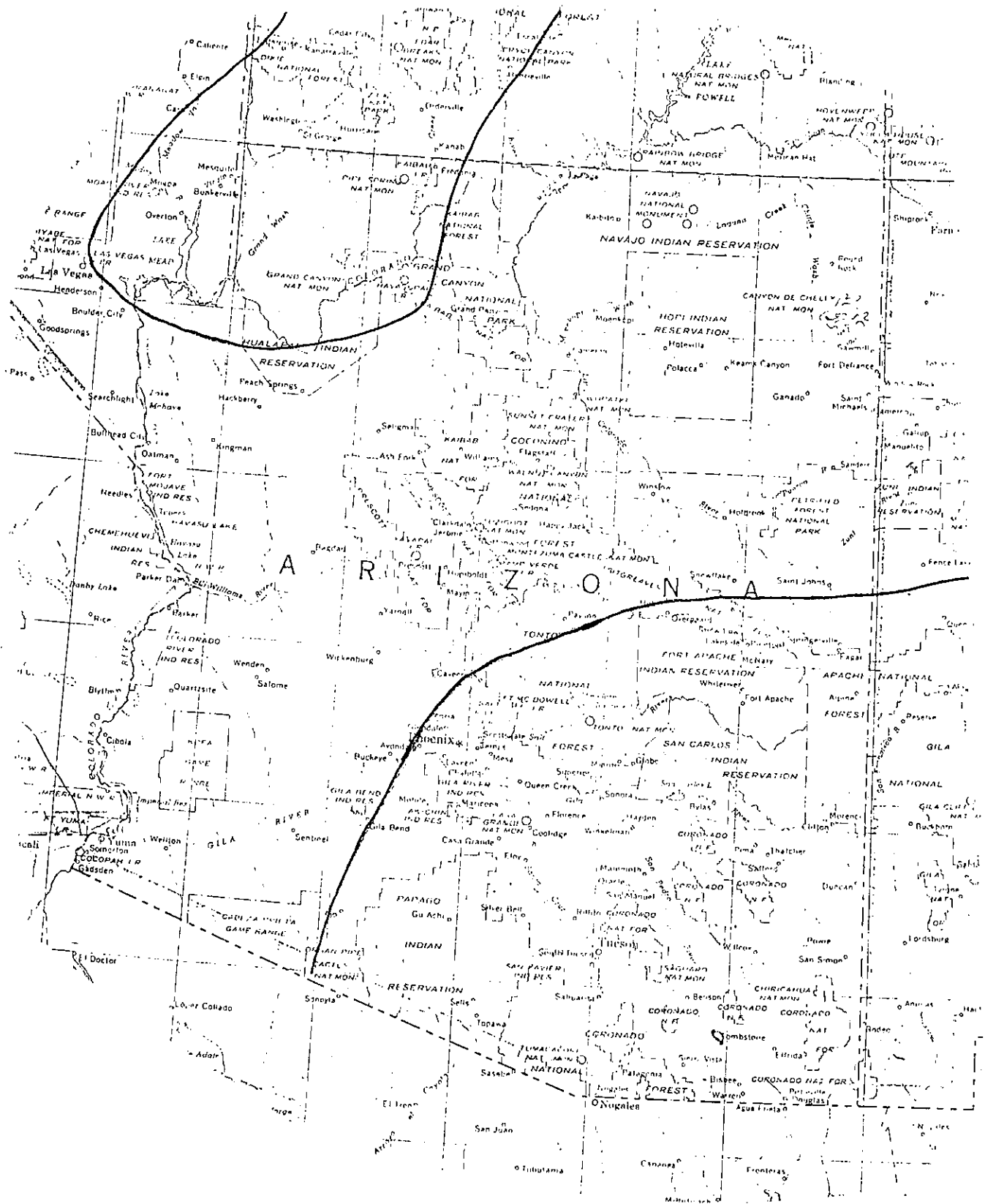
U  
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FIGURE 2.--FAULT TRENDS IN SOUTHERN ARIZONA.

## SELECTED REFERENCES

- Arizona Bureau of Mines and University of Arizona, 1962, Folio of geologic and mineral maps of Arizona: University of Arizona Press, Tucson.
- Arizona Bureau of Mines, 1969, Mineral and water resources of Arizona: Univ. of Ariz. Bull. 180.
- Conley, J.N., June 1971, Index of samples of drill-bit cuttings and/or cores of wells drilled in Arizona: Oil & Gas Conserv. Comm. State of Ariz.
- Damon, P.E., 1959, Geochemical dating of igneous and metamorphic rocks in Arizona: Ariz. Geol. Soc., Guidebook II, n. 16-20.
- \_\_\_\_\_, 1967, Potassium-Argon dating of igneous and metamorphic rocks with applications to the Basin Ranges of Arizona and Sonora: Radiometric Dating for Geologists, Intersci. Pub.
- Damon, P.E. and Bikerman, Michael, November 1964, K-AR Dating of post-laramide plutonic and volcanic rocks within the Basin and Range Province of southeastern Arizona and adjacent areas: Ariz. Geol. Soc. Digest, vol. VII, p. 63-78.
- Damon, P.E., and Giletti, B.J., 1961, The age of the basement rocks of the Colorado Plateau and adjacent areas: N.Y. Acad. Sci., Annals vol. 91, art. 2, p. 443-453.
- Damon, P.E., Mauger, R.L. and Bikerman, Michael, 1964, K-AR Dating of laramide plutonic and volcanic rocks within the Basin and Range Province of Arizona and Sonora: Proceedings, XXII Inter. Geol. Cong., New Delhi, India, p.45-55.
- Damon, Paul E. and Mauger, R.L., 1966, Epeirogeny-Orogeny viewed from the Basin and Range Province: Soc. of Min. Engrs. Trans., vol. 235, p. 99-112.
- Darton, N.H., 1920, Geothermal data on the U.S.: U.S. Geol. Sur. Bull. 70.
- Erickson, Rolfe, Welded intrusive volcanic breccia in the Dos Cabezas Mountains, Arizona: Submitted to Bulletin Volcanologique.
- Gilluly, James, 1949, Distribution of mountain building in geologic times: Geol. Soc. Am. Bull., v. 60, p. 561-590.
- \_\_\_\_\_, 1963, The tectonic evolution of the western United States: Quart. Jour. Geol. Soc. London, p. 113-174.
- Haigler, L.B., 1969, Geothermal resources in mineral and water resources of Arizona: Ariz. Bur. Mines Bull. 180.
- Harshbarger, J.W., 1958, Use of ground water in Arizona: in Climate and man in the Southwest - a symposium: Univ. Ariz. Bull., v. 28, no. 4, p. 51-68
- \_\_\_\_\_, 1960, (Contributor to section on water); in Arizona-its people and resources: Univ. Ariz. Press, Tucson.
- Harshbarger, J.W., Lewis, D.D., Skibitzke, H.E., Heckler, W.L., and Kister, L.R. (revised by H.L. Baldwin), 1966, Arizona water: U.S. Geol. Survey, Water-Supply Paper 1648.
- Hunt, C.B., 1956, Cenozoic geology of the Colorado Plateau: U.S. Geol. Survey, Prof. Paper 279.
- Jahns, R.H., 1959, Collapse depressions of the Pinacate volcanic field, Sonora, Mexico: Ariz. Geol. Soc., Guidebook II, p. 165-184.
- Johnson, P.W., 1959, Test holes in southern Arizona valleys: Ariz. Geol. Soc., Guidebook II, p. 62-65.
- Kuhn, T.H., 1941, Pipe deposits of the Copper Creek area, Arizona: Econ. Geol., v. 36., p. 512-538.
- Lasky, S.G., and Webber, B.N., 1949, Manganese resources of the Artillery Mountains region, Mohave County, Arizona: U.S. Geol. Survey Bull. 961.
- Lindgren, Waldemar, 1905, Description of the Clifton quadrangle: U.S. Geol. Survey Geol. Atlas, Folio 129, 13 p.
- Longwell, C.R., 1926, Structural studies in southern Nevada and western Arizona: Geol. Soc. Am. Bull., v. 37, p. 551-583; (absts): p. 165 (1926); Pan-Am. Geol., v. 45, p. 164.
- \_\_\_\_\_, 1950, Tectonic theory viewed from the Basin Ranges: Geol. Soc. Am. Bull., v. 61, p. 413-433.
- Lovering, T.S., 1948, Geothermal gradients, recent climatic changes, and rate of sulphide oxidation in the San Manuel District, Arizona: Economic Geol. Bull, v. 43, No. 1, p. 1-20.
- McKee, E.D., 1951, Sedimentary basins of Arizona and adjoining areas: Geol. Soc. America, Bull. vol. 62, p. 481-506.
- McNitt, J.R., 1963, Exploration and development of geothermal power in California: Calif. Div. Mines and Geol., Spec. Rept. 75, 45 p.
- \_\_\_\_\_, 1965, Review of geothermal resources in Terrestrial heat flow: Am. Geophys. Un. Geophys. Mon. ser. no. 8, p. 240-266.
- Mayo, E.B., 1958, Lineament tectonics and some ore districts of the southwest: Min. Engr. v. 10, no. 11, p. 1169-1175.
- \_\_\_\_\_, 1959, Volcanic geology of the northern Chiricahua Mountains: Ariz. Geol. Soc., Guidebook II, p. 135-138.
- Peirce, H.W., Keith, S.B., and Wilt, J.C., 1970, Coal, oil, natural gas, helium and uranium in Arizona: Ariz. Bur. of Mines, Bull. 182.

- Richard, K.E., and Courtright, J.H., 1960, Some Cretaceous-Tertiary relationships in southeastern Arizona and New Mexico: Ariz. Geol. Soc., Digest vol. III, p. 1-7.
- Robinson, H.H., The San Franciscan volcanic field, Arizona: U.S. Geol. Survey, Prof. Paper 76.
- Titley, S.R., 1959, Igneous rocks of the Basin and Range Province in Arizona: Ariz. Geol. Soc., Guidebook II, p. 85-88.
- Waring, G.A., 1915, Thermal springs of the U.S. and other countries of the world-a summary: U.S. Geol. Survey Prof. Paper 492.
- Wilson, E.D., 1931, New mountains in the Yuma Desert, Arizona: Geog. Rev., v. 21, p. 221-228.
- \_\_\_\_\_, 1933, Geology and mineral deposits of southern Yuma County, Arizona: Ariz. Bur. Mines Bull. 134, 236 p.
- Wilson, E.D. and Moore, R.T., 1959, Structure of Basin and Range Province in Arizona: Ariz. Geol. Soc., Guidebook II, p. 89-105.
- Wright, J.J., 1971, The occurrence of thermal ground-water in the Basin and Range Province of Arizona: *in* Hydrology and Water Resources in the Southwest, Proceedings, Ariz. Sec. Am. Water Resources Assoc. and Ariz. Acad. Sci., Tempe.



## COLORADO

Most of Colorado's 200 hot springs occur in central and southwestern mountainous parts of the state in the Tertiary volcanics. The Late Cenozoic Rio Grande Rift Valley extends northward from New Mexico through the Rockies of central Colorado; and together with the San Juan volcanic field of southwestern Colorado, represents a region of above-normal heat flow (from 1.52 to 2.4 HFU). There are no KGRA's in Colorado. The hottest areas:

Poncha Springs  
Temperature 75°C (167°F)

Waunita Hot Springs  
Temperature 72°C (162°F)

Pagosa Hot Springs  
Temperature 72°C (162°F)

Greatest flow occurs at Glenwood Springs  
3,000 gpm at 41°C to 65°C (106-149°F)

Steamboat Springs  
2,000 gpm at 39°C to 65°C (102-149°F)

Alamosa  
An area of hot oil and gas exploration wells under investigation by the USGS.

Table 1-- Chemical analyses of water from selected springs in Colorado.  
(Analytical results in parts per million)

No.	Name	Discharge (gpm)c	Temp. (°F)d	Silica (Si) (Na)	Potassium (K)	Calcium (Ca)	Magnesium (Mg)	Chloride (Cl)	Sulphate (SO <sub>4</sub> )	Bicarbonate (HCO <sub>3</sub> )
1	Routt	130a	150a	88	11	8	-	137	44	146
2	Steamboat Springs	2,000b	150b	85	155	113	32	1346	486	3253
3	Eldorado	10a	70a	17	1	10	4	7	38	46
4	Glenwood Springs	400a	150b	37	460	431	91	11025	1194	437
5	Buena Vista (Cottonwood)	150a	144a	61	34	5	3	28	108	87
6	Mount Princeton	300a	150a	60	32	11	T	12	62	86
7	Poncha Spgs	500b	168b	80	21	18	2	55	199	221
8	Waunita (Tomichi)	1,000b	160a	86	15	12	3	20	179	175
9	Powderhorn	100b	114a	80	75	133	49	120	132	1107
10	Ridgway	300b	132b	58	102	274	21	103	1287	278
11	Wagon Wheel Gap	100b	131a	94	276	68	18	205	210	1048
12	Ouray	200b	158a	52	45	383	11	51	1016	37
13	Pagosa Spgs.	600a	158a	68	370	247	17	180	1504	636

23

Remarks:

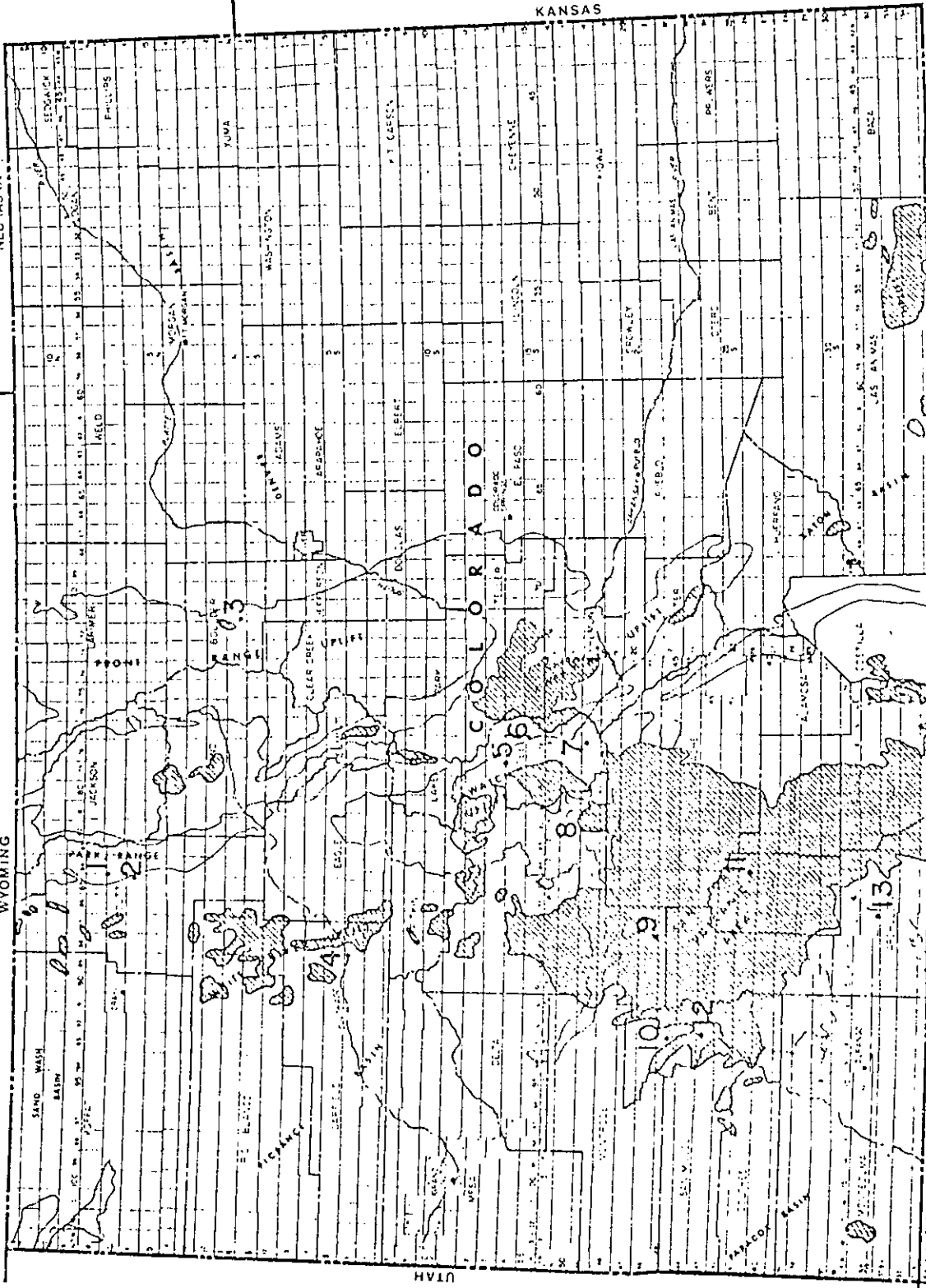
- a. Colorado Geological Survey Bull. 11. Discharge value given is maximum for a group of springs.
- b. U. S. Geological Survey Prof. Paper 492. Discharge value given is total for a group of springs. Analytical analyses from Colo. Geological Survey Bull. 11.
- c. Table extracted from: Pearl, R. H., Geothermal resources of Colorado - A Summary, Geothermal Overview of the Western U.S., p. D1-D-8 (1972)



COLORADO GEOLOGICAL SURVEY

WYOMING

COLORADO



NEBRASKA KANSAS OKLAHOMA NEW MEXICO

3 HOT SPRINGS

CENOZOIC VOLCANISM

Figure 1--Map showing selected hot springs and areas of Cenozoic volcanism

## CALIFORNIA

### Summary

The chemical composition and geologic setting of six potential geothermal areas are described by Reed in a forthcoming report of the California Division of Mines, Oil and Gas Commission. The areas described are: Imperial Valley, Long Valley, Sierra Valley, Honey Lake Valley, Surprise Valley, and Modoc Plateau. In most areas, calculated temperatures are below 150°C, but in Long Valley temperatures between 200°C and 250°C are indicated and a potential for generation of electricity seems to exist. Boron content in excess of 2.0 ppm is common in all areas, although a few scattered samples show lower values. Dissolved solids range from about 1,000 to 3,500 ppm except at Honey Lake and Modoc Plateau, where the values average less than 1,000 ppm.

### Tabular Data

Data concerning thermal spring or well location, temperature, pH, chemical analysis, flow volume, dissolved solids, etc., are presented in table form for each thermal area discussed.

## Imperial Valley Area

### Geologic Setting

The Imperial Valley geothermal region is the landward extension of the Gulf of California rift. It is roughly aligned along the East Pacific rise and the San Andreas fault system. The depression narrows and topographically rises in the Coachella Valley north of the Salton Sea and is terminated by the junction of the San Jacinto Range with the Transverse Ranges. To the southeast, it gradually widens into the Gulf of California. The Salton - Imperial rift is generally bounded by faults and diagonally transected by normal and strike-slip faults that are genetically associated with active, modern seismicity. The trough is filled with up to six km of water-saturated, weakly consolidated sand and silt of the Colorado River Delta and coarse detritus of local marginal origin which have been accumulating on granitic basement since earliest Pliocene time. The Salton Sea geothermal brines contain 220,000 to 260,000 ppm total weight of dissolved solids which are rich in chlorides of sodium, calcium and potassium and have significant amounts of lithium, cesium, manganese, silver, iron, strontium, barium, lead, zinc and copper. This is one of the highest natural concentrations of salts known in the world. Major problems of brine deposition, corrosion and disposal have retarded development at the Salton Sea field.

The Colorado River Delta sediments consist of dominant quartz and calcite, with lesser amounts of dolomite, plagioclase, potassium feldspars, mica and clay; and gypsum occurs sporadically throughout the sediments (Muffler and Doe, 1968). Metamorphic reactions occur in the sediments, and greenschist facies minerals are forming at temperatures up to 300°C at depths of 900 m or more (Muffler and White, 1969).

There is little rainfall in the Imperial Valley (average 1.5 cm/yr), and water from the Colorado River maintains the ground water table at or near the surface in undeveloped areas of the valley. For agricultural use, the

land is drained to lower the water table three to four miles below the surface. Many artesian wells have been drilled along the eastern side of the valley for domestic or stock use.

Crustal thinning has occurred in the Salton Trough (Biehler and others, 1964), and the geothermal activity in the Imperial Valley appears to be related to heat from the mantle moving through the thinner crust.

#### Water Chemistry

The chemistry and isotopic composition of water samples from the Imperial Valley shows a dominating influence by Colorado River water. Subsurface flow of the Colorado River water appears to have entered the valley from the southeast and moved to all parts of the area sampled, but the age of the ground water is unknown. Runoff water from the local mountains seems to have its greatest contribution in the northeastern-most samples where isotopic and chemical differences are greatest.

Water from 30 of the wells was of the sodium-chloride type, and the remainder was of the sodium-bicarbonate type. The occurrence of the sodium-bicarbonate water appears to be random within the area sampled, but is restricted to depths of 150 to 300 m. Sulfate is a significant anion in some of the samples and may reflect the occurrence of gypsum and anhydrite in the sediments. Generally, higher concentrations of major constituents correlate with higher temperature and greater depth. The chemistry of the subsurface water is primarily controlled by the solubility of minerals within the upper part of the sedimentary section, and the chemical indicators of temperature suggest that the water had never been in equilibrium with the rock of a high temperature system.

There seems to be little contribution from the deep geothermal system thought to exist in the area near Brawley. High bicarbonate content of certain samples, especially samples 2 and 3, could be indicative of the

metamorphism of calcite described by Muffler and White (1968). The high concentration of boron found in many analyses may also be a contribution from deeper, high temperature water.

The discharge volume of thermal springs is generally low, ranging up to 240 liters per minute but averaging less than 100 l/min. Calculated dissolved solids range from 1020 ppm to 3810 ppm, and calculated temperatures do not exceed 116°C. The hot brine system of the Salton Sea area was not included in Reed's report, but dissolved solids in excess of 300,000 ppm have been reported.

## Long Valley

### Geologic Setting

The Long Valley Caldera formed by crustal collapse 680,000 years ago following the explosive eruption of the rhyolitic Bishop Tuff, and intermittent volcanic activity continued in the caldera up to 650 years ago (Bailey and others, 1973). It is estimated from geophysical evidence that between 2700 m and 4000 m of accumulated sediments, flows, and pyroclastic material overlie granitic basement in the eastern half of the caldera. The deepest well drilled in the caldera penetrated 324 m of sediment and pyroclastic rock in the Casa Diablo area. The hydrothermal activity in the area may be related to a cooling magma chamber beneath the caldera, and the recent history of activity makes a magmatic heat source most likely for the geothermal system.

### Water Chemistry

The chemical constituents in the water appear to be controlled by mineral solution reactions at temperatures between 200°C and 250°C. The agreement of several of the chemical indicators of temperature implies that the water reached equilibrium in some reservoir and moved rapidly to the surface without re-equilibration. The great depth of the basin fill makes

it possible that a deeper, hotter reservoir exists with no communication directly to the surface. Boron content ranges from 0.18 ppm to 15.0 ppm. Calculated dissolved solids range from 1290 ppm to 1810 ppm, but two samples show anomolous values of 202 ppm and 235 ppm. Flow rates are in the range of 100 to 400 liters per minute, with the exception of sample locality 4. The flow rate there is 3200 l/min. Calculated water temperatures exceed 100°C at all but one of the sampled locations. Temperatures in excess of 200°C are indicated at two locations.

## Sierra Valley

### Geologic Setting

Sierra Valley is an intermountain basin of the Basin and Range Structural Province and is situated just east of the crest of the Sierra Nevada Mountains. Late Pliocene (post-Warner basalt) block faulting created the valley, and its form has changed little since that time (Durrell, 1966). The Sierra Valley was the site of a large lake from late Pliocene until late Pleistocene, when it and the Mohawk Lake (Durrell, 1966) downstream were drained by the Feather River. Headward erosion by the river has not yet reached Sierra Valley and the lake sediments have not been dissected. A geothermal test well, drilled in the center of the valley, penetrated 400 m of lake sediments and 250 m of rhyolite tuff and breccia (possibly Miocene Delleker formation) before entering the biotite granite basement.

The elevation of Sierra Valley is approximately 1500 m, and the shoreline of the former lake lies at about 1550 m. The valley is bordered by structural blocks of Cretaceous granitic rocks and Tertiary andesite and basalt. Several northwest trending faults cut the valley sediments and the igneous rocks on both sides. It is not known if the faulting is still active; but faults through the valley are visible from the air, because they disrupt the ground water flow and the vegetation on the surface. The valley has an

average precipitation of about 50 cm/yr and collects a great deal of runoff from the surrounding mountains. The ground water table is near the surface in most of the valley, and extensive marshes exist in the central and northern portions of the valley.

#### Water Chemistry

Sample 9, from Campbell Hot Springs, is a sodium-bicarbonate chloride type water issuing from Tertiary andesite. Sample 8, from a cold artesian well, is a sodium-bicarbonate water thought to represent local meteoric water. The remaining samples, from hot artesian wells, are sodium-chloride type water.

All of the samples from the hot artesian wells show a very similar chemistry, indicating that they are only various dilutions of the same thermal source. It appears that deep circulating ground water is rising along a northwest striking fault in the area of sample 5. Hot water (above 130°C) then moves into the permeable sand which is tapped by the well at 5. Leakage from the fault into other aquifers may feed the other wells.

There is little likelihood of a significant quantity of high temperature water in this area. Thermal energy seems localized along a bedrock fracture zone, and the deep circulation of ground water appears to be the major means of heat transfer to the surface.

Boron content ranges from 0.07 to 8.8 ppm. Only two localities show values of less than 7.3 ppm. Dissolved solids exceed 980 ppm for all but two samples, which are less than 370 ppm. Discharge rates are very low. Calculated temperatures are generally grouped in the range of 110 to 140°C as determined by quartz solubility.

## Honey Lake Valley

### Geologic Setting

Honey Lake Valley is a part of the Basin and Range Structural Province. The valley is bounded on the southwest by the Honey Lake Fault and the uplifted granitic and volcanic Diamond Mountains block. There appears to be more than 2400 m of vertical offset on the Honey Lake Fault since the late Pliocene (post-Warner basalt). The valley is bordered on the northeast by the Pliocene basalts of the Skedaddle Mountains and on the north by the Pliocene and Pliestocene basalts of the Modoc Plateau.

An unpublished gravity survey for the California Department of Water Resources indicates that granite basement lies at a depth of  $1600 \text{ m} \pm 100 \text{ m}$  below the center of the valley. A dry gas test well drilled northwest of Herlong penetrated 250 m of Quaternary lake deposits, 820 m of interbedded sandstone and shale, and 140 m of andesite breccia and flows. Long lines of calcareous tufa deposits lead back toward the Skedaddle Mountains from both Wendel and Amedee Hot Springs on the northeast shore of Honey Lake. The tufa deposits trace the existence of two faults by marking the former positions of hot springs.

The valley is a closed drainage basin, and the entire basin has an average precipitation of about 6 cm/yr. The local basalts have a high permeability and supply significant recharge to the ground water in storage. Honey Lake, an alkali lake, occupies the lowest part of the basin throughout most of the year.



## Water Chemistry

The dilute water in samples 1 and 4 are of the sodium-bicarbonate type, and the remaining three samples are sodium-sulfate type water. The sodium-bicarbonate type water is thought to represent local meteoric water. The chemical geothermometers seem to be unreliable in this area. Because of extensive volcanism in the area, it is possible that a significant amount of volcanic ash is contained in the valley sediments. If the silica in solution were in equilibrium with cristobalite, this would explain higher concentrations of silica than expected from quartz solubility. The ash could also supply a higher than normal amount of potassium to the solution and affect the calculated temperatures from the cation ratios.

Samples from both Wendel and Amedee Springs appear to be from the same thermal reservoir containing water above 140°C. Deep circulation of ground water seems to be the source of this thermal system.

Flow volume is greatest at Wendel Hot Springs where a discharge of 1200 l/min was measured. Boron content ranges from less than 0.02 ppm to 5.5 ppm. Dissolved solids range from 233 ppm to 1040 ppm.

## Surprise Valley

### Geologic Setting

Surprise Valley is a graben lying between the horsts of the Warner Mountains on the west and the Hays Canyon Range on the east. At least 2000 m of sediments and volcanic debris have accumulated in the valley, and the crest of the Warner Mountains is up to 1500 m above the valley floor. Vertical displacement on the Surprise Valley Fault, which marks the west edge of the valley, is thus greater than 3500 m. Scarps in alluvium indicate recent motion, and radiometric ages suggest faulting began less than 15 million years ago (Duffield and Fournier, 1974).

Quaternary rhyolite flows and obsidian bodies crop out in the Fandango Valley of the northern Warner Mountains and also near the south end of the range (Duffield and Fournier, 1974). The possibility exists that the volcanic rocks are related to a shallow crustal magmatic heat source in the area.

Surface and ground water is supplied by runoff mainly from the Warner Mountains; and streams, springs and artesian wells are numerous on the western side of the valley. The east side is an alkali desert. Surprise Valley forms a closed basin, and lakes have occupied the basin since the late Tertiary. At present, three shallow, alkali lakes exist on the valley floor during most of the year.

### Water Chemistry

All but one of the samples from Surprise Valley are sodium-sulfate type water. The sample from Fort Bidwell is a sodium-bicarbonate type water. The thermal water appears to rise along several faults which are subparallel to the sides of the valley.

The chemical compositions and temperature indicators of the samples seem to reflect shallow aquifers with water up to 170°C. The greater depth of the valley fill makes it possible that a deeper geothermal reservoir exists but is not reflected in water chemistry. Duffield and Fournier (1974) have applied various mixing models to chemical analyses from Surprise Valley wells and springs, and they calculate temperatures of 220°C for the Leonards Hot Springs and Fort Bidwell areas.

Boron content ranges from 0.61 ppm to 7.6 ppm. Calculated temperatures generally exceed 90°C, but are less than 180°C. Dissolved solids range from 370 ppm to 1210 ppm. Discharge volumes do not exceed 500 l/min.

## Modoc Plateau, Modoc, Lassen, & Shasta Counties

### Geologic Setting

The Modoc Plateau is an area of extensive flood basalts and pervasive normal faulting. Large quantities of Tertiary volcanic rocks covered the area, and volcanism and faulting have continued into the Holocene with a diminishing rate of activity (MacDonald, 1966). The area has been generally uplifted to elevations between 1500 m and 1800 m, and a very large number of north to northwest trending faults cut the plateau.

High permeability, flood basalts control the hydrology in the Modoc Plateau. Through most of the region the water table is at an altitude of 1220 m to 1250 m, and most of the surface flow is drained to the water table in areas over 1250 m elevation (MacDonald, 1966). The plateau receives an average precipitation of about 25 cm/yr, but in some areas there is no surface flow at all.

The Likely Fault Zone is a large, northwest striking, structural feature cutting across the Modoc Plateau southwest of Alturas. The fault appears to have normal vertical offset, but it is unclear if any horizontal offset is present (Duffield and Fournier, 1974). Kelley Hot Spring (sample 1) may be associated with the Likely Fault. A geothermal test well drilled near the spring penetrated 977 m of volcanic and sedimentary rock, and the bottom of the well is possibly in the Cedarville Series of Miocene age.

Hunt Hot Springs (sample 7) issue from the contact between Eocene conglomerate of the Montgomery Creek Formation and the Jurassic, Bagley andesite. Samples 2 through 6 came from valleys covered with Quaternary continental sediments or pyroclastic rocks.

## Water Chemistry

Sample 2, from the Williams Ranch well, is a dilute, sodium-bicarbonate type water; and the remaining samples are more concentrated sodium-sulfate type water. The chemical indicators of temperature give values below 155°C for all the samples. It is possible that all the samples in this group are mixtures of cold near-surface water with deeper thermal water.

Several of these samples could come from hot water rising along faults. Faulting is pervasive in the plateau; but, except for Kelley Hot Spring, none of the sampling localities is close to mapped faults. Because of the highly permeable basalts underlying most of the region, it is probable that all the hot springs are rising along structural barriers to the ground water flow.

Boron content ranges from 0.22 ppm to 12.8 ppm. Calculated dissolved solids range from 231 ppm to 1220 ppm. Discharge volumes are less than 300 l/min, except for Kelley Hot Spring, where a flow of 1250 l/min was measured.

## Conclusions

In evaluating the results of the six areas studied, Reed concludes that the thermal waters generally show the effects of dilution by cold, near-surface water, which adversely affects the use of chemical geothermometers. Many of the samples showed the effects of reacting with the surrounding rocks at above-normal temperatures; but the indicated temperatures were, in most cases, below 150°C.

Table 1a. Imperial Valley Artesian Wells

Sample Number	Location	Owner or Operator	Date Sampled (mo/dy/yr)	Producing Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (mmho/cm)	Discharge (l/min)	SO <sub>4</sub> <sup>18</sup> (% <sub>v</sub> )	SiH <sub>2</sub> <sup>2</sup> (% <sub>v</sub> )
1	NW/SE Sec. 23 T. 12S., R. 15E., S.B.	D. Brownell	12/15/70	86.9-99.1	-22.8	31.5	7.85	3.17			
2	SE/SE Sec. 27 T. 12S., R. 15E., S.B.	G. Brownell	12/15/70	125.0-131.1	-25.9	34.1	8.12	2.75		-7.3	-63
3	NE/NE Sec. 35 T. 12S., R. 15E., S.B.	Cowell	12/15/70	92.7-104.9	-19.8	33.0	8.13	2.61			
4	SW/SE Sec. 31 T. 12S., R. 16E., S.B.	P. Reblak	12/16/70	281.9	-7.6	39.1	7.79	6.08	20		
5	SE/SE Sec. 1 T. 13S., R. 15E., S.B.	Taylor	12/15/70	325.2-331.9	-19.8	55.2	7.66	6.90	64	-9.3	
6	SW/SE Sec. 3 T. 13S., R. 15E., S.B.	Malberry School	12/14/70	271.3	-34.4	40.3	8.12	2.70			
7	NW/NE Sec. 5 T. 13S., R. 15E., S.B.	J. Williams	12/14/70	259.4-264.0	-43.3	35.7	7.80	2.89	40		
8	NE/NE Sec. 5 T. 13S., R. 15E., S.B.	Wiest Store	12/14/70	235.3-247.5	-42.1	37.7	7.83	2.55			
9	SW/SE Sec. 16 T. 13S., R. 15E., S.B.	M. Lunceford	12/14/70	231.6	-36.0	39.4	8.29	2.58	40		
10	SW/SE Sec. 23 T. 13S., R. 15E., S.B.	J. Ratliff	12/13/70	396.2	-25.0	55.7	7.74	6.15	160	-8.8	
11	SW/SE Sec. 24 T. 13S., R. 15E., S.B.	V. Butters	12/13/70	213.4	-22.6	42.8	8.36	2.84	60	-7.6	
12	NW/NE Sec. 32 T. 13S., R. 15E., S.B.	T. Shank	12/12/70	304.8	-38.7	43.5	7.82	4.76			

Table extracted from: Reed, M. J., Chemistry of Thermal Water in Selected Geothermal Areas of California, 38 pgs., (1974)

Table 1a. Imperial Valley Artesian Wells

Sample Number	Location	Owner or Operator	Date Sampled (mo/day/yr)	Producing Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (µmho/cm)	Discharge (l/min)	60.18 (µmho/cm) $\times 10^{-2}$
13	NE/NE Sec. 33 T. 13S., R. 15E., S.B.	Magnolia School	12/12/70	386.8-423.4	-33.5	50.9	7.70	6.61	140	
14	NW/SW Sec. 34 T. 13S., R. 15E., S.B.	M. Phegley	12/13/70	285.3-290.8	-31.4	44.3	8.17	3.14	40	-8.9
15	SE/SE Sec. 6 T. 13S., R. 16E., S.B.	F. Schoneman	12/16/70	187.7	-11.6	33.3	8.16	2.83		
16	SE/SW Sec. 6 T. 13S., R. 16E., S.B.	T. Olesh	12/16/70	91.4	-15.2	32.0	8.12	3.14	20	
17	NW/NE Sec. 6 T. 14S., R. 15E., S.B.	N. Fifield	12/12/70	384.2-393.2	-40.2	51.4	7.77	6.72		
18	SW/SW Sec. 9 T. 14S., R. 15E., S.B.	J. Birger	12/11/70	117.3	-34.4	31.1	7.79	4.79	20	
19	NW/NW Sec. 11 T. 14S., R. 15E., S.B.	Moiola Feed Lot	12/13/70	198.1	-26.8	42.0	8.20	2.80	28	
20	SW/SW Sec. 12 T. 14S., R. 15E., S.B.	Mendiburu Feed Lot	12/13/70	356.9-375.8	-21.9	51.7	7.63	5.71		
21	NW/NE Sec. 15 T. 14S., R. 15E., S.B.	A. Gislser	12/11/70	355.1	-29.0	47.5	8.08	3.56	24.0	-9.9
22	NW/SW Sec. 23 T. 14S., R. 15E., S.B.	J. Birger	12/11/70	228.6	-25.9	39.3	8.16	2.59	20	
23	NE/NE Sec. 27 T. 14S., R. 15E., S.B.	J. Birger	12/11/70	121.9	-26.8	31.8	7.87	2.51	16	
24	NW/NE Sec. 34 T. 14S., R. 15E., S.B.	Jenson	12/11/70	108.8	-26.8	29.5	8.01	2.83	120	



Table 1a. Imperial Valley Artesian Wells

Sample Number	Location	Owner or Operator	Date Sampled (mo/dy/yr)	Producing Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (mho/cm)	Discharge (l/min)	$\text{SO}_4^{2-}$ (mg/l)	$\text{SiO}_2$ (mg/l)
25	SW/SE Sec. 34 T. 14S., R. 15E., S.B.	Geddis	12/11/70	185.9	-24.4	35.5	7.95	3.07	60	-9.9	
26	SW/SE Sec. 4 T. 14S., R. 16E., S.B.	F. Borchard	12/10/70	118.9-139.0	-4.6	37.3	8.35	1.99			
27	SW/SE Sec. 4 T. 14S., R. 16E., S.B.	F. Borchard	12/10/70	139.3	-4.6	38.4	9.38	1.96	48		
28	SE/NE Sec. 11 T. 14S., R. 16E., S.B.	U.S.-B.L.M.	12/16/70	86.9	7.6	34.5	8.23	2.26		-7.2	-62
29	SW/SE Sec. 16 T. 14S., R. 16E., S.B.	M. Axler	12/12/70	121.9	-5.2	25.4	8.17	1.41	4	-7.2	
30	SW/NE Sec. 21 T. 14S., R. 16E., S.B.	S. Stacey	12/10/70	133.2	-4.9	32.5	8.32	1.85	20		
31	SW/NW Sec. 22 T. 14S., R. 16E., S.B.	C. Singh	12/10/70	212.8-216.1	-2.1	41.7	7.91	4.95	20	-9.9	
32	SW/NE Sec. 10 T. 15S., R. 15E., S.B.	Shawner	12/3/70	140.2	-22.6	31.6	7.77	3.08	108	-7.8	
33	SE/NE Sec. 14 T. 15S., R. 15E., S.B.	C. Allen	12/3/70	263.3	-13.7	40.0	8.14	2.60	40	-9.1	
34	SW/NE Sec. 26 T. 15S., R. 15E., S.B.	J. DePaoli	11/29/70	240.8-289.6	-7.6	40.1	7.97	2.58		-9.4	-90
35	NE/NE Sec. 35 T. 15S., R. 15E., S.B.	Holtville Ice Co.	11/29/70	335.3	-5.5	44.6	7.96	4.33	116	-10.5	-94
36	SW/NW Sec. 36 T. 15S., R. 15E., S.B.	City of Holtville	11/28/70	255.4-259.7	-4.6	29.0	7.77	1.71	12	-7.8	

Table 1a. Imperial Valley Artesian Wells

Sample Number	Location	Owner or Operator	Date Sampled (mo/dy/yr)	Producing Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (mmho/cm)	Discharge (l/min)	SO <sub>4</sub> <sup>2-</sup> (%)	SH <sup>2+</sup> %
37	SE/SE Sec. 7 T. 15S., R. 16E., S.B.	Hooke	12/3/70	202.4-211.8	-11.3	35.6	8.25	2.20	12		
38	SW/NE Sec. 8 T. 15S., R. 16E., S.B.	G. Hoyt	12/3/70	144.8-148.7	-10.4	30.8	8.12	2.27	4		
39	SE/SW Sec. 15 T. 15S., R. 16E., S.B.	R. Garowal	12/2/70	243.6	0	32.0	8.39	1.92			-7.2
40	SW/NE Sec. 19 T. 15S., R. 16E., S.B.	F. Strahm	12/2/70	254.2	-8.2	35.6	8.36	1.53	40		-7.4
41	NE/SW Sec. 22 T. 15S., R. 16E., S.B.	D. Starr	12/2/70	228.6	0.3	34.7	8.15	2.88	12		-9.8
42	SE/NE Sec. 23 T. 15S., R. 16E., S.B.	L. Foster	12/2/70	137.8-165.2	4.6	34.1	8.38	1.75	100		-8.2
43	SW/SE Sec. 29 T. 15S., R. 16E., S.B.	A. Fusi	11/30/70	163.7-187.8	-3.0	30.6	8.26	1.19	8		-6.9
44	SW/SE Sec. 30 T. 15S., R. 16E., S.B.	A. Fusi, J.	11/30/70	270.7-285.9	-3.0	39.7	8.17	2.26			
45	SE/NE Sec. 4 T. 16S., R. 16E., S.B.	C. Anstiel	11/30/70	268.2-285.9	0	34.5	8.22	2.96			
46	NE/NE Sec. 14 T. 16S., R. 16E., S.B.	Watton Labor Camp	12/1/70	343.8	5.2	43.0	8.09	3.29			-10.0
47	NE/NE Sec. 15 T. 16S., R. 16E., S.B.	Alamo School (abd.)	12/1/70	263.3-267.3	3.7	37.2	8.33	2.21	16		-9.0
48	NE/NE Sec. 9 T. 17S., R. 16E., S.B.	L. Bornt	12/9/70	184.1-216.4	9.4	37.2	8.13	2.35			-10.3

Table 1b. Chemical Constituents (in mg/l)

Sample Number	Location	Cations										Anions				Other		Calculated Dissolved Solids
		Li	Na	K	Mg	Ca	Sr	Fe	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B			
1	NW/SW Sec. 23	0.50	670	5.5	23	32	1.0	0.24	1.9	620	360	2.3	450	32	2.6	2230		
2	SE/SE Sec. 27	0.24	530	3.7	9.7	19	0.4	0.09	1.5	510	290	3.4	300	36	2.7	1710		
3	NE/NE Sec. 35	0.28	510	3.8	11	19	0.57	0.05	1.6	520	290	3.5	280	33	2.7	1680		
4	SH/SW Sec. 31	0.68	1200	7.7	11	37	1.7	0.25	1.3	1700	290	2.0	60	27	8.9	3350		
5	SE/NE Sec. 1	0.63	1100	9.4	14	43	1.6	0.03	1.3	1200	400	2.5	620	33	9.7	3440		
6	SH/SW Sec. 3	0.17	500	3.0	4.2	9.7	0.25	0.19	2.0	450	480	6.0	120	38	2.4	1620		
7	NW/NW Sec. 5	0.32	620	5.4	11	11	0.25	0.20	1.4	230	1300	7.7	100	36	2.6	2350		
8	NW/NW Sec. 5	0.20	550	4.5	8	8.7	0.18	0.28	1.4	200	1200	8.7	55	43	5.8	2500		
9	SW/SE Sec. 16	0.18	490	3.1	3.3	9.9	0.14	0.10	1.9	400	540	9.8	120	30	2.4	1610		
10	SH/SE Sec. 23	0.56	980	8.7	10	29	0.95	0.12	1.3	960	450	3.2	540	33	3.7	3020		
11	SH/SE Sec. 24	0.17	500	2.9	3.3	8.7	0.16	0.05	1.4	450	470	10	130	29	1.8	1610		
12	NW/NW Sec. 32	0.13	900	4.5	12	19	1.2	0.17	1.2	820	680	4.9	160	30	5.7	2660		
13	NE/NE Sec. 33	0.51	1100	8.9	14	29	1.4	0.19	1.0	1400	550	3.5	270	32	2.2	3410		
14	NW/SW Sec. 34	0.14	570	3.4	4.0	8.8	0.21	0.12	1.9	380	830	13	100	44	4.9	1560		
15	SE/SE Sec. 6	0.32	560	3.8	10	21	0.87	0.03	1.2	610	250	3.2	250	22	2.5	1730		
16	SE/SW Sec. 6	0.41	580	4.1	9.3	23	0.83	0.04	1.0	660	270	3.1	240	31	4.7	1830		
17	NW/NE Sec. 6	0.23	1200	6.9	15	40	1.2	0.32	1.0	1300	890	6.9	310	35	2.0	3210		
18	SH/SW Sec. 9	0.17	1000	5.9	45	52	1.3	0.19	1.2	1200	560	3.3	490	36	3.7	3400		

Table 1b. Chemical Constituents (in mg/l)

Sample Number	Location	Li	Cations							Anions				Other		Calculated Dissolved Solids
			Na	K	Mg	Ca	Sr	Fe	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B	
19	NW/NW Sec. 11	0.12	520	3.1	4.2	10	0.13	0.20	2.0	340	810	12	81	31	5.5	1820
20	SH/SH Sec. 12	0.42	940	8.0	10	33	0.75	0.11	1.2	1000	380	2.3	550	33	6.8	2970
21	NW/NE Sec. 15	0.15	610	3.7	4.2	22	0.47	0.19	1.6	490	800	11	130	36	13	2120
22	NW/SH Sec. 23	0.11	500	3.0	3.9	8.7	0.07	0.12	2.3	350	740	10	87	37	2.6	1740
23	NE/NE Sec. 27	0.09	530	4.2	15	26	0.49	0.24	1.7	330	700	4.4	280	24	3.7	1920
24	NW/NE Sec. 34	0.11	590	4.2	13	29	0.59	0.60	1.3	530	740	6.3	190	33	2.8	2140
25	SH/SE Sec. 34	0.11	620	4.6	17	30	1.3	0.34	1.3	420	650	5.4	370	24	2.7	2150
26	SH/SE Sec. 4	0.12	390	2.0	2.1	6.2	0.20	0.05	2.9	200	610	12	120	23	1.1	1770
27	SH/SE Sec. 4	0.11	390	2.1	2.0	4.8	0.12	0.07	3.3	170	640	14	130	23	7.6	1390
28	SE/NE Sec. 11	0.25	420	3.0	5.2	12	0.36	0.03	1.3	500	200	2.9	130	30	9.0	1310
29	NW/SE Sec. 16	0.08	310	2.2	5.3	13	0.53	0.10	1.6	300	310	3.1	57	18	0.8	1020
30	NW/NE Sec. 21	0.08	380	2.0	3.0	7.6	0.31	0.06	2.6	300	440	7.4	100	28	2.8	1270
31	NW/NW Sec. 22	0.36	900	5.7	8.2	31	0.61	0.22	1.7	1200	270	2.4	130	28	2.7	2580
32	SH/NE Sec. 10	0.12	670	4.9	17	22	0.95	0.12	1.8	600	670	3.4	190	37	1.9	2220
33	SE/NE Sec. 14	0.11	490	2.9	4.0	11	0.50	0.16	2.0	320	790	10	84	33	2.5	1750
34	SH/NE Sec. 26	0.12	500	3.2	4.7	11	0.47	0.19	2.0	300	790	6.9	110	26	3.6	1760
35	NE/NE Sec. 35	0.21	790	4.6	6.1	20	0.72	0.08	1.4	950	370	3.6	150	29	3.6	2330
36	NW/NW Sec. 36	0.04	350	2.5	7.9	15	0.44	0.32	1.7	290	490	2.1	79	23	3.2	1270

Table 1b. Chemical Constituents (in mg/l)

Sample Number	Location	Li	Cations							Anions				Other		Calculated Dissolved Solids
			Na	K	Mg	Ca	Sr	Fe	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B	
37	SE/SE Sec. 7	0.11	460	2.5	3.5	9.1	0.15	0.12	2.8	260	710	11	110	38	4.4	1610
38	SW/NW Sec. 8	0.08	510	3.3	8.6	16	0.39	0.42	1.4	280	880	9.4	130	24	6.7	1670
39	SE/SW Sec. 15	0.12	420	2.2	2.7	7.6	0.10	0.10	3.9	220	600	12	160	25	2.7	1460
40	SW/NW Sec. 19	0.05	320	1.8	2.7	6.5	0.11	0.16	3.2	72	650	12	100	35	1.2	1230
41	NE/SW Sec. 22	0.16	570	3.7	5.5	17	0.44	0.14	2.2	600	430	5.4	170	22	2.9	1650
42	SE/NW Sec. 23	0.13	370	2.1	2.0	6.7	0.09	0.04	3.4	160	580	12	140	23	3.5	1350
43	SW/SE Sec. 29	0.04	260	2.0	2.5	7.1	0.25	0.40	2.9	40	580	7.6	110	21	1.9	1040
44	SW/SE Sec. 30	0.09	450	2.5	3.3	9.0	0.18	0.20	2.3	260	690	9.4	130	41	6.7	1600
45	SE/NW Sec. 4	0.12	610	3.5	5.2	11	0.31	0.24	1.6	490	850	13	74	22	3.2	2080
46	NE/RE Sec. 14	0.25	560	3.6	3.5	17	0.57	0.05	2.1	630	310	3.8	120	26	3.4	1680
47	NW/NE Sec. 15	0.10	420	2.4	2.7	10	0.41	0.05	2.3	310	510	9.3	130	25	2.2	1420
48	NE/NW Sec. 9	0.11	450	2.9	4.2	17	0.70	0.12	1.7	400	370	4.4	180	29	1.5	1450

Table 1c. Calculated Temperatures, Imperial Valley (in °C)

Sample Number	Location	Quartz Solubility (regular) (with steam loss)	Na/K	Na, K, Ca (B = 1/3) (B = 4/3)	Measured Temperature
1	NW/SW Sec. 23	92	6	89 90	31.5
2	SE/SE Sec. 27	87	-1	84 85	34.1
3	NE/NE Sec. 35	83	2	85 86	33.0
4	SW/SW Sec. 31	75	-5	86 106	39.1
5	SE/NE Sec. 1	83	8	94 108	55.2
6	SW/SW Sec. 3	89	-7	82 93	40.3
7	NW/NW Sec. 5	87	8	96 115	35.7
8	NW/NW Sec. 5	95	7	95 111	37.7
9	SW/SE Sec. 16	79	-5	83 93	39.4
10	SW/SE Sec. 23	83	9	97 114	55.7
11	SW/SW Sec. 24	78	-9	81 94	42.8
12	NW/NW Sec. 32	79	-14	78 99	43.5
13	NE/NE Sec. 33	82	5	95 116	50.9
14	NW/SW Sec. 34	96	-8	83 101	44.3
15	SE/SE Sec. 6	67	-2	83 85	33.3
16	SE/SW Sec. 6	81	-1	84 86	32.0
17	NW/NE Sec. 6	86	-9	82 100	51.4
18	SW/SW Sec. 9	87	-8	79 86	31.1
19	NW/NW Sec. 11	81	-8	82 94	42.0

Table 1c. Calculated Temperatures, Imperial Valley (in °C)

Sample Number	Location	Quartz Solubility (regular)	Quartz Solubility (with steam loss)	Na/K	Na, K, Ca (B = 1/3)	Na, K, Ca (B = 4/3)	Measured Temperature
20	SW/SW Sec. 12	83		7	94	107	51.7
21	NW/NE Sec. 15	87		-7	80	84	47.5
22	NW/SW Sec. 23	88		-7	82	96	39.3
23	NE/NE Sec. 27	70		4	85	83	31.8
24	NW/NE Sec. 34	83		0	83	81	29.5
25	SW/SE Sec. 34	70		1	85	84	35.5
26	SW/SE Sec. 4	69		-13	76	85	37.3
27	SW/SE Sec. 4	69		-12	79	94	38.4
28	SE/NE Sec. 11	79		0	84	85	34.5
29	NW/SE Sec. 16	60		-1	80	71	25.4
30	NW/NE Sec. 21	77		-12	76	82	32.5
31	NW/NW Sec. 22	77		-5	83	95	41.7
32	NW/NE Sec. 10	88		1	87	95	31.6
33	SE/NE Sec. 14	83		-8	80	89	40.0
34	SW/NE Sec. 26	74		-5	83	92	40.1
35	NE/NE Sec. 35	78		-9	82	97	44.6
36	NW/NW Sec. 36	69		0	81	73	29.0
37	SE/SE Sec. 7	89		-11	78	87	35.6
38	SW/NW Sec. 8	70		-4	82	85	30.8

Table 1c. Calculated Temperatures, Imperial Valley (in °C)

Sample Number	Location	Quartz Solubility		Na, K, Ca		Measured Temperature	
		(regular)	(with steam loss)	Na/K (B = 1/3)	(B = 4/3)		
39	SE/SW Sec. 15	72		-13	77	86	32.0
40	SW/NW Sec. 19	86		-10	77	80	35.6
41	NE/SW Sec. 22	67		-4	83	89	34.7
42	SE/NW Sec. 23	69		-10	78	86	34.1
43	SW/SE Sec. 29	65		3	84	79	30.6
44	SW/SE Sec. 30	93		-10	78	87	39.7
45	SE/NW Sec. 4	67		-9	82	96	34.5
46	NE/NE Sec. 14	74		-5	82	88	43.0
47	NW/NE Sec. 15	72		-9	78	83	37.2
48	NE/NW Sec. 9	78		-4	80	78	37.2



Table 2a. Long Valley Springs and Wells

Sample Number	Location	Name	Date Sampled (mo/dy/yr)	Altitude (m)	Water Temperature (°C)	pH	Conductivity @ 25°C (mmho/cm)	Discharge (l/min)	SO <sup>18</sup> (‰)	SH <sup>2</sup> (‰)
1	NE/NE Sec. 25 T. 2S., R. 27E., M.D.	Big Springs	5/21/72	2220	11	6.83	0.182	380	-15.89	-115.4
2	SW/NW Sec. 13 T. 3S., R. 28E., M.D.	Little Hot Creek hot spring	5/13/72	2135	79	6.51	1.95	280	-15.34	-121.8
3	NE/NE Sec. 25 T. 3S., R. 28E., M.D.	Hot Creek hot spring	8/29/73	2145	90	7.89	1.77	400		
4	SW/NW Sec. 12 T. 3S., R. 28E., M.D.	"Endogenous" 5 Magma Power Co. (123.4 m deep)	5/19/72	2231	94	9.25	1.92	3260 (kg/min) mixed steam & water	-14.16	-115.8
5	SW/NW Sec. 35 T. 3S., R. 28E., M.D.	hot pool	5/24/72	2165	60	7.15	1.80	0	-12.44	-111.7
6	NE/NW Sec. 13 T. 3S., R. 29E., M.D.	artesian well (24.4 m deep)	5/23/72	2075	10	8.82	0.191	25	-17.07	-129.5
7	SW/SW Sec. 21 T. 3S., R. 29E., M.D.	hot spring	5/22/72	2105	56	6.53	1.79	103	-16.17	-123.9
8	SE/NE Sec. 28 T. 3S., R. 29E., M.D.	hot spring	5/22/72	2095	49	6.60	1.90	200	-15.85	-123.4
9	NE/NE Sec. 31 T. 3S., R. 29E., M.D.	hot spring	5/20/72	2130	58	7.53	1.50	190	-15.23	-121.2
10	NW/SE Sec. 34 T. 3S., R. 29E., M.D.	hot spring	5/23/72	2090	41	6.64	1.63	150	-16.08	-124.9

Table 2b. Chemical Constituents (in mg/l)

Sample Number	Location	Cations										Anions				Other		
		Li	Na	K	Rb	Cs	Mg	Ca	Sr	F	Cl	Br	I	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B
1	NE/NE Sec. 25	0.04	23	4.0	0.01	<0.05	5.9	5.1	0.07	0.5	5.7	0.20	90	0	8.1	58	0.37	<0.
2	SW/NW Sec. 13	2.8	410	30	0.26	0.30	0.6	50	0.60	8.4	200	0.80	735	0.3	96	110	10.5	2.
3	NE/NE Sec. 25	2.3	400	24	0.22	0.2	0.1	1.6	9.6	225			580	25.9	100	150	10.5	2.
4	SW/NW Sec. 32	2.8	390	45	0.48	0.60	0.1	0.9	0.14	12	280	1.10	450	29.6	130	340	15	10
5	SW/NW Sec. 35	2.5	380	25	0.28	0.45	0.1	3.3	0.10	11	230	0.80	466	0.7	120	300	13	1.
6	NE/NW Sec. 13	0.14	38	1.3	<0.01	<0.05	0.2	5.3	0.04	0.6	3.0	0.03	111	2.8	3.7	64	0.18	3.
7	SW/SW Sec. 21	1.5	310	37	0.11	0.10	0.6	25	0.20	4.6	150	0.50	828	0.3	68	250	7.7	0.
8	SZ/SE Sec. 28	1.7	400	43	0.14	0.05	0.6	22	0.14	4.8	170	0.60	845	0.3	69	240	8.8	0.
9	NE/NE Sec. 31	2.0	310	22	0.19	0.20	0.4	15	0.28	7.5	170	0.60	515	1.9	81	150	7.9	0.
10	NW/SE Sec. 34	1.6	320	28	0.08	0.10	1.2	23	4.6	150	0.50	0.4	695	0.3	59	205	3.1	0.

57  
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Table 2c. Trace Constituents (in mg/l)

Sample Number	Location	Al	Mn	Fe	Ni	Cu	Zn	As	Se	Sb	Hg	N (NH <sub>3</sub> )	P (PO <sub>4</sub> )	Calculated Dissolved Solids
1	NE/NE Sec. 25	0.002	< 0.02	< 0.05	< 0.05	< 0.03	0.045	0.02		< 0.1	< 0.0001	0.13	1.0	202
2	SW/NW Sec. 13	0.006	0.24	0.15	0.10	0.07	0.44	0.74	0.005	< 0.1	< 0.0001	0.40	0.15	1660
3	NE/NE Sec. 25	0.060	< 0.02	< 0.05	< 0.05	0.02	< 0.01				0.0013	0.50		1510
4	SW/NW Sec. 32		< 0.02	0.05	< 0.05	< 0.03	0.19	2.2	0.001	0.2	0.0001	0.40	0.18	1710
5	SW/NW Sec. 35	0.057	< 0.02	< 0.05	< 0.05	0.02	0.045	0.34	0.004	0.3	0.0003	0.15	0.71	1570
6	NE/NW Sec. 13	0.003	< 0.02	< 0.05	< 0.05	< 0.03	< 0.01	0.02	0.002	< 0.1	0.0001	0.35	0.15	235
7	SW/SW Sec. 21	0.002	0.10	0.23	< 0.05	0.04	0.18	0.46	0.005	< 0.1	< 0.0001	0.20	0.28	1890
8	SE/NE Sec. 23	0.003	0.08	< 0.05	< 0.05	0.03	0.12	0.34	0.004	< 0.1	0.0001	0.10	0.46	1810
9	NE/NE Sec. 31	0.006	< 0.02	< 0.05	< 0.05	< 0.03	0.02	0.84		< 0.1	0.0001	0.09	0.43	1290
10	NW/SE Sec. 34	< 0.002	0.085	0.45	< 0.05	< 0.03	0.10	0.36		< 0.1	< 0.0001	0.15	0.18	1500

Other trace constituents measured but below limit of detection:

- Cd < 0.01
- Co < 0.06
- Au < 0.1
- Pb < 0.1
- Ag < 0.04

Table 2d. Calculated Temperatures, Long Valley (in °C)

Sample Number	Location	Quartz Solubility		Na/K	Na, K, Ca		Measured Temperature
		(regular)	(with steam loss)		(B = 1/3)	(B = 4/3)	
1	NE/NE Sec. 25	109		259	187	82	11
2	SW/NW Sec. 13	143		150	171	138	79
3	NE/NE Sec. 25	161	153	131	191	257	90 (boiling)
4	SW/NW Sec. 32	219	200	201	238	344	94 (boiling)
5	SW/NW Sec. 35	209	192	140	189	225	60 (evaporation)
6	NE/NW Sec. 13	114		86	117	52	10
7	SW/SW Sec. 21	196		205	200	166	56
8	SE/NE Sec. 28	193		192	200	183	49
9	NE/NE Sec. 31	161		147	175	157	58
10	NW/SE Sec. 34	182		169	183	155	41

Table 3a. Sierra Valley, Artesian Wells and Spring

Sample Number	Location	Owner	Date Sampled (mo/dy/yr)	Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (mmho/cm)	Discharge (l/min)	SO <sub>4</sub> (%)	SiH <sup>2</sup> (%)
1	NW/SE Sec. 13 T. 22N., R. 14E., M.D.	Marble Hot Springs	6/20/73	104	1486	73.2	8.19	3.87	100		
2	SW/SE Sec. 13 T. 22N., R. 14E., M.D.	Marble Hot Springs	6/20/73	99	1486	70.3	7.96	3.74	80		
3	NW/NE Sec. 32 T. 22N., R. 15E., M.D.	W. Hagge	6/27/73	213	1486	39.8	7.61	2.39	2.5		
4	NW/NE Sec. 32 T. 22N., R. 15E., M.D.	W. Hagge	6/28/73	182	1486	38.6	7.48	1.62	2		
5	NE/SW Sec. 32 T. 22N., R. 15E., M.D.	G. Filipini	6/20/73	335	1487	94.2	7.97	4.72	50		
6	SE/SE Sec. 32 T. 22N., R. 15E., M.D.	W. Hagge	6/28/73	274	1487	44.1	8.00	2.55	4		
7	SW/NW Sec. 5 T. 21N., R. 15E., M.D.	G. Filipini	6/22/73	244	1489	51.3	7.42	2.71	8		
8	SW/SW Sec. 9 T. 21N., R. 15E., M.D.	A. Genascl	6/22/73	122	1498	18.2	7.68	0.198	12		
9	NW/NE Sec. 19 T. 20N., R. 15E., M.D.	Campbell Hot Springs	6/19/73	spring	1530	36.8	10.13	0.596	1		

Table 3b. Sierra Valley, Chemical Constituents (in mg/l)

Sample Number	Location	Cations										Anions					Others		Calculated Dissolved Solids
		Li	Na	K	Rb	Ce	Mg	Ca	Mn	Fe	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B		
1	NW/SE Sec. 13	0.42	410	7.6	0.06	0.2	0.1	37	0.022	< 0.06	2.2	460	41	0	340	60	7.4	1370	
2	SW/SE Sec. 13	0.40	400	7.4	0.05	0.2	0.1	38	0.03	< 0.06	2.2	460	51	1	330	62	7.3	1360	
3	NW/NE Sec. 32	0.50	400	6.3	0.05	0.1	1.3	26	0.10	0.63	1.5	440	76	0	270	80	7.8	1310	
4	NW/NE Sec. 32	0.36	300	5.3	0.03	0.1	1.4	12	0.075	1.43	1.0	300	87	0	190	77	6.2	980	
5	NE/SW Sec. 32	0.65	450	13	0.11	0.2	0.1	39	0.01	< 0.06	2.6	540	50	1	370	98	8.3	1570	
6	SE/SE Sec. 32	0.48	420	8.5	0.05	0.2	0.6	20	0.06	< 0.06	1.9	450	72	1	270	92	6.2	1110	
7	SW/NW Sec. 5	0.38	380	5.6	0.04	0.1	0.6	20	0.09	0.09	1.8	390	103	0	220	85	7.4	1210	
8	SW/SW Sec. 9	< 0.01	38	3.9	< 0.01	< 0.05	4.0	5.5	0.013	0.15	< 0.1	5.8	130	1	13	65	0.07	266	
9	NW/NE Sec. 19	0.06	92	1.1	< 0.01	< 0.05	< 0.1	2.0	< 0.01	< 0.06	< 0.1	52	86	25	27	82	1.6	369	

Trace Constituents Below Detection:

- Co < 0.04
- Ni < 0.04
- Cu < 0.01
- Zn < 0.01
- Cd < 0.01
- Pb < 0.06

Table 3c. Calculated Temperatures, Sierra Valley (in °C)

Sample Number	Location	Quartz Solubility		Na, K, Ca		Measured Temperature	
		(regular)	(with steam loss)	(B = 1/3)	(B = 4/3)		
1	NW/SE Sec. 13	111		46	112	92	73.2
2	SW/SE Sec. 13	112		46	111	90	70.3
3	NW/NE Sec. 32	125		37	107	93	39.8
4	NW/NE Sec. 32	123		44	113	102	38.6
5	NE/SW Sec. 32	136	132	74	131	112	94.2 (boiling)
6	SE/SE Sec. 32	133		51	120	112	44.1
7	SW/NW Sec. 5	128		34	106	95	51.3
8	SW/SW Sec. 9	115		187	167	85	18.2
9	NW/NE Sec. 19	126		23	95	76	36.8

Table 4a. Honey Lake Valley Wells and Springs

Sample Number	Location	Name	Date Sampled (mo/dy/yr)	Producing Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (micro/cm)	Discharge (l/min)	SO <sub>4</sub> <sup>2-</sup> (%)	SiO <sub>2</sub> (g/cc)
1	NE/NE Sec. 6 T. 29N., R. 12E., M.D.	Roosevelt Swimming Pool	7/18/73	90 (?)	1295	35.8	8.01	0.254	pumped		
2	SE/NE Sec. 6 T. 29N., R. 12E., M.D.	Latter Day Saints Church	7/18/73	169-181	1268	48.8	7.87	1.07	800 pumped		
3	SE/SE Sec. 23 T. 29N., R. 15E., M.D.	Wendel Hot Springs	7/17/73	spring	1231	95.6	8.38	3.34	1200		
4	NE/SW Sec. 30 T. 29N., R. 16E., M.D.	Southern Pacific Railroad	7/17/73	93	1223	28.2	8.33	0.332	300 pumped		
5	NW/NE Sec. 8 T. 28N., R. 16E., M.D.	Amadee Hot Springs	7/17/73	spring	1219	95.1	8.43	2.86	500		



Table 4b. Chemical Constituents (in mg/l)

Sample Number	Location	Cations										Anions		Others		Calculated Dissolved Solids
		Li	Na	K	Rb	Mg	Ca	Zn	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B	
1	NE/NE Sec. 6	<0.01	20	3.8	<0.01	3.4	19	0.043	<0.1	2.0	120	1	11	53	<0.02	233
2	SE/NE Sec. 6	0.05	140	4.6	0.02	1.6	24	0.009	1.2	64	68	1	190	62	1.4	556
3	SW/SE Sec. 23	0.12	280	7.5	0.04	<0.1	18	0.015	4.1	190	50	1	360	120	5.5	1040
4	NE/SW Sec. 30	0.01	58	8.0	0.01	2.2	6.0	<0.005	0.2	17	112	1	32	42	0.22	279
5	NW/NE Sec. 8	0.08	250	5.5	0.02	<0.1	14	<0.005	4.4	160	44	2	300	95	4.0	879

Trace Constituents Below Detection:

Cs < 0.1, Mn < 0.01, Fe < 0.06, Cd < 0.01, Co < 0.05, Cu < 0.02, Ni < 0.04, Pb < 0.1

Table 4f. Calculated Temperatures, Honey Lake Valley (in °C)

Sample Number	Location	Quartz Solubility		Na, K, Ca		Measured Temperature
		(regular)	(with steam loss)	(B = 1/3)	(B = 4/3)	
1	NE/NE Sec. 6	105		177	53	35.8
2	SE/NE Sec. 6	112		123	73	48.8
3	SW/SE Sec. 23	148	142	126	104	95.6 (boiling)
4	NE/SW Sec. 30	94		190	115	28.2
5	NW/NE Sec. 8	134	130	113	98	95.1 (boiling)

Table 5a. Surprise Valley Springs and Artesian Wells

Sample Number	Location	Name	Date Sampled (mo/dy/yr)	Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (mmho/cm)	Discharge (l/min)	SO <sub>4</sub> <sup>18</sup> (%)	SiO <sub>2</sub> (%)
1	NW/NE Sec. 17 T. 46N., R. 16E., M.D.	Fort Bidwell Reservation	7/26/73		1414	45.1	7.85	0.733	400		
2	SW/NE Sec. 24 T. 44N., R. 15E., M.D.	Lake City Mud Explosion	8/16/73	pool	1366	96.5	7.44	3.74	0		
3	NW/NE Sec. 12 T. 43N., R. 16E., M.D.	Seyferth Hot Springs	7/26/73	spring	1417	85.4	7.66	3.28	500		
4	NE/NE Sec. 13 T. 43N., R. 16E., M.D.	Leonards Hot Springs	7/25/73	spring	1390	61.8	7.82	2.59	150		
5	NE/SH Sec. 6 T. 42N., R. 17E., M.D.	Hot Springs Motel	7/27/73	27	1372	98.1	8.40	2.66	300		
6	NE/NE Sec. 7 T. 39N., R. 17E., M.D.	Menlo Hot Springs	8/23/73	spring	1384	57.4	8.91	0.808	500		

Table 5b. Chemical Constituents (in mg/l)

Sample Number	Location	Cations										Anions				Other		Calculated Dissolved Solids
		Li	Na	K	Rb	Mg	Ca	Mn	Zn	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>	B		
1	NW/NE Sec. 17	0.03	110	9.5	0.01	0.1	4.2	< 0.01	< 0.005	2.2	31	131	1	86	82	0.61	458	
2	SW/NE Sec. 24	0.24	320	15	0.08	< 0.1	7.7	< 0.01	< 0.005	7.6	220	112	0	320	200	6.3	1210	
3	NW/NW Sec. 12	0.15	300	9.0	0.04	< 0.1	28	0.01	< 0.005	5.4	220	63	0	370	110	7.6	1110	
4	NE/NE Sec. 13	0.13	330	8.5	0.03	0.6	26	0.03	< 0.005	5.2	220	82	1	390	110	7.6	1180	
5	NE/SW Sec. 6	0.10	280	5.5	0.03	< 0.1	16	< 0.01	0.014	5.1	200	57	2	320	100	5.7	991	
6	NE/NE Sec. 7	< 0.02	100	1.4	< 0.01	< 0.1	5.1	< 0.01	< 0.05	3.8	25	27	34	120	53	0.93	370	

Trace Constituents Below Detection:

- Cs < 0.1
- Fe < 0.06
- Cd < 0.01
- Co < 0.05
- Cu < 0.02
- Ni < 0.04
- Pb < 0.1

Table 5c. Calculated Temperatures, Surprise Valley (in °C)

Sample Number	Location	Quartz Solubility		Na, K, Ca		Measured Temperature	
		(regular)	(with steam loss)	(B = 1/3)	(B = 4/3)		
1	NW/NE Sec. 17	126		167	179	141	45.1
2	SW/NE Sec. 24	180	168	110	160	161	96.5 (boiling)
3	NW/NW Sec. 12	143		76	129	101	85.4
4	NE/NE Sec. 13	143		66	124	102	61.8
5	NE/SH Sec. 6	137	133	50	115	96	98.1 (boiling)
6	NE/NE Sec. 7	105		31	96	65	57.4

Table 6a. Modoc Plateau Hot Springs and Artesian Well

Sample Number	Location	Name	Date Sampled (mo/dy/yr)	Producing Depth (m)	Altitude (m)	Water Temperature (°C)	pH	Conductivity (µmho/cm)	Discharge (l/min)	$80^{18}(\%)$	$Si^{2}(\%)$
1	NE/NE Sec. 29 T. 42N., R. 10E., M.D.	Kelley Hot Spring	7/23/73	spring	1326	91.5	8.08	2.77	1250		
2	SW/NE Sec. 31 T. 40N., R. 13E., M.D.	Williams Ranch Well	7/24/73	30-38	1347	43.8	8.40	0.333	pumped		
3	NW/SW Sec. 9 T. 39N., R. 5E., M.D.	Little Hot Springs	8/22/73	spring	1082	75.7	7.59	2.49	300		
4	NW/NE Sec. 29 T. 39N., R. 14E., M.D.	West Valley Reservoir Hot Springs	7/24/73	spring	1460	77.3	7.79	2.98	12		
5	NW/SE Sec. 12 T. 38N., R. 7E., M.D.	Bassett Hot Springs	8/14/73	spring	1265	79.0	8.53	2.33	200		
6	SW/SE Sec. 15 T. 38N., R. 8E., M.D.	Kellog Hot Springs	8/22/73	spring	1277	78.4	8.63	2.59	15		
7	NW/SW Sec. 25 T. 37N., R. 14., M.D.	Hunt Hot Springs	8/21/73	spring	503	57.6	8.75	2.55	32		

Table 6b. Chemical Constituents (in mg/l)

Sample Number	Location	Cations							Anions				Others		Calculated Dissolved Solids	
		Li	Na	K	Rb	Mg	Ca	Mn	F	Cl	HCO <sub>3</sub>	CO <sub>3</sub>	SO <sub>4</sub>	SiO <sub>2</sub>		B
1	NE/NW Sec. 29	0.15	250	6.5	0.02	< 0.1	20	< 0.01	2.1	160	45	1	300	110	3.8	899
2	SW/NW Sec. 31	< 0.01	49	3.4	< 0.01	< 0.1	4.2	< 0.01	0.4	11	80	3	28	52	0.22	231
3	NW/SW Sec. 9	0.17	230	5.2	0.02	0.2	44	0.01	1.9	120	49	0	400	87	3.9	941
4	NW/NE Sec. 29	0.40	330	11	0.06	< 0.1	19	0.012	4.0	150	63	0	510	130	4.5	1220
5	NW/SE Sec. 12	0.10	220	3.2	0.01	< 0.1	30	< 0.01	2.0	93	30	1	370	68	2.5	820
6	SW/SE Sec. 15	0.12	240	5.9	0.02	< 0.1	30	< 0.01	2.6	110	31	2	370	85	3.2	880
7	NW/SW Sec. 25	0.12	300	4.3	0.02	< 0.1	52	< 0.01	3.6	140	49	6	520	47	12.8	1130

Constituents Below Detection:

- Cs < 0.1
- Fe < 0.06
- Cd < 0.01
- Co < 0.05
- Cu < 0.02
- Ni < 0.04
- Pb < 0.1
- Zn < 0.005

Table 6c. Calculated Temperature, Modoc Plateau (in °C)

Sample Number	Location	Quartz Solubility		Na, K, Ca		Measured Temperature
		(regular)	(with steam loss)	(B = 1/3)	(B = 4/3)	
1	NE/NW Sec. 29	143	137	123	95	91.5 (boiling)
2	SW/NW Sec. 31	104		154	90	43.8
3	NW/SW Sec. 9	130		111	69	75.7
4	NW/NE Sec. 29	152		138	120	77.3
5	NW/SE Sec. 12	117		96	62	79.0
6	SW/SE Sec. 15	128		117	82	78.4
7	NW/SW Sec. 25	99		96	63	57.6



## References Cited

- Bailey, R. A., Lanphere, M. A., and Dalrymple, G. B., 1973, Volcanism and geochronology of Long Valley Caldera, Mono County, California: *abs.* : *Am. Geophys. Union Trans.*, v. 54, n. 11, p. 1211.
- Barnes, I., 1964, Field measurement of alkalinity and pH: *U. S. Geol. Survey Water-Supply Paper 1535-H*, 17 p.
- Barnes, I., and O'Neil, J. R., 1969, The relationship between fluids in some fresh Alpine-type ultramafics and possible modern serpentinization, western United States: *Geol. Soc. America Bull.*, v. 80, p. 1947-1960.
- Biehler, S., Kovach, R. L., and Allen, C. R., 1964, Geophysical framework of northern end of Gulf of California structural province, *in* van Andel, T. H. and Shor, G. G., Jr., eds. *Marine geology of the Gulf of California*: *Am. Assoc. Petroleum Geologists Mem.* 3, p. 126-143.
- Bigeleisen, J., Perlman, M. L., and Prosser, H. C., 1952, Conversion of hydrogenic materials to hydrogen for isotope analysis: *Anal. Chemistry*, v. 24, p. 1356-1357.
- Brown, E., Shougstad, M. W., and Fishman, M. J., 1970, Methods for collection and analysis of water samples for dissolved minerals and gases: *U.S. Geol. Survey Techniques of Water-Resources Inv.*, book 5, chap. A-1, 160 p.
- Coplen, T. B., 1972, Origin of geothermal waters in the Imperial Valley of southern California, *in* Rex, R. W., and others, *Cooperative investigation of geothermal resources in the Imperial Valley area and their potential value for desalting of water and other purposes*: *Univ. Calif. Riverside, Inst. Geophys. and Planetary Phys. Pub.* 72-33, sec. E, 33 p.
- Craig, H., 1961, Standard for reporting concentrations of deuterium and oxygen - 18 in natural waters: *Science*, v. 133, p. 1833-1834.
- Duffield, W. A., and Fournier, R. O. 1974, Reconnaissance study of the geothermal resources of Modoc County, California: *U. S. Geol. Survey Open File Rpt.*, 19 p.
- Durrell, C., 1966, Tertiary and Quaternary geology of the northern Sierra Nevada, *in* Bailey, E. H., ed., *Geology of northern California*: *Calif. Div. of Mines and Geology Bull.* 190, p. 185-197.
- Ellis, A. J., 1970, Quantitative interpretation of chemical characteristics of hydrothermal systems: *Geothermics, Special Issue 2*, v. 2, part 1, p. 516-528.
- Epstein, S., and Mayeda, T., 1953, Variations of the O-18 content of waters from natural sources: *Geochim. et Cosmochim. Acta*, v. 4, n. 5, p. 213-224.

- Fournier, R. O., and Rowe, J. J., 1966, Estimation of underground temperatures from the silica content of water from hot springs and wet-steam wells: *Am. Jour. Sci.*, v. 264, p. 685-697.
- Fournier, R. O., and Truesdell, A. H., 1973, An empirical Na-K-Ca geothermometer for natural waters: *Geochim. et Cosmochim. Acta*, v. 37, p. 1255-1275.
- Fournier, R. O., and Truesdell, A. H., 1974, Geochemical indicators of subsurface temperature, Pt. II: Estimation of temperature and fraction of hot water mixed with cold water: *U. S. Geol. Survey Jour. of Research*, v. 2, N. 3,
- Fournier, R. O., White, D. E., and Truesdell, A. H., 1974, Geochemical indicators of subsurface temperature, Pt. I: Basic assumptions: *U. S. Geol. Survey Jour. of Research*, v. 2, n. 3,
- Macdonald, G. A., 1966, Geology of the Cascade range and Modoc Plateau, in Bailey, E. H., ed., *Geology of northern California: Calif. Div. of Mines and Geology Bull.* 190, p. 65-96.
- Muffler, L. J. P., and Doe, B. R., 1968, Composition and mean age of detritus of the Colorado River Delta in the Salton Trough, southeastern California: *Jour. Sed. Petrology*, v. 38, n. 2, p. 384-399.
- Muffler, L. J. P., and White, D. E., 1968, Origin of CO<sub>2</sub> in the Salton Sea geothermal system, southeastern California, U.S.A.: 23rd. *Internat. Geol. Cong.*, Prague, v. 17, symp. 2 proc., p. 185-194.
- Muffler, L. J. P., and White, D. E., 1969, Active metamorphism of Upper Cenozoic sediments in the Salton Sea geothermal field and the Salton Trough, southeastern California: *Geol. Soc. America Bull.*, v. 80, n. 2, p. 157-182.
- Waring, G. A., 1965, Thermal springs of the United States and other countries of the world - a summary, revised by Blankenship, R. R., and Bentall, R.: *U. S. Geol. Survey Prof. Paper* 492, 383p.
- Willey, L. M., O'Neil, J. R., and Rapp, J. B., Chemistry of thermal waters in Long Valley, Mono County, California: *U. S. Geol. Survey Open File Rpt.*, 19 p.

## IDAHO

### Summary

At least 380 hot springs and wells are known to occur throughout the central and southern parts of Idaho. One hundred twenty-four of these were inventoried as a part of a study by Young and Mitchell (1973). Thermal waters flow from rocks ranging in age from Precambrian to Holocene and from a wide range of rock types--igneous, metamorphic, and both consolidated and unconsolidated sediments. Twenty-eight sites occur on or near fault zones, while a greater number probably related to faulting. Most are low temperature with no steam. Idaho is underlain by large areas of granitic rocks of the Idaho Batholith (of Cretaceous age), where most of the springs occur, and by Tertiary volcanic rocks of the Snake River Plain. Heat flow in the state averages 2.0 HFU, which is similar to the Basin and Range Province as a whole. Most geothermal water (with the exception of that south-east of the Snake River Plain) has a specific conductance of less than 1,500 microhms/cm at 25°C. Virtually all spring and well water is alkaline, with pH's ranging from about 7 to more than 9. More than 200 homes in Boise are heated by hot water (77°C [170°F]) and there are two KGRA's in the state. Typical chemical content of water from granitic rocks in most areas north of the Snake River Plain as averaged by Ross, (1972).

SiO <sub>2</sub> -62 ppm	carbonate - 22 ppm
calcium - 4 ppm	bicarbonate - 59 ppm
magnesium - 0.3 ppm	sulfate - 14 ppm
sodium - 60 ppm	chloride - 9 ppm
potassium - 1.3 ppm	fluoride - 12.1 ppm
carbonate - 22 ppm	nitrate - 0.2 ppm
total dissolved solids - 224 ppm	pH - 9.2

Hottest springs:

Vulcan Hot Spring

Maximum temperature - 89°C (192°F)

Discharge - 450 gpm

Dissolved solids (calculated) - 362 ppm

pH - 8.5

Boiling Springs

Maximum temperature 85°C (185°F)

Discharge - 165 gpm

Dissolved solids-270 ppm

PH - 8.8

Bonneville Hot Springs

Maximum temperature 85°C (185°F)

Discharge-363 ppm

Dissolved solids - 306 ppm

pH - 8.1

Big Creek Hot Springs

Maximum temperature 93°C (199°F)

Discharge - 75 gpm

Dissolved solids - 727 ppm

pH - 7.5

Bridge Spring

Maximum temperature 93°C (199°F)

Discharge - 26 gpm

Water quality of geothermal water in Idaho is good except in the southeastern part of the state, where the waters are uniformly high in total dissolved

Measured water temperatures at 124 wells and springs inventoried ranged from 12° to 93°C and averaged 50°C. Estimated aquifer temperatures, calculated using the silica and the sodium-potassium-calcium geochemical thermometers, range from 5° to 370°C and averaged 110°C. Estimated aquifer temperatures in excess of 140°C exist at 42 sites. No areal patterns to the distribution of temperatures either at the surface or subsurface are apparent.

Generally, the quality of the thermal waters is good. Dissolved solids concentrations range from 14 to 13,700 ppm and averaged 812 ppm, with higher values occurring in the southeastern part of the state.

At the Frazier KGRA in Raft River Valley, water temperatures at the surface are above 90°C at two wells. Geochemical thermometers indicate temperatures of 135° to 145°C may exist at depths. Dissolved solids concentrations in waters issuing from the two wells were 1720 and 3360 ppm. The minerals being deposited by these waters consist chiefly of halite (NaCl) and calcite (CaCO<sub>3</sub>).

#### Well and Spring Numbering System

The numbering system used by the U.S. Geological Survey in Idaho indicates the location of wells or springs within the official rectangular subdivision of the public lands, with reference to the Boise base line and meridian. The first two segments of the number designate the township and range. The third segment gives the section number, followed by three letters and a numeral, which indicate the quarter section, the 40-acre tract, the 10-acre tract, and the serial number of the well within the tract, respectively. Quarter sections are lettered a, b, c, and d in counterclockwise order from the northeast quarter of each section (Fig. 2). Within the quarter sections, 40-acre and 10-acre tracts are lettered in the same manner. Well 6S-5E-10ddd1 is in the SE1/4 SE1/4 SE1/4 sec.10, T.6S., R.5E., and was the first well inventoried in that tract. Springs are designated by the letter "S" following the last numeral; for example, 4S-13E-30adb1S.

#### Generalized Geologic Setting of Idaho

The state of Idaho is underlain by rocks of igneous, metamorphic, and sedimentary origins. These formations range in age from Precambrian to Holocene and represent a varied and complex geologic history. Large scale igneous activity has occurred throughout most of the state. Cenozoic lava flows ranging in composition from rhyolite to basalt are exposed in most of the western, central, and southern parts of the state, while Mesozoic

and Cenozoic granitic rocks are the predominant rock type of large areas of central Idaho. Marine sedimentary rocks of Paleozoic age are the principal rock type of southeastern Idaho, while metamorphic rocks of Precambrian age are exposed in northern and east-central Idaho.

Although the occurrence of thermal activity and its association to a particular rock type in Idaho is obscure, known thermal anomalies are limited to the central and southern parts of the state. The occurrence and associated rock type of sampled springs and wells is discussed in the following sections.

A brief description of the geology, including the age and lithology of the spring vent or aquifer and, where possible, the controlling structure and the active deposition at each spring and well is given in Table 1. These descriptions indicate that thermal springs and wells throughout the state issue from a great diversity of rock types of nearly all ages. However, the lithology and age of the spring vent or aquifer may not be indicative of the aquifer from which the thermal waters originate. Many thermal springs in central Idaho occur in association with fault zones in Cretaceous and Tertiary granitic and related rocks, whereas springs and wells along the margins of the Snake River Plain occur in Cenozoic basaltic and rhyolitic lava flows and associated sedimentary rocks. In southeastern Idaho, springs and wells are primarily associated with fault zones in Paleozoic marine sedimentary rocks that may in places be overlain by unconsolidated valley fill.

Although nearly one-fifth of the sampled springs issue from known faults, a greater number are thought to be associated with faulting.

Active deposition of minerals from water discharged by thermal springs and wells occurs throughout the state. Minerals deposited include: gypsum, halite, various carbonates, and silicates. Carbonate deposits were identified using diluted hydrochloric acid, while siliceous deposits were identified by hardness and visual examination.

#### Water Chemistry

The chemical analyses of thermal spring and well waters sampled by Young and Mitchell (1973) are given in Table 2. The aquifer temperatures estimated by the silica method are given in Table 3.

Most thermal waters in Idaho are low in dissolved solids, with concentrations in sampled waters ranging from 14 to 13,700 ppm. Thermal waters in the southeastern part of Idaho are higher in dissolved solids than thermal waters in other parts of the state. Waters which are high in dissolved solids generally give high Na-K-Ca temperatures relative to silica temperatures (Table 3), whereas waters low in dissolved solids give high silica temperatures relative to low Na-K-Ca temperatures.

Measured temperatures of sampled waters ranged from 12°C in northern Fremont County to 93.0°C in Cassia and Lemhi Counties and averaged 50°C for all sampled springs and wells. Examination of the temperature data collected does not reveal any correlation of temperature with location, rock type, or structure.

Estimated aquifer temperatures for the waters sampled ranged from 5° to 370°C, as estimated by the sodium-potassium-calcium geochemical thermometer and from less than 35° to 170°C, as estimated by the silica geochemical thermometer.

Estimate aquifer temperatures in excess of 140°C were found at 42 sites. Generally, for waters high in dissolved solids, the Na-K-Ca geochemical thermometer indicated higher aquifer temperatures than did the silica geochemical thermometer; whereas for waters low in dissolved solids, the silica geochemical thermometer indicated highest temperatures.



TABLE 2  
CHEMICAL ANALYSES OF THERMAL WATERS FROM SELECTED SPRINGS AND WELLS IN IDAHO  
(Chemical constituents in milligrams per liter)

Spring or well Identification Number	Reported Well Depth Surface (feet)	Sample Collection Date	Discharge (gpm)	Temperature (°C)	Silica (Si)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved Solids (Calculated)	Dissolved Solids (measured)	Hardness		Specific Conductance	pH (field)	Alkalinity as CaCO <sub>3</sub>	Percent Sodium	Sodium Absorption Ratio	Area No.	
																		as CaCO <sub>3</sub>	Non- carbonate							
ADA COUNTY																										
5N 1E 35acal		5-31-72	22	40.0	33	4.3	0	49	3.2	112	1	23	0.03	4.9	11	0.05	193	0.26	11	0	285	7.5	84	89	7.4	
4N 2E 29acdl	1,195	5-31-72	-	47.0	16	4.5	3	55	2.4	145	2	21	0.02	4.4	10	0.06	225	0.31	14	0	311	7.1	122	89	7.1	
3N 2E 12cddl	400	5-31-72	-	75.0	78	2	0	75	1.3	141	4	23	.01	9.3	24	0.08	299	0.41	9	0	386	7.3	122	95	13	
ADAMS COUNTY																										
White Licks Hot Springs		6-29-72	30	65.0	110	39	0.3	420	17	71	0	660	.05	150	8.8	0.07	1,440	1.96	99	40	2,050	7.6	56	88	18	1
16N 2E 33Bcals																										
Zim's Resort Hot Springs		6-28-72	-	65.0	64	12	1	190	3.6	47	9	330	.03	32	2.3	.07	666	.91	30	0	940	8.5	54	92	15	
20N 1E 26dcbis																										
Krigbaum Hot Springs		6-29-72	40	43.0	73	5.3	2	140	3.3	81	9	190	.05	26	2.8	.05	490	.67	14	0	668	8.8	81	94	16	
19N 2E 22cals																										
Starkey Hot Springs		6-27-72	130	56.0	56	4.5	0	86	1.6	60	6	150	.05	14	.9	.05	369	.50	12	0	502	8.6	58	94	12	
18N 1W 34dcbis																										
BANKS COUNTY																										
5S 34E 26dcb1	582	7-27-72	15	40.5	20	70	25	150	21	478	0	95	0	87	5.2	.02	705	.96	280	0	1,170	7.7	392	52	3.9	2
Lava Hot Springs		8-15-72	-	44.5	32	120	32	170	39	542	0	110	.04	190	.7	.38	962	1.31	430	0	1,580	6.6	445	43	3.6	3
9S 38E 21dcbis																										
Donata Hot Springs		5-17-72	490	43.0	29	43	15	20	9.1	214	0	18	0	20	.4	.5	262	.56	170	0	413	6.7	176	19	.7	
12S 37E 12cdbl1																										
BEAR LAKE COUNTY																										
Bear Lake Hot Springs		5- 9-72	-	47.5	35	210	55	180	61	256	0	800	.01	79	7.1	.56	1,560	2.12	750	540	2,040	6.6	210	32	2.9	4
15S 44E 13cals																										
BLAINE COUNTY																										
1S 17E 23ab1	260	6-21-72	15	70.5	100	22	1.3	330	19	766	0	60	.04	83	13	.06	1,010	1.37	60	0	1,500	6.4	628	89	19	5
Gover Hot Springs		7-11-72	1,000	70.5	86	2.9	0	84	2.1	51	25	72	.02	11	16	.06	324	.44	7	0	423	8.0	83	95	14	
4N 17E 15acis																										
Clarendon Hot Springs		7-11-72	100	47.0	80	2.2	.1	81	1.7	29	30	68	.01	11	15	.06	303	.41	6	0	400	8.2	74	96	15	
3N 17E 37dcbis																										
Hailey Hot Springs		7-11-72	70	69.0	85	2	0	68	1.5	88	0	51	.02	10	12	.07	273	.37	5	0	337	8.7	72	96	13	
2N 18E 15dcbis																										
Goodie Hot Springs		8- 8-72	346	52.0	28	56	11	63	17	560	0	28	.01	14	1.7	.05	396	.54	190	0	653	7.3	395	40	2	
1S 21E 14dcbis		8- 8-72	420	44.0	26	60	12	48	8.9	284	0	63	.03	6.5	2.5	.03	371	.5	200	0	591	7.3	241	33	1.5	
1S 22E 14dcbis																										
BOISE COUNTY																										
Bonneville Hot Springs		8-18-72	365	85.0	100	2.2	.1	67	2.9	58	21	55	.03	7.2	17	.02	306	.42	6	0	377	8.1	83	94	15	6
10N 10E 31cals																										

Table extracted from: Yours, H. W. and Mitchell, J. C., Geothermal Investigations in Idaho, Idaho Department of Water Administration, Bulletin No. 30, Part 1, 43 pgs., (1973)

TABLE 2 (Cont'd.)

Spring or Well Identification Number	Reported Well Bottom Surface (feet)	Sample Location (well section)	Discharge (gpm)	Temperature (°C)	Silica (SI)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Dichromate (NO <sub>2</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Thiosulfate (T)	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved Solids (Calculated)	Dissolved Solids (Total)	Hardness		Specific Conductance (µmhos/cm-cm)	pH (field)	Alkalinity as CaCO <sub>3</sub>	Percent Sodium	Sodium Adsorption Ratio	Fig. 6 Area No.	
																			CaCO <sub>3</sub>	Non-carbonate							
BOISE COUNTY (Cont'd.)																											
9N 3E 25ba1S		8-4-72	20	80.0	120	4.5	0	130	4.8	160	0	79	0.02	34	13	0.04	464	0.63	11	0	600	8.1	131	94	17	7	
Kirshan Hot Springs																											
9N 8E 33ca1S		7-14-72	250	65.0	69	1.9	.1	66	1.3	46	21	45	.02	3	15	.06	245	.33	5	0	322	7.8	75	95	13		
8N 5E 10ba1S																											
8N 5E 10ba1S		6-8-72	82	40.0	48	2.4	.1	66	.9	85	1	42	.01	5.3	3.1	.25	216	.29	6	0	317	8.8	71	95	11		
8N 5E 10ba1S		8-18-72	70	55.0	59	1.9	0	68	1.1	40	30	38	.02	5.6	14	.04	237	.32	5	0	336	8.6	83	96	14		
BONNEVILLE COUNTY																											
3N 4SE 9ba1S		8-10-72	270	25.0	11	440	96	1,110	120	1,200	0	390	.04	1,900	1.7	.05	4,650	6.32	1,500	510	7,950	6.3	984	59	12	8	
BUTTE COUNTY																											
3N 2SE 32ca1S	360	8-9-72	12	41.0	55	74	24	72	21	322	0	170	.02	21	3.2	.12	599	.61	260	19	898	6.3	264	34	1.9		
3N 27E 20ba1S	475	8-9-72	-	65.0	55	64	24	31	7.7	315	0	56	.02	22	.8	.96	398	.54	260	0	648	7.2	256	20	.8		
CANYON COUNTY																											
Wardson Hot Springs		6-20-72	193	66.0	73	1.4	0	54	3	51	37	12	.03	5.1	4.1	.07	215	.29	4	0	252	8.0	103	94	15	5	
1N 13E 22ba1S		7-10-72	466	81.0	96	1.8	0	69	1.9	51	28	35	.02	5	15	.07	277	.38	5	0	328	7.3	88	96	14		
Worswick Hot Springs		6-21-72	115	53.5	63	2.3	0	87	1.4	82	15	48	.02	25	19	.06	302	.41	5	0	441	8.2	92	96	17		
1N 15E 14ba1S		6-20-72	15	31.0	36	6	0	32	.3	31	26	3.1	.04	2.1	.8	.05	116	.16	2	0	150	9.2	69	97	11		
1S 12E 31ca1S	400	6-20-72	4	35.0	76	3.2	.1	92	1.3	210	0	6.4	.04	12	11	.04	308	.42	8	0	413	7.4	177	95	14		
1S 13E 27ca1S	190	6-20-72	31	70.0	77	3.6	.1	99	2.5	226	0	13	.04	15	14	.08	337	.46	10	0	471	7.5	185	94	14		
Barron's Hot Springs		6-20-72	31	70.0	77	3.6	.1	99	2.5	226	0	13	.04	15	14	.08	337	.46	10	0	471	7.5	185	94	14		
1S 13E 28ca1S		6-20-72	31	70.0	77	3.6	.1	99	2.5	226	0	13	.04	15	14	.08	337	.46	10	0	471	7.5	185	94	14		
2N 2W 34ba1S	318	6-9-72	200	51.0	38	3.5	.1	110	.8	279	0	59	.04	11	4.1	.15	384	.52	9	0	589	7.5	229	97	19		
6S 41E 19ba1S		5-15-72	1,300	42.0	24	660	260	94	240	2,500	0	980	.05	40	1.9	.04	3,550	4.8	2,700	670	4,590	6.8	2,050	6	.8	9	
Seda Springs		8-15-72	-	31.0	29	640	170	12	25	2,290	0	800	.07	4.9	.5	.05	3,120	4.24	3,000	1,110	3,990	6.3	1,880	1	.1		
95 41E 12ba1S		8-15-72	-	31.0	29	640	170	12	25	2,290	0	800	.07	4.9	.5	.05	3,120	4.24	3,000	1,110	3,990	6.3	1,880	1	.1		
CASSIA COUNTY																											
1S5 24E 23ba1S	414	5-18-72	58	93.0	90	53	4	560	22	55	0	57	0	900	5.7	.54	1,720	2.34	130	89	3,050	7.4	45	88	21	10	
1S5 26E 23ba1S	540	5-18-72	60	90.0	97	130	4	1,110	35	36	0	61	.01	1,900	14	.57	3,360	4.57	330	300	6,090	7.7	30	87	27	10	
115 25E 11ca1S	447	7-26-72	2,090	60.0	60	8.2	.5	110	3.9	135	0	59	0	55	14	0	372	.51	25	0	574	7.7	103	90	10		
145 21E 34ba1S		7-26-72	280	43.0	47	14	1.1	44	9.6	134	0	15	.01	7	1.3	.01	210	.29	59	0	282	6.0	118	65	5		
Oakley Warm Spring		10-26-72	10	47.0	70	2.7	0	87	2.2	43	29	22	.03	53	8	.04	295	.4	7	0	421	9.6	84	95	15		
145 22E 27ca1S		7-25-72	100	38.0	44	37	9.3	70	3.1	169	0	33	.03	80	2.9	.56	365	.5	150	0	606	7.4	139	53	2.7		
155 24E 23ba1S	500	7-25-72	100	38.0	44	37	9.3	70	3.1	169	0	33	.03	80	2.9	.56	365	.5	150	0	606	7.4	139	53	2.7		
CLACK COUNTY																											
Warm Springs		8-28-72	1,920	29.0	17	54	19	9.9	2.9	209	0	62	.02	5.3	1	.12	274	.37	210	42	457	7.0	171	9	.3		
1N 31E 10ba1S		8-28-72	1,920	29.0	17	54	19	9.9	2.9	209	0	62	.02	5.3	1	.12	274	.37	210	42	457	7.0	171	9	.3		

Reported Spring or Well Identification Number	Well Depth	Surface Area (feet)	Sample Collection Date	Discharge (gpm)	Temperature (°C)	Silica (SI)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Phosphate (P)	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved Solids (Calculated)	Dissolved Solids (Total)	Hardness		Specific Conductance	pH (Field)	Alkalinity as CaCO <sub>3</sub>	Percent Sodium	Sodium Sulfate	Fig. 6
																			as CaCO <sub>3</sub>	Non-carbonate						

CLARK COUNTY (Cont'd.)

Lady Hot Springs 9X 33E 206C1S			8-25-72	250	50.0	34	87	16	27	15	179	0	190	0.03	8	6	0.02	471	0.64	280	140	691	6.3	147	16	0.17		
8X 17E 326A1S			7-12-72	425	51.0	43	21	5.5	100	13	234	0	94	.02	26	8.4	.06	425	.58	72	0	651	6.7	192	71	5.1	11	
14X 19E 344A1S	3,000		7-12-72	50	40.0	23	55	21	45	7.6	226	0	130	.01	4	1.1	.1	398	.54	220	38	625	7.3	185	30	1.3		
Sunbeam Hot Springs 11X 15E 19C1S			7-12-72	444	76.0	91	1.5	0	85	2.4	319	0	54	.02	12	15	.06	320	.44	4	0	413	8.5	98	96	19		
Sullivan Hot Springs 13X 17E 270A1S			7-12-72	70	41.0	38	49	11	170	35	554	0	26	.02	57	1.8	.06	640	.87	170	0	1,070	7.0	454	66	5.7		
Barney Hot Springs 13X 25E 236A1S			7-13-72	170	63.5	18	37	20	9	1.5	181	0	35	.03	4	.5	.25	215	.29	170	26	364	7.8	148	10	.3		
Stanley Hot Springs 10X 11E 36A1S			7-12-72	110	41.0	55	2.2	.1	60	.5	30	28	31	.01	5	1.4	.05	211	.29	6	0	293	8.8	71	95	11		
Slate Creek Hot Springs 10X 10E 30A1S			7-11-72	185	50.0	86	8.1	.1	83	4.5	110	0	110	.02	7	8.7	.03	362	.49	21	0	432	8.0	90	97	5.1		
55 SE 340A1	1,320		7- 5-72	2	34.0	58	9.1	1	320	11	797	0	6.5	.04	59	2.2	.04	859	1.17	27	0	1,340	7.7	654	94	27	12	
Weinaver Hot Springs 5X 7E 24B1S			8-17-72	349	76.0	100	1.1	.1	67	1.8	5	51	31	.03	2.9	10	.02	267	.36	3	9	293	8.5	59	56	16		
Dutch Frank's Springs 5X 9E 7B1S			8-17-72	300	65.0	72	2.2	.2	57	1.2	17	40	30	.05	2.4	10	.02	223	.3	6	0	268	8.6	81	94	9.9		
Paradise Hot Springs 3X 10E 35B1S	600		9-29-72	-	56.0	69	1.5	.1	50	1	45	35	17	.05	2.6	3.1	.04	200	.27	4	0	252	9.2	94	96	11		
35 SE 360A1			9-12-72	200	68.0	86	1.5	0	67	.8	74	50	14	.04	4.5	17	.06	297	.4	4	0	362	8.5	144	98	20		
Larry Hot Springs 35 10E 310B1S			7- 5-72	-	65.0	100	.4	0	54	1.7	90	35	10	.04	2.7	7	.07	248	.34	1	0	243	8.4	117	98	29		
45 SE 360A1	1,900		6- 6-72	8	35.0	86	3.2	.2	160	3.7	447	0	5.4	.05	10	3	.06	491	.67	10	0	703	7.8	264	96	22		
45 SE 8A11	1,003		8-29-72	-	62.0	85	.9	0	82	.8	81	41	14	.03	3.2	16	.05	283	.38	2	0	387	9.2	155	98	24		
55 10E 7A11	1,300		6-19-72	-	53.0	42	2.5	0	79	.9	115	16	12	.03	6.1	20	.03	235	.32	6	0	267	8.5	121	96	24		
55 10E 370B1	935		6-22-72	54	37.5	46	2.5	.2	130	.9	270	5	2.5	.05	2.9	13	.06	363	.5	6	0	500	7.9	235	97	22		

FRANKLIN COUNTY

Maple Grove Hot Springs 135 41E 7A2A1S			5-10-72	350	76.0	55	89	24	490	110	491	0	260	.04	630	1.1	.07	1,900	2.58	320	0	3,160	7.3	403	70	12	15	
145 39E 360A1	40		5-11-72	-	44.5	80	25	7.1	360	24	524	0	15	.05	320	10	1.5	1,110	1.51	92	0	1,490	7.5	430	8.7	16	23	
Wayland Hot Springs 155 39E 340A1S			5- 9-72	2,000	77.0	80	160	16	3,100	600	699	0	50	.06	5,400	12	.81	9,830	13.4	470	0	16,500	7.0	573	84	65	15	

TABLE 2 (Cont'd.)

Spring or Well Identification Number	Reported Kell Depth Below Land Surface (feet)	Sample Collection Date	Discharge (gpm)	Temperature (°C)	Silica (SI)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Nitrate (NO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Phosphate (P)	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved Solids (Calculated)	Dissolved Solids (Total)	Hardness		Specific Conductance	pH (Field)	Alkalinity as CaCO <sub>3</sub>	Percent Sodium Absorption Ratio	Area No.	
																			CaCO <sub>3</sub>	Non-carbonate						
FRANKLIN COUNTY (Cont'd.)																										
FRANKLIN COUNTY																										
155 39E 17bed1	22	5-10-72	25	82.0	130	250	23	4,300	880	735	0	54	0.68	7,700	7	1.6	13,700	18.6	720	170	27,200	7.8	601	84	70	13
Ashton Mm. Springs																										
90 42L 23ab1S		8-28-72	22	41.0	110	1.1	.1	36	1.6	92	0	4.7	.05	2.9	2.2	.54	205	.28	3	0	166	7.6	75	94	8.8	14
Big Springs																										
14N 44E 38bb1S		8-28-72	92,000	12.0	47	5.6	.6	14	3	46	0	3.2	.03	2.5	3.1	.05	102	.14	16	0	102	6.4	38	60	2.5	
Lily Pad Lake																										
10N 45E 55ab1S		8-30-72	-	67.5	.3	2.6	.4	.5	1	11	0	2.2	.03	1.1	.1	.44	14	.02	8	0	19	7.2	9	10	.1	
7N 41E 35cd1d1																										
350		8-9-72	-	36.0	75	28	6.3	78	8.6	240	0	33	.02	24	5.4	.79	380	.52	95	0	538	7.9	197	61	3.5	
GENE COUNTY																										
Keystone Hot Springs																										
7N 1E 8da1S		11-24-72	20	65.0	120	8.7	.6	160	7.7	187	0	110	.04	62	16	0	577	.78	24	0	799	7.5	153	91	14	15
7N 1E 9cd1S																										
8-4-72		-	45.0	94	15	2.4	99	5.3	169	0	57	.02	30	8	.67	397	.54	47	0	529	7.6	159	80	6.3		
45 13E 26ab1																										
160		6-21-72	-	47.0	92	9.8	1.2	100	5.9	278	0	19	.05	8.2	12	.49	373	.51	30	0	497	7.0	207	85	7.9	
White Arrow Hot Springs																										
45 13E 30ab1S		5-26-72	850	65.0	97	1.2	0	91	1.6	141	22	15	.03	6.6	12	.11	316	.43	3	0	407	7.5	152	98	23	
55 12L 3aa1																										
692		6-19-72	-	43.0	62	1.6	.1	90	.8	83	42	19	.03	8.4	19	.17	284	.39	4	0	413	8.6	138	97	19	
IDARHO COUNTY																										
Weir Creek Hot Springs																										
36N 11E 13b1S		9-13-72	40	47.5	49	3.3	0	29	.5	21	22	15	.03	2.1	2.2	.05	134	.18	8	0	145	8.5	54	35	4.4	
Jerry Johnson Hot Springs																										
36N 15E 15a1S		8-25-72	300	48.0	49	2.7	.2	37	.4	24	25	25	.04	1.9	1.6	.03	135	.21	8	0	156	8.7	61	91	5.9	
Red River Hot Springs																										
28N 10L 341S		8-21-72	35	55.0	76	2.7	0	81	1.6	36	36	44	.01	4.4	23	.04	266	.39	6	0	330	8.6	89	95	14	
Riggins Hot Springs																										
24N 2E 14ac1S		9-1-72	250	42.0	72	6.2	.1	160	3.4	11	25	300	.02	8	2.1	.62	582	.79	16	0	812	8.6	51	95	17	
Burjard Hot Springs																										
22N 4E 1buc1S		8-1-72	162	45.0	73	2.3	0	49	.8	19	41	18	.02	3	2	.03	199	.27	6	0	218	8.1	84	94	8.9	
JEFFERSON COUNTY																										
Hess Hot Springs																										
4N 40E 256b1S		7-27-72	200	49.0	30	450	82	1,500	190	1,100	0	740	.04	2,400	3.1	.1	5,940	8.05	1,500	500	8,540	6.7	902	66	17	8
LEFLORE COUNTY																										
Big Creek Hot Springs																										
23N 18L 22c1S		7-13-72	275	93.0	150	5.3	.2	220	14	488	0	53	.05	29	15	.07	727	.99	14	0	1,010	7.5	400	84	26	16
Salmon Hot Springs																										
20N 22E 35ab1S		5-24-72	145	45.0	33	23	11	190	28	565	0	34	.04	50	1.8	.93	649	.88	100	0	1,000	6.3	463	75	8.2	17

TABLE 2 (Cont'd.)

Spring or Well Identification Number	Reported Well Depth Below Land Surface (feet)	Sample Collection Date	Discharge (gpm)	Temperature (°C)	Silica (ppm)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Phosphate (P)	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved Solids (Calc'd)	Dissolved Solids (measured)	Hardness		Specific Conductance	pH (field)	Alkalinity as CaCO <sub>3</sub>	Percent Sodium	Sodium Adsorption Ratio	Area No.	
																			as CaCO <sub>3</sub>	Non-carbonate							
LEWIS COUNTY (Cont'd.)																											
MADISON COUNTY																											
Sharkey Hot Springs 20N 24E 34cc61S	8-24-72	8	52.0	91	7.3	0.6	270	17	470	0	160	0.02	51	12	0.08	840	1.14	21	0	1,270	7.4	386	95	26	17		
16N 21E 18dc1S	8-24-72	20	46.0	37	11	1.4	160	11	319	0	66	.04	26	7	.06	486	.66	33	0	757	7.4	278	88	12	18		
ONEIDA COUNTY																											
Green Canyon Hot Springs SN 43E 06ca1S	8-9-72	-	64.0	25	140	32	3.9	3.6	167	0	330	.01	1.7	1.6	.13	621	.84	480	340	846	6.8	137	2	.1			
14S 36E 27ca1S	5-16-72	44	25.0	19	240	79	1,200	210	958	0	25	0	2,100	.4	.95	4,350	5.92	920	140	7,580	6.5	786	69	17	19		
ONYHIE COUNTY																											
Pleasantview Warm Springs 15S 35E 38ab1S	5-16-72	3,810	25.0	21	110	33	280	29	333	0	110	0	470	.7	1.5	1,220	1.66	410	140	2,190	6.8	271	58	6	19		
Woodruff Hot Springs 16S 36E 10bb1S	5-11-72	-	27.0	29	130	45	910	87	454	0	58	.03	1,600	.6	1.4	3,090	4.2	510	140	5,570	7.3	372	76	18	19		
12S 34E 36bb1S	5-17-72	189	24.0	33	56	19	15	4.3	226	0	18	0	35	.3	.73	295	.40	220	33	479	6.7	385	13	.4			
4S 2E 32cc1	6-6-72	30	42.0	94	4.1	.7	150	8.8	390	0	2.1	.08	15	7.7	.65	479	.65	13	0	859	8.2	320	93	15	20		
5S 3E 26bb1	6-12-72	380	84.5	110	1.6	0	90	1.5	74	38	4	.02	14	30	.28	456	.05	1	0	562	7.0	121	82	13	20		
6S 3E 2cc1	6-12-72	489	58.0	92	1.3	0	120	3.5	129	25	.02	14	10	.05	346	.05	3	0	502	8.1	121	82	13	20			
6S 3E 10bb1	6-14-72	1	38.5	100	2.3	.1	170	7.5	165	21	.24	.07	15	28	.13	266	.50	6	0	439	8.1	121	82	13	20		
6S 3E 29cc1	6-14-72	3	34.0	100	6.8	0	92	7	140	0	56	.07	15	15	.03	361	.49	17	0	459	8.0	115	89	9	20		
6S 6E 12cc1	6-15-72	990	37.0	100	30	.5	170	14	460	0	3.6	.06	18	5.6	.06	548	.75	37	0	833	7.3	377	89	14	20		
7S 5E 7ab1	6-14-72	-	39.0	81	6.3	.1	50	7.2	94	1	18	.04	8.3	9.7	.33	230	.31	16	0	278	8.1	80	81	5.4	20		
INDIAN BATHS COUNTY																											
Indian Baths Hot Springs 8S 6E 38dd1S	7-3-72	458	39.0	76	5.9	.4	54	7.3	124	2	15	.04	8	8.8	.79	242	.33	16	0	287	8.2	105	32	5.8	20		
Murphy Hot Springs 16S 9L 24bb1S	5-23-72	270	51.0	83	.6	0	30	2.0	67	1	4.7	.1	2.3	3.6	.64	163	.22	2	0	137	7.1	57	94	11	21		
1S 4N 12dbb1	6-13-72	410	35.5	40	2.2	0	110	.3	214	0	8.6	.01	28	7.9	.04	302	.41	5	0	483	7.2	176	98	20			
1S 2N 7ccb1	6-5-72	189	52.5	32	1.9	0	140	1.7	105	33	.46	.03	12	11	.04	334	.45	5	0	545	8.1	124	88	22			
1S 2E 34bd1	7-24-72	1,060	46.0	83	1.1	.2	100	.8	103	33	.45	.23	13	14	.04	339	.46	3	0	459	7.9	138	98	23	20		
1S 1E 18cc1	7-24-72	30	49.5	68	1.5	0	87	.6	60	54	20	.02	11	5.8	.04	277	.38	4	0	364	8.2	139	98	20			
7S 6E 9bd1	6-15-72	153	50.0	93	1.6	0	99	2.8	72	40	27	.06	9.7	22	.05	331	.45	4	0	446	8.2	126	9	22			
POKEE COUNTY																											
Indian Hot Springs 12S 7E 33cc1S	6-2-72	1,730	69.0	75	1.5	0	75	.6	67	30	24	.04	8.4	14	.06	262	.36	4	0	360	8.0	165	97	17			
Indian Springs 8S 31E 18db1S	7-27-72	1,540	32.0	20	76	19	110	10	254	0	19	.02	220	.7	.13	600	.82	270	80	1,100	7.5	208	46	2.9			
10S 30E 13cc1S	7-27-72	418	38.0	22	92	35	62	.44	160	0	23	.02	250	.5	.82	576	.78	370	230	1,110	7.6	131	26	1.4			
TAIN FALLS COUNTY																											
Miracle Hot Springs 8S 14E 31ac1S	5-24-72	3,350	54.0	93	2.2	0	120	1.5	63	54	29	.03	35	20	.50	336	.53	5	0	560	9.0	142	97	22			

TABLE 2 (Cont'd.)

Spring or Well Identification Number	Reported Well Depth Below Surface (feet)	Sample Collection Date	Discharge (gpm)	Temperature (°C)	Silica (SI)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Phosphate (P)	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved Solids (Calculated)	Dissolved Solids (tons per ac-ft)	Hardness		Specific Conductance	pH (field)	Alkalinity as CaCO <sub>3</sub>	Percent Sodium	Sodium Absorption Ratio	Area No. Fig. 6	
																			as CaCO <sub>3</sub>	Non-carbonate							
TWIN FALLS COUNTY (Cont'd.)																											
85 14E 33ba1	210	5-24-72	60	59.0	97	1.1	0	100	1.5	88	38	26	0.03	27	15	0.54	351	0.48	3	0	479	8.5	135	98	26		
115 19E 33add1	620	5-25-72	1,930	33.0	63	27	3.9	17	8.6	118	0	12	.04	15	.3	1	209	.28	83	0	266	6.6	97	28	.8		
Nat.-Pop.-Pax Warm Springs																											
125 17E 31ba1S	775	7-25-72	30	36.0	19	34	14	43	11	266	0	18	.01	8	1.9	.02	280	.38	140	0	469	7.6	218	37	1.6		
125 19E 1bb1a1		7-25-72	543	58.0	67	18	2	16	6	95	0	9.3	.26	8	.6	.63	176	.24	53	0	198	7.6	78	36	1		
Magic Hot Springs																											
165 17E 31ac1S		5-23-72	385	45.5	23	30	8.9	13	4.5	162	0	15	.03	3.8	.3	.42	180	.24	110	0	281	6.4	133	19	.5		
VALLEY COUNTY																											
Vulcan Hot Springs																											
14N 6E 11ba1S		8-2-72	500	87.0	120	1.8	.1	94	3	120	0	43	.02	17	24	.05	362	.49	5	0	451	8.5	98	96	18	22	
Hot Creek Springs																											
15N 3E 13bc1S		8-2-72	798	634.0	60	1.3	.1	60	.6	17	45	16	.02	16	2.6	0	210	.29	4	0	279	9.8	89	97	14		
Molly's Hot Springs																											
15N 6E 14acc1S		8-2-72	20	59.0	87	2	0	70	1.5	48	30	17	.02	10	17	.03	258	.55	5	0	326	7.7	89	96	14		
14N 3E 36abd1	50	8-3-72	-	42.5	45	1.6	0	58	.4	62	22	17	.04	15	3.8	.09	194	.26	4	0	275	9.2	87	97	13		
Cabarton Hot Springs																											
13N 4E 31cb1S		8-3-72	270	70.5	78	1.7	0	100	1.9	70	26	46	.02	49	11	.05	348	.47	4	0	511	7.7	101	97	21		
Boiling Springs																											
12N 5E 22bc1S		8-3-72	165	85.0	94	1.9	.1	71	1.7	81	24	12	.02	12	13	.04	270	.37	5	0	331	8.8	106	95	14		
WASHINGTON COUNTY																											
14N 3N 3dd1	925	6-28-72	-	25.5	70	2.6	.2	73	6.8	157	16	15	.04	3.8	1	.04	266	.36	7	0	309	8.7	155	91	12	1	
13N 3N 8ccc1	963	6-28-72	-	28.0	84	5.7	.8	73	23	225	0	14	.04	3.1	.7	.04	318	.45	25	0	338	8.3	185	74	6.4	1	
11N 6N 10cca1	400	6-28-72	1/3	70.0	170	2.7	0	160	5.1	92	19	150	.03	55	4.6	.07	612	.83	7	0	698	8.2	107	96	27	23	
11N 3N 7bdb1S		6-30-72	10	87.0	170	27	.7	300	19	198	0	270	.31	190	2.9	.06	1,080	1.47	70	0	1,480	6.8	162	87	16	23	
14N 3N 19cdd1S		6-27-72	58	50.0	55	8	.8	80	1.9	81	1	110	.05	15	.8	.30	314	.43	23	0	406	8.5	68	87	7.2		
14N 2N 6bba1S		6-28-72	431	70.0	72	17	.1	200	3.8	24	20	200	.09	140	1.9	.06	667	.91	43	0	1,000	7.8	53	90	13		
13N 4N 13ba1	1,350	6-28-72	-	28.0	73	3.5	.2	86	.7	188	20	14	.03	3.2	.7	.04	294	.4	10	0	375	6.5	188	95	12		

a Discharge estimated.

b Measured temperature is probably lower than at point of discharge.

TABLE 3  
ESTIMATED AQUIFER TEMPERATURES AND ATOMIC RATIOS OF SELECTED CHEMICAL CONSTITUENTS

Spring or Well Identification Number	Discharge (gpm)	Water Temperature at Surface (°C)	Aquifer Temperatures from Geochemical Thermometers °C aSilica bSodium-Potassium-Calcium	Sodium Potassium (NS/K)	Calcium Bicarbonate (Ca/HCO <sub>3</sub> )	Magnesium Calcium (Mg/Ca)	Atomic Ratios				Area Number Fig. 6	
							Sodium Calcium (Na/Ca)	Bicarbonate plus Carbonate (C <sub>1</sub> /HCO <sub>3</sub> + CO <sub>3</sub> )	Chloride Fluoride (Cl/F)	Calcium Sodium (Ca/Na)		
ADA COUNTY												
5K 1E 35ca1	22	40.0	85	26.0	0.058	-	19.9	0.075	0.239	0.154		
4N 2E 29acd1	-	47.0	95	59	0.047	0.11	21.3	.051	.14	.14		
3N 2E 12cdd1	-	75.0	125	98.1	0.022	-	65.4	.11	.208	.008		
ADAMS COUNTY												
White Licks Hot Springs												
16N 2E 33bec1S	30	65.0	145	42	.836	.013	18.8	3.64	9.13	.054		1
Zim's Resort Hot Springs												
20N 1E 26ddb1S	-	65.0	115	89.8	.389	.014	27.6	.981	7.46	.066		
Krighbaum Hot Springs												
19N 2E 22cca1S	40	43.0	120	72.1	.1	.062	46	.496	4.98	.06		
Starkey Hot Springs												
18N 1N 34bbb1S	130	56.0	110	91.4	.114	-	33.3	.364	8.34	.09		
BANSOCK COUNTY												
55 34E 26dab1	15	40.5	65	12.1	.223	.589	3.74	.313	14.6	.203		2
Lava Hot Springs												
9S 38E 21dd1S	-	44.5	80	7.41	.337	.439	2.47	.603	145	.234		3
Downata Hot Springs												
12S 37E 12cdc1S	c490	43.0	80	3.74	.306	.575	.811	.161	26.8	1.19		
BEAR LAKE COUNTY												
Bear Lake Hot Springs												
15S 44E 13cca1S		47.5	85	5.02	1.25	.432	1.49	.531	5.96	.292		4
BLAINE COUNTY												
1S 17E 23aab1	15	70.5	135	29.5	.044	.097	26.1	.186	3.42	.052		5
Guver Hot Springs												
4N 17E 15aac1S	c1,000	70.5	130	63	.087	-	50.5	.248	.368	.074		
Clarendon Hot Springs												
3N 17E 27dcb1S	100	47.0	125	81	.115	.075	64.2	.318	.393	.066		
Halliey Hot Springs												
2N 18E 18bbb1S	70	49.0	130	77.1	.035	-	59.3	.196	.447	.076		
Condit Hot Springs												
1S 21E 14dd1S	346	52.0	80	6.3	.237	.324	1.96	.067	4.41	.431		
1S 22E 1da1S	c20	44.0	75	9.17	.311	.33	1.39	.038	1.51	.586		

TABLE 3 (Cont'd.)

Spring or Well Identification Number	Discharge (gpm)	Water Temperature at Surface (°C)	Aquifer Temperatures from Geochemical Thermometers °C (rounded to 5°C)	Silica	Sodium Potassium (Na/K)	Calcium Bicarbonate (Ca/HCO <sub>3</sub> )	Magnesium Calcium (Mg/Ca)	Atomic Ratios			Chloride Fluoride (Cl/F)	Sulfate Sodium (SO <sub>4</sub> /Na)	Area Number F.R. #
								Sodium Calcium (Na/Ca)	Bicarbonate plus Carbonate (Cl/HCO <sub>3</sub> + CO <sub>3</sub> )	Chloride Bicarbonate plus Carbonate (Cl/HCO <sub>3</sub> + CO <sub>3</sub> )			
<u>BOISE COUNTY</u>													
Bonneville Hot Springs	363	85.0	140	135	39.3	0.058	0.075	53.1	0.156	0.227	0.08	6	
10N 10E 31c1S													
9N 3E 28bac1S	20	80.0	140	150	46.1	.043	-	50.4	.366	1.4	.059	7	
Kirkham Hot Springs	c250	65.0	80	115	86.3	.063	.087	60.6	.077	.107	.076		
9N 8E 32cac1S													
8N 5E 1bc1S	52	40.0	65	100	125	.043	.069	47.9	.102	.882	.085		
8N 5E 10b1S	70	55.0	75	110	105	.072	-	62.4	.137	.214	.074		
<u>BONNEVILLE COUNTY</u>													
2N 43E 9bb1S	c70	25.0	55	190	15.6	.558	.36	4.36	2.72	599	.069	8	
<u>BUTTE COUNTY</u>													
3N 25E 32cd1	12	41.0	90	105	5.83	.35	.534	1.7	.112	3.52	.434		
3N 27E 9abb1	-	35.0	55	85	6.85	.309	.618	.644	.12	14.7	.937		
<u>CANON COUNTY</u>													
Wardrop Hot Springs	193	66.0	120	155	30.6	.042	-	67.2	.099	.667	.06	5	
3N 13E 32abb1S													
Horswick Hot Springs	466	81.0	95	135	61.8	.054	-	66.8	.108	.179	.071		
3N 14E 28ca1S													
Elk Creek Hot Springs	c15	53.5	80	115	106	.043	-	65.9	.442	.705	.063		
1N 15E 14ada1S													
1S 12E 31cb1	15	31.0	50	85	183	.029	.052	93	.063	1.41	.085		
1S 13E 27cb1	4	35.0	70	120	120	.023	.062	50.1	.096	.585	.071		
Barron's Hot Springs	31	70.0	90	125	67.3	.024	.046	47.9	.114	.574	.07		
1S 13E 34bc1S													
<u>CANYON COUNTY</u>													
2N 2N 34ab1	c700	51.0	55	85	234	.019	.047	54.8	.068	1.44	.062		
6S 41E 19baa1S	c1,300	42.0	370	70	.666	.402	.649	.248	.028	11.3	.995	9	
Soda Springs	-	31.0	80	35	.887	.425	.438	.053	.004	5.25	7.66		
9S 41E 12ada1S													
<u>CASSIA COUNTY</u>													
15S 26E 13bb1	58	93.0	145	135	43.3	1.47	.012	18.4	28.2	84.6	.047	10	
15S 26E 23ad1	60	90.0	140	135	53.4	5.5	.005	14.8	90.8	72.7	.038	10	
11S 23E 11cc1	2,090	60.0	90	110	48	.10	.1	23.4	.757	2.31	.095		
14S 21E 34bd1	c50	43.0	95	95	7.79	.148	.129	5.48	.084	2.89	.309		
Oakley Warm Springs	c10	47.0	90	115	67.3	.096	-	56.2	1.26	3.55	.069		
14S 21E 27dc1S													



TABLE 3 (Cont'd.)

Spring or Well Identification Number	Discharge (gpm)	Water Temperature at Surface (°C)	Aquifer Temperatures from Geometrical Intersectors of Silica <sup>a</sup> Sodium-Potassium-Calcium	Sodium Potassium (Na/K)	Calcium Bicarbonate (Ca/HCO <sub>3</sub> )	Magnesium Calcium (Mg/Ca)	Atomic Ratios			Chloride Fluoride (Cl/F)	$\sqrt{\frac{\text{Calcium Sodium}}{\text{Ca/Na}}}$	Area Number Fig. 6
							Sulfate Plus Carbonate (SO <sub>4</sub> +CO <sub>3</sub> )	Sulfate Plus Carbonate (SO <sub>4</sub> +CO <sub>3</sub> )	Sulfate Plus Carbonate (SO <sub>4</sub> +CO <sub>3</sub> )			
CASSIA COUNTY (Cont'd.)												
155 24E 22dd81	100	38.0	95	45	38.4	0.333	0.414	3.3	0.835	14.8	0.316	
CLARK COUNTY												
Warm Springs 11N 32E 25aac1S	1,920	29.0	60	25	5.81	.393	.58	.52	.044	2.84	2.7	
Lidy Hot Springs 9N 33E 2bbc1S	c250	450.0	85	65	3.06	.74	.303	.541	.077	.714	1.25	
CUSTER COUNTY												
8N 17E 33ba1S	c25	51.0	90	185	13.1	.137	.432	8.3	.191	1.66	.166	11
14N 19E 34da1	50	40.0	70	60	10.1	.371	.629	1.43	.03	1.95	.598	
Sumbeam Hot Springs 11N 15E 19c1S	444	76.0	135	130	60.2	.019	-	98.8	.174	.429	.052	
Sullivan Hot Springs 11N 17E 27bd1S	70	41.0	85	100	19.3	.135	.37	6.05	.177	17	.15	
Barney Hot Springs 11N 25E 23ca1S	170	428.5	60	15	10.2	.311	.891	.424	.038	4.29	2.45	
Stanley Hot Springs 10S 13E 3cab1S	110	41.0	105	45	204	.112	.075	47.5	.147	.191	.09	
ELMORE COUNTY												
Slate Creek Hot Springs 10N 16E 30a1S	185	50.0	130	90	31.4	.112	.02	17.9	.11	.431	.125	
5S 8E 34bdc1	2	34.0	110	145	49.5	.017	.181	61.3	.127	14.4	.034	12
Weinmeyer Hot Springs 5N 7E 24b1S	349	76.0	155	125	63.3	.335	.15	106	.068	.355	.057	
Dutch Frank's Spring												
5N 9E 7b1S	c300	65.0	120	70	80.8	.197	.15	45.2	.072	.129	.094	
Paradise Hot Springs												
3N 10E 35bd1S	-	56.0	115	75	85	.051	.11	58.1	.056	.449	.089	
5S 8E 36cda1	c700	68.0	130	70	185	.031	-	101	.062	.142	.051	
Latty Hot Springs												
5S 10E 31cd1S	-	455.0	135	135	54	.007	-	235	.038	.207	.043	
4S 8E 36bba1	8	38.0	130	125	73.5	.011	.103	87.2	.038	1.79	.041	
4S 9E 8ab1	-	62.0	130	80	174	.017	.045	159	.042	1.07	.042	
5S 10E 7acd1	-	32.0	90	65	149	.033	-	55.1	.08	.163	.073	
5S 10E 32bdb1	54	37.5	95	70	246	.014	.132	90.7	.179	1.2	.044	
FRANKLIN COUNTY												
Maple Grove Hot Springs 13S 41E 7aca1S	350	76.0	105	235	7.58	.276	.444	9.6	2.21	307	.07	13
14S 39E 36ada1	-	44.5	125	170	25.5	.073	.468	25.1	1.05	17.1	.05	13

TABLE 3 (Cont'd.)

Spring or Well Identification Number	Discharge (gpm)	Water Temperature at Surface (°C)	Aquifer Temperatures from Geophysical Thermometers °C (rounded to 5°C) Silica Sodium-Potassium-Calcium	Sodium Potassium (Na/K)	Calcium Bicarbonate (Ca/HCO <sub>3</sub> )	Magnesium Calcium (Mg/Ca)	Atomic Ratios			Area Number Fig. 6	
							Sodium Calcium (Na/Ca)	Bicarbonate plus Carbonate (Cl/HCO <sub>3</sub> + CO <sub>3</sub> )	Chloride Fluoride (Cl/F)		Calcium Sodium (Ca/Na)
FRANKLIN COUNTY (Cont'd.)											
FREMONT COUNTY											
Wayland Hot Springs 155 39E 8bc1S	25	77.0	125	7.99	0.348	0.165	33.8	13.3	241	0.015	13
155 39E 17bc1	25	82.0	155	8.31	.519	.152	30.	18.1	589	.013	13
ASHTON KAYM											
9N 42E 23db1S	2	41.0	145	38.3	.018	.15	57.1	.054	.706	.106	14
8N 44E 34bb1S	92,000	12.0	95	7.94	.185	.177	4.36	.094	.432	.614	
Lily Pad Lake 10N 45E 35ab1S	-	17.0	35	.85	.36	.254	.335	.172	5.89	11.7	
7N 41E 35cd1	-	36.0	120	15.4	.178	.371	4.86	.172	2.38	.246	
GEN COUNTY											
GOODING COUNTY											
Roystone Hot Springs 7N 1E 8da1S	20	65.0	150	35.3	.071	.114	32.1	.571	2.08	.067	15
7N 1E 9cd1S	-	45.0	135	31.8	.135	.264	11.5	.305	2.01	.142	
4S 13E 28ab1	-	47.0	135	28.8	.054	.202	17.8	.051	.366	.114	
White Arrow Hot Springs 4S 13E 30ad1S	826	65.0	135	96.7	.013	-	132	.07	.295	.044	
5S 12E 3aa1	-	43.0	115	191	.029	.103	98.1	.115	.237	.051	
IDAHO COUNTY											
Weir Creek Hot Springs 36N 11E 13b1S	40	47.5	100	98.6	.239	-	15.3	.083	.512	.227	
Jerry Johnson Hot Springs 36N 13L 18a1S	300	48.0	100	157	.171	.122	23.9	.066	.636	.161	
Red River Hot Springs 28N 10E 3d1S	35	55.0	120	86.1	.114	-	52.3	.104	.103	.074	
Riggins Hot Springs 24N 2E 14da1S	50	42.0	120	80	.858	.027	45	.378	2.04	.057	
Burgdorf Hot Springs 22N 4E 1bd1S	162	45.0	120	104	.184	-	37.1	.095	.804	.112	
JEFFERSON COUNTY											
Heise Hot Springs 4N 40E 25dc1S	205	49.0	80	13.4	.623	.3	5.81	3.75	.415	.051	8

TABLE 3 (Cont'd.)

Spring or Well Identification Number	Discharge (gpm)	Water Temperature at Surface (°C)	Aquifer Temperatures from Geochronical Thermometers °C (rounded to 3°C)	Silica	Sodium-Potassium-Calcium	Sodium Potassium (Na/K)	Calcium Bicarbonate (Ca/HCO <sub>3</sub> )	Magnesium Calcium (Mg/Ca)	Ionic Ratios			Chloride Fluoride (Cl/F)	Sulfate Sodium V(Ca/Na)	Area Number Fig. 6
									Sulfate Sodium V(Ca/Na)	Chloride Fluoride (Cl/F)	Chloride Fluoride plus Carbonate (Cl/HCO <sub>3</sub> + CO <sub>3</sub> )			
<b>LEWIS COUNTY</b>														
Big Creek Hot Springs 23N 15E 22c15	275	93.0	160	175	26.7	0.017	0.062	72.4	0.102	1.04	0.038	16		
Salmon Hot Springs 20N 22E 34bd15	145	45.0	80	205	11.5	.062	.788	14.4	.152	14.9	.092	17		
Sharkey Hot Springs 20N 24E 34cc15	8	45.0	135	175	27	.024	.135	64.5	.187	2.28	.056	17		
16N 21E 18ad15	20	46.0	85	165	24.7	.049	.21	25.3	.132	1.99	.075	18		
<b>MADISON COUNTY</b>														
Green Canyon Hot Springs 5N 43E 06ca15	-	44.0	70	5	1.84	1.28	.377	.049	.018	.569	.11	11		
<b>ONEIDA COUNTY</b>														
14S 36E 27ca15	44	25.0	65	230	9.72	.381	.542	8.72	.377	2,810	.047	19		
Pleasantview Hot Springs 15S 35E 34bd15	3,810	25.0	65	175	16.4	.506	.494	4.44	2.44	360	.156	19		
Woodruff Hot Springs 16S 36E 10bb15	-	27.0	80	190	17.8	.436	.57	12.2	6.06	1,430	.046	19		
12S 24E 36cb15	189	24.0	85	35	5.93	.377	.559	.467	.267	62.5	1.81	19		
<b>ONYIEF COUNTY</b>														
4S 2E 31bc1	30	42.0	135	165	29	.016	.281	65.8	.066	1.04	.049	20		
5S 3E 26cb1	280	84.5	145	90	102	.037	-	87.2	.214	.25	.054	20		
6S 3E 26cc1	489	55.0	135	150	39.2	.013	-	121	.164	.556	.046	20		
6S 5E 10dd1	4	38.5	115	145	45.4	.023	.066	85.7	.139	.287	.043	20		
6S 5E 29dc1	3	34.0	135	165	22.4	.074	-	23.6	.184	.556	.103	20		
6S 6E 12cd1	-	37.0	135	175	20.6	.033	.082	29.6	.067	1.72	.066	20		
7S 5E 7ab1	-	39.0	125	190	11.8	.1	.026	13.8	.147	.459	.182	20		
Indian Bath Hot Springs 8S 6E 34bd15	458	39.0	120	185	12.6	.072	.112	16	.109	.487	.165	20		
Murphy Hot Springs 16S 9E 45bd15	270	51.0	125	160	25.5	.014	-	87.2	.058	.342	.094	21		
1N 4W 12db1	410	55.5	85	40	624	.016	-	87.2	.225	1.9	.049	21		
1S 2W 7cb1	169	45.5	90	85	170	.015	-	110	.364	.826	.042	21		
4S 1E 34bd1	-	75.0	125	75	238	.016	.3	135	.146	.556	.039	21		
5S 1E 24d1	1,040	66.0	125	90	213	.017	.137	145	.164	.498	.04	21		
5S 2E 1bb1	30	49.5	115	65	247	.038	-	101	.165	1.02	.031	21		
7S 6E 9bd1	153	50.0	135	130	60.1	.034	-	108	.148	.256	.046	21		
Indian Hot Springs 12S 7E 53c15	1730	69.0	120	60	213	.034	-	87.2	.148	.322	.059	21		

TABLE 3 (Cont'd.)

Spring or Well Identification Number	Discharge (gpm)	Water Temperature at Surface (°C)	Aquifer Temperatures from Geochemical Thermometers or (rounded to 5°C) Silica Sodium-Potassium-Calcium	Sodium Potassium (Na/K)	Calcium Bicarbonate (Ca/HCO <sub>3</sub> )	Magnesium Calcium (Mg/Ca)	Atomic Ratios			Area Number Fig. 6	
							Sodium Calcium (Na/Ca)	Bicarbonate plus Carbonate (C/HCO <sub>3</sub> + CO <sub>3</sub> )	Chloride Fluoride (Cl/F)		
POWER COUNTY											
TWIN FALLS COUNTY											
Indian Springs 85 31E 18ab1S	1,540	52.0	65	70	18.7	0.456	0.412	2.52	1.49	168.0	0.288
105 30E 13cd1S	418	38.0	70	70	7.53	.875	.591	1.17	2.69	167	.562
MIRACIE HOT SPRINGS											
85 14E 31ab1S	350	54.0	135	85	136	.053	-	95.1	.511	.938	.045
85 14E 33cb1	60	59.0	115	110	113	.019	-	158	.367	.965	.038
115 19E 33dd1	1,930	33.0	115	70	3.36	.348	.238	1.1	.219	26.8	1.11
NAT-POO-PAX WARM SPRINGS											
125 17E 31ab1S	30	36.0	65	80	6.65	.195	.679	2.2	.052	2.26	.492
125 19E 1bb1	543	38.0	115	65	4.54	.288	.183	1.55	.145	7.14	.963
165 17E 31ac1S	385	45.5	70	45	4.91	.282	.489	.755	.04	6.79	1.53
VALLEY COUNTY											
Vulcan Hot Springs 14N 6E 11bd1S	600	87.0	150	135	53.3	.023	.092	91	.244	.38	.052
Hot Creek Springs 15N 5E 13bc1S	798	34.0	110	60	170	.116	.127	80.5	.439	3.3	.069
Molly's Hot Springs 15N 6E 14ac1S	20	59.0	130	85	79.4	.063	-	61	.215	.315	.073
14N 3E 36ab1	-	42.5	95	45	247	.039	-	63.2	.306	2.12	.079
Cabarton Hot Springs 13N 4E 31ac1S	70	70.5	125	100	89.5	.037	-	103.0	.974	2.39	.047
Boiling Springs 12N 5E 21bc1S	165	85.0	135	90	71	.036	.087	65.1	.196	.495	.07
WASHINGTON COUNTY											
14N 3W 5ddc1	-	25.5	115	180	18.3	.025	.127	48.9	.038	2.04	.08
13N 3W 3ccc1	-	28.0	130	240	3.4	.039	.152	14.6	.024	2.37	.147
11N 6W 10cc1	175	70.0	170	140	53.4	.045	-	105	.85	6.41	.037
11N 3W 7db1S	10	97.0	170	165	26.9	.208	.043	19.4	1.65	35.1	.063
14N 3W 19cb1S	58	50.0	105	65	71.6	.15	.165	17.4	.315	10	.128
14N 2W 0bb1S	431	70.0	120	80	89.5	1.08	.01	30.5	5.43	39.5	.075
13N 4W 13ba1	-	28.0	120	50	209	.028	.094	42.8	.026	2.45	.079

a Using curve A (equilibrium with quartz) Fournier and Truesdell, 1970.

b Fournier and Truesdell, 1973.

c Discharge estimated.

d Measured temperature is probably lower than temperature at point of discharge.

## SELECTED REFERENCES

- Anderson, A. L., 1931, Geology and mineral resources of eastern Cassia County, Idaho: Idaho Bur. Mines and Geology Bull. 14, 169 p.
- \_\_\_\_\_ 1947, Geology and ore deposits of Boise Basin, Idaho: U. S. Geol. Survey Bull. 944-C, p. 119-319.
- \_\_\_\_\_ 1957, Geology and mineral resources of the Baker quadrangle, Lemhi County, Idaho: Idaho Bur. Mines and Geology, Pamph. 112, 71 p.
- Armstrong, F. C., 1969, Geologic map of the Soda Springs quadrangle, southeastern Idaho: U. S. Geol. Survey Misc. Geol. Inv. Map I-557, 2 sheets.
- Barnes, H. L., ed., 1967, Geochemistry of hydrothermal ore deposits: New York; Holt, Rinehart, and Winston, Inc., 670 p.
- Blackwell, D. D., 1969, Heatflow determinations in the northwestern United States: Jour. Geophys. Research, v. 74, no. 4, p. 992-1007.
- Bodvarsson, G., 1970, Evaluation of geothermal prospects and the objectives of geothermal exploration: *Geoexploration*, 8, 7.
- Burnham, W. L., Harder, A. H., and Dion, N. P., 1969, Availability of ground water for large-scale use in the Malad Valley-Bear River areas of southeastern Idaho - an initial assessment: U. S. Geol. Survey Open-File Report, 40 p.
- Chasteen, A. J., 1972, Geothermal energy - growth spurred on by powerful motives: *Mining Engineering*, v. 24, no. 10, p. 100.
- Choate, Raoul, 1962, Geology and ore deposits of the Stanley area: Idaho Bur. Mines and Geology Pamph. 126, 122 p.
- Crosthwaite, E. G., 1957, Ground-water possibilities south of the Snake River between Twin Falls and Pocatello, Idaho: U. S. Geol. Survey Water-Supply Paper 1460-C, p. 99-145.
- \_\_\_\_\_ 1969<sub>a</sub>, Water resources of the Goose Creek-Rock Creek area, Idaho, Utah and Nevada: Idaho Dept. of Reclamation Water Information Bull. 8, 73 p.
- \_\_\_\_\_ 1969<sub>b</sub>, Water resources of the Salmon Falls Creek Basin, Idaho-Nevada: U. S. Geol. Survey Water-Supply Paper 1879-D, 33 p.
- Dion, N. P., 1969, Hydrologic reconnaissance of the Bear River Basin in southeastern Idaho: Idaho Dept. of Reclamation Water Information Bull. 13, 66 p.
- Dion, N. P., and Griffith, M. L., 1967, A ground-water monitoring network for southwestern Idaho: Idaho Dept. of Reclamation Water Information Bull. 2, 16 p.

## SELECTED REFERENCES (Cont'd.)

- Ellis, A. J., 1970, Quantitative interpretation of chemical characteristics of hydrothermal system, in Proceedings United Nations Symp. on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, Part 1 Geothermics, Spec. Issue 2, p. 516-528.
- Fenneman, N. M., 1931, Physiography of Western United States: New York, McGraw-Hill Book Co., 534 p.
- Forrester, J. D., 1956 Geology and mineral resources of the Salmon quadrangle, Lemhi County, Idaho: Idaho Bur. Mines and Geology Pamph. 106, 102 p.
- Fournier, R. O., and Rowe, J. J., 1966, Estimation of underground temperatures from the silica content of water from hot springs and wet steam wells: Am. Journ. Sci., v. 264, p. 685-695.
- Fournier, R. O., and Truesdell, A. H., 1970, Chemical indicators of subsurface temperature applied to hot waters of Yellowstone National Park, Wyo., U. S. A., in Proceedings United Nations Symp. on the Development and Utilization of Geothermal Energy, Pisa, 1970, v. 2, Part 1 Geothermics, Spec. Issue 2, p. 529-535.
- \_\_\_\_\_ 1973, An empirical Na-K-Ca geothermometer for natural waters: Geochim. et Cosmochim. Acta. (in press).
- Godwin, L. H., Haigler, L. B., Rioux, R. L., White, D. E., Muffler, L. J. P., and Wayland, R. G., 1971, Classification of public lands valuable for geothermal steam and associated geothermal resources: U. S. Geol. Survey Circ. 647, 17 p.
- Greenberg, S. A., and Price, E. W., 1957, The solubility of silica in solutions of electrolytes: J. Phys. Chem. 61, p. 1539-1541.
- Grose, L. T., 1971, Geothermal energy: geology, exploration, and developments; Part 1: Colorado School Mines Research Inst. Min. Industries Bull., v. 14, no. 6, 14 p.
- Hamilton, Warren, 1965, Geology and petrogenesis of the Island Park Caldera of rhyolite and basalt, eastern Idaho, in Shorter Contributions to General Geology, 1964: U. S. Geol. Survey Prof. Paper 504-C, p. C1-C37.
- \_\_\_\_\_ 1969, Reconnaissance geologic map of the Riggins quadrangle, west-central Idaho: U. S. Geol. Survey Misc. Inv. Map I-579, 1 sheet.
- Healy, J., 1970, Pre-investigation geological appraisal of geothermal fields, in Proceedings United Nations Symp. on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, Part 1 Geothermics, Spec. Issue 2, p. 571-577.
- Holland, H. D., 1965, Some applications of thermochemical data to problems of ore deposits, II. Mineral assemblages and the composition of ore-forming fluids: Econ.

## SELECTED REFERENCES (Cont'd.)

- Geology v. 60, p. 1101-1166.
- Jobin, D. A., and Schroeder, M. L., 1964, Geology of the Conant Valley quadrangle, Bonneville County, Idaho: U. S. Geol. Survey Mineral Inv. Field Studies Map MF-277, 1 sheet.
- Kirkham, V. R. D., 1924, Geology and oil possibilities of Bingham, Bonneville, and Caribou Counties, Idaho: Idaho Bur. Mines and Geology Bull. 8, 108 p.
- Littleton, R. T., and Crosthwaite, E. G., 1957, Ground-water geology of the Bruneau-Grandview area, Owyhee County, Idaho: U. S. Geol. Survey Water-Supply Paper 1460-D, p. 147-198.
- Livingston, D. C., and Laney, F. B., 1920, The copper deposits of the Seven Devils and adjacent districts including Heath, Hornet Creek, Hoodoo, and Deer Creek: Idaho Bur. Mines and Geology Pamph. 1, 105 p.
- Mahon, W. A. J., 1970, Chemistry in the exploration and exploitation of hydrothermal systems, in Proceedings United Nations Symp. on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, Part 2 Geothermics, Spec. Issue 2.
- Malde, H. E., Powers, H. A., and Marshall, C. H., 1963, Reconnaissance geologic map of west-central Snake River Plain, Idaho: U. S. Geol. Survey Misc. Geol. Inv. Map I-373, 1 sheet.
- Malde, H. E. and Powers, H. A., 1972, Geologic map of the Glens Ferry-Hagerman area, west-central Snake River Plain, Idaho: U. S. Geol. Survey Misc. Geol. Inv. Map I-696, 2 sheets.
- Mansfield, G. R., 1927, Geography, geology, and mineral resources of part of southeastern Idaho: U. S. Geol. Survey Prof. Paper 152, 453 p.
- Meinzer, O. E., 1924, Origin of the thermal springs of Nevada, Utah, and southern Idaho: Jour. Geology, v. 32, no. 4, p. 295-303.
- Nace, R. L., and others, 1961, Water resources of the Raft River Basin, Idaho-Utah: U. S. Geol. Survey Water-Supply Paper 1582, 138 p.
- Newcomb, R. C., 1970, Tectonic structure of the main part of the basalt of the Columbia River Group, Washington, Oregon, and Idaho: U. S. Geol. Survey Misc. Geol. Inv. Map I-587, 1 sheet.
- Norvitch, R. F., and Larson, A. L., 1970, A reconnaissance of the water resources in the Portneuf River Basin, Idaho: Idaho Dept. of Reclamation Water Information Bull. 16, 58 p.

## SELECTED REFERENCES (Cont'd.)

- Piper, A. M., 1923, Geology and water resources of the Goose Creek Basin, Cassia County, Idaho: Idaho Bur. Mines and Geology Bull. 6, 78 p.
- \_\_\_\_\_ 1924, Possibilities of petroleum in Power and Oneida Counties, Idaho: Idaho Bur. Mines and Geology Pamph. 12, 24 p.
- Ralston, D. R., and Chapman, S. L., 1968, Ground-water resources of the Mountain Home area, Elmore County, Idaho: Idaho Dept. of Reclamation Water Information Bull. 4, 63 p.
- \_\_\_\_\_ 1969, Ground-water resources of northern Owyhee County, Idaho: Idaho Dept. of Reclamation Water Information Bull. 14, 85 p.
- Ross, C. P., 1937, Geology and ore deposits of the Bayhorse region, Custer County, Idaho: U. S. Geol. Survey Bull. 877, 161 p.
- \_\_\_\_\_ 1963, Geology along U. S. Highway 93 in Idaho: Idaho Bur. Mines and Geology Pamph. 130, 98 p.
- Ross, C. P., and Forrester, J. D., 1947, Geologic map of the State of Idaho: U. S. Geol. Survey and Idaho Bur. Mines and Geology, 1 map.
- Ross, S. H., 1971, Geothermal Potential of Idaho: Idaho Bur. Mines and Geology Pamph. 150, 72 p.
- Savage, C. N., 1958, Geology and mineral resources of Ada and Canyon Counties: Idaho Bur. Mines and Geology County Report 3, 94 p.
- Siever, R., 1962, Silica solubility, 0°-200°C, and diagenesis of siliceous sediments: Jour. Geol. v. 70, p. 127-150.
- Smith., R. O., 1959, Ground-water resources of the middle Big Wood River-Silver Creek area. Blaine County, Idaho: U. S. Geol. Survey Water-Supply Paper 1478, 64 p.
- Stearns, H. T., Bryan, L. L., and Crandall, Lynn, 1939, Geology and water resources of Mud Lake Region, Idaho, including the Island Park area: U. S. Geol. Survey Water-Supply Paper 818, 125 p.
- Stearns, H. T., Crandall, Lynn, and Steward, W. G., 1938, Geology and ground-water resources of the Snake River Plain in southeastern Idaho: U. S. Geol. Survey Water-Supply Paper 774, 268 p.
- Stearns, N. D., Stearns, H. T., and Waring, G. A., 1937, Thermal springs in the United States: U. S. Geol. Survey Water-Supply Paper 679-B, p. 59-206.



## SELECTED REFERENCES (Cont'd.)

- Umpleby, J. B., 1913, Ore deposits in the Sawtooth quadrangle, Blaine and Custer Counties, Idaho, in *Contributions to Economic Geology: U. S. Geol. Survey Bull.* 580, p. 221-249.
- Umpleby, J. B., Westgate, L. G., and Ross, C. P., 1930, *Geology and ore deposits of the Wood River region, Idaho: U. S. Geol. Survey Bull.* 814, 250 p.
- Walker, E. H., and Sisco, H. G., 1964, *Ground-water in the Midvale and Council areas, Upper Weiser River Basin, Idaho: U. S. Geol. Survey Water-Supply Paper* 1779-Q, 26 p.
- Walton, W. C., 1962, *Ground-water resources of Camas Prairie, Camas and Elmore Counties, Idaho: U. S. Geol. Survey Water-Supply Paper* 1609, 57 p.
- Waring, G. H., (revised by R. R. Blankenship and Ray Bentall), 1965, *Thermal springs of the United States and other countries of the world - a summary: U. S. Geol. Survey Prof. Paper* 492, 383 p.
- White, D. E., 1970, *Geochemistry applied to the discovery, evaluation, and exploitation of geothermal energy resources, in Proceedings United Nations Symp. on the Development and Utilization of Geothermal Energy, Pisa, 1970, v. 1, Part 2 Geothermics, Spec. Issue 2.*
- White, D. E., Muffler, L. J. P., and Truesdell, A. H., 1971, *Vapor-dominated hydrothermal systems compared with hot-water systems: Econ. Geol., v. 66, no. 1, p. 75-97.*

## MONTANA

### Summary

The western third of the state is block faulted and composed in part of Tertiary granitic and volcanic rocks where the thermal springs occur. Approximately 40 thermal springs occur in the state, and their temperatures range up to 80°C. There is one KGRA near Yellowstone National Park in the southern part of the state. Flow rates range up to 50,000 gpm. The water quality is good in the areas underlain by granitic rocks of the Idaho Batholith, but water quality is variable in the region of Yellowstone Park.

The steepest geothermal gradient measured is in the area around Marysville, where conducted heat flow range is from 3.2 to 19.5 u cal/cm<sup>2</sup> sec. At one locality, a temperature of 58°C was measured at a depth of only 220 meters. Drilling has indicated that the immediate source of the high heat flow is an unexposed reservoir of thermal fluids rather than a shallow still cooling magma chamber. No water of substantial quantity has been located. The drilling confirms the convective circulation of water within the fractured granite to depths of 6000 feet, and thus casts doubts on the relative abundance of hot "dry" geotherman systems. Reference: Blackwell, D. D. and Baag, C. G., Heat Flow in a "Blind" Geothermal Area near Marysville, Montana, Geophysics, Vol. 38, No. 5, p. 941-956 (1973).

## NEVADA

### Summary

The chemical composition and minimum thermal reservoir temperatures of 55 hot springs in northern and central Nevada are described in a U.S.G.S. open file report issued in May, 1974 (Mariner, et.al.).

Major element constituents: sodium, potassium, calcium and silica suggest minimum thermal aquifer temperatures of 140°C or more at 16 of the hot spring complexes. At least five of the hot springs issue mixed waters, and true thermal aquifer temperatures may be significantly higher than indicated by the water chemistry.

Sodium is the principal cation in these waters, while the anions may be bicarbonate, chloride or a mixture of chloride, bicarbonate and sulfate. The pH values range from 6.47 to 9.38. Boron content ranges from less than .02 ppm to 47 ppm, with 18 springs having less than 1.0 ppm and eight springs having from 1.0 to 2.0 ppm. Boron was not reported for 13 springs. Specific conductance values range from 321 to 6,910. The sampled hot springs have chemical compositions characteristic of hot water-dominated systems.

Most of the thermal springs are along permeable zones associated with faults. Quaternary alluvium and fine-grained sediments cover the fault lines along which many of the springs issue.

Thirteen KGRA's have been designated in Nevada, and most of the north-western part of the state is of potential geothermal importance.

## Geologic Setting

Nevada, part of the Basin and Range Province, consists of roughly parallel fault-block mountain ranges separated by alluvial-filled valleys. Exposed rocks range in age from Precambrian to Quaternary. Precambrian and Paleozoic rocks crop out in the mountain ranges of eastern Lander, Eureka and southern Elko Counties (Montgomery, 1965). Mesozoic sedimentary and volcanic rocks are widespread in Pershing and southeastern Humboldt Counties; Mesozoic granitic intrusive rocks are common in western Pershing and northern Humboldt Counties, as well as near the Sierra Nevada of California. Cenozoic volcanic rocks and related sedimentary rocks are predominant in Washoe, northern Humboldt, northern Elko, Churchill, western Lander and Nye Counties.

The types and ages of rock exposed near the springs, as well as selected references on the geology of the area around the spring, are listed in Table 4. Most of the thermal springs are along permeable zones associated with faults. Quaternary alluvium and lacustrine rocks cover the fault lines along which many of the springs issue.

Eight of the thermal springs having estimated thermal-aquifer temperatures of 140°C or more have a poorly-developed regional trend. This regional trend extends from Wabuska Hot Springs in Lyon County, northeast through Churchill, Pershing, and into southeastern Humboldt County. The Stillwater Range of Churchill County and East Range of Pershing County parallel this trend.

## Tabular Data

Table 1 lists the springs by name, location and topographic map coverage.

Table 2 contains measured temperature, specific conductance, pH values and major chemical constituents.

Table 3 contains the estimated thermal-aquifer temperatures and flow rates. The underlined values represent the "best" estimate of thermal-aquifer temperature.

Table 4 lists the types and ages of rock exposed near the springs and selected references on the geology of the surrounding areas.

## Water Chemistry

The chemical analyses (Table 2) show that most of the thermal springs discharge sodium bicarbonate water. However, thermal springs along the western side of the state, Washoe County, discharge sodium chloride waters. Some sodium-mixed anion waters occur around the Black Rock Desert in Humboldt County. Sodium and calcium occur in approximately equal molar amounts in the water from Diana's Punch Bowl, the springs at Potts Ranch and East Ruby Marsh, and Walti Hot Springs. Water from Nile Spring is a calcium magnesium bicarbonate type.

The waters range in pH from 6.47 at Walti Hot Springs to 9.38 at the Beowawe "steam" well. Specific conductance is highest for sodium chloride waters, such as the discharge from Great Boiling Spring at Gerlach, and lowest for the sodium calcium bicarbonate or calcium magnesium bicarbonate waters mentioned above.

### Thermal-Aquifer Temperature Estimates

Thermal-aquifer temperatures estimated from the chemical analyses indicate 16 different spring complexes where the waters have circulated through rock having a temperature of at least 140°C. The flowing well in Stillwater, Beowawe Hot Springs, Wabuska Hot Springs, Leach Hot Springs, Great Boiling Spring and Steamboat Springs all have estimated minimum thermal-aquifer temperatures of at least 140°C and are in areas designated as known geothermal resource areas by Godwin and others (1971). Other areas of geothermal potential not in known geothermal resource areas are: Lee Hot Springs and Dixie Valley Hot Springs in Churchill County, Sulphur Hot Springs and two unnamed hot springs near Wells in Elko County, Pinto Hot Springs, Baltazor Hot Spring, and an unnamed hot spring (Hot Springs Ranch) in Humboldt County, an unnamed hot spring (Smith Creek Valley) in Lander County, as well as Kyle Hot Springs and an unnamed hot spring (Jersey Valley) in Pershing County. Dixie Valley Hot Springs, Mineral Hot Springs, an unnamed hot spring (near Wells), Double Hot Spring and Dyke Hot Spring may be mixed waters. The thermal-aquifer temperatures estimated from the water compositions may be significantly below the true thermal-aquifer temperature.

## SUMMARY OF TEMPERATURE DATA

The major portion of Nevada's hot springs are found in the northern 1/2 of the state. The Basin and Range physiographic province has the highest heat flow, and the west central and north central portions of the state have the highest hot spring temperatures. The Battle Mountain high in north-central Nevada has a heat flow of 3 HFU. The average heat flow for Nevada is general is 2 HFU. Spring temperatures range up to 208°F (98°C). Discharge ranges up to 12,000 gpm.

### Hottest areas:

#### Steamboat Springs

Max. Spring Temp - 203°F (95°C)

Max Well Temp - 369°F (187°C)

Hot water present with 5-10% steam flashover.

Numerous homes are heated from warm water wells in the Reno area.

#### The Needles (Pyramid Lake)

Max. Spring Temp - 208°F (98°C)

Max. Well Temp - 240°F (116°C)

Large flow

#### Wards Hot Springs (Fly Ranch)

Max Spring Temp - 203°F (95°C)

Max. Well Temp - 220°F (104°C)

Largest hot springs in northwestern Nevada

#### Beowawe Geysers

Max Spring Temp - 205°F (96°C)

Max Well Temp - 414°F (212°C)

Hot water with 5-10% steam flashover.

Problems of scaling and cold water inflow.

#### Wabuska Hot Springs

Max Spring Temp - 162°F (72°C)

Max Well Temp - 222°F (106°C)

Hot water used for greenhouse heating

#### Darrough Hot Springs

Max Spring Temp - 207°F (97°C)

Max Well Temp - 265°F (129°C)

Very large flow of hot water, little steam.

#### Brady's Hot Springs

Max Spring Temp - 194°F (90°C)

Max Well Temp - 418°F (214°C)

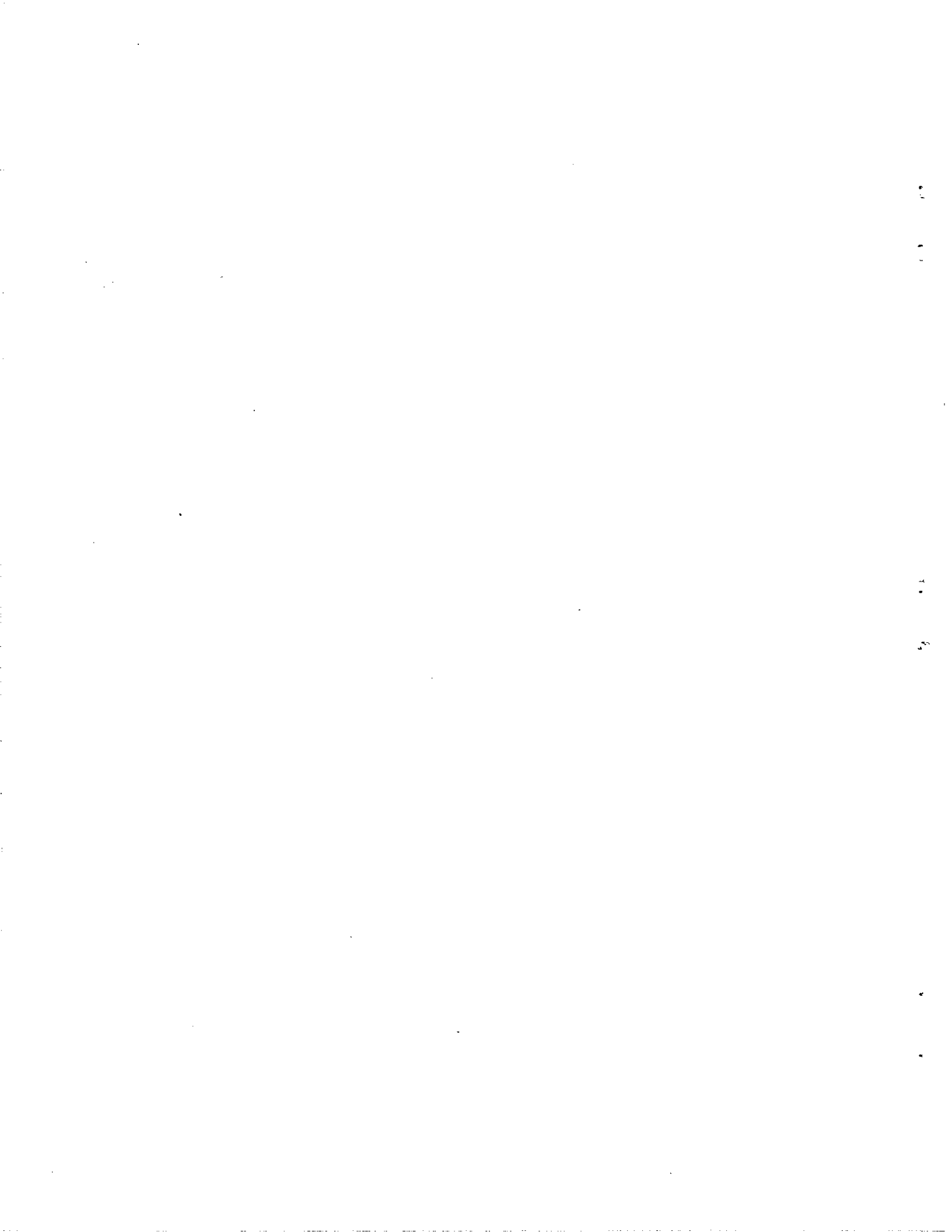
Hot water with 5% steam flashover.

Problem of scaling.

#### Stillwater

Max Well Temp - 240°F (116°C)

Some water wells drilled in this area encountered hot water and steam which have been used for space heating. No springs or other surface features.





NEVADA BUREAU OF MINES AND GEOLOGY

TABLE I  
Exploratory geothermal drilling in Nevada through 1973<sup>1</sup>

Operator	Name	API No. <sup>2</sup>	Location	Depth	Maximum Spring Temp. (°F)	Maximum Well Temp. (°F)	Completion Date	Remarks
1. Steamboat Hot Springs Nevada Thermal Power Co. Nevada Thermal Power Co. Nevada Thermal Power Co. Nevada Thermal Power Co. Nevada Thermal Power Co. Nevada Thermal Power Co.	Steamboat No. 1 <sup>3</sup>	27-031-90000	NW/4, NE/4, S28, T18N, R20E	1830	203	369	1954	Hot water present with 5-10% steam flashover. Eight core holes drilled by the U. S. Geological Survey (1950), for a total footage of 3316 feet. Also several wells for hot water baths, etc. Numerous homes are heated from warm water wells in the Reno area. For more information see White (1968).
	Steamboat No. 2 <sup>3</sup>	27-031-90001	SE/4, SW/4, S28, T18N, R20E	964			1959	
	Steamboat No. 3 <sup>3</sup>	27-031-90002	NW/4, NE/4, S32, T18N, R20E	1263			1960?	
	Steamboat No. 4 <sup>3</sup>	27-031-90003	NE/4, NW/4, S32, T18N, R20E	520?			1960	
	Steamboat No. 5 <sup>3</sup>	27-031-90004	NW/4, NW/4, S32, T18N, R20E	826			1961	
	Steamboat No. 6 <sup>3</sup>	27-031-90005	NW/4, NW/4, S32, T18N, R20E	716			1961	
2. The Needles (Pyramid Lake) Western Geothermal Inc. Western Geothermal Inc. Western Geothermal Inc.	Needles No. 1	27-031-90006	NW/4, SW/4, SW/4, S6, T26N, R21E	5888	208	240	1965	Large flow of hot water.
	Needles No. 2(?) <sup>3</sup>	27-031-90007	C/W/2, NE/4, S12, T26N, R21E	~4000?			1964	
	Needles No. 3(?) <sup>3</sup>	27-031-90008	NW/4, SW/4, SW/4, S6, T26N, R21E	?			1964	
3. Wards Hot Springs (Fly Ranch) Western Geothermal Inc. Western Geothermal Inc.	Fly Ranch No. 1(?) <sup>3</sup>	27-031-90009	SW/4, S2, T34N, R23E	1000+	203	>220	1964	Largest hot springs in northwestern Nevada.
	Granite Creek Ranch No. 1(?) <sup>3</sup>	27-031-90010	S35(?) <sup>3</sup> , T34N, R23E	800	193	190	1965?	
4. Monte Neva Hot Springs Magma Power Co.	Monte Neva No. 1(?) <sup>3</sup>	27-033-90000	S24(?) <sup>3</sup> , T31N, R63E	402	205	414	1965	Melvin (Goodrich) Hot Springs
	Beowawe No. 1	27-011-90000	NE/4, SE/4, NW/4, S17, T32N, R48E	1918			1959?	
5. Beowawe Geysers Magma Power Co. Vulcan Thermal Power Co. Vulcan Thermal Power Co. Vulcan Thermal Power Co. Vulcan Thermal Power Co. Vulcan Thermal Power Co. Vulcan Thermal Power Co. Vulcan Thermal Power Co. Sierra No. 1 Sierra No. 2 Sierra No. 3 Sierra No. 4 Chervon-American Thermal Res.	Beowawe No. 2	27-011-90001	SE/4, SW/4, NW/4, S17, T32N, R48E	715			1959?	
	Vulcan No. 1	27-011-90002	NW/4, SW/4, NW/4, S17, T32N, R48E	715?			1961	
	Vulcan No. 2	27-011-90003	C, SE/4, NW/4, S17, T32N, R48E	655?			1961	
	Vulcan No. 3	27-011-90004	NE/4, SW/4, NW/4, S17, T32N, R48E	795 or 715			1961	
	Vulcan No. 4	27-011-90005	S17, T32N, R48E	767			1961	
	Vulcan No. 5	27-011-90006	S17, T32N, R48E	237			1963?	
	Vulcan No. 6	27-011-90007	NW/4, SW/4, NE/4, S17, T32N, R48E	478			1963	
	Sierra No. 1	27-011-90008	S17, T32N, R48E	927			1964?	
	Sierra No. 2	27-011-90009	S17, T32N, R48E	397			1964?	
	Sierra No. 3	27-011-90010	NW/4, SE/4, NW/4, S17, T32N, R48E	2052			1964?	
	Sierra No. 4	27-011-90011	NW/4, NE/4, NW/4, S17, T32N, R48E	1005			1964?	
	Chervon-American Thermal Res.	Ginn No. 1-13	27-015-90000	C, SE/4, SE/4, S13, T31N, R47E	-			-
	6. Hot Springs Point (Crescent Valley) Magma Power Co.	Hot Springs Point No. 1(?) <sup>3</sup>	27-011-90012	S1, 2, or 11, T29N, R48E	410	122	166	1965

Table extracted from: Garside, L. J., Geothermal Exploration and Development in Nevada through 1973, Mackay School of Mines at University of Nevada, Report #21, 12 pgs. (1974).

GEOHERMAL EXPLORATION AND DEVELOPMENT IN NEVADA THROUGH 1973

TABLE 1 (Continued)  
Exploratory geothermal drilling in Nevada through 1973<sup>1</sup>

Operator	Name	API No. <sup>2</sup>	Location	Depth	Maximum Spring Temp. (°F)	Maximum Well Temp. (°F)	Completion Date	Remarks
7. Wabuska Hot Springs Magma Power Co. Magma Power Co. Magma Power Co.	Wabuska No. 1	27-019-90000	S167, T15N, R25E	488	162	222	1959	Hot water used for green-house heating.
	Wabuska No. 2	27-019-90001	SE/4, NE/4, SW/4, S16, T15N, R25E	532?			1959	
	Wabuska No. 3	27-019-90002	NE/4, SE/4, SW/4, S16, T15N, R25E	2223			1959	
8. Fernley (Hazen) Magma Power Co. Magma Power Co. Magma Power Co.	Hazen No. 1(?) <sup>3</sup>	27-019-90003	SW/4, S187, T20N, R26E	750	?	270	1962	Patua Hot Springs.
	Hazen No. 2(?) <sup>3</sup>	27-019-90004	S187, T20N, R26E	~300?			1962	
	Hazen No. 3(?) <sup>3</sup>	27-019-90005	S187, T20N, R26E	~300?			1962	
9. Hind's Hot Springs U. S. Steel Corp. U. S. Steel Corp. U. S. Steel Corp.	Hind's No. 1(?) <sup>3</sup>	27-019-90006	SW/4, SE/4, S16, T12N, R23E	?	144	?	1962?	Hot water in wells cooler than springs at surface.
	Hind's No. 2(?) <sup>3</sup>	27-019-90007	SW/4, SE/4, S16, T12N, R23E	?			1962?	
	Hind's No. 3(?) <sup>3</sup>	27-019-90008	SW/4, SE/4, S16, T12N, R23E	?			1962?	
10. Darrough Hot Springs Magma Power Co.	Darrough No. 1(?) <sup>3</sup>	27-023-90000	S177, T11N, R43E	812	207?	265	1962	Very large flow of hot water, little steam.
11. Brady's Hot Springs Magma Power Co. Magma Power Co. Magma Power Co. Magma Power Co. Magma Power Co. Magma Power Co. Magma Power Co. Magma Power Co. Earth Energy Inc.	Brady No. 1	27-001-90000	NE/4, NE/4, SW/4, S12, T22N, R26E	700?	194	418	1959?	Hot water with 57 steam flashover. Problem of scaling.
	Brady No. 2	27-001-90001	NE/4, NE/4, SW/4, S12, T22N, R26E	241			1959?	
	Brady No. 3	27-001-90002	SE/4, SE/4, NW/4, S12, T22N, R26E	610			1961?	
	Brady No. 4	27-001-90003	SE/4, SE/4, NW/4, S12, T22N, R26E	723			1961?	
	Brady No. 5	27-001-90004	NW/4, SW/4, NE/4, S12, T22N, R26E	593			1961?	
	Brady No. 6	27-001-90005	NW/4, SW/4, NE/4, S12, T22N, R26E	770			?	
	Brady No. 7	27-001-90006	NW/4, SW/4, NE/4, S12, T22N, R26E	250			?	
	R. Brady EE No. 1	27-001-90007	S127, T22N, R26E	5062?			1964	
	Brady Pros. No. 1	27-001-90008	S127, T22N, R26E	1758?			1965?	
12. Stillwater O'Neill Geothermal, Inc.	Joseph I. O'Neill, Jr. Reynolds No. 1	27-001-90009	NE/4, SW/4, SW/4, S6, T19N, R31E	~4200?	-	240	1964	Some water wells drilled in this area encountered hot water and steam which have been used for space heating. No springs or other surface features.
13. Wally's Hot Springs U. S. Steel Corp. U. S. Steel Corp.	Wally's No. 1	27-005-90000	SE/4, NW/4, NW/4, S22, T13N, R19E	1268	160	181	1962	Twenty-six shallow holes were also drilled to measure the temperature gradient.
	Wally's No. 1	27-005-90001	SW/4, SW/4, NW/4, S22, T13N, R19E	499			1962	

<sup>1</sup>Listing does not include thermal water wells or wells drilled to exploit thermal waters for spas, swimming pools, space heating, etc.

<sup>2</sup>The American Petroleum Institute Unique well number system has been applied to geothermal wells as well as oil and gas wells, and is recommended for the unique identification of wells by all agencies of industry and government.

<sup>3</sup>Name assigned by Nevada Bureau of Mines and Geology; original name unknown.

Table 1.--Location and topographic map coverage of selected hot springs and wells

Spring or well	Location	Topographic map coverage
Churchill County		
1 Lee Hot Springs	Unsurveyed (lat. 39°12' N., long. 118°43' W)	Allen Springs, Nev. (15'); Reno, Nev. (2°)
2 Dixie Valley Hot Springs	SE 1/4 sec. 5 and NE 1/4 sec. 8, T. 22N., R. 35E.	Dixie Hot Springs, Nev. (15'); Reno, Nev. (2°)
3 Flowing well in Stillwater	SW 1/4 sec. 7, T. 19N., R. 31E.	Stillwater, Nev. (15'); Reno, Nev. (2°)
Douglas County		
1 Walleys Hot Springs	NE 1/4 sec. 22, T. 13N., R. 19E.	Minden, Nev.-Calif. (7-1/2'); Walker Lake, Calif.-Nev. (2°)
Elko County		
1 Hot Hole	NE 1/4 sec. 21, T. 34N., R. 55E.	Elko, west, Nev. (7-1/2'); Elko, Nev.-Utah (2°)
2 Sulphur Hot Springs	NW 1/4 sec. 11, T. 31N., R. 59E.	Lamoille, Nev. (15'); Elko, Nev.-Utah (2°)
3 Unnamed hot spring (Hot Creek)	NW 1/4 sec. 12, T. 28N., R. 52E.	Pine Valley, Nev. (15'); Winnemucca, Nev. (2°)
4 Nile Spring	SW 1/4 sec. 30, T. 47N., R. 70E.	Goose Creek, Nev.-Utah-Idaho (15'); Wells, Nev.-Utah-Idaho (2°)
5 Mineral Hot Spring	sec. 16, T. 45N., R. 64E.	Delaplain, Nev.-Idaho (15'); Wells, Nev.-Utah-Idaho (2°)
6 Unnamed hot spring near Wells	sec. 20, T. 38N., R. 62E.	Oxley Peak, Nev. (7-1/2'); Wells, Nev.-Utah-Idaho (2°)
7 Unnamed hot spring near Wells	NE 1/4 sec. 17, T. 38N., R. 62E.	Oxley Peak, Nev. (7-1/2'); Wells, Nev.-Utah-Idaho (2°)
8 Unnamed hot spring (Wild Horse Reservoir)	SE 1/4 sec. 4, T. 43N., R. 55E.	Wild Horse, Nev. (15'); Wells, Nev.-Utah-Idaho (2°)
9 Unnamed hot springs (SSE Patsville)	Unsurveyed (lat. 41°5' N., long. 115°55' W)	Mountain City, Nev.-Idaho (15'); Wells, Nev.-Utah-Idaho (2°)
10 Hot Sulphur Springs	NE 1/4 sec. 8, T. 41N., R. 52E.	Tuscarora, Nev. (15'); McDermitt, Nev.-Ore.-Idaho (2°)
11 Unnamed hot spring near Carlin	sec. 33, T. 33N., R. 52E.	Carlin, Nev. (15'); Winnemucca, Nev. (2°)
12 Unnamed hot spring near Ruby Marsh	NW 1/4 sec. 2, T. 27N., R. 58E.	Ruby Lake NW, Nev. (7-1/2'); Elko, Nev.-Utah (2°)
Eureka County		
1 Waiti Hot Springs	SW 1/4 sec. 33, T. 24N., R. 48E.	Waiti Hot Spring, Nev. (15'); Millet, Nev. (2°)
2 Hot Springs Point	NE 1/4 sec. 11, T. 29N., R. 48E.	Crescent Valley, Nev. (15'); Winnemucca, Nev. (2°)
3 Beowawe "steam" well	NW 1/4 sec. 17, T. 31N., R. 48E.	Dunphy, Nev. (15'); Winnemucca, Nev. (2°)
4 Beowawe Hot Spring	SE 1/4 sec. 8, T. 31N., R. 48E.	Dunphy, Nev. (15'); Winnemucca, Nev. (2°)
5 Bartholomae Hot Springs	SE 1/4 sec. 28, T. 18N., R. 50E.	Antelope Peak, Nev. (15'); Millet, Nev. (2°)

Table 1. Location and topographic map coverage of selected hot springs and wells Continued

Spring or well	Location	Topographic map coverage
	Humboldt County	
1 Unnamed hot spring (Hot Springs Ranch)	SE 1/4 sec. 5, T. 33N., R. 40E.	Edna Mtn., Nev. (15'); Winnemucca, Nev. (2°)
2 Unnamed hot spring near Colconda	SE 1/4 sec. 29, T. 36N., R. 40E.	Edna Mtn., Nev. (15'); Winnemucca, Nev. (2°)
3 Double Hot Springs	sec. 4, T. 36N., R. 26E.	; Vya, Nevada-Oregon (2°)
4 Unnamed hot springs in Soldier Meadows	sec. 23, T. 40N., R. 24E.	; Vya, Nevada-Oregon (2°)
5 West Pinto Hot Spring (well)	Unsurveyed (lat. 41°20'N., long. 118°48'W)	; Vya, Nevada-Oregon (2°)
6 East Pinto Hot Spring	Unsurveyed (lat. 41°21'N., long. 118°47'W)	; Vya, Nevada-Oregon (2°)
7 Dyke Hot Spring	SE 1/4 sec. 25, T. 43N., R. 39E.	Duffer Peak, Nev. (15'); Vya, Nevada-Oregon (2°)
8 Flowing well near Bakazor Hot Spring	NW 1/4 sec. 13, T. 46 N., R. 28E.	Denio, Nev.-Ore. (15'); Vya, Nevada-Oregon (2°)
9 Baltazor Hot Springs	NW 1/4 sec. 13, T. 46N., R. 28E.	Denio, Nev.-Ore. (15'); Vya, Nevada-Oregon (2°)
10 Bog Hot Springs	NW 1/4 sec. 18, T. 46N., R. 28E.	Railroad Point, Nev.-Ore. (15'); Vya, Nevada-Oregon (2°)
11 Hot Pot	SW 1/4 sec. 11, T. 35N., R. 43E.	Hot Pot, Nev. (7-1/2'); Winnemucca, Nev. (2°)
12 Howard Hot Springs	NE 1/4 sec. 4, T. 44N., R. 31E.	Duffer Peak, Nev. (15'); Vya, Nevada-Oregon (2°)
13 The Hot Springs	NE 1/4 sec. 20, T. 41N., R. 41E.	Hot Springs Peak, Nev. (15'); McDermitt, Nev.-Ore.-Idaho (2°)
	Lander County	
1 Spencer Hot Springs	Unsurveyed (lat. 39°49' N., long. 116°51' W)	Spencer Hot Springs, Nev. (15'); Millet, Nev. (2°)
2 Unnamed hot spring (Valley of the Moon)	NE 1/4 sec. 23, T. 27N., R. 43E.	The Cedars, Nev. (15'); Winnemucca, Nev. (2°)
3 Unnamed hot spring (Smith Creek Valley)	sec. 11, T. 17N., R. 39E.	; Millet, Nev. (2°)
4 Buffalo Valley Hot Springs	SE 1/4 sec. 23, T. 29N., R. 41E.	Buffalo Springs, Nev. (15'); Winnemucca, Nev. (2°)
	Lyon County	
1 Wabuska Hot Springs	SE 1/4 sec. 16, T. 15N., R. 25E.	Wabuska, Nev. (15'); Tonopah, Nev. (2°)
2 Nevada Hot Springs	SE 1/4 sec. 16, T. 12N., R. 23E.	Wellington, Nev. (15'); Walker Lake, Calif.-Nev. (2°)
	Mineral County	
1 Soda Springs	SE 1/4 sec. 29, T. 6N., R. 35E.	Sodaville, Nev. (7-1/2'); Walker Lake, Calif.-Nev. (2°)
	Nye County	
1 Darrough "steam" well	sec. 8, T. 11N., R. 43E.	; Tonopah, Nev. (2°)
2 Darrough Hot Springs	sec. 8, T. 11N., R. 43E.	; Tonopah, Nev. (2°)
3 Diana's Punch Bowl	SE 1/4 sec. 22, T. 14N., R. 47E.	Diana's Punch Bowl, Nev. (15'); Millet, Nev. (2°)
4 Hot spring near Diana's Punch Bowl	SE 1/4 sec. 22, T. 14N., R. 47E.	Diana's Punch Bowl, Nev. (15'); Millet, Nev. (2°)
5 Pott's Ranch Hot Spring	NE 1/4 sec. 2, T. 14N., R. 47E.	Diana's Punch Bowl, Nev. (15'); Millet, Nev. (2°)
6 Unnamed warm spring near Warm Springs	SW 1/4 sec. 20, T. 4N., R. 50E.	Warm Springs, Nev. (15'); Tonopah, Nev. (2°)

Table 1.--Location and topographic map coverage of selected hot springs and wells--Continued

Spring or well	Location	Topographic map coverage
Pershing County		
1 Unnamed hot spring (Jersey Valley)	SW 1/4 sec. 28, T. 27N., R. 40E.	Mt. Moses, Nev. (15'); Winnemucca, Nev. (2°)
2 Kyle Hot Springs	SW 1/4 sec. 1, T. 29N., R. 36E.	Kyle Hot springs, Nev. (15'); Winnemucca, Nev. (2°)
3 Sou Hot Springs	SE 1/4 sec. 29, T. 26N., R. 38E.	Cain Mtn., Nev. (15'); Winnemucca, Nev. (2°)
4 Unnamed hot springs (Lower Ranch)	NW 1/4 sec. 16, T. 25N., R. 39E.	Cain Mtn., Nev. (15'); Winnemucca, Nev. (2°)
5 Unnamed hot spring near Trego	Unsurveyed (lat. 40°46' N., long. 119°7' W)	; Lovelock, Nev.-Calif. (2°)
6 Unnamed hot spring near Black Rock	Unsurveyed (lat. 40°57' N., long. 118°58' W)	; Lovelock, Nev.-Calif. (2°)
7 Leach Hot Springs	SE 1/4 sec. 36, T. 32N., R. 38E.	Leach Hot Springs, Nev. (15'); Winnemucca, Nev. (2°)
Washoe County		
1 Steam Geyser (Needle Rocks)	Unsurveyed (lat. 40°9' N., long. 119°40' W)	The Needle Rocks, Nev. (7-1/2'); Lovelock, Nev.-Calif. (2°)
2 Great Boiling Spring	NW 1/4 sec. 15, T. 32N., R. 23E.	Gerlach, Nev. (15'); Lovelock, Nev.-Calif (2°)
3 Flowing well near Gerlach	sec. 2, T. 34N., R. 23E.	; Lovelock, Nev.-Calif. (2°)
4 Steamboat Springs	NE 1/4 sec. 33, T. 18N., R. 20E.	Steamboat, Nev. (7-1/2'); Reno, Nev. (2°)

# NEVADA

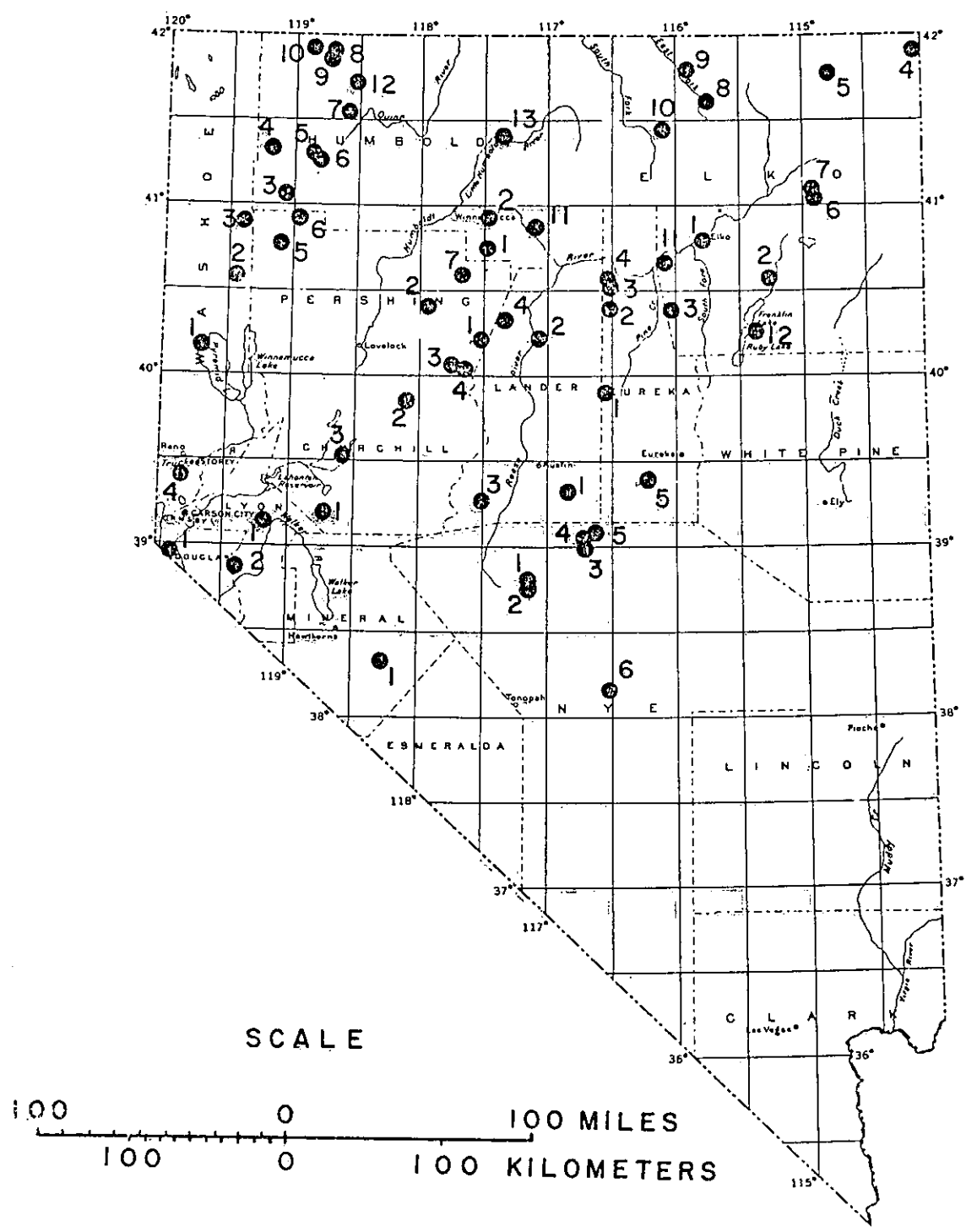


Figure 1. Map of the State of Nevada showing the location of sampled thermal springs and wells. The numbered dots correspond to sampled springs and wells listed by county in tables 1, 2, 3, and 4 of the text.

Table 2.--Chemical analyses of selected hot springs and wells  
 [Concentrations in milligrams per liter; parentheses indicate supplementary samples; n.a. indicates not available]

Spring or well	Temperature (°C)	pH	Specific conductance	Silica (SiO <sub>2</sub> )	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Lithium (Li)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Boron (B)
Churchill County															
1 Lee Hot Springs	88	7.36	2,430	180	44	0.6	450	26	0.70	114	<1	470	380	7.9	2.4
2 Dixie Valley Hot Springs	72	8.59	914	115	3.6	.02	190	6.5	.38	111	11	111	126	16.3	.89
3 Flowing well in Stillwater	96	7.57	6,910	170	108	1.7	1,480	42	1.94	90	<1	190	2,200	5.0	15
Douglas County															
1 Walleys Hot Springs	61	8.77	726	58	10	.01	145	3.6	.20	50	9	235	44	4.9	1.2
Elko County															
1 Hot Hole	56	7.21	908	65	60	15.5	120	39	.33	488	1	72	16	1.9	.70
2a Sulphur Hot Springs	93	8.53	601	210	1.0	.03	135	8.9	.46	244	15	40	23	17.7	.20
2b Sulphur Hot Springs	45	8.63	652	230	1.6	.02	150	9.8	.51	247	12	40	4	19.0	.23
3 Unnamed hot spring (Hot Creek)	26	7.30	408	20	46	23.5	10	2.1	.02	226	1	27	4.6	<0.1	.03
(4) Mile Spring	43	7.2	321	31	40	11.5	10	5.6	<.2	149	--	37	8.7	.4	<.02
(5) Mineral Hot Springs	60	9.1	344	83	1.6	<.01	75	2.2	<.2	108	--	45	15	8.9	.47
(6) Unnamed hot spring near Wells	50	7.3	753	165	12	.3	160	16	.8	345	--	61	22	10	1.2
(7) Unnamed hot spring near Wells	61	7.3	1,650	105	75	37	300	31	.8	1,135	--	32	27	7.2	.89
(8) Unnamed hot spring (Wild Horse Reservoir)	54	7.2	818	40	48	12	130	22	.5	482	--	40	14	5.2	.67
(9) Unnamed hot spring (SSE Patsville)	41	7.4	624	23	29	7.7	110	8.3	.4	380	--	36	4.4	3.4	.22
(10) Hot Sulphur Springs	90	7.0	1,760	84	49	13	390	41	.7	1,180	--	18	40	7.2	.77
(11) Unnamed hot spring near Carlin	79	7.6	625	70	60	15	45	16	n.a.	335	--	52	12	n.a.	n.a.
(12) Unnamed hot spring near Ruby Marsh	65	8.0	600	50	45	12	58	14	n.a.	377	--	24	6.5	n.a.	n.a.
Eureka County															
1 Malti Hot Springs	72	6.47	592	68	56	12	44	14	.3	264	<1	64	12	2.5	.12
2 Hot Springs Point	54	6.63	1,730	67	53	35	230	58	1.1	913	<1	7	1	6.6	2.1
3 Beowave "steam" well	--	9.38	1,490	500	1.3	.2	250	38	2.1	505	81	64	70	<.05	2.5
4 Beowave Hot Spring	98	8.98	1,020	320	1.0	<.1	230	16	1.3	321	32	130	69	17	2.1

Table 2.--Chemical analyses of selected hot springs and wells--Continued

Spring or well	Temperature (°C)	pH	Specific Conductance	Silica (SiO <sub>2</sub> )	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Lithium (Li)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Boron (B)
(5) Bartholomae Hot Springs	54	9.25	295	85	1	-0.1	64	0.7	n.a.	144	--	18	6.3	n.a.	n.a.
Humboldt County															
1 Unnamed hot spring (Hot Springs Ranch)	85	8.36	1,060	125	16	.9	200	18	1.2	385	--	140	41	n.a.	2.6
2 Unnamed hot spring near Golconda	74	6.53	810	66	33	6.8	130	22	.36	429	<.1	36	18	1.8	1.1
3 Double Hot Springs	80	7.93	902	105	4.8	.1	180	4.5	.06	261	2	120	59	10	1.8
4 Unnamed hot springs in Soldier Meadows	54	8.55	363	63	3.1	<.1	74	1.1	.17	92	3	41	18	12	.64
5 West Pinto Hot Springs (well)	92	7.65	1,320	160	4.6	.1	320	25	.45	436	2	130	160	14	6.9
6 East Pinto Hot Springs	93	7.14	1,560	150	14	.4	330	23	.45	495	1	120	160	12	7.5
7 Dyke Hot Spring	66	8.86	666	85	1.8	<.1	150	4.3	.09	243	17	82	21	8.0	1.0
8 Flowing well near Baltaax Hot Spring	90	7.50	934	150	10	.1	180	8.2	.22	156	<.1	230	47	6.8	2.1
9 Baltaax Hot Springs	80	8.00	947	160	8.4	<.1	180	8.7	n.a.	139	2	220	48	7.1	2.9
10 Bog Hot Springs	54	9.05	356	57	.2	<.1	81	1.0	.03	116	11	45	15	1.7	.91
(11) Hot Pot	58	8.0	1,400	80	29	5	288	33	.72	823	--	60	28	n.a.	n.a.
(12) Howard Hot Spring	56	9.2	400	85	3	<.1	88	1.7	n.a.	127	--	62	10	n.a.	n.a.
(13) The Hot Springs	58	8.0	1,340	55	10	8	296	36	n.a.	881	--	36	26	n.a.	n.a.
Lander County															
1 Spencer Hot Springs	72	6.49	1,180	77	43	9.4	200	36	1.8	672	<.1	51	22	4.7	2.6
(2) Unnamed hot spring (Valley of the Moon)	53	8.0	700	40	20	9	118	21	n.a.	333	--	64	21	n.a.	n.a.
3 Unnamed hot spring (Smith Creek Valley)	86	7.72	737	110	4.8	.06	170	8.4	.38	246	5	102	22	8.9	.66
4 Buffalo Valley Hot Springs	49	6.53	1,530	80	45	4.9	250	34	.80	813	<.1	110	29	4.8	2.3
Lyon County															
(1) Habuska Hot Springs	97	8.5	1,550	115	38	.2	277	15	n.a.	70	--	580	46	n.a.	n.a.
2 Nevada Hot Springs	61	8.65	509	52	4.5	.01	102	2.5	.08	54	7	169	17	3.1	.19
Mineral County															
1 Soda Springs	35	7.60	1,640	46	.40	3.3	305	16	.65	112	<.1	597	87	7.4	2.3



Table 2.--Chemical analyses of selected hot springs and wells--Continued

Spring or well	Temperature (°C)	pH	Specific conductance	Silica (SiO <sub>2</sub> )	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Lithium (Li)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Boron (B)
Nye County															
1 Darrough "steam" well	94	8.29	499	105	1.4	0.1	110	2.9	0.3	165	3	55	12	15	0.24
2 Darrough Hot Springs	95	8.29	479	98	1.3	.1	110	2.6	.3	146	3	53	12	14	.22
3 Diana's Punch Bowl	59	7.14	605	46	50	11	55	15	.4	277	<1	59	8	2.8	.21
4 Hot spring near Diana's Punch Bowl	51	6.73	589	46	47	11	57	15	0.4	270	<1	59	8	2.8	0.23
5 Pott's Ranch Hot Spring	45	6.62	561	36	52	11	47	13	.3	249	<1	57	10	2.0	.17
(6) Unnamed warm spring near Warm Springs	61	8.1	1,250	60	43	24	175	24	n.a.	714	--	120	32	n.a.	n.a.
Pershing County															
1 Unnamed hot spring (Jersey Valley)	29	7.10	1,040	110	36	4.4	180	20	1.2	374	<1	150	40	7.8	1.9
2 Kyle Hot Springs	77	6.50	3,220	150	95	25.5	540	80	3.1	544	<1	51	770	5.7	3.8
(3) Sou Hot Springs	73	8.1	1,407	65	110	22	165	26	n.a.	312	--	370	75	n.a.	n.a.
(4) Unnamed hot spring (Lower Ranch)	40	8.1	850	42	31	15	143	12	n.a.	456	--	63	29	n.a.	n.a.
(5) Unnamed hot spring near Trejo	86	8.4	2,300	85	25	.2	463	9.3	n.a.	154	--	86	520	n.a.	n.a.
(6) Unnamed hot spring near Black Rock	90	8.1	6,590	120	35	4	1,500	20	n.a.	932	--	290	787	n.a.	n.a.
7 Leach Hot Spring	92	7.40	811	135	8.8	.5	160	13	1.7	366	1	53	29	7.8	1.2
Washoe County															
1 Steam Geyser (Needle Rocks)	56	8.43	6,200	110	260	.1	1,100	160	.61	24	1	340	1,900	3.0	6.1
2 Great Boiling Spring	86	7.15	7,610	165	68	1.2	1,400	130	1.6	83	<1	400	2,200	4.5	9.9
3 Flowing well near Gerlach	80	7.91	1,800	82	31	4.2	340	17	.46	458	4	46	240	7.0	1.9
4 Steamboat Springs	94	7.19	3,340	270	16	0.7	680	66	7.5	364	2	73	837	2.1	47

Table 3.--Estimated thermal-aquifer temperatures of selected hot springs  
 [Spring deposits: X, calcium carbonate or silica apron; T, traces of calcium carbonate or silica; --, calcium carbonate or silica not detected]

Spring or well	Spring temperature (°C)	Flow (lpm)	Spring deposits		Comments	Estimated reservoir temperatures				
			CaCO <sub>3</sub>	Silica		Silica conductive	Silica adiabatic	Na-K	Na-K-1/3Ca	
Churchill County										
1 Lee Hot Springs	88	130	X	X	High chloride, near boiling	<u>173</u>	162	128	128	<u>162</u>
2 Dixie Valley Hot Springs <sup>1/</sup>	72	200	--	--	Low calcium, moderate chloride, moderate silica	<u>165</u>	139	86	86	<u>144</u>
3 Flowing well in Stillwater	96	--	--	--	High chloride, boiling	169	<u>159</u>	72	72	<u>140</u>
Douglas County										
1 Walleys (Genoa) Hot Springs	61	75	X	--	Na-K-4/3Ca estimate near the spring temperature	109	109	64	64	119
Elko County										
1 Hot Hole (Elko Hot Springs)	56	75	X	--	Calcite ppt., Na-K-Ca meaningless	<u>114</u>	113	380	380	234
2a Sulphur Hot Springs	93	75	--	X	Boiling spring, low chloride and calcium	<u>183</u>	171	140	140	<u>181</u>
2b Sulphur Hot Springs <sup>2/</sup>	45	500	--	X	May be boiling at the orifice (?), large lake with a large cooling surface	<u>190</u>	176	139	139	<u>178</u>
3 Unnamed hot spring (Hot Creek)	26	6,000	--	--	Na-K-4/3Ca estimate near the spring temperature	63	69	282	282	161
4 Nile Spring	43	Supplementary sample			Na-K-4/3Ca estimate near the spring temperature	81	84	543	543	220
5 Mineral Hot Springs <sup>1/</sup>	60	Supplementary sample			Low calcium, magnesium, and chloride (?), moderate silica	<u>127</u>	124	75	75	<u>129</u>
6 Unnamed hot spring near Wells <sup>1/</sup>	50	Supplementary sample			Low calcium and chloride (?), high silica	<u>167</u>	158	184	184	<u>184</u>
7 Unnamed hot spring near Wells	61	Supplementary sample			High calcium and magnesium, low chloride, mixed (?)	<u>140</u>	135	188	188	<u>181</u>
8 Unnamed hot spring (Wild Horse Reservoir)	54	Supplementary sample			High calcium and bicarbonate, low chloride	<u>92</u>	94	255	255	197
9 Unnamed hot spring (SSE Patsville)	41	Supplementary sample			Na-K-4/3Ca estimate near the spring temperature	69	74	153	153	156
10 Hot Sulphur Springs	90	Supplementary sample			High calcium and bicarbonate, low chloride, mixed (?)	<u>128</u>	125	190	190	191
11 Unnamed hot spring near Carlin	79	Supplementary sample			Na-K-4/3Ca estimate near the spring temperature	118	117	395	395	216
12 Unnamed hot spring near Ruby Marsh	65	Supplementary sample			Na-K-4/3Ca estimate near the spring temperature	102	102	314	314	202
Eureka County										
1 Walti Hot Springs	72	300	X	--	Na-K-4/3Ca estimate near the spring temperature	116	115	375	375	212
2 Hot Springs Point	54	125	X	--	High calcium, very low chloride, mixed water (?)	<u>115</u>	115	325	325	233
3 Beowave "steam" well	--	--	--	X	Boiling, very low calcium	252	226	238	238	242

Table 3.--Estimated thermal-aquifer temperatures of selected hot springs--Continued

Spring or well	Spring temperature (°C)	Flow (lpm)	Spring deposits		Comments	Estimated reservoir temperatures (°C)				
			CaCO <sub>3</sub>	Silica		Silica conductive	Silica adiabatic	Na-K	Na-K-1/3Ca	Na-K-4/3Ca
4 Beowawe Hot Spring	98	100	--	X	Boiling, very low calcium, superheated	214	196	145	194	237
5 Bartholomae Hot Springs	54	Supplementary sample			Na-K-4/3Ca estimate near the spring temperature	128	125	19	92	73
Humboldt County										
1 Unnamed hot spring (Hot Springs Ranch)	85	100	X	T	Calcite precipitating, mixed (?)	150	144	172	180	139
2 Unnamed hot spring near Golconda	74	750	X	--	High calcium, bicarbonate, and magnesium, possible mixing (?)	115	114	255	201	121
3 Double Hot Spring <sup>1/</sup>	80	175	T	T	Low calcium, bicarbonate, and magnesium, mixed	140	135	64	127	113
4 Unnamed hot springs in Soldier Meadows	54	50	--	--	Na-K-4/3Ca estimate near the spring temperature	112	112	34	98	65
5 West Pinto Hot Spring (well)	92	100	X	X	Low calcium and magnesium, spring near boiling	165	156	157	192	208
6 East Pinto Hot Spring	93	500	X	X	Low calcium and magnesium, spring near boiling	162	153	165	176	163
7 Dyke Hot Spring <sup>1/</sup>	66	100	T	--	Low calcium and magnesium, spring not boiling	128	126	73	137	136
8 Flowing well near Baltazor Hot Spring	90	25	--	--	Low calcium, nearly boiling, low flow rate	162	153	107	168	118
9 Baltazor Hot Spring	80	100	T	T	Low calcium, may be a mixed water although chloride and magnesium are low	165	156	111	152	100
10 Bog Hot Spring	54	4,000	--	--	Low TDS, deep circulation meteoric or mixed (?)	108	109	25	109	128
11 Hot Pot	58	Supplementary sample			High calcium, bicarbonate, calcite ppt. (?), indeterminate	125	122	200	195	155
12 Howard Hot Spring	56	Supplementary sample			Na-K-4/3Ca estimate near the spring temperature	128	125	49	110	81
13 The Hot Springs	58	Supplementary sample			Calcite ppt. (?), low chloride, indeterminate	106	106	208	209	197
Lander County										
1 Spencer Hot Springs	72	50	X	--	Low flow rate, low chloride, high calcium and bicarbonate	123	121	264	210	141
2 Unnamed hot spring (Valley of the Moon)	53	Supplementary sample			Very low silica, probably deep circulation meteoric	92	94	263	207	132
3 Unnamed hot spring (Smith Creek Valley)	86	75	T	--	Spring near boiling, low calcium, low chloride (?)	143	137	114	157	139
4 Buffalo Valley Hot Springs	49	10	T	--	High calcium and bicarbonate, low flow rate	125	122	223	198	140

Table 3.--Estimated thermal-aquifer temperatures of selected hot springs--Continued

Spring or well	Spring temperature (°C)	Flow (lpm)	Spring deposits		Comments	Estimated reservoir temperatures (°C)				
			CaCO <sub>3</sub>	Silica		Silica conductive	Silica adlabatic	Na-K	Na-K-1/3Ca	Na-K-4/3Ca
Lyon County										
1 Mabuska Hot Springs	97	Supplementary sample			Boiling	145	139	120	152	111
2 Nevada Hot Springs	61	200 X	--	--	Na-K-4/3Ca estimate near the spring temperature	104	104	64	119	86
Mineral County										
1 Soda Springs	35	100 T	--	--	High calcium, deep circulation meteoric or mixed water (?)	98	99	122	154	116
Nye County										
1 Darrough "steam" well	94	300	--	--	Boiling	140	135	68	131	122
2 Darrough Hot Springs	95	350 X	T		Boiling	136	132	61	127	120
3 Diana's Punch Bowl	59	-- X	--	--	Na-K-4/3Ca estimate near the spring temperature	99	100	341	208	86
4 Hot springs near Diana's Punch Bowl	51	200 X	--	--	Na-K-4/3Ca estimate near the spring temperature	99	100	334	207	88
5 Pott's Ranch Hot Spring	45	125 X	--	--	Na-K-4/3Ca estimate near the spring temperature	91	92	344	205	79
6 Unnamed warm spring near Warm Springs	61	Supplementary sample			High calcium, magnesium, and bicarbonate, probably low temperature	110	110	225	192	122
Pershing County										
1 Unnamed hot spring (Jersey Valley)	29	20 X	X		Low flow rate, qualitatively high aquifer temperature	142	137	196	182	119
2 Kyle Hot Springs	77	20 X	T		Low flow rate, qualitatively high aquifer temperature	171	161	199	194	154
3 Sou Hot Springs	73	Supplementary sample			Na-K-4/3Ca estimate near spring temperature	114	113	244	190	100
4 Unnamed hot spring (Lower Ranch)	40	Supplementary sample			High calcium, magnesium, and bicarbonate	94	96	162	164	100
5 Unnamed hot spring near Trego	86	Supplementary sample			High chloride, near boiling, flow rate (?)	128	125	51	120	111
6 Unnamed hot spring near Black Rock	90	Supplementary sample			High chloride, near boiling, flow rate (?)	148	142	28	117	151
7 Leach Hot Springs	92	200 T	T		Near boiling, low calcium and magnesium	155	147	161	176	139

Table 3.--Estimated thermal-aquifer temperatures of selected hot springs--Continued

Spring or well	Spring temperature (°C)	Flow (lpm)	Spring deposits		Comments	Estimated reservoir temperatures (°C)				
			CaCO <sub>3</sub>	Silica		Silica conductive	Silica adiabatic	Na-K	Na-K-1/3Ca	Na-K-4/3Ca
1 Steam Geyser (Needle Rocks)	56	--	X	--	Washoe County High chloride, calcium, calcite ppt. (?), boiling in well (?)	143	137	232	214	184
2 Great Boiling Spring	86	--	T	X	Boiling, high chloride	167	158	175	205	230
3 Flowing well near Gerlach	80	500	X	--	Calcite ppt., Na-K-Ca estimate too high	125	124	115	154	125
4 Steamboat Springs	94	50	--	X	Boiling, high chloride	201	186	180	208	233

1/ Mixed waters

2/ Temperature measured at the outlet of the lake not in the orifice of the spring

Table 4.--Age and type of rock near each spring

Spring or well	Age and type of rock	Geologic reference
Churchill County		
1 Lee Hot Springs	Miocene to Pliocene volcanic rocks	Williden and Speed (1968)
2 Dixie Valley Hot Springs	Quaternary alluvium, Tertiary volcanic rocks, and possibly late Mesozoic intrusive and metamorphic rocks	Page (1965)
3 Flowing well in Stillwater	Quaternary alluvium and Tertiary basalt(?)	Williden and Speed (1968)
Douglas County		
1 Walleys Hot Springs	Triassic and Jurassic metavolcanic rocks of greenschist facies	Noore (1969)
Elko County		
1 Hot Hole	Tertiary limestone, lacustrine rocks and volcanic rocks	Granger, Mendell, Simmons, and Lee (1957)
2 Sulphur Hot Springs	Quaternary alluvium, late Mesozoic granites, and Paleozoic to Precambrian metamorphic rocks	Granger, Mendell, Simmons, and Lee (1957)
3 Unnamed hot spring (Hot Creek)	Paleozoic limestone	Smith and Kerner (1972)
4 Mile Spring	Tertiary lacustrine rocks	Granger, Mendell, Simmons, and Lee (1957)
5 Mineral Hot Spring	Tertiary lacustrine rocks, granite(?) and volcanic flows	Granger, Mendell, Simmons, and Lee (1957)
6 Unnamed hot spring near Wells	Tertiary lacustrine rocks	Granger, Mendell, Simmons, and Lee (1957)
7 Unnamed hot spring near Wells	Tertiary lacustrine rocks	Granger, Mendell, Simmons, and Lee (1957)
8 Unnamed hot spring (Wild Horse Reservoir)	Tertiary volcanic and lacustrine rocks	Granger, Mendell, Simmons, and Lee (1957)
9 Unnamed hot spring (SSE Patsville)	Tertiary volcanic rocks and Paleozoic limestone	Granger, Mendell, Simmons, and Lee (1957)
10 Hot Sulphur Springs	Tertiary volcanic rocks and Paleozoic limestone	Granger, Mendell, Simmons, and Lee (1957)
11 Unnamed hot spring near Carlin	Quaternary alluvium and Tertiary volcanic rocks	Granger, Mendell, Simmons, and Lee (1957)
12 Unnamed hot spring near Ruby Marsh	Quaternary alluvium and Paleozoic(?) marine sedimentary rocks	Granger, Mendell, Simmons, and Lee (1957)
Eureka County		
1 Walthi Hot Springs	Quaternary alluvium, late Mesozoic to early Cenozoic granite, and Paleozoic sedimentary rock	Roberts, Montgomery, and Lehner (1967)
2 Hot Springs Point	Late Miocene and early Pliocene basalts, and Ordovician quartzite and cherts	Gilluly and Gates (1965)
3 Beowave "steam" well	Miocene basalt and andesite flows	Gilluly and Gates (1965)
4 Beowave Hot Spring	Miocene basalt and andesite flows	Gilluly and Gates (1965); Stewart and McKee (1970)
5 Bartholomae Hot Springs	Quaternary alluvium and Tertiary volcanic rocks	Roberts, Montgomery, and Lehner (1967)

Table 4.--Age and typ. of rock near each spring--Continued

Spring or well	Age and type of rock	Geologic reference
	Humboldt County	
1 Unnamed hot spring near Hot Springs Ranch	Cambrian phyllitic shale	Willden (1964)
2 Unnamed hot spring near Golconda	Quaternary alluvium, Cambrian quartzite, and Tertiary volcanic rocks	Ferguson, Roberts, and Muller (1952)
3 Double Hot Springs	Quaternary alluvium, Tertiary basalt and ash-flow rhyolite	Willden (1964)
4 Unnamed hot spring in Soldier Meadows	Quaternary alluvium, Tertiary flows and tuffs	Willden (1964)
5 West Pinto Hot Spring (well)	Cretaceous or Tertiary granodiorite, and Tertiary basalt	Willden (1964)
6 East Pinto Hot Spring	Cretaceous or Tertiary granodiorite	Willden (1964)
7 Dyke Hot Spring	Quaternary alluvium, Triassic and Jurassic metamorphic rocks	Willden (1964)
8 Flowing well near Baltazor Hot Spring	Quaternary alluvium, Tertiary volcanic rocks, and Cretaceous to Tertiary granodiorite	Willden (1964)
9 Baltazor Hot Spring	Quaternary alluvium, Tertiary volcanic rocks, and Cretaceous to Tertiary granodiorite	Willden (1964)
10 Bog Hot Springs	Quaternary alluvium, Pliocene volcanic and sedimentary rocks	Willden (1964)
11 Hot Pot	Quaternary alluvium, Tertiary basalt(?), and Cambrian quartzite(?)	Willden (1964)
12 Howard Hot Spring	Quaternary alluvium and Tertiary flows	Willden (1964)
13 The Hot Springs	Tertiary sedimentary rocks and flows	Willden (1964)
	Lander County	
1 Spencer Hot Springs	Quaternary alluvium, Oligocene or Miocene ash-flow tuff, Jurassic "granite", Ordovician cherts or quartzites	Stewart and McKee (1970); McKee (1968)

Table 4.--Age and type of rock near each spring--Continued

Spring or well	Age and type of rock	Geologic reference
2 Unnamed hot spring (Valley of the Moon)	Quaternary alluvium covering Tertiary volcanic rocks	Stewart and McKee (1970)
3 Unnamed hot spring (Smith Creek Valley)	Quaternary alluvium, Oligocene or Miocene (?) ash-flow rhyolites	McKee (1968)
4 Buffalo Valley Hot Springs	Quaternary alluvium, Quaternary basalts, and Tertiary tuffs	Stewart and McKee (1970)
1 Wabuska Hot Springs	Lyon County Quaternary alluvium, Miocene to Pleistocene basalt and andesite, Triassic and Jurassic metavolcanic rocks	Moore (1969)
2 Nevada Hot Springs	Cretaceous intrusives of granitic to mafic composition	Moore (1969)
1 Soda Springs	Mineral County Quaternary alluvium, Quaternary basalt, and Tertiary tuffaceous rocks	Ross (1961)
1 Darrough "steam" well	Nye County Quaternary alluvium and Paleozoic rhyolite	Kleinhampl and Ziony (1967)
2 Darrough Hot Springs	Quaternary alluvium and Paleozoic rhyolite	Kleinhampl and Ziony (1967)
3 Diana's Punch Bowl	Quaternary alluvium and Tertiary ash-flow rhyolite	Kleinhampl and Ziony (1967)
4 Hot spring at Diana's Punch Bowl	Quaternary alluvium and Tertiary ash-flow rhyolite	Kleinhampl and Ziony (1967)
5 Pott's Ranch Hot Spring	Tertiary ash-flow rhyolite	Kleinhampl and Ziony (1967)
6 Unnamed warm spring near Warm Springs	Tertiary volcanics and Paleozoic sedimentary rocks	Kleinhampl and Ziony (1967)
1 Unnamed hot spring (Jersey Valley)	Pershing County Quaternary alluvium, Tertiary tuffs and flows	Tatlock (1969)
2 Kyle Hot Springs	Quaternary alluvium and Paleozoic metamorphic rocks	Tatlock (1969)
3 Sou Hot Springs	Quaternary alluvium, Tertiary flows and volcanic derived sedimentary rocks	Tatlock (1969)
4 Unnamed hot spring (Lower Ranch)	Quaternary alluvium, Tertiary rhyolite, and metamorphosed Triassic rocks	Tatlock (1969)



Table 4.---Age and type of rock near each spring--Continued

Spring or well	Age and type of bedrock	Geologic reference
5 Unnamed hot spring (Trego)	Quaternary dune sands and Cretaceous granite	Tatlock (1969)
6 Unnamed hot spring (Black Rock)	Quaternary playa sediments, Tertiary volcanic and sedimentary rocks	Tatlock (1969)
7 Leach Hot Springs	Quaternary alluvium, Tertiary sedimentary rocks, basalt of unknown age, Paleozoic metamorphic rocks	Tatlock (1969)
1 Steam Geyser (Needle Rocks)	Washoe County	
	Quaternary tufa and alluvium, Tertiary olivine basalt	Bonham (1969)
2 Great Boiling Spring	Cretaceous or Tertiary granodiorite,	Bonham (1969)
	Quaternary alluvium and lake sediments	
3 Flowing well near Garlach	Quaternary alluvium, late Tertiary basalts, tuffs, and volcanic sandstone	Bonham (1969)
4 Steamboat Springs	Cretaceous granodiorite	Thompson and White (1964)

#### REFERENCES CITED

- Barnes, Ivan, 1964, Field measurement of alkalinity and pH: U.S. Geol. Survey Water- Supply Paper 1535-H, 17 p.
- Bonham, H. F., 1969, Geology and mineral deposits of Washoe and Storey Counties, Nevada: Nevada Bur. Mines Bull. 70, 140 p.
- Ellis, A. J., 1970, Quantitative interpretation of chemical characteristics of hydrothermal systems, in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, part 1: Geothermics Spec. Issue 2, p. 516-528.
- Ferguson, H. G., Roberts, R. J., and Muller, S. W., 1952, Geology of the Golconda quadrangle, Nevada: U.S. Geol. Survey Geologic Map GQ-15, scale 1:125,000.
- Fournier, R. O., and Rowe, J. J., 1966, Estimation of underground temperatures from the silica content of water from hot springs and wet steam wells: Am. Jour. Sci., v. 264, p. 685-697.
- Fournier, R. O., and Truesdell, A. H., 1970, Chemical indicators of subsurface temperature applied to hot waters of Yellowstone National Park, Wyo., U.S.A., in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, part 1: Geothermics Spec. Issue 2, p. 529-535.
- \_\_\_\_\_ 1973, An empirical Na-K-Ca geothermometer for natural waters: Geochim. Cosmochim. Acta, v. 37, p. 1255-1275.

- \_\_\_\_\_ 1974, Geochemical indicators of subsurface temperature, Part II: Estimation of temperature and fraction of hot water mixed with cold water: U.S. Geol. Survey Jour. Research, v. 2, no. 3 (in press).
- Fournier, R. O., White, D. E., and Truesdell, A. H., 1974, Geochemical indicators of subsurface temperature, Part I: Basic assumptions: U.S. Geol. Survey Jour. Research, v. 2, no. 3 (in press).
- Gilluly, James, and Gates, Olcott, 1965, Tectonic and igneous geology of the northern Shoshone Range, Nevada: U.S. Geol. Survey Prof. Paper 465, 153 p.
- Godwin, L. H., Haigler, L. B., Rioux, R. L., White, D. E., Muffler, L. J. P., and Wayland, R. G., 1971, Classification of public lands valuable for geothermal steam and associated geothermal resources: U.S. Geol. Survey Circ. 647, 17 p.
- Granger, A. E., Mendell, M. B., Simmons, G. C., and Lee, Florence, 1957, Geology and mineral resources of Elko County, Nevada: Nevada Bur. Mines Bull. 54, 190 p.
- Kleinhampl, F. J., and Ziony, J. I., 1967, Preliminary geologic map of northern Nye County, Nevada: U.S. Geol. Survey open-file map.
- Mahon, W. A. J., 1970, Chemistry in the exploration and exploitation of hydrothermal systems, in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, part 2: Geothermics Spec. Issue 2, p. 1310-1322.
- McKee, E. H., 1968, Geologic map of the Spencer Hot Springs quadrangle, Lander County, Nevada: U.S. Geol. Survey Quad. Map GQ-770.

- Montgomery, K. M., 1965, Preliminary geologic map of Nevada: U.S. Geol. Survey open-file map.
- Moore, J. G., 1969, Geology and mineral deposits of Lyon, Douglas, and Ormsby Counties, Nevada: Nevada Bur. Mines Bull. 75, 45 p.
- Page, B. M., 1965, Preliminary geologic map of a part of the Stillwater Range, Churchill County, Nevada: Nevada Bur. Mines Map 28.
- Roberts, R. J., Montgomery, K. M., and Lehner, R. E., 1967, Geology and mineral resources of Eureka County, Nevada: Nevada Bur. Mines Bull. 64, 152 p.
- Ross, D. C., 1961, Geology and mineral deposits of Mineral County, Nevada: Nevada Bur. Mines Bull. 58, 98 p.
- Smith, J. F., and Ketner, K. B., 1972, Generalized geologic map of the Carlin, Dixie Flats, Pine Valley, and Robinson Mountain quadrangles, Elko and Eureka Counties, Nevada: U.S. Geol. Survey Miscellaneous field studies map MF-481, scale 1:125,000.
- Stewart, J. H., and McKee, E. H., 1970, Geologic map of Lander County, Nevada: U.S. Geol. Survey open-file map.
- Tatlock, D. B., 1969, Preliminary geologic map of Pershing County, Nevada: U.S. Geol. Survey open-file map.
- Thompson, G. A., and White, D. E., 1964, Regional geology of the Steamboat Springs area, Washoe County, Nevada: U.S. Geol. Survey Prof. Paper 458-A, 51 p.
- Waring, G. A., 1965, Thermal springs of the United States and other countries of the world--a summary: U.S. Geol. Prof. Paper 492, 833 p.

- White, D. E., 1965, Saline waters of sedimentary rocks, in Fluids in subsurface environments--a symposium: Am. Assoc. Petroleum Geologists Mem. 4, 342-366.
- \_\_\_\_\_ 1970, Geochemistry applied to the discovery, evaluation, and exploitation of geothermal energy resources, in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 1, part 2: Geothermics Spec. Issue 2 (in press).
- White, D. E., Muffler, L. J. P., and Truesdell, A. H., 1971, Vapor-dominated hydrothermal systems compared with hot-water systems: Econ. Geol., v. 66, no. 1, p. 75-97.
- Willden, Ronald, 1964, Geology and mineral resources of Humboldt County, Nevada: Nevada Bur. Mines Bull. 59, 154 p.
- Willden, Ronald, and Speed, R. C., 1968, Preliminary geologic map of Churchill County, Nevada: U.S. Geol. Survey open-file map, scale 1:200,000.

## NEW MEXICO

New Mexico has approximately 60 thermal areas above 90°F (32°C). Thermal areas are concentrated in the block-faulted volcanic terrain of the southwest and along the faulted western margin of the Rio Grande Rift Valley. Heat flow in this region is up to 2.77 HFU. The Rio Grande trough is bounded on the east and west by discontinuous fault zones. Warm and hot springs are found along the faults which border the trough; most are on the west side of the trough.

Average temperature 90-100°F (32-38°C); temperatures ranged up to 240°F (116°C)  
Total solids average 1,000 to 2,000 ppm  
pH average 7.5, except Sandoval County with a 1.9 average pH

### Best potential areas:

Animas Valley in Hidalgo County  
Cliff-Gila-Riverside area in Grant County  
Southern Rio Grande trough  
Upper Jemez River Basin

### Lemonade Spring

150°F (66°C)  
216 ppm SiO<sub>2</sub>  
pH 1.9  
1,950 ppm total solids

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts)

gpm	*F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>3</sub>
<b>Bernalillo</b>															
Source: Spring ("Clear Water Spring")			Location: 9N.4E.24.1												
Remarks: About 50 yards north of Iron Spring															
—	69	7-25-45	—	—	—	—	—	—	218	48	—	—	234	890	0
<b>Catron</b>															
Source: Spring			Location: 2S.14W.17.410												
Remarks: Across road from old church and near court house upper end of Mangas															
450	72	7-17-63	—	—	—	—	—	—	<1	—	—	62	—	115	22
Source: Aragon Springs			Location: 5S.16W.3.300												
Remarks: "a" taken upstream from springs; "b" taken several hundred yards downstream from springs. Numerous springs in meadow															
2000	70	11-20-52	44	—	—	—	—	—	—	—	—	20	—	138	0
1000	68	11- 8-54	42	0.1	0.05	0.00	—	—	21	6.6	—	19	3.3	139	0
1500	70	7-16-63	—	—	—	—	—	—	22	6	—	—	13	124	0
"a"	—	7-16-63	—	—	—	—	—	—	22	6	—	—	24	154	0
"b"	—	7-16-63	—	—	—	—	—	—	28	6	—	—	14	146	0
Source: Frisco Hot Springs (upper)			Location: 5S.19W.34.200												
Remarks: Discharge from alluvium collected at old bath house (FRH); runs out of Gila Conglomerate about 100 ft above floor of San															
—	98	5-22-58	58	—	—	—	0.00	0.01	—	—	2.2	66	0.5	57	55
Source: Frisco Hot Springs (lower)			Location: 12S.20W.23.100												
Remarks: Discharges from lava of late Tertiary age, at temperatures ranging from 80°-124°F (S); temperature=130°F (P)															
20+	117.0	5-16-53	85	—	—	—	—	—	—	—	—	—	333	130	0
—	109	6-13-58	76	—	—	—	0.00	0.00	—	—	0	280	16	132	0
—	115	7- 6-59	—	—	—	—	—	—	49	40	—	289	—	127	0
<b>Dona Ana</b>															
Source: Well			Location: 19S.5E.5.100												
Remarks: Two dug wells 65 ft deep															
—	69	—	—	—	—	—	—	—	120	106	—	—	858	171	104
Source: Radium Springs (Radium Hot Springs)			Location: 21S.1W.10.213												
Remarks: Springs issue at base of rhyolite hill on east border of lowland of the Rio Grande; temperatures = 168°, 185°F (S); rhyolite															
—	—	5-17-48	71	—	—	—	—	—	142	23	—	—	1160	427	—
—	128	11-17-54	75	0.1	0.15	0.40	—	—	126	12	—	1100	161	417	0
10	120+	4-14-58	66	—	—	—	0.0	0.25	—	—	0.0	1100	155	424	0
—	—	5- 4-62	—	—	0.0	0.1	—	—	131	15	—	1100	163	416	—
—	—	8-31-22	60	—	1.2	—	—	—	138	17	—	1164	111	429	0
Source: Well			Location: 21S.1W.20.200												
Remarks: (3 miles SW of Radium Springs) Drilled well, 2½ inches in diameter, 250 ft deep, cased to 150 ft, water-bearing formation =															
8-9	"hot"	5-25-58	46	—	—	—	0.01	0.18	—	—	0.0	227	6.0	534	0
Source: Cleolas Spring			Location: 21S.1E.23.200												
Remarks: Discharges from andesite tuff															
0.5	65	4-24-58	39	—	—	—	0.02	0.00	—	—	0.0	66	5.5	216	0
Source: Well			Location: 21S.1W.31.200												
Remarks: Drilled well 2½ inches in diameter, 51 ft deep, cased to 24 ft, water level 24 ft below land surface, water-bearing formation 20-															
4	68	4-24-58	56	—	—	—	—	—	—	—	—	95	6.6	310	0
Source: Well			Location: 28S.1W.31.400												
Remarks: Drilled well 1200 ft deep, water level at 597 ft below land surface, water sample from 1030-1200 ft															
13.2	90	2-19-55	19	0.02	—	—	—	—	110	1.1	—	—	928	55	—
Source: Wildcat oil well			Location: 28S.2W.24.213												
Remarks: Water was struck at 675 ft; sample from waste water pumped by well during drilling; contains detergent and possibly other															
500	113	11-25-61	—	—	—	—	—	—	—	—	—	1380	—	934	0
<b>Grant</b>															
Source: Gila Hot Springs			Location: 13S.13W.5.241												
Remarks: Discharge from lava (S)															
25	147	6-23-57	33	—	0.00	—	—	—	11	0.2	—	—	129	109	0
100	147	7-25-62	68	0.31	0.00	0.00	—	—	12	0	—	121	3.6	106	0
Source: Hot Spring			Location: 13S.13W.10.121												
10	126	6-23-57	—	—	—	—	—	—	—	—	—	—	—	108	0

New Mexico tables extracted from: Summers, W. K., A Preliminary Report on New Mexico's Geothermal Energy Resources, State Bureau of Mines and Mineral Resources, New Mexico Institute of Mining and Technology, Circular 80, 41 pgs., (1965)

WELLS INCLUDING CHEMICAL ANALYSES

pc. million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (μmhos at 25°C)	pH	H <sub>2</sub> S	Total β-γ activity (μμc/l)	Ra (μμc/l)	U (μg/l)	Ref.
<b>County</b>																	
93	300	—	—	—	—		1330		742	—	232	—	—	—	—	—	4086
<b>County</b>																	
2	1	—	—	—	—				0.0	—	304	9.3	—	—	—	—	FRH
along east edge of canyon floor (B); issues from lake beds—sand, gravel, clay—of extinct Lake San Agustin (S&B)																	
7.0	5	0.4	0.5	—	—				86	0	236	—	—	—	—	—	21145
2.9	4.5	0.6	1.5	—	0.04		175		80	—	234	8.2	—	<7	<0.1	1.1	S&B
2	3	—	—	—	—				78	—	235	7.6	—	—	—	—	FRH
2	3	—	—	—	—				80	—	267	7.4	—	—	—	—	FRH
2	3	—	—	—	—				94	—	264	7.9	—	—	—	—	FRH
Francisco River bed, 130°F at 30 gpm (Sc)																	
6.6	5.0	1.0	0.8	—	0.04				10	0	284	9.7	—	—	—	—	38851
45	512	1.6	1.5	—	—				157	50	1930	—	—	—	—	—	22690
47	434	1.8	1.3	—	0.32				142	34	1660	7.6	—	—	—	—	38864
	460	—	—	—	—		1020		286	182	1780	7.8	—	—	—	—	42605
<b>County</b>																	
241	1541	—	—	—	—		3019		174	—	—	—	—	—	—	—	M&H
dikes intruded into latitic tuffs overlain by alluvium (S&B)																	
265	1680	4.6	2.0	—	—			3540	449	—	6060	—	—	—	—	—	C
255	1650	4.8	1.1	—	—			3620	364	—	6100	7.2	—	170	0.6	18	S&B
277	1660	5.7	1.4	—	0.32				380	32	5540	7.2	—	—	—	—	38867
269	1677	5.3	0.26	—	—		3660		390	—	6100	7.2	—	—	—	—	NMPL
253	1752	—	—	—	—		3738	3707	352	—	—	—	—	—	—	—	2058
melsite breccia																	
126	32	2.2	28	—	0.13				142	0	1130	8.2	—	—	—	—	38863
100	42	1.0	0.3	—	0.04				188	11	627	7.8	—	—	—	—	38862
1 ft = latite tuff																	
49	10	5.2	16	—	.17				126	—	612	7.8	—	—	—	—	38847
927	910	—	3.3	—	—		2930		279	—	4640	8.6	—	—	—	—	H&K
drilling chemical contaminants; lime of Cretaceous age																	
456	1610	—	—	—	—				736	0	7380	7.3	—	—	—	—	48858
<b>County</b>																	
46	104	12	0.5	—	0.07		369		28	—	653	8.2	—	—	—	—	36105
45	102	9	0.07	0.00	—			414	421	0	638	7.5	—	—	—	—	4897
22	59	—	—	—	—				15	0	432	8.1	—	—	—	—	36104



TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts)

gpm	*F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>2</sub>	
Grant																
Source: Drilled well			Location: 15S.17W.27.111													
Remarks: Gila Conglomerate; water level 17.5 ft below surface; well 300 ft deep; producing from 75-85 ft and 170-180 ft																
10	92	7-14-62	48	—	0.00	—	—	—	3.0	0.1	—	—	—	150	130	21
Source: Spring			Location: 15S.17W.30.224													
Remarks: Gila Conglomerate:																
30	77	9-14-55	—	—	—	—	—	—	—	—	—	—	—	—	123	14
Source: Allen Springs			Location: 16S.15W.26.412													
Remarks: Issues from fault zone in limestone																
80	77.5	4- 2-54	18	—	—	—	—	—	78	39	—	—	18	387	0	
Source: Drilled well			Location: 16S.17W.9.242													
Remarks: 36 ft deep, diameter 8", cased to 36 ft; water level 19 ft below surface																
—	86	6- 8-55	40	—	0.09	—	—	—	7.5	1.0	—	—	—	126	241	0
Source: Well			Location: 16S.17W.34.212													
Remarks: Pipe in orifice; volcanic rock (latite ?); water level +60 ft below surface																
90	84	4-26-55	32	—	0.04	—	—	—	18	3.0	—	—	—	92	232	0
Source: Spring			Location: 16S.18W.34.314													
Remarks: Volcanic tuff of Tertiary age and sandstone																
0.75	68	7-28-55	—	—	—	—	—	—	—	—	—	—	—	—	212	0
Source: Spring			Location: 16S.21W.20.300 (?)													
Remarks: Spring in bed of Bitter Creek																
1	69	9-20-41	—	—	—	—	—	—	536	67	—	—	—	62	164	0
Source: Ash Spring			Location: 17S.15W.20.222													
Remarks: Granite of Precambrian age																
0.25	72	8- 54	15	—	—	—	—	—	71	21	—	—	—	11	200	0
Source: Spring			Location: 17S.21W.18.200 (?)													
Remarks: Spring at fault.																
1	69	9-17-41	—	—	—	—	—	—	56	93	—	—	—	83	415	12
Source: Spring			Location: 18S.9W.31.340													
5	71.5	6-10-52	—	—	—	—	—	—	—	—	—	—	—	—	160	0
Source: Goat Spring			Location: 18S.9W.34.124													
Remarks: Mimbres conglomerate; water issues from joints in conglomerate striking N. 45 W., dipping 30° S																
20	66	3-21-57	—	—	—	—	—	—	—	—	—	—	—	—	210	0
Source: Mimbres Hot Springs			Location: 18S.10W.18.100													
Remarks: "a" spring upstream from Mimbres spring; "b," Green Horse Spring; "c," Mimbres Hot Spring. Discharge 100 gpm at 135° and																
"a"	10	79	6- 5-52	—	—	—	—	—	—	—	—	—	—	—	83	14
"b"	10	135.53	6- 5-52	—	—	—	—	—	—	—	—	—	—	—	75	20
"c"	20	137	6- 5-52	53	—	—	—	—	12	2.6	—	—	—	86	113	—
Source: Faywood Hot Springs			Location: 20S.11W.20.243													
Remarks: Several springs discharging 120 gpm @ 142°F from base of lava slope; issues from top of travertine mound 20 ft high (S); (see																
15-20	129.2	6- 5-52	—	—	—	—	—	—	—	—	—	—	—	91	282	0
50	128	4-19-57	43	—	0.0	0.1	0.00	—	—	38	7.3	—	—	85	7.8	278
Include with data below: Pb = 0.00																
50	128	11- 9-54	—	—	—	—	—	0.0	0.0	37	8.5	—	—	—	282	0
Source: Well			Location: 23S.15W.31.110													
Remarks: Drilled well 470 ft deep, water level 443 ft below surface																
—	82	5-16-55	34	—	0.15	—	—	—	44	13	—	—	—	123	229	0
Guadalupe																
Source: Spring			Location: 8N.21E.1.333													
Remarks: Blue Hole at outlet to U.S. Fish Hatchery at Santa Rosa; yield of 500-1000 gpm at 65.5°F (B)																
500-	1000	65.5	5- 6-59	16	—	—	—	—	620	62	—	—	—	33	181	0
Hidalgo																
Source: Blowing" well			Location: 22S.21W.3.312													
Remarks: Well of unknown depth with water level 446 ft below the land surface, discharges water from the Santa Fe formation (H&K);																
5	88	7- 8-55	—	—	—	—	—	—	—	—	—	—	—	—	431	—

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
 (Concentration in mg/l unless noted)

Well No.	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (µmhos at 25°C)	pH	H <sub>2</sub> S	Total β-γ activity (µµc/l)	Ra (µµc/l)	U (µµc/l)	Ref.
County (Continued)																	
103	18	21	0.1	---	---		431	435	8	0	665	9.0	---	---	---	---	50019
	5.2	---	---	---	---				72	0	256	8.7	---	---	---	---	31368
20	38	0.8	2.4	---	---		404		355	38	621	---	---	---	---	---	26035
51	16	8.0	1.6	---	---		363	370	22	0	551	7.9	---	---	---	---	29790
38	8.5	6.0	0.1	---	---		311	312	58	0	472	7.9	---	---	---	---	29750
	8.8	---	---	---	---				158	0	389	7.2	---	---	---	---	31374
19	21	1.1	0.8	---	---		2288		1613	---	255	---	---	---	---	---	1559
10	4	1.2	1.2	---	---		332		264	100	526	---	---	---	---	---	27635
33	21	0.9	20	---	---		773		522	---	118	---	---	---	---	---	1569
	9.0	5.2	---	---	---				66	---	347	---	---	---	---	---	19655
	9.0	---	---	---	---				132	0	353	7.4	---	---	---	---	38212
Flow from Mimbres fault zone (S); about 30 springs with a combined flow of 100+ gpm from latite and rhyolite at the surface (EBu)																	
	17	16	---	---	---				11	---	451	---	---	---	---	---	19647
	16	16	---	---	---				9	---	450	---	---	---	---	---	19645
5	17	16	0	---	---		308		40	0	457	---	---	---	---	---	19646
So JoEBu)																	
11	18	7.0	0.1	---	---				129	0	606	---	---	---	---	---	19824
2	16	6.8	0.2	0.00	---			384	125	---	605	7.4	---	19	29	0.1	S&B
	17	---	---	---	---				128	0	600	---	---	---	---	---	27917
15	23	2.8	3.5	---	---		549	553	164		820	8.0	---	---	---	---	29797
County																	
40	48	0.5	0.1	---	---		2460		1800	1650	2620	7.3	---	---	---	---	42376
Temperature is 95°F, depth is 449 ft (R)																	
	102	---	---	---	---				128		1590	7.8	---	---	---	---	H&K

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts)

gpm	°F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>2</sub>
<b>Hidalgo</b>															
Source: Wells			Location: 25S.19W.7.234												
Remarks: A group of three irrigation wells in alluvium, ranging from 83 to 106 ft deep with casings perforated 42-90, 2, & 50-82 ft, drilled															
—	—	2- 1-49	—	—	—	—	—	—	24	1.5	—	324	—	146	0
—	240	4-28-49	141	—	—	—	—	—	19	1.2	—	329	—	181	0
—	—	7-30-51	—	—	—	—	—	—	—	—	—	—	—	163	7
—	—	3-28-52	—	—	—	—	—	—	—	—	—	—	—	163	7
—	210	4-27-54	138	0.1	0.07	0.00	—	—	21	0.7	—	324	21	145	6
—	—	4-10-55	135	—	—	—	—	—	22	1.5	—	319	—	157	0
<b>Rio Arriba</b>															
Source: Springs at Ojo Caliente			Location: 24N.8E.												
Soda Spring															
15	95	10- 1-47	60	—	—	—	—	—	23	8.7	—	1040	—	2200	0
Soda Spring															
—	—	10- 6-49	66	—	—	—	—	—	25	9.0	—	997	29	2180	0
Sodium Sulfate Spring															
0.25	90	10- 1-47	56	—	—	—	—	—	25	8.7	—	1040	—	2210	0
Soda Spring															
—	115	10- 6-49	60	0.3	1.2	—	—	—	23	9.5	Trace	996	31	2230	—
Bath House															
10	105	10- 6-49	63	—	0.02	—	—	—	24	7.6	—	933	34	2160	0
Arsenic Spring															
—	113	10- 6-49	63	—	0.01	—	—	—	25	8.9	—	928	30	2160	0
Source: Spring			Location: 25N.8E.25												
Remarks: Three almost contingent springs; from Dakota Sandstone (?)															
Field 1															
—	97	9- 5-52	—	—	—	—	—	—	—	—	—	—	—	692	0
Field 2															
—	97	9- 5-52	—	—	—	—	—	—	—	—	—	—	—	698	0
Field 3															
—	97	9- 5-52	—	—	—	—	—	—	—	—	—	—	—	694	0
Field 4															
—	97	9- 5-52	22	—	—	—	—	—	145	59	—	187	—	698	0
Field 5															
—	65	9-15-52	15	—	0.01	—	—	—	44	11	—	11	—	73	0
<b>Sandoval</b>															
Source: Spring			Location: 13N.4E.36.323												
Remarks: Seep in arroyo bottom at fault															
0.5	68	8- 9-62	—	—	—	—	—	—	—	—	—	10	1.2	291	0
Source: San Ysidro Warm Springs			Location: 15N.1E.3, 9, and 10												
Remarks: A group of several springs north of Rio Salado, deposits of calcareous tufa around spring (Ri); several springs issuing from															
—	68	9-15-24	15	—	3.0	—	—	—	368	85	—	2219	—	1757	—
—	—	3- 6-45	—	—	—	—	—	—	322	84	—	1830	—	1780	0
—	—	3- 6-45	—	—	—	—	—	—	306	73	—	2080	—	2000	0
—	—	3- 2-45	—	—	0.48	—	—	—	324	85	—	1850	—	1820	0
5-10	72	9-29-48	16	—	—	—	—	—	300	74	—	2100	—	2020	0
Source: San Ysidro Hot Springs			Location: 15N.1E.8.400												
Remarks: Issues from faulted beds of Triassic age (S); calcareous tufa around spring (Ri)															
—	86	9-15-24	15	—	2.0	—	—	—	497	91	—	3310	—	1969	—
Source: "Hot well"			Location: 16N.1W.1 (unsurveyed)												
Remarks: Drilled well 12" in diameter, 550 ft deep (Ri), 2008 ft deep [Aqua Zarca, 600 ft; San Andres, 870 ft; Abo, 1535 ft; Magdalena,															
—	115	9-29-26	18	—	2.3	—	—	—	400	73	—	3450	—	1498	—
Include with data below: Mn = 0.00; Pb = 0.6; Li = 6.9; As = 0.60; Br = 0.3; I = 4.6; Se = 0.00															
1500	130	3-14-64	31	2.6	3.9	0.01	0.04	1.5	345	56	—	3550	87	1450	0
Source: Spring and oil test well			Location: 16N.1W.1.410												
Remarks: The flowing abandoned well is an oil test, rich in H <sub>2</sub> S and other gases; the oil test is 2000 ft deep															
—	180	4- 3-56	35	—	—	—	—	—	328	76	—	3500	—	1470	0
Well	—	7-29-58	31	—	0.00	—	—	—	301	67	—	3590	—	1410	0
—	140	9-29-48	27	—	—	—	—	—	368	73	—	3640	—	1470	0
Source: Swimming pool spring			Location: 16N.1E.20 (unsurveyed)												
—	70	9-11-24	30	—	0.60	—	—	—	260	70	—	2400	—	1301	—
Source: Indian Spring			Location: 16N.2E.29.142												
2	95	8-30-62	48	—	0.03	—	—	—	100	8.6	—	1240	—	1280	0

10 WELLS INCLUDING CHEMICAL ANALYSES (cont)  
 (million unless noted)

W	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (µmhos at 25°C)	pH	H.S.	Total dry activity (µmhos/l)	Ra (µmhos/l)	U (µmhos/l)	Ref.	
County (Continued)																		
into "hard rock" and "solid rock" categories																		
9	85	—	5	—	—	—	—	1020	68	0	1550	—	—	—	—	—	—	R
10	78	11	0.2	—	0.45	—	—	1130	32	0	1540	—	—	—	—	—	—	WSSA
—	81	—	—	—	—	—	—	—	—	0	1650	—	—	—	—	—	—	N
—	82	—	0.4	—	—	—	—	—	55	—	1600	8.2	—	—	—	—	—	R
74	83	9.9	0.3	—	—	—	—	1160	56	—	1580	8.4	—	12	0.3	0.2	—	S&B
79	80	13	0.2	—	—	—	—	1410	61	0	1510	7.6	—	—	—	—	—	R
County																		
68	238	16	1.6	—	1.7	—	—	2640	106	0	3890	—	—	—	—	—	—	8977
62	240	16	0.9	—	—	—	—	2620	100	0	3910	7.2	—	—	—	—	—	13378
65	245	16	1.7	—	4.6	—	—	2650	106	0	3890	—	—	—	—	—	—	8978
1	231	0.84	0.9	0.2	1.2	—	—	—	—	—	—	7.2	—	—	—	—	—	WH&W
66	238	16	0.5	—	1.7	—	—	2540	91	0	3920	6.9	—	—	—	—	—	13192
66	238	16	0.8	—	1.6	—	—	2530	99	0	3930	7.1	—	—	—	—	—	13193
—	107	—	—	—	—	—	—	—	—	—	1730	—	—	—	—	—	—	26032
—	111	—	—	—	—	—	—	—	—	—	1760	—	—	—	—	—	—	26033
—	108	—	—	—	—	—	—	—	—	—	1740	—	—	—	—	—	—	26034
70	110	1.4	0.2	—	—	—	—	1140	604	32	1740	—	—	—	—	—	—	26035
10	2.5	0.2	0.2	—	—	—	—	230	115	95	347	7.0	—	—	—	—	—	26119
County																		
57	6.4	—	0.6	—	—	—	—	—	294	56	563	7.4	—	—	—	—	—	50288
alted crest of anticline (S)																		
12	1940	—	Trace	—	—	—	—	7320	—	—	—	—	—	—	—	—	—	Ri
60	1710	—	—	—	BO <sub>3</sub> =60	—	—	6020	1150	0	903	—	—	—	—	—	—	3478
60	1920	—	—	—	BO <sub>3</sub> =60	—	—	6560	1060	0	980	—	—	—	—	—	—	3479
10	1700	—	—	—	—	—	—	6100	1160	0	896	—	—	—	—	—	—	3548
10	1880	3.8	5.1	—	10.3	—	—	6610	1050	0	9750	6.5	—	—	—	—	—	10983
91	2500	—	Trace	—	50	—	—	10960	1608	—	—	—	—	—	—	—	—	Ri
90 ft] (B)																		
15	2660	—	0	—	—	—	—	11120	1299	—	—	—	—	—	—	—	—	Ri
60	2990	2.8	0.2	—	4.8	—	—	11000	1090	0	15300	7.3	—	—	—	—	—	54144
20	2900	—	—	—	—	—	—	10900	1130	0	15000	6.6	—	—	—	—	—	32740
10	2970	4.5	8.1	—	6.6	—	—	11000	1030	0	14900	7.2	—	—	—	—	—	39231
10	3010	2.6	—	—	6.6	—	—	11400	1220	14	15400	6.8	—	—	—	—	—	10984
28	2330	—	Trace	—	—	—	—	7510	—	937	—	—	—	—	—	—	—	Ri
86	1140	7.3	0.3	—	6.1	—	—	3470	285	0	5680	8.0	—	—	—	—	—	50248

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts

gpm	°F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>3</sub>	
Sandoval																
Source: Spring at Ojo del Espiritu ranch house (CCC camp on San Ysidro quadrangle)										Location: 17N.1W.15 (unsurveyed)						
—	60	9-22-24	30	—	0.30	—	—	—	90	12	12	—	12	259	—	
Source: Soda Dam springs (The Sulphurs)										Location: 18N.2E.14 (unsurveyed)						
Remarks: Large tufa deposit in channel of Jemez River from flow of about 10 gpm at 75°-105°F (S); at contact of granite of Precambrian																
—	104	8-21-24	48	—	0.10	—	—	—	328	23	—	1000	—	1440	—	
40	96	6-28-49	47	—	—	—	—	—	327	27	—	830	—	1400	0	
Sulphur Pool																
—	100	8-2-49	—	—	—	—	—	—	—	—	—	—	—	1500	0	
Sulphur Pool																
10	102	6-24-49	47	—	—	—	—	—	330	29	—	1170	—	1540	0	
Sulphur Pool																
1	—	1-20-50	42	—	—	—	—	—	221	29	—	1020	197	1200	0	
Sulphur Pool																
—	—	12-20-49	—	—	—	—	—	—	—	—	—	—	—	1520	0	
Geyser Spring																
—	—	10-10-50	—	—	—	—	—	—	—	—	—	—	—	1550	0	
Geyser Spring																
—	—	6-20-50	—	—	—	—	—	—	—	—	—	—	—	1550	0	
Geyser Spring																
—	110	8-2-49	—	—	—	—	—	—	—	—	—	—	—	1580	0	
Geyser Spring																
5	—	6-28-49	47	—	—	—	—	—	344	29	—	1140	—	1580	0	
Geyser Spring																
—	—	12-20-49	—	—	—	—	—	—	—	—	—	—	—	1590	0	
—	—	8-31-46	—	—	—	—	—	—	—	—	—	—	—	883	0	
Hole-in-Rock Spring																
—	—	6-14-49	35	—	—	—	—	—	304	32	—	932	—	1560	0	
Sulphur Pool																
—	—	8-31-49	48	—	0.04	—	—	—	332	33	—	1000	183	1530	0	
—	104	8-21-24	48	—	0.10	—	—	—	328	23	—	1000	—	1440	0	
2	81	6-26-49	43	—	—	—	—	—	326	30	—	1200	—	1550	0	
25	95	8-2-49	44	—	—	—	—	—	326	27	—	986	—	1440	0	
—	—	8-31-46	—	—	—	—	—	—	346	33	—	1230	—	1530	0	
—	—	8-31-46	—	—	—	—	—	—	314	32	—	989	—	1560	0	
0.5	102	6-14-49	47	—	—	—	—	—	340	31	—	1170	—	1590	0	
Well																
12.5	—	12-13-57	—	—	—	—	—	—	—	—	—	—	—	338	5	
Dug Pit																
40	96	6-28-49	47	—	—	—	—	—	327	27	—	830	—	1400	0	
Source: Jemez Hot Springs										Location: 18N.2E.23 (unsurveyed)						
Remarks: About 10 springs flowing 200 gpm with temperatures ranging from 94° to 168°F from faults in redbeds of Permian age (S)																
Bath House																
—	125	8-21-24	91	—	1.2	—	—	—	166	9.0	—	645	—	791	—	
—	160	4-3-56	86	—	—	—	—	—	136	10	—	618	70	716	0	
Main Spring No. 1																
—	—	4-15-47	47	—	0.04	—	—	—	18	6.2	—	12	3.6	94	0	
Main Spring No. 2																
—	—	4-15-47	60	—	0.01	—	—	—	47	14	—	14	3.0	228	0	
No. 3																
—	—	4-15-47	51	—	0.01	—	—	—	34	10	—	39	3.8	232	0	
—	164	8-1-47	64	—	0.0	—	—	—	137	4.4	—	701	—	750	0	
25	—	6-14-49	91	—	—	—	—	—	137	9.0	—	677	—	740	0	
Behind Bath House																
10	150	6-14-49	92	—	—	—	—	—	140	9.4	—	680	—	758	0	
10	150	8-31-49	93	—	0.03	—	—	—	138	6.6	—	572	70	735	0	
Behind Bath House																
20	152	10-24-51	—	—	—	—	—	—	—	—	—	—	—	727	0	
Source: Hot spring (McCauley Spring)										Location: 18N.3E.4						
Remarks: Issues at base of recent volcanic flow on contact with red beds of Permian age; no obvious mineralization																
110	98	8-1-47	53	—	0	—	—	—	11	4.2	—	23	—	87	0	
Source: Sulphur Springs on Sulphur Creek										Location: 19N.3E.4 (unsurveyed)						
Remarks: Thermal waters issue from volcanics of late Tertiary age (Ri); about 8 springs discharging 500 gpm at 86°-167°F from andesites																
Ladies Bath House; Acidity as SO <sub>4</sub> = 2304																
—	110	8-31-24	276	195	369	—	Trace	—	321	24	—	304	—	0	0	
Alum Spring; Acidity as SO <sub>4</sub> = 2328																
—	76	8-31-24	146	421	72	—	—	—	316	51	—	127	—	0	0	
Alum Spring; Acidity as SO <sub>4</sub> = 2570																
—	61	8-13-47	170	501	2.8	—	—	—	256	35	—	—	—	0	0	
Laxative Spring; Acidity as SO <sub>4</sub> = —																
—	—	8-31-49	42	—	0.64	—	—	—	168	23	—	14	—	0	0	

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
per million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (μmhos at 25°C)	pH	H <sub>2</sub> S	Total H <sup>+</sup> activity (μμc/l)	Ra (μμc/l)	U (μS/l)	Ref.	
County (Continued)																		
99	4	—	0.25	—	—	396			274	—	—	—	—	—	—	—	—	Ri
and Magdalena Group (Ri) (see Jo)																		
40	1320	—	Trace	—	—	3458			914	—	—	—	—	—	—	—	—	Ri
51	1080	2.4	1.3	—	8.7		3060		927	—	5160	6.8	—	—	—	—	—	12208
39	1540	—	—	—	—				—	—	6610	—	—	—	—	—	—	12529
42	1540	3.2	1.0	—	14		3920		942	0	6600	6.9	—	—	—	—	—	12210
39	1510	3.3	4.3	—	—		3660		670	0	6150	—	—	—	—	—	—	13845
—	1530	—	—	—	—				—	—	6530	—	—	—	—	—	—	13484
—	1550	—	—	—	—				—	—	6620	—	—	—	—	—	—	15097
—	1550	—	—	—	—				—	—	6590	—	—	—	—	—	—	15096
40	1510	—	—	—	—				—	—	6590	—	—	—	—	—	—	12530
42	1500	3.2	1.3	—	13		3880		978	0	6520	6.9	—	—	—	—	—	12211
—	1520	—	—	—	—				—	—	6530	—	—	—	—	—	—	13485
912	1570	—	—	—	—				—	—	5970	—	—	—	—	—	—	7124
60	1110	3.2	1.6	—	9.8		3250		890	0	5410	6.9	—	—	—	—	—	12146
41	1550	3.6	1.4	—	12		3950		964	0	6570	6.9	—	—	—	—	—	13080
40	1320	—	Trace	—	BO <sub>3</sub> =2.5		3471		914	—	3458	—	0	—	—	—	—	3227
10	1580	3.6	1.2	—	12		3990		937	0	6720	7.0	—	—	—	—	—	12209
5	1300	2.8	1.7	—	9.1		3440		924	0	5860	—	—	—	—	—	—	12531
263	1520	3.5	1.5	—	—		4150		999	0	651	—	—	—	—	—	—	7125
53	1220	2.8	1.4	—	—		3380		915	0	563	—	—	—	—	—	—	7126
37	1540	4.0	2.0	—	12.5		3950		976	0	6620	6.7	—	—	—	—	—	12147
—	9.0	—	—	—	—				266	0	563	8.3	—	—	—	—	—	—
51	1080	2.4	1.3	—	8.7		3060		927	0	5160	6.8	—	—	—	—	—	12208
(see also Ri, Jo)																		
42	820	—	5.0	—	2.5		2184		452	—	—	—	—	—	—	—	—	Ri, 3226
44	870	5.2	0.5	—	—		2190		380	0	3860	6.7	—	—	—	—	—	32741
15	4	0.8	0.4	—	0.0		153		70	0	1840	—	—	—	—	—	—	8252
15	4	0.8	0.3	—	0.0		270		175	0	3510	—	—	—	—	—	—	8253
17	4	0.8	0.3	—	BO <sub>3</sub> =0.0		274		126	0	3640	—	—	—	—	—	—	8254
44	855	7.1	0.4	—	6.33		2180		360	0	3700	7.2	—	—	—	—	—	8928
51	835	4.9	1.7	—	—		2170		379	0	3640	7.6	—	—	—	—	—	12139
51	835	4.9	1.5	—	—		2190		388	0	3670	7.7	—	—	—	—	—	12140
49	795	5.2	0.8	—	11		2150		372	0	3560	7.2	—	—	—	—	—	13079
—	—	—	—	—	—				—	—	3680	—	—	—	—	—	—	17772
8.0	8.0	1.6	0.4	—	BO <sub>3</sub> =0		152		45	0	19.8	8.1	—	—	—	—	—	8929
and rhyolites of Tertiary age (S) (see also GPH)																		
3560	294	—	0	77	—		6270	5420	—	—	—	—	8.2	—	—	—	—	3224
3159	1.0	—	Trace	41	—		4344	4344	—	—	—	—	2.1	—	—	—	—	3221
—	170	0.7	0.4	—	—				—	—	826	1.8	—	—	—	—	—	8937
614	8.0	0.0	0.4	—	—		967		514	514	1270	3.1	—	—	—	—	—	13257

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts

gpm	*F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>2</sub>
Sandoval															
Source: Sulphur Springs on Sulphur Creek (cont)															
Lemonade Spring	150	8-31-49	216	56	33	3.3	—	—	185	52	—	6.7	24	0	0
Men's Bath House	110+	8-31-49	219	205	217	—	—	—	66	17	—	—	42	0	0
Bath House; As = 0.05; Li = 0.07; Pb = 0.12; Se = 0.03	188	11-4-63	190	36	115	0.33	—	—	6.9	9.7	—	24	31	0	0
Foot Bath	99	8-31-24	259	90	252	—	—	—	41	16	—	—	52	0	—
Lemonade Spring	115	8-13-47	162	—	1.8	—	—	—	150	73	—	—	—	0	0
Lemonade Spring	130	7-28-49	—	—	—	—	—	—	—	—	—	—	—	—	—
Lemonade Spring; Se = 0.00	—	1-20-50	—	—	—	—	—	—	—	—	—	—	—	—	—
Lemonade Spring	—	10-24-51	—	—	—	—	—	—	—	—	—	—	—	—	—
Men's Bath House; Acidity as SO <sub>4</sub> = 5400	—	8-31-24	324	303	1250	—	—	—	303	33	—	—	157	0	0
Ladies' Bath House	—	8-31-49	237	172	93	—	—	—	110	11	—	24	—	0	0
Mud Bath	99	7-28-49	—	—	—	—	—	—	—	—	—	—	—	—	—
Hot Sulphur Mud Bath	80-98	8-13-47	—	694	22	1.1	—	—	154	42	—	—	12	—	—
Mud Foot Bath	97	8-13-47	122	469	12	—	—	—	7.2	19	—	—	—	0	0
Footbath Spring	72	7-28-49	—	—	—	—	—	—	—	—	—	—	—	—	—
Mud Bath	105	8-31-49	174	104	92	—	—	—	45	12	—	13	—	0	0
Electric Spring	97	8-13-47	166	—	3.4	—	—	—	140	161	—	—	—	0	0
Electric Spring	15	102	7-28-49	—	—	—	—	—	—	—	—	—	—	—	—
Electric Spring	1	8-31-49	206	194	81	—	—	—	101	23	—	9.6	42	0	0
Alum and Boric Spring	72	8-13-47	—	361	17	—	—	—	372	43	—	—	47	—	—
Sulphur Creek below Sulphur Springs	—	10-22-49	83	76	40	—	—	—	164	24	—	16	—	0	0
Source: Hot spring (Natural Bath-Tub) Location: 19N.3E.28.310															
Remarks: Issues at base of recent volcanic flow on contact with red beds of Permian age; no observable mineralization; elevation 7350 ft															
—	100	8-1-47	71	—	0	—	—	—	7.5	2.2	—	—	56	139	0
Source: Steam well Location: 20N.3E.35 (unsurveyed)															
Remarks: Analyses reported to be of steam condensate which would enter stream as runoff															
—	—	6-13-63	—	—	12.6	0.5	—	—	28	0	—	83	27	623	—
—	—	6-18-63	—	—	1.05	0.03	—	—	2	1.2	—	83	31	854	—
Source: Alamo Canyon Spring Location: 20N.3E.35															
Remarks: Over area approximately 50 × 100 ft are numerous points evolving gas, some sulfurous gas but largely CO <sub>2</sub> ; elevation 8700 ft;															
—	—	8-1-47	87	—	0.41	—	—	—	32	22	—	—	—	0	0
Source: Spring on Rio Antonio (warm spring) Location: 20N.4E.7 (unsurveyed)															
Remarks: Issues from rhyolite of Tertiary age (He); 50 gpm at 120°F (S) (see also P)															
25	101	8-1-47	103	—	0	—	—	—	6.5	1.1	—	—	40	77	0
San Miguel															
Source: Spring Location: 16N.16E.6															
1	106	3-11-52	68	—	0.02	—	—	—	4.5	1.0	—	—	179	77	16
5	123	3-11-52	59	—	0.03	—	—	—	4.5	1.1	—	—	179	66	22
Source: Montezuma Hot Springs Location: 16N.16E.6															
—	—	5-16-39	—	—	—	—	—	—	14	8.3	—	—	141	—	—
—	130	7-2-40	—	—	—	—	—	—	8.5	0.7	—	—	173	92	11
—	—	8-20-40	—	—	—	—	—	—	—	—	—	—	—	82	13
—	—	8-20-40	—	—	—	—	—	—	—	—	—	—	—	80	15
Sierra															
Source: Oil test, Victoria Land and Cattle Company No. 2 Location: 10S.2W.25.100															
Remarks: Flow from 1328 to 1347 ft from San Andres Limestone (H&K)															
900	94	—	—	—	—	—	—	—	—	—	—	—	—	636	—

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
per million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (μmhos at 25°C)	pH	H <sub>2</sub> S	Total activity (μμe/l)	Ra (μμe/l)	U (μg/l)	Ref.
County (Continued)																	
1570	3.5	1.1	0	—	—	1950			676	0	4570	1.9	—	—	—	—	He, 13260
1250	45	0.9	0.5	—	—		3630		234	234	13900	1.4	—	—	—	—	13253
1100	24	1.2	0.0	2.4	—				57	57	13800	2.0	—	—	—	—	5300
1337	20	—	0	154	—	2562	3221		168	—	—	—	1.6	—	—	—	Ri, 3225
1190	65	1.0	0.3	—	—				—	—	—	2.0	—	—	—	—	8935
1590	—	—	—	—	—				—	—	4570	2.2	—	—	—	—	12524
—	—	—	—	—	—				—	—	3560	2.2	—	—	—	—	13558
—	—	—	—	—	—				—	—	2710	2.3	—	—	—	—	17774
6156	54	—	Trace	37	—	7887	8617		—	—	—	—	—	—	—	—	3223
2740	20	0.5	0.6	—	—	2690			320	320	8510	1.6	—	—	—	—	13254
750	—	—	—	—	—				—	—	6100	1.9	—	—	—	—	12520
1430	241	—	—	—	—				—	—	1410	1.9	—	—	—	—	9171
1190	649	—	0.3	—	—				—	—	1660	1.5	—	—	—	—	8934
2230	—	—	—	—	—				—	—	13900	1.7	—	—	—	—	12519
1440	—	—	0.7	—	—	1730			162	162	4370	1.9	—	—	—	—	13258
15	410	1.5	0.3	—	—				—	—	1170	1.5	—	—	—	—	8936
880	—	—	—	—	—				—	—	12500	1.7	—	—	—	—	12522
820	2.5	1.0	0.0	—	—	3160			346	346	12700	1.4	—	—	—	—	13259
1560	74	—	—	—	—		364		—	—	794	2.9	—	—	—	—	9172
1160	4.0	0.3	0.0	—	—	1600			508	508	2270	2.5	—	—	—	—	13262
17	11	0.8	0.2	—	0.8		234		28	0	28.3	7.3	—	—	—	—	8933
335	121	24	0.00	—	—	2970			70	—	2225	8.0	—	—	—	—	NMPHL
177	114	24	0.00	—	—	1700			10	—	2070	8.3	—	—	—	—	NMPHL
acidity as H <sub>2</sub> SO <sub>4</sub> = 153																	
242	3.0	0.3	0.6	—	BO <sub>3</sub> =0.2				—	—	72.1	2.9	—	—	—	—	8930
15	17	1.6	0.4	—	0		222	230	20	0	16.7	6.7	—	—	—	—	He
County																	
42	155	20	0.1	—	—	528	524		15	0	876	8.8	—	—	—	—	18610
42	155	20	0.1	—	—	530	515		16	0	876	9.0	—	—	—	—	18609
66	154	—	—	—	—	554			—	—	878	—	—	—	—	—	2083
49	158	—	0.2	—	—	537			—	—	870	—	—	—	—	—	4801
—	160	—	—	—	—	531			—	—	878	—	—	—	—	—	5233
—	159	—	—	—	—	523			—	—	872	—	—	—	—	—	5233
County																	
1660	22	—	—	—	—				1850	—	1850	7.2	—	—	—	—	H&K



TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts)

gpm	°F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>3</sub>	
<b>Sierra</b>																
Source: Warm spring			Location: 12S.5W.28													
Remarks: In Cuchillo Canyon, 10 miles NW of T or C and 15 miles W of Cuchillo (Th)																
—	85.6	2-9-10-39	37	—	0.00	—	—	—	109	10	—	386	26	212	0	
Source: Drilled wells at Truth or Consequences			Location: 13S.4W.33													
Remarks: (for detailed location, see Th) Magdalena Limestone (Th); "discharge," respectively, from 27 ft, 105 ft, and 100 ft deep																
—	112.3	2-9-10-39	38	—	0.10	—	—	—	150	16	—	692	45	218	—	
—	114	2-9-10-39	—	—	0.07	—	—	—	152	18	—	740	—	210	—	
—	111.2	2-9-10-39	—	—	0.42	—	—	—	155	17	—	772	—	214	—	
Source: Springs at Truth or Consequences			Location: 13S.4W.33													
Remarks: Discharge from alluvium (for detailed description, see Th)																
1.3	103-106	9-9-10-39	36	—	0.1	—	—	—	154	14	—	731	39	215	0	
1	99-103	2-9-10-39	37	—	0.08	—	—	—	148	16	—	678	36	212	0	
Old Gov't Spring			Location: 13S.4W.34.310													
—	106	7-23-38	—	—	—	—	—	—	168	36	—	687	53	217	0	
Source: Drilled well in Truth or Consequences			Location: 13S.4W.34.310													
Remarks: Valley fill; well 120 ft deep (Th)																
—	70	2-9-10-39	18	—	3.30	—	—	—	460	76	—	661	21	152	—	
Source: Palomas Spring			Location: 13S.5W.31.130													
Remarks: Santa Fe Formation																
—	—	6-12-58	—	—	—	—	—	—	—	—	—	—	—	171	0	
Source: Wells of Truth or Consequences			Location: 14S.4W.4													
Remarks: (for detailed location, see Th) Magdalena Limestone (Th); "discharge," respectively, from 212 ft and 208 ft deep																
—	—	2-9-10-39	—	—	0.08	—	—	—	154	16	—	730	—	121	—	
—	—	2-9-10-39	32	—	0.53	—	—	—	153	15	—	714	43	219	—	
Source: Yucca Baths at Truth or Consequences (Ponce de Leon Spring)			Location: 14S.4W.4													
Remarks: (for detailed location, see Th) Alluvium samples from 14 ft deep drive point into spring head (Th)																
—	110	2-9-10-39	—	—	0.08	—	—	—	154	19	—	751	—	218	—	
2	110	3-31-52	39	—	0.01	—	—	—	174	25	—	692	—	221	—	
1.1	109	5-28-54	41	0.1	0.02	0.19	—	—	154	21	—	735	61	216	0	
Well 14 ft deep			Location: 14S.4W.4													
—	109.4	4-28-43	—	—	—	—	—	—	—	—	—	—	—	—	—	
0.5	—	7-12-54	—	—	—	—	—	—	156	20	—	749	—	220	0	
0.75	110	8-2-55	—	—	—	—	—	—	—	—	—	—	—	221	0	
0.75	110.5	9-17-56	—	—	—	—	—	—	—	—	—	—	—	216	0	
Flowing Well			Location: 14S.4W.4													
8.5	108	8-5-57	—	—	—	—	—	—	—	—	—	—	—	228	0	
Bath House Spring			Location: 14S.4W.4													
—	104	4-15-58	—	—	—	—	—	—	—	—	—	—	—	227	0	
North Spring			Location: 14S.4W.4													
—	107	8-3-59	—	—	—	—	—	—	—	—	—	—	—	221	0	
—	104	4-4-60	—	—	—	—	—	—	—	—	—	—	—	220	0	
—	107.5	8-13-62	—	—	—	—	—	—	—	—	—	—	—	220	0	
Source: City well No. 2 at Truth or Consequences			Location: 14S.4W.6													
Remarks: (for detailed location, see Th) Alluvium; well 200 ft deep (Th)																
—	77	2-9-10-39	21	—	2.20	—	—	—	58	5	—	131	9	125	—	
Source: Drilled well			Location: 15S.6W.31.34													
Remarks: 193 ft deep, diameter 6"; alluvium; water level at 100 ft																
15	70.2	8-20-46	—	—	—	—	—	—	63	21	—	36	—	242	0	
Source: Drilled well at Hot Springs			Location: 16S.5W.22													
40	75	6-14-46	—	—	—	—	—	—	21	4.4	—	59	—	169	0	
Source: Derry Warm Springs			Location: 17S.4W.29.340													
Remarks: Issues from limestone bluff on east side of Rincon Valley (C); small deposits of travertine; flows approximately 50 gpm from 2																
50	93	4-17-48	—	—	—	—	—	—	52	19	—	303	—	370	0	
5-10	93.2	5-7-52	32	—	—	—	—	—	48	20	—	293	—	372	0	
10-15	93	4-30-57	—	—	—	—	—	—	—	—	—	—	—	368	0	
<b>Socorro</b>																
Source: Springs			Location: 1N.1E.31.320													
Remarks: Sand and gravel; flow in area may be as much as 100 gpm; (a) pipe from side of hill at pond on railroad near San Acacia; (b)																
(a)	25	7-9-63	—	—	—	—	—	—	368	143	—	900	—	178	0	
(b)	<1	7-9-63	—	—	—	—	—	—	320	118	—	900	—	163	0	

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
 per million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (μmhos at 25°C)	pH	H <sub>2</sub> S	Total activity (μmC/l)	Ra (μmC/l)	U (μg/l)	Ref.
<b>County (Continued)</b>																	
79	650	2.4	25.0	—	0.5	1428	1429		314	—	—	—	—	—	—	—	Th
102	1230	3.4	2	—	—	2486			441	—	—	—	—	—	—	—	Th
74	1280	2.6	—	—	—				—	—	—	—	—	—	—	—	Th
73	1330	3.2	—	—	—				457	—	—	—	—	—	—	—	Th
79	1300	3	10	—	—	2560	2472		442	—	452	—	—	—	—	—	Th
81	1210	3	5	—	—	2418	2318		236	—	429	—	—	—	—	—	Th
95	1314	—	—	—	—				—	—	459	—	—	—	—	—	684
1193	1120	1.6	4	—	—	3720			1462	—	—	—	—	—	—	—	Th
23	114	—	—	—	—				158	18	664	7.7	—	—	—	—	38787
75	1250	2.8	—	—	—				451	—	—	—	—	—	—	—	Th
105	1240	3.2	6.2	—	—	2437			444	—	—	—	—	—	—	—	Th
86	1290	3.2	—	—	—				463	—	—	—	—	—	—	—	Th
98	1240	2.8	2.7	—	—		2380	2640	537	—	4430	7.2	—	—	—	—	S&B
93	1290	3.3	2.0	—	—			2670	470	—	4510	7.3	—	100	0.7	3.3	S&B
—	1285	—	—	—	—				—	—	438	—	—	—	—	—	653
95	1290	—	—	—	—		2420		471	290	4420	—	—	—	—	—	27039
91	1300	—	—	—	—				490	309	4450	7.4	—	—	—	—	31082
91	1290	—	—	—	—				525	348	4450	7.4	—	—	—	—	33929
—	1280	—	—	—	—				470	283	4400	7.2	—	—	—	—	38827
—	1280	—	—	—	—				475	289	4460	7.2	—	—	—	—	38930
—	1290	—	—	—	—				—	—	4450	7.2	—	—	—	—	43037
96	1300	—	—	—	—				470	290	4450	7.5	—	—	—	—	44734
—	1310	—	—	—	—				510	330	4480	7.2	—	—	—	—	50301
52	212	0.6	1	—	—	556			166	—	—	—	—	—	—	—	Th
80	26	1.2	1.2	—	—		348		244	45	60.9	—	—	—	—	—	6870
36	13	1.2	0.8	—	0.0		219		70	0	36.0	—	—	—	—	—	6135
levels: (1) 5 ft above floor at 92°F, (2) 6 ft above at 66°F from alluvial material above Derry fault block																	
309	160	5.8	2.0	—	—		1030		208	—	1650	—	—	—	—	—	C
306	141	6	1.3	—	—		1030		207	0	1660	—	—	—	—	—	18725
303	158	—	—	—	—				192	0	1660	7.4	—	—	—	—	35977
<b>County</b>																	
bank of railroad just north of (a)																	
1010	1816	—	—	—	—				1510	—	6970	7.4	—	—	—	—	FRH
920	1526	—	—	—	—				1284	—	6080	7.6	—	—	—	—	FRH

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts

gpm	*F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Na	K	HCO <sub>3</sub>	CO <sub>3</sub>	Socorro		
Source: Spring Location: 1N.2W.7.100																	
Remarks: Madera Limestone (see H&K)																	
500	70	11-30-49	—	—	—	—	—	—	138	59	—	—	887	420	0		
Source: Artesian spring Location: 1N.2W.7.132																	
Remarks: (mouth of box on Rio Salado) Madera Limestone																	
—	72	11-19-61	—	—	—	—	—	—	—	—	—	—	—	398	0		
Source: Well Location: 1N.2E.34.130																	
Remarks: 33 ft deep																	
—	65	—	21	—	—	—	—	—	562	130	—	—	21	142	—		
Source: Ojitos Springs Location: 2S.1W.19.431 and 2S.1W.19.414																	
Remarks: A group of springs discharging from rhyolite breccia fault zone																	
106	—	1952	—	—	—	—	—	—	50	12	—	—	67	308	—		
—	68	7- 9-63	—	—	—	—	—	—	29	11	—	—	38	195	0		
—	—	7- 9-63	—	—	—	—	—	—	40	14	—	—	76	332	0		
Source: Cook Spring Location: 3S.1W.15.313; temperature in pond																	
10-15	66	3-20-58	28	—	—	—	—	0.00	0.00	—	—	—	66	3.0	175	0	
Include with data below: Li = 0.33																	
—	70	9-24-64	26	—	0.58	—	—	—	13	4	—	—	68	3.4	158	3	
Source: Well (Blue Canyon) Location: 3S.1W.16.323																	
Remarks: Drilled well 300 ft deep, water level 210 ft below surface in rhyolite breccia																	
20	90.4	7-24-56	26	—	—	—	—	—	—	—	—	—	53	145	8		
—	88	12-20-61	—	—	—	—	—	—	18	5	—	—	55	166	0		
—	89	4-10-65	27	—	—	—	—	—	20	4.6	—	—	56	3	163	0	
Source: Socorro warm springs Location: 3S.1W.22.113																	
Remarks: (S) Several springs flowing from lake beds of Tertiary age lying against lava hills flowing 500 gpm at 91°F (S&B); flow from																	
—	—	2-17-36	—	—	—	—	—	—	19	4	—	—	55	5	168	—	
—	—	12- 4-36	—	—	—	—	—	—	18	5	—	—	—	53	156	—	
—	91	1952	—	—	—	—	—	—	19	5	—	—	—	53	163	—	
353	90	1-24-57	27	0.00	0.00	0.00	—	—	18	3.9	—	—	52	2.8	154	0	
220	90	3-20-58	39	—	—	—	—	0.00	0.04	—	—	1.6	55	3.0	160	5	
—	91	12-12-61	—	—	—	—	—	—	18	5	—	—	50	—	163	0	
—	91	2- 5-63	—	—	—	—	—	—	13	5	—	—	—	—	156	0	
—	92	4-10-65	—	—	—	—	—	—	18	4.4	—	—	—	—	155	0	
Source: Sedillo Spring Location: 3S.1W.22.131																	
Remarks: Issues from rhyolite breccia																	
240	90	3-20-58	27	—	—	—	—	0.00	0.04	—	—	—	0.00	54	2.9	159	0
—	88	12-12-61	—	—	—	—	—	—	18	5	—	—	—	50	—	154	5
Source: Ojo de las Cañas, east of Rio Grande Location: 3S.2E.19.323																	
Remarks: Controlled by prominent east-dipping sandstone bed in Yeso Formation, with fault control downstream (FRH)																	
—	79	6-13-62	—	—	—	—	—	—	552	141	—	—	—	38	—	193	0
10	—	3-15-63	—	—	—	—	—	—	488	141	—	—	—	—	—	142	0
Source: Well Location: 4N.2E.35.220																	
Remarks: 187 ft deep																	
—	73	2-24-50	32	—	—	—	—	—	24	6.3	—	—	—	34	—	115	—
Source: Well Location: 4N.3E.23.430																	
Remarks: 370 ft deep																	
—	72	3-29-50	17	—	—	—	—	—	20	6.0	—	—	—	28	—	98	—
Source: Well Location: 5S.1E.36.440																	
Remarks: Alluvium																	
—	80	10- 5-62	—	—	—	—	—	—	—	—	—	—	—	—	—	258	0
Source: Sawmill Spring Location: 5S.3W.4.231																	
Remarks: Sawmill Canyon above Birris Ranch																	
—	66	8- 7-63	—	—	—	—	—	—	33	3.9	—	—	—	6	—	124	0
Source: Spring Location: 6S.1W.6.440																	
Remarks: (Diamond A Ranch) Volcanics in fault zone (?)																	
25	70	6-18-63	—	—	—	—	—	—	22	2.4	—	—	—	22	—	112	0
Source: Springs Location: 8S.7W.31.300																	
Remarks: Issues from Gila Conglomerate (?); (a) from developed spring area, below ruins, east of Alamosa River at Ojo Caliente; (b) from																	
(a)	4.5	82	12-13-63	—	—	—	—	—	44.8	1.4	—	—	—	—	—	117.1	0.
(b)	—	—	—	—	—	—	—	—	37.6	1.9	—	—	—	—	—	122	0.

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
per million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (µmhos at 25°C)	pH	H <sub>2</sub> S	Total β-γ activity (µµe/l)	Ra (µµe/l)	U (µµe/l)	Ref.
County (Continued)																	
611	1160	—	4.9	—	—		3110		714	—	5023	—	—	—	—	—	13430
—	1080	—	—	—	—				—	—	4750	7.6	—	—	—	—	48607
1770	10	0.6	10	—	—	2590			1940	—	2760	—	—	—	—	—	SP
38	20	—	—	—	—				174	—	—	—	—	—	—	—	—
20	14	—	—	—	—				118	—	499	7.7	—	—	—	—	FRH
20	22	—	—	—	—				158	—	703	8.2	—	—	—	—	FRH
44	14	1.0	1.1	—	0.08				62	0	393	8.1	—	—	—	—	38856
42	14	0.8	0.8	—	0.13	254	250		49	0	391	8.4	—	—	—	—	55434
37	14	0.6	1.0	—	0.76				78	—	380	—	—	—	—	—	33772
32	17	—	—	—	—				68	—	390	8.0	—	—	—	—	FRH
36	14	—	1.1	—	—				69	0	375	7.6	—	—	—	—	56677
rhyolite agglomerate (Sp, Sd, Ho, FRH, Jo)																	
30	14	—	1.3	—	—				63	—	340	—	—	—	—	—	Sd
	13	1.0	0.6	—	—				64	—	318	—	—	—	—	—	Sd
	13	—	—	—	—	234			70	—	8.2	—	—	—	—	—	
28	15	0.6	1.2	0.15	—			224	61	—	348	7.8	—	<11	0.2	1.8	S&B
33	16	0.7	1.1	—	0.06				74	0	362	8.1	—	—	—	—	38854
28	8	—	—	—	—				64	—	370	8.1	—	—	—	—	FRH
20	12	—	—	—	—				52	—	356	7.8	—	—	—	—	FRH
—	—	—	—	—	—				63	0	346	7.8	—	—	—	—	56678
33	14	0.8	1.3	—	0.05				63	0	318	8.2	—	—	—	—	38853
24	10	—	—	—	—				64	—	370	8.4	—	—	—	—	FRH
1776	24	—	—	—	—				1960	—	3030	7.6	—	—	—	—	FRH
1568	20	—	—	—	—				1800	—	2800	—	—	—	—	—	FRH
49	9	0.8	1.6	—	—	214			86	—	310	—	—	—	—	—	Sp
40	7	0.6	1.6	—	—	168			74	—	263	—	—	—	—	—	Sp
170	1060	—	—	—	—				1700	1490	6740	7.0	—	—	—	—	50501
4	4	—	—	—	—				98	—	246	7.0	—	—	—	—	FRH
12	4	—	2.2	—	—				66	—	244	7.9	—	—	—	—	FRH
ravine, east of Alamosa River at Ojo Caliente																	
80	154.0	—	—	—	—				112	—	910	7.9	—	—	—	—	FRH
76.0	108.0	—	—	—	—				94	—	772	8.1	—	—	—	—	FRH

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts

gpm	°F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>3</sub>	Valencia
Source: Coyote Spring Location: 5N.3W.20.441																
Remarks: Issues from limestone at wide gap in hogback upstream from fault; precipitate covers estimated 30 acres of arroyo flow (Ti)																
3	64	1941	59	—	0.80	—	—	—	284	245	—	10000	—	2700	—	—
Source: Spring Location: 6N.2W.6.340																
Remarks: Issues from Magdalena Limestone; a Lucero fault zone spring (Wr)																
0.1	78	—	21	—	0.02	—	—	—	534	448	—	3650	38	1385	—	—
Source: Spring Location: 6N.3W.35.340																
Remarks: Issues from Magdalena Limestone; a Lucero fault zone spring (Wr); issues from limestone at gap in hogback upstream from																
—	71	1941	15	—	0.57	—	—	—	704	356	—	4570	79	2049	—	—
—	58	5-1-57	16	—	—	—	—	—	823	460	—	5830	—	2990	0	—
10	15.3	11-30-63	18	—	0.01	—	—	—	809	419	—	5730	99	2120	0	—
Source: Laguna Pueblo springs Location: 7N.2W.6.210 & .444																
Remarks: All issues from limestone in gap in hogback upstream from fault; precipitate covers arroyo floor (Ti)																
(.210)	—	1941	—	—	—	—	—	—	227	185	—	11400	—	2050	—	—
(.444)	68	4-30-57	25	—	—	—	—	—	681	314	—	7450	—	2440	—	—
Source: Springs Location: 7N.2W.6.400																
Remarks: Two springs with large travertine deposits issue from sandstone in fault zone																
0.5	80	1941	—	—	—	—	—	—	312	133	—	9460	—	2100	—	—
3	58	2-20-56	—	—	—	—	—	—	92	126	—	9670	—	2440	—	—
Source: Springs Location: 7N.2W.7.100 and .340																
Remarks: Issue from sandstone and shale along fault zone (Wr)																
(.100)	0.5	76	—	37	—	0.08	—	—	108	138	—	9760	283	1713	—	—
(.340)	0.05	—	—	—	—	—	—	—	324	152	—	—	—	2214	—	—
Source: Springs Location: 7N.2W.7.124 and .320																
Remarks: Issue from shale and sandstone at fault zone (Ti)																
(.320)	0.1	—	1941	—	—	—	—	—	324	152	—	9250	—	2210	—	—
(.124)	3	76	8-25-41	37	—	0.08	—	—	108	138	—	10000	—	1710	—	—
Source: Laguna Pueblo Seeps Location: 7N.2W.18.140, .312, and .313																
Remarks: Issue from sandstone and shale in fault zone (Ti)																
(.313)	0.02	82	1941	—	—	—	—	—	560	188	—	—	—	2030	—	—
(.312)	0.05	—	1941	—	—	—	—	—	560	188	—	—	—	2030	—	—
(.110)	0.2	—	1941	—	—	—	—	—	418	187	—	10700	—	1610	—	—
Source: Unnamed springs and seeps, Laguna Pueblo Location: 7N.2W.30																
Remarks: (see Wr and Ti for exact locations) A series of springs along the Lucero fault zone issuing from limestone, sandstone, and																
0.05	86	—	—	—	—	—	—	—	297	224	—	—	—	1374	—	—
0.35	75	1941	32	—	15	—	—	—	702	214	—	6470	165	2174	—	—
0.02	82	—	31	—	1.8	—	—	—	918	220	—	10950	283	2840	—	—
5	72	1941	20	—	0.09	—	—	—	516	163	—	6820	—	1340	—	—
Source: Spring Location: 7N.2W.31.140																
Remarks: Issues from sandstone in Abo Formation along fault zone (Wr; see also H&K)																
0.05	80	1941	20	—	0.18	—	—	—	603	271	—	5230	118	1615	—	—
Source: Spring Location: 7N.4W.3.430																
Remarks: Issues from shale and sandstone (Wr)																
0.01	65	—	—	—	—	—	—	—	606	150	—	—	—	408	—	—
Source: Spring Location: 7N.5W.20.340																
Remarks: Issues from sandstone and shale (Wr)																
3	68	—	—	—	—	—	—	—	604	130	—	—	—	428	—	—
Source: Ojo Escondido Spring Location: 8N.2W.19.421																
Remarks: Issues from sandstone along fault zone (Ti)																
20	73	9-8-41	12	—	0.02	—	—	—	33	20	—	23	5.6	220	—	—

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
per million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (μmhos at 25°C)	pH	H <sub>2</sub> S	Total βγ activity (μμc/l)	Ra (μμc/l)	U (μμc/l)	Ref.	
County																		
1510	12500	1.6	—	—	—	29500			1720	0	—	—	—	—	—	—	—	Ti
2640	5160	1.0	—	—	—	13540			3170	—	—	—	—	—	—	—	—	Wr
fault; water cascades through dams built by precipitates (Ti)																		
2660	6210	—	—	—	—	15630			3220	1540	—	—	—	—	—	—	—	Wr-Ti
3250	7900	1.8	—	—	—	19790			3940	1860	26700	6.5	—	—	—	—	—	Ti
3210	7790	1.1	1.3	—	3.2	19300			3740	1760	26800	6.4	—	—	—	—	—	53068
7800	11600	—	—	—	—	32400			1330	—	—	—	—	—	—	—	—	Ti
3490	9610	—	—	—	—	22700			2990	990	31000	7	—	—	—	—	—	Ti
6380	9600	—	—	—	—	—	27100		1330	—	—	—	—	—	—	—	—	Ti
6300	9400	—	—	—	—	—	—		748	—	35200	7.7	—	—	—	—	—	Ti
6710	9940	5.8	—	—	—	29500			1716	—	—	—	—	—	—	—	—	Wr
6670	9070	—	—	—	—	26700			—	—	—	—	—	—	—	—	—	Wr
8870	9070	—	—	—	—	26700			1430	—	—	—	—	—	—	—	—	Ti
6710	9940	5.8	—	—	—	27900			837	—	—	—	—	—	—	—	—	Ti
390	9210	—	—	—	—	27800			2180	521	—	—	—	—	—	—	—	Ti
390	9210	—	—	—	—	27800			2180	521	—	—	—	—	—	—	—	Ti
3040	10200	—	—	—	—	30000			1810	492	—	—	—	—	—	—	—	Ti
gypsum (Ti)																		
6710	8880	—	—	—	—	25700			1660	540	—	—	—	—	—	—	—	Wr-Ti
6710	6560	2.8	—	—	—	20920			2630	853	—	—	—	—	—	—	—	Wr-Ti
6710	11120	3.4	—	—	—	33900			3260	—	—	—	—	—	—	—	—	Wr-Ti
6740	8170	4.3	—	—	—	20900			1960	860	—	—	—	—	—	—	—	Wr-Ti
380	5120	1.2	—	—	—	17540			2620	—	—	—	—	—	—	—	—	Wr
369	146	—	—	—	—	3440			—	—	—	—	—	—	—	—	—	Wr
610	113	—	—	—	—	3500			—	—	—	—	—	—	—	—	—	Wr
32	5.6	0.7	0.1	—	—	239			164	—	—	—	—	—	—	—	—	Ti, Wr

TABLE 3. INFORMATION ABOUT THERMAL SPRINGS  
(All constituents in parts

gpm	*F	Date	SiO <sub>2</sub>	Al	Fe	Mn	Cu	Zn	Ca	Mg	Ba	Na	K	HCO <sub>3</sub>	CO <sub>2</sub>	Valencia
Source: Spring Location: 8N.2W.30.340																
Remarks: Issues from fault zone (Wr)																
0.03	73	—	—	—	—	—	—	—	227	185	—	—	—	2017	—	
0.3	72	9-3-41	20	—	0.09	—	—	—	516	163	—	6630	194	1340	—	
Source: Suwanee Spring (Laguna Pueblo) Location: 8N.3W.10.224																
Remarks: Issues from alluvium and lava-filled shallow valley (Ti)																
30	62	4-2-58	29	—	—	—	—	—	258	90	—	—	558	224	—	
Source: Dipping Vat Spring Location: 8N.3W.12.342																
Remarks: Issues from sandstone (Ti)																
400	60	4-2-58	30	—	0.26	—	—	—	270	109	—	—	609	222	—	
Source: Spring near Mesa Redondo Location: 8N.3W.15.413																
25	68.2	11-29-63	11	—	0.05	—	—	—	676	174	—	3500	116	2370	0	
Source: Springs Location: 8N.3W.35.100																
Remarks: Issue from shale, sandstone, and basalt slump blocks (Ti)																
1	65	—	28	—	0.09	—	—	—	516	163	—	6630	194	1345	—	
1	65	9-3-41	28	—	0.02	—	—	—	65	18	—	43	3.9	377	0	
Source: Ojo Caliente Spring Location: 8N.20W.21.140																
—	80	12-20-33	—	—	—	—	—	—	145	44	—	—	50	342	0	
450	71	11-25-57	19	—	—	—	—	—	143	37	—	—	66	344	0	
—	—	10-19-61	16	—	0.01	—	—	—	224	21	—	3.3	—	119	0	
Source: Spring Location: 10N.3W.26.330 (?)																
Remarks: From uncontaminated Todiito gypsum (?)																
1/16	72	6-30-58	18	—	—	—	—	—	174	15	—	—	9.4	599	0	

12345=U.S. Geol. Surv. laboratory number; analyses probably previously unpublished  
 B=unpublished data in the files of the N. Mex. Inst. Min. and Tech., State Bur. Mines and Mineral Res.  
 Bu=Bushman, 1955  
 C=Conover, 1954  
 E=Elston, 1957  
 FRH=unpublished data from the files of Francis R. Hall (or Hall, 1963)  
 He=Hem, 1959  
 H&K=Hood and Kister, 1962

AND WELLS INCLUDING CHEMICAL ANALYSES (cont)  
per million unless noted)

SO <sub>4</sub>	Cl	F	NO <sub>3</sub>	PO <sub>4</sub>	B	Total solids	Total solids (sum)	Total solids (evap)	Hardness as CaCO <sub>3</sub>	Noncarb. hardness as CaCO <sub>3</sub>	Sp. Cond. (μmhos at 25°C)	pH	H <sub>2</sub> S	Total β <sup>-γ</sup> activity (μμc/l)	Ra (μμc/l)	U (μg/l)	Ref.	
County (Continued)																		
7800	11560	—	—	—	—	32400			—	—	—	—	—	—	—	—	—	Wr
6540	6170	4.3	—	—	—	20900			1960	—	—	—	—	—	—	—	—	H&K
1530	344	—	3.6	—	—	3020			1090	807	3790	7.7	—	—	—	—	—	Ti
1640	384	0.7	4.7	—	0.85	3270			1120	940	4030	7.7	—	—	—	—	—	Ti
4220	2740	2.3	0.7	—	6.3		12600		2400	458	15800	7.3	—	—	—	—	—	53069
6540	6170	4.3	—	—	—	20900			1958	—	—	—	—	—	—	—	—	Wr
13	3.1	0.2	0.83	—	—	355	361		236	—	—	—	—	—	—	—	—	Ti
310	34	0.0	0.10	—	—		752		543	—	—	—	—	—	—	—	—	12573
303	37	0.7	0.0	—	—	809	775		509	227	1120	7.1	—	—	—	—	—	38875
463	54	0.5	11	—	0.23		1120		644	—	1560	7.2	—	—	—	—	—	48325
13	5.5	0.6	2.3	—	—		533		496	4	906	7.7	—	—	—	—	—	40498

Jo= Jones, 1904  
 K= Kelley and Silver, 1952  
 L= Meinzer and Hall, 1915  
 NMPHL= N. Mex. Pub. Health Lab.  
 P= Peck, 1886  
 R= Reeder, 1957  
 Ri= Renick, 1931

S= Stearns et al., 1937  
 Sc= Personal communication from the Work Unit Conservationists, U.S. Soil Conserv. Serv.  
 Sd= Schofield, 1938  
 Sp= Spiegel, 1957  
 S&B= Scott and Barker, 1962  
 Ti= Titus, 1963  
 Wr= Wright, 1946  
 WH&W= White et al., 1963



TABLE 2. INFORMATION ABOUT THERMAL SPRINGS AND WELLS FOR WHICH NO CHEMICAL ANALYSES ARE AVAILABLE

COUNTY	SOURCE	LOCATION	TEMP. (°F)	DISCHARGE (GPM)	DISCHARGE FROM (AQUIFER)	DEPTH (FEET)	REMARKS (SEE FOOTNOTES AT END OF TABLE)
Bernalillo	well	10N.2W.21.343	90	—	—	1180	L
	spring	11N.2W.32	68	3	Mancos Shale	—	S
	spring	11S.12W.30.100	80	50	lava agglomerate	—	S
	spring	11S.14W.25	160	900	—	—	T
Catron	spring	12S.13W.7	—	—	—	—	F
	spring	12S.13W.11	131	30	lava	—	S
	spring	12S.13W.14	100	—	—	—	P
	spring	12S.13W.24	—	—	—	—	F
	spring	19S.2W.9.120	120	100	—	110	Sc
	well	23S.1E.7	—	—	—	—	P
Dona Ana	Agua Caliente	27S.1W.8	100	—	—	100	LRe
	Kilbourne Hole	12S.20W.26	—	—	—	—	L
Grant	spring	13S.13W.20	—	30	lava	—	S
	spring	14S.16W.3	—	20	lava	—	S
	spring	14S.16W.16	—	20	lava	—	S
	hot spring area	15S.17W.9&10	—	—	—	—	U
	flowing well	15S.17W.29.442	68.5	2	—	410	U; cased to 40 ft.
	warm spring	16S.12W.22	150	—	—	—	P
	well	16S.17W.9.223	85	—	—	22	L
	well	16S.17W.10.433	90	—	—	—	U
	well	16S.17W.19.213	67	—	—	180	U
	spring	17S.17W.34	—	30	Gila Conglomerate	—	Sc
Guadalupe	Hudson's Hot Spring	18S.10W.4.100	142	—	lava	—	S
	Apache Tejo Warm Spring	19S.12W.19.300	89.97	2000	—	—	SPPa
	warm spring	20S.11W.18.300	—	dry	near rhyolite plug	—	PaU
	well	20S.19W.15.400	—	—	—	—	L
	Fuller's Ranch well	20S.19W.15.400	81	—	—	361	U
	Rock Lake	8N.21E.14	65	2700	—	—	T
	well	20S.19W.19.321	81	—	—	361	R
	well	21S.18W.18.180	83	—	—	—	R
	well	21S.20W.1.410	82	—	—	—	R
	well	22S.19W.23.130	85	—	—	—	R
Hidalgo	well	24S.18W.32	—	—	—	—	well used to heat greenhouse
	well	25S.19W.7.143	98	—	—	—	R
	well	25S.19W.7.134	84.5	—	—	74	R
	well	30S.19W.7	—	—	—	—	Sc; hot stock well
	spring	30S.19W.7	—	—	—	—	Sc; thermal pipe by road
	Humble Oil and Refining Co. No. 1	32S.16W.25	—	—	—	14588	BH 320°F
	spring	16N.18W.35	—	—	—	—	P
	Togay Spring	19N.15W.33	65	20	—	—	S
	Spense Hot Springs	14N.3E.28	110	100	—	—	F
	spring	18N.1E.24	—	—	—	—	T
McKinley	San Antonio Spring	20N.3E.29	130	150	—	—	SBaSc
	Westates Petroleum No. 1	20N.3E.35	—	—	—	3675	BH 160+°F
	spring	24N.18W.3	—	—	—	—	PJo
	spring	25N.18W.34	65	3	—	—	S
Sandoval	spring	25N.18W.	67	7	—	—	S

TABLE 2. INFORMATION ABOUT THERMAL SPRINGS AND WELLS FOR WHICH NO CHEMICAL ANALYSES ARE AVAILABLE (cont)

COUNTY	SOURCE	LOCATION	TEMP. (°F)	DISCHARGE (GPM)	DISCHARGE FROM (AQUIFER)	DEPTH (FEET)	REMARKS (see FOOTNOTES AT END OF TABLE)
San Miguel Sierra	spring well	26N.18W.10	68	3	—	—	SCPJo
	Victoria Land and Cattle Co. No. 2	16N.16E.5.211	—	—	granite	168	—
	Victoria Land and Cattle Co. No. 1	10S.1W.25	—	—	—	6039	BH 122°F
	16 wells	10S.1W.27	—	—	Magdalena Limestone	6332	BH 170°F
	spring	13S.4W.33	98.6-112.5	—	—	20-250	Th; Cl = 1320-1440
	Ojo Caliente	13S.4W.33	—	—	—	—	Th
	8 wells	14S.1W.28	—	—	Magdalena Limestone	107-225	P
	springs	14S.4W.4	107-116	—	—	—	Th; Cl = 1370-1400
	Cabello Springs	14S.5W.12.4	94-110.8	—	—	—	Th; Cl = 1340-1380
	Barney Iorio No. 1 fee	14S.5W.25	136	—	—	—	P
Socorro	Sunray-Midcontinent No. 1	15S.2W.23	90	30	—	1530	K
	Ojo Caliente	8S.7W.30	89	1350	—	9774	BH 236°F
	Grater well	9S.1E.18.320	—	—	basalt	—	SPSC
	Warm Sulphur Springs	—	68	—	—	—	VW
	Glen-Woody Camp Springs	24N.11E.28	—	—	—	—	P
	Ponce de Leon Hot Spring	24N.13E.7	98	100	—	—	F
	Mamby's Hot Spring	26N.11E.1	100	—	—	—	Sc
	Warmisley Hot Spring	27N.12E.31	—	—	—	—	HSc
	Arsenic Springs	28N.12E.8.100	—	—	—	—	Jo
	springs	29N.12E.12	—	—	—	—	FWi
Valencia	warm spring	5N.1W.16	64-82	—	Rio Puerto fault	—	F
	spring	7N.2W.8	65	3	—	—	S
	spring	7N.2W.16	67	7	sandstone and shale intruded by porphyry	—	S
	Quelities Mineral Spring	8N.2W.17	80	3	—	—	S

B = open-file data of the N. Mex. Inst. Min. and Tech., State Bur. Mines and Mineral Res.

Ba = Bailey, 1961

BH = bottom-hole temperature

F = personal communications April 1965 from District Rangers, U.S. Forest Service

G = Gilbert, 1875

H = Herron, 1915, plate IV

J = Jones Springs quadrangle, 1948

Jo = Jones, 1904

K = Kelley and Silver, 1956

L = Lee, 1907b

P = Peale, 1886

Pa = Paige, 1916

R = Reeder, 1957

Re = Reiche, 1940

S = Stearns, 1937

Sc = Personal communications May 1965 from the Work Unit Conservatoinists, U.S. Soil Conserva. Serv.

T = Personal communication April 1965 from Fred A. Thompson, N. Mex. Dept. Game and Fish

Th = Theis et al., 1941

U = Unpub. data in the files of the Ground-Water Branch, U.S. Geol. Surv.

V = Valverde quadrangle, 1958

W = Weber, 1963

WI = Winograd, 1959

## OREGON

### Summary

A definitive study of water quality and predicted thermal reservoir temperatures for 32 thermal springs was completed by the U.S.G.S. in 1974 (Mariner, et.al.). Chemical composition of the spring waters suggests aquifer temperatures which range from 43°C to 181°C. Four springs have aquifer temperatures exceeding 150°C, as estimated by both silica and Na-K or Na-K-Ca geothermometers. Five other springs have estimated aquifer temperatures of 140°C-150°C. Chemical compositions of the springs are indicative of hot water-dominated geothermal systems.

Sodium is the principal cation in all of the spring waters. The pH's range from 6.53 to 9.68. Boron content ranges from 0.11 to 30 ppm. Only five springs have a boron content of 1.0 ppm or less, and three springs have 1.0-2.0 ppm boron conductances for the springs with estimated aquifer temperatures of at least 150°C range from 1,010 to 4,590.

The areas which have thermal springs are underlain principally by Cenozoic basalts, andesites and related tuffs, and sedimentary rocks. Almost all of the springs issue along faults and are generally located at or near the edge of an alluvium-filled valley.

Seven KGRA's have been designated in the state.

## Generalized Geologic Setting

Areas of Oregon which have thermal springs are underlain principally by Cenozoic volcanic rocks (basalts or andesites), as well as related tuffs and sedimentary rocks. Late Cenozoic basalt and andesite flows crop out in the Cascade Range. Southeastern Oregon, part of the Basin and Range Province, consists chiefly of flows of basalt and andesite with some rhyolitic flows and domes interbedded with ash-flow tuffs and continental sedimentary rocks. The principal valleys all contain Quarternary alluvium and lake sediments, usually with some fluvial glacial deposits and pumice. Rocks exposed in the northeastern part of the state are largely Cenozoic basalts and andesites with some interbedded sedimentary rocks. A window of Mesozoic and Paleozoic rock crops out in Grant County. These pre-Cenozoic rocks include gabbro and granite in addition to sedimentary and volcanic rocks. Some of the sedimentary and volcanic rocks have been strongly metamorphosed. A similar window of pre-Cenozoic rocks is exposed in southwestern Oregon (Corcoran, 1969).

Rock types which crop out near the spring or may be present at depth are listed in Table 4. Recent papers on the geology around the various springs are listed for reference in Table 4. Almost all of the springs issue along faults. Generally, the springs are located at or near the edge of an alluvium-filled valley.

## Tabular Data

Table 1 contains the name, location and topographic map coverage of the 32 thermal springs and wells reported on by the U.S. Geological Survey (Mariner et.al.).

Table 2 contains the results of water analysis reported for the major chemical constituents.

Table 3 contains a description of the rock type and estimated thermal aquifer temperatures of each hot spring. The underlined values represent the "best" estimates of thermal aquifer temperature.

Table 4 lists rock types and references to recent papers on the geology around each hot spring.

## Water Quality

All of the hot springs and wells show sodium as the principal cation. Chloride is the principal anion of all four hot springs in the Cascade Range. Bicarbonate is the principal anion at Weberg, Blue Mountain, Ritter Crane, and the two unnamed hot springs in southern Malheur County. Sulfate is the principal anion at Olene Gap and Medical Hot Springs. The other springs are mixed anion waters with varying proportions of bicarbonate, sulfate and chloride. The thermal waters from the Cascade Range and southern Harney County are the most saline. Alvord Hot Spring in southern Harney County and Belknap Hot Spring in the Cascade Range are very saline, with total dissolved solids of 2,980 and 3,020 mg/l (milligrams per liter), respectively. The pH's range from 6.53 at Weberg Hot Spring to 9.68 at Ritter Hot Spring; both springs are in Grant County. Lithium, boron and, to a lesser extent, fluoride concentrations are high in thermal waters from southern Harney County. Kah-Ne-Tah Hot Spring in Wasco County is very high in fluoride, 20.5 mg/l.

## Summary

Nine of the sampled thermal springs have chemical compositions which indicate aquifer temperatures of at least 140°C. Possible contamination with meteoric water of low specific conductance is not considered, and therefore the estimated thermal aquifer temperatures are minimum values. The chemical compositions of the springs are typical for springs associated with hot water-dominated systems. Specific conductances for the springs with estimated aquifer temperatures of at least 140°C are high to very high. Areas underlain by thermal aquifers of 140°C or more include: the Alvord area of southern Harney County, the Lakeview area of southern Lake County, and the Vale area of northern Malheur County. The Lakeview and Vale areas have been designated as KGRA's. Other KGRA's include: Breitenbush Hot Spring, Crump Geyser, Mt. Hood, Carey Hot Spring (Austin Hot Spring), and Klamath Falls.

Thermal springs are spread eastward from the west edge of the Cascade Mountains. Springs are concentrated along the linear belt that marks the boundary between the High Cascades of Pliocene to Holocene age and the western Cascades of early Tertiary age. Hot springs are scarce in the high Cascades, but there is some fumarolic activity near the summit of Mt. Hood. The southern and southeastern quadrants of the state lie within the Basin and Range Province where there are several thermal areas associated with many of the grabens. Greater than 40% of Oregon, mainly the southeastern lava plains and the Cascade Range, is underlain by Late Cenozoic volcano-tectonic and caldera collapse structures. Oregon has the youngest volcanism in the continental U.S.

Best known areas:

Klamath Falls  
Temperature 185°F (85°C)

Lakeview  
On the southern edge of the Goose Lake-Summer Lake graben complex

Warner Valley

Alvord Valley

Vale - a geological extension of the Bruneau-Oreana-Grandview geothermal trend  
Average temperature of the major areas is 85°F (29°C)  
Temperatures range up to 198°F (92°C)  
Flows are up to 5,000 gpm

Blue Mountain Region - Walla Walla, WA, to Milton Freewater, Oregon  
High temperature found in water wells  
Thermal Spring  
Wallowa-Olympic Lineament (major structural discontinuity)

Table 1.--Location and topographic map coverage of selected hot springs

Spring	Location	Topographic map coverage
1 Radium Hot Springs(well)	Baker County NE 1/4 sec. 28, T. 7S., R. 39E.	Haines, Ore. (7-1/2'); Baker, Oregon-Idaho (2°)
1 Austin Hot Springs	Clackamas County NW 1/4 sec. 30, T. 6S., R. 7E.	Fish Creek Mtn. Ore., (15'); Vancouver, Ore.-Wash. (2°)
1 Weberg Hot Spring	Grant County sec. 18, T. 18S., R. 26E.	; Burns, Oregon (2°)
2 Blue Mountain Hot Springs	S 1/2 sec. 13, T. 14S., R. 34 E.	Prairie City, Ore. (15'); Canyon City, Oregon (2°)
3 Ritter Hot Springs	NW 1/4 sec. 8, T. 8S., R. 30E.	Ritter, Ore. (15'); Canyon City, Oregon (2°)
1 Unnamed hot spring (Trout Creek)	Harney County sec. 16, T. 39S., R. 37E.	; Adel, Oregon (2°)
2 Hot Lake	sec. 15, T. 37S., R. 33E.	; Adel, Oregon (2°)
3 Unnamed hot spring (near Hot Lake)	sec. 15, T. 37S., R. 33E.	; Adel, Oregon (2°)
4 Alvord Spring (Indian Spring)	sec. 33, T. 34S., R. 34E.	; Adel, Oregon (2°)
5 Mickey Springs	sec. 13, T. 33S., R. 35E.	; Adel, Oregon (2°)
6 Unnamed hot spring (near Harney Lake)	sec. 36, T. 27S., R. 29-1/2E.	; Burns, Oregon (2°)
7 Crane Hot Springs	S 1/2 sec. 34, T. 24S., R. 33E.	Crane, Ore. (15'); Burns, Oregon (2°)
1 Olene Gap Hot Springs	Klamath County SW 1/4 sec. 14, T. 39S., R. 10E.	Merrill, Ore.-Calif. (15'); Klamath Falls, Ore.-Calif. (2°)
1 Fisher Hot Springs	Lake County NW 1/4 sec. 10, T. 38S., R. 25E.	Crump Lake, Ore. (7-1/2'); Adel, Oregon (2°)
2 Crump (Charles Crump's Spring)	sec. 27, T. 38S., R. 24E.	; Adel, Oregon (2°)
3 Barry Ranch Hot Springs	SE 1/4 sec. 27, T. 39S., R. 20E.	Lakeview NE, Ore. (7-1/2'); Klamath Falls, Ore.-Calif. (2°)
4 Hunters Hot Springs	NW 1/4 sec. 4, T. 39S., R. 20E.	Lakeview NE, Ore. (7-1/2'); Klamath Falls, Ore.-Calif. (2°)
5 Summer Lake Hot Spring	NE 1/4 sec. 12, T. 33S., R. 17E.	Slide Mtn. Ore. (7-1/2'); Klamath Falls, Ore.-Calif. (2°)

Tables extracted from: Mariner, et al., USGS (1974)



Table 1.--Location and topographic map coverage of selected hot springs--Continued

Spring	Location	Topographic map coverage
1 Belknap Hot Springs	Lane County NE 1/4 sec. 11, T. 16S., R. 6E.	McKenzie Bridge, Ore. (15'); Salem, Oregon (2')
2 Cougar Reservoir Hot Spring	sec. 7, T. 17S., R. 5E.	McKenzie Bridge, Ore. (15'); Salem, Oregon (2')
1 Unnamed hot springs (near McDermitt)	Malheur County sec. 25, T. 40S., R. 42E.	; Jordan Valley, Ore.-Idaho (2')
2 Unnamed hot springs (at Three Forks)	sec. 3, T. 35S., R. 45E.	; Jordan Valley, Ore.-Idaho (2')
3 Unnamed hot spring (near Riverside)	sec. 20, T. 24S., R. 37E.	; Burns, Oregon (2')
4 Beulah Hot Springs	SE 1/4 sec. 2, T. 19S., R. 37E.	Beulah, Oregon (15'); Burns, Oregon (2')
5 Neal Hot Springs	NW 1/4 sec. 9, T. 18S., R. 43E.	Jamleson, Oregon (15'); Baker, Idaho-Oregon (2')
6 Unnamed hot springs (near Little Valley)	NW 1/4 sec. 30, T. 19S., R. 43E.	Harpert, Oregon (15'); Boise, Idaho-Oregon (2')
7 Mitchell Butte Hot Spring	NE 1/4 sec. 12, T. 21S., R. 45E.	Mitchell Butte, Ore. (7-1/2'); Boise, Idaho-Oregon (2')
1 Breitenbush Hot Springs	Marion County NE 1/4 sec. 20, T. 9S., R. 7E.	Breitenbush Hot Spring, Ore. (15'); Canyon City, Ore. (2')
1 Lehman Springs	Umatilla County NE 1/4 sec. 12, T. 5S., R. 33E.	Lehman Springs, Ore. (7-1/2'); Pendleton, Ore.-Wash. (2')
1 Medical Hot Springs	Union County NE 1/4 sec. 25, T. 6S., R. 41E.	Flagstaff Butte, Ore. (7-1/2'); Grangeville, Idaho-Ore.-Wash. (2')
2 Hot Lake	SE 1/4 sec. 5, T. 4S., R. 39E.	Craig Mtn., Ore. (7-1/2'); Grangeville, Idaho-Ore.-Wash. (2')
1 Kahneeta Hot Springs (Kah-Ne-Tah)	Wasco County E 1/2 sec. 20, T. 8S., R. 13E.	Eagle Butte, Ore. (7-1/2'); Bend, Oregon (2')

Table 2.--Chemical analyses of selected hot springs

Spring	Temperature (°C)	pH	Specific conductance	Silica (SiO <sub>2</sub> )	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Lithium (Li)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Boron (B)
Baker County															
1 Radium Hot Springs (well)	58	9.56	290	78	1.5	0.1	58	1.1	0.01	86	27	34	17	1.3	0.42
Clackamas County															
1 Austin Hot Springs	86	7.63	1,720	81	35	.1	300	7.1	.4	56	<1	140	430	1.4	2.5
Grant County															
1 Weberg Hot Springs	46	6.53	2,570	82	38	7.8	610	36	.7	1,710	1	13	50	3.9	15
2 Blue Mountain Hot Springs	58	7.96	610	47	2.2	.2	140	3.3	.07	323	3	11	15	10.6	1.6
3 Ritter Hot Springs	41	9.68	319	70	1.4	<.05	72	.82	.01	86	28	9	29	4.0	2.6
Harney County															
1 Unnamed hot spring (Trout Creek)	52	6.77	1,168	105	18	.8	270	10.8	.68	439	1	204	24	12.8	.89
2 Hot Lake	36	7.28	2,410	190	16	.3	500	31	.65	420	1	350	300	9.0	16.6
3 Unnamed hot spring (near Hot Lake)	96	7.30	2,020	160	14	.3	450	28	.51	374	4	434	250	7.2	15
4 Alvord Spring (Indian Spring)	76	6.73	4,590	120	13	2.2	960	69	2.1	1,196	1	220	780	10.2	30
5 Mickey Springs	73	8.05	2,490	200	.9	.1	550	35	1.1	774	11	230	240	16	10.5
6 Unnamed hot spring (near Harney Lake)	68	7.26	2,970	92	12	1.8	630	13	.45	566	1	140	590	3.3	11.3
7 Crane Hot Springs	78	8.10	810	83	3.7	.1	170	3.9	.09	202	3	86	79	9.0	7.9
Klamath County															
1 Olene Gap Hot Springs	74	7.68	1,140	98	40	.2	190	7.2	.15	53	<1	400	59	1.2	1.0
Lake County															
1 Fisher Hot Springs	68	7.93	513	77	8.4	1.0	92	7.9	.04	105	1	59	56	3.5	2.2
2 Crump (Charles Crump's Spring)	78	7.26	1,490	180	16	.2	280	11	.4	153	1	200	240	4.9	13.6
3 Barry Ranch Hot Springs	88	7.76	1,370	130	-8.8	.1	280	9.0	.15	232	2	240	170	5.4	11.2
4 Hunters Hot Springs	96	7.77	1,120	140	13	.1	210	8.5	.15	79	1	260	120	4.4	6.9
5 Summer Lake Hot Spring	43	8.43	1,790	94	2.1	.1	390	4.6	.15	406	10	120	280	2.2	6.9

Table 2.--Chemical analyses of selected hot springs--Continued

Spring	Temperature (°C)	pH	Specific conductance	Silica (SiO <sub>2</sub> )	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Lithium (Li)	Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Boron (B)
Lane County															
1 Belknap Hot Springs	71	7.62	4,300	96	210	.2	690	15	.95	17	<1	170	1,300	1.2	6.4
2 Cougar Reservoir Hot Spring	44	7.76	2,890	50	225	.1	392	6.3	.52	19	<1	260	788	.8	5.1
Malheur County															
1 Unnamed hot springs (near McDermit)	52	8.79	598	72	.6	<.1	130	1.0	.06	237	13	52	14	6.6	1.1
2 Unnamed hot springs (at Three Forks)	34	8.11	338	40	10.5	.7	61	1.2	.04	108	1	34	18	4.2	.11
3 Unnamed hot springs (near Riverside)	63	7.43	1,330	110	34	.5	240	9.7	.27	160	<1	290	140	4.8	6.6
4 Beulah Hot Springs	60	7.56	1,090	170	24	.2	200	6.0	.24	161	1	290	55	4.7	4.7
5 Neal Hot Springs	87	7.32	1,010	180	8.8	.2	190	16	.3	198	<1	120	120	9.4	4.1
6 Unnamed hot springs (near Little Valley)	70	8.71	740	115	3.2	<.05	160	3.2	.11	127	1	110	74	6.8	4.7
7 Mitchell Butte Hot Spring	62	8.69	559	94	4.6	<.1	110	1.6	.03	72	3	130	28	10.4	.49
Marion County															
1 Breitenbush Hot Springs	92	7.31	4,030	83	100	1.3	720	31	1.8	152	<1	140	1,300	3.4	4.1
Umatilla County															
1 Lehman Springs	61	9.18	252	44	.9	.1	53	.7	.03	101	13	23	5.4	2.1	.12
Union County															
1 Medical Hot Springs	60	8.23	1,173	80	72	.2	190	7.0	.05	26	<1	400	77	1.2	2.2
2 Hot Lake	80	9.21	688	48	4.9	<.1	130	2.7	.03	75	12	56	140	1.7	2.9
Wasco County															
1 Kahneeta Hot Springs (Kah-Ne-Tah)	52	8.32	1,370	104	3.2	<.05	325	3.4	.52	493	9	34	155	21	2.6

Table 3.-Geologic setting and estimated thermal aquifer temperatures of selected hot springs--Continued

Spring	Spring temperature (°C)	Flow (lpm)	Gas	Spring deposits		Rock type at the spring	Estimated thermal aquifer temperature				
				CaCO <sub>3</sub>	Silica		Silica conductive	Silica adiabatic	Na-K	Na-K-1/3ca	Na-K-4/3ca
Lane County											
1 Belknap Hot Springs**	71	300	--	--	--	Olivine basalt	135	131	56	114	82
2 Cougar Reservoir Hot Spring**	44	200	--	--	--	Andesite, basalt, and basic tuff-breccia	89	92	38	95	49
Malheur County											
1 *Unnamed hot springs (near McDermitt)	52	750	T	--	--	Basalt	118	118	3	91	105
2 Unnamed hot springs (at Three Forks)**	34	4,000	--	--	--	Basalt	95	97	52	100	44
3 *Unnamed hot springs (near Riverside)	63	200	T	T	--	Andesite	142	137	98	138	97
4 Beulah Hot Springs	60	50	T	T	X	Vitric tuff	169	159	76	125	86
5 Neal Hot Springs	87	90	T	T	X	Basalt	173	162	164	181	151
6 *Unnamed hot springs (near Little Valley)***	70	550	T	T	T	Basalt and andesite	145	139	51	119	109
7 Mitchell Butte Hot Spring**	62	60	--	T	--	Volcanic arkose	133	130	33	100	72
Marion County											
1 Breitenbush Hot Springs	92	3,400	--	X	--	Andesite	127	124	103	149	128
Umatilla County											
1 Lehman Springs***	61	275	X	T	--	Andesite	98	98	28	97	73
Union County											
1 Medical Hot Springs**	60	200	X	T	--	Basalt	125	123	91	125	67
2 Hot Lake	80	1,500	T	T	--	Basalt and mylonite	102	103	53	115	90
Wasco County											
1 *Kahneeta Hot Springs (Kah-Ne-Tah)	52	200	T	T	--	Rhyolite, andesite, basalt, and tuffs	139	135	17	103	121

\* Mixed waters

\*\* Temperature estimates based on the solubility of cristobalite improve the agreement between the silica and cation geothermometers at Radium Hot Springs (72°C), Ritter Hot Springs (70°C), Olene Gap Hot Springs (84°C), Belknap Hot Springs (84°C), Cougar Reservoir Hot Springs (51°C), the unnamed hot springs near Three Forks (39°C), Mitchell Butte Hot Springs (83°C), and Medical Hot Springs (74°C). Solubility data from Fournier and Rowe (1962).

\*\*\* Temperature estimates based on the solubility of chalcedony improve the agreement between the silica and cation geothermometers at Austin Hot Springs (95°C), the unnamed hot springs near Little Valley (112°C), Summer Lake Hot Springs (107°C), and perhaps, Lehman Hot Springs (68°C). Solubility data from Fournier (1973).

Table 3.--Geologic setting and estimated thermal aquifer temperatures of selected hot springs

Spring	Spring Temperature (°C)	Flow (lpm)	Spring deposits			Rock type of the spring	Estimated thermal aquifer temperature				
			Gas	CaCO <sub>3</sub>	Silica		Silica conductive	Silica adlabatic	Na-K	Na-K-1/3Ca	Na-K-4/3Ca
<b>Baker County</b>											
1 Radium Hot Springs (well)**	58	1,100	X	--	--	Alluvium, quartzdiorite, and basalt	123	122	48	109	77
<b>Clackamas County</b>											
1 Austin Hot Springs ***	86	950	T	--	--	Olivine basalt, basaltic andesite and pyroxene andesite	124	123	61	118	88
<b>Grant County</b>											
1 Weberg Hot Spring	46	40	X	T	--	Arkosic sandstone	125	124	130	170	162
2 *Blue Mountain Hot Springs	58	250	--	T	--	Andesite	99	101	61	126	118
3 Ritter Hot Springs **	41	130	--	--	--	Basalt	118	117	20	92	71
<b>Harney County</b>											
1 *Unnamed hot spring (Trout Creek)	52	200	T	--	--	Andesite, basalt, and rhyolite	140	135	97	144	118
2 Hot Lake	36	3,500	T	--	X	Alluvium, andesite, and basalt	176	165	134	176	181
3 Unnamed hot spring (near Hot Lake)	96	15	X	T	X	Alluvium, andesite, and basalt	165	156	134	176	178
4 *Alvord (Indian) Hot Spring	76	500	T	T	X	Rhyodacite, andesite, and basalt	148	142	148	199	254
5 *Hickey Springs	73	100	X	T	X	Andesitic tuff-breccia	179	168	136	207	330
6 *Unnamed hot spring (near Harney Lake)	68	550	T	T	--	Basaltic tuff, and olivine basalt	132	129	52	130	151
7 Crane Hot Springs	78	550	T	--	--	Augite andesite	127	125	59	124	114
<b>Klamath County</b>											
1 Olene Gap Hot Springs **	74	200	T	--	--	Andesite, basalt, and andesitic tuff-breccia	136	132	93	130	80
<b>Lake County</b>											
1 Fisher Hot Springs	68	70	T	T	--	Alluvium and olivine basalt	127	121	167	170	112
2 Crump (Charles Crump's Spring)	78	0-50	T	T	X	Alluvium and olivine basalt	173	162	96	144	123
3 *Barry Ranch Hot Springs	88	200	X	T	T	Andesite, andesitic tuff-breccia, and rhyolite	153	146	81	140	131
4 Hunters Hot Springs	96	2,300	T	T	T	Alluvium, andesite, and andesitic tuff-breccia	157	149	98	143	114
5 *Summer Lake Hot Spring ***	43	75	T	T	--	Alluvium, andesite, andesitic tuff-breccia	134	130	22	112	149

Table 4.--Age of bedrock and geologic coverage of each spring

Spring	Age of bedrock	Geologic reference
1 Radium Hot Springs (well)	Baker County Quaternary alluvium, Late Cretaceous diorite, and Permian greenstone	Gilluly (1937)
1 Austin Hot Springs	Clackamas County Pliocene to Holocene mafic flows and perhaps pyroclastic rocks	Peck, Griggs, Schlicker, Wells, and Dole (1964)
1 Weberg Hot Spring	Grant County Lower and Middle Jurassic sandstone and volcanic rocks	Brown and Thayer (1966)
2 Blue Mountain Hot Springs	Miocene and Pliocene andesite flows	Brown and Thayer (1966)
3 Ritter Hot Springs	Miocene and Pliocene basalt flows	Brown and Thayer (1966)
1 Unnamed hot spring (Trout Creek)	Harney County Quaternary alluvium, Miocene to Pliocene basalt, andesite, and rhyolite flows	Walker and Repenning (1965)
2 Hot Lake	Quaternary alluvium and playa deposits	Walker and Repenning (1965)
3 Unnamed hot spring (near Hot Lake)	Quaternary alluvium and playa deposits	Walker and Repenning (1965)
4 Alvord (Indian) Spring	Miocene rhyodacite, basalt, and andesite	Walker and Repenning (1965)
5 Mickey Springs	Miocene andesitic tuff-breccia, basalts, and andesites	Walker and Repenning (1965)
6 Unnamed hot spring (near Harney Lake)	Pliocene basalts, tuffs, and welded tuffs	Walker and Swanson (1967); Greene, Walker and Corcoran (1972)
7 Crane Hot Springs	Quaternary alluvium, Pliocene and Pleistocene pyroclastic rocks, and Pliocene basalt and andesite	Leonard (1970); Greene, Walker, and Corcoran (1972)
1 Olene Gap Hot Springs	Klamath County Pliocene and Pleistocene basalts and associated pyroclastic rocks	Peterson and Groh (1967)
1 Fisher Hot Springs	Lake County Quaternary alluvium and Miocene to Pliocene olivine basalt	Walker and Repenning (1965)
2 Crump (Charles Crump's Spring)	Quaternary alluvium and Miocene to Pliocene olivine basalt	Walker and Repenning (1965); Peterson (1959)
3 Barry Ranch Hot Springs	Oligocene(?) and Miocene basalt or andesite flows, tuff-breccia, tuff, and tuffaceous rocks	Walker (1963)

Table 4.--Age of bedrock and geologic coverage of each spring--Cont'd

Spring	Age of bedrock	Geologic reference
4 Hunter Hot Springs	Quaternary alluvium, Quaternary to late Tertiary basalts and andesites, middle Tertiary tuffs	Walker (1963)
5 Summer Lake Hot Spring	Tertiary and Quaternary sedimentary rocks overlying Tertiary andesite flows	Walker (1963)
1 Belknap Hot Springs	Lane County Pliocene to Holocene basic volcanic flows and pyroclastic rocks	Peck, Griggs, Schlicker, Wells, and Dole (1964)
2 Cougar Reservoir Hot Spring	Miocene mafic to intermediate flows, tuffs, and tuff-breccias	Peck, Griggs, Schlicker, Wells, and Dole (1964)
1 Unnamed hot springs (near McDermit)	Malheur County Quaternary alluvium, Tertiary and Quaternary pediment gravels, and Miocene volcanic rocks	Walker and Repenning (1956)
2 Unnamed hot springs (at Three Forks)	Miocene and Pliocene volcanic flows and tuffs	Walker and Repenning (1966)
3 Unnamed hot springs (near Riversdale)	Miocene basalt	Walker and Repenning (1965)
4 Beulah Hot Springs	Miocene and Pliocene vitric tuff	Greene, Walker, and Corcoran (1972); Bowen (1956)
5 Neal Hot Springs	Miocene(?) volcanic flows	Walker (1973)
6 Unnamed hot springs (near Little Valley)	Pliocene basalt and sedimentary volcanic rocks	Corcoran, Doak, Forter, Pritchett, and Privratsky (1962)
7 Mitchell Butte Hot Spring	Pliocene conglomerate, sandstone, and siltstones	Corcoran, Doak, Forter, Pritchett, and Privratsky (1962)
1 Breitenbush Hot Springs	Marion County Miocene basalt flows, tuff-breccias, and tuffs, near an area of propylitically altered rock	Peck, Griggs, Schlicker, Wells, and Dole (1964)

# OREGON

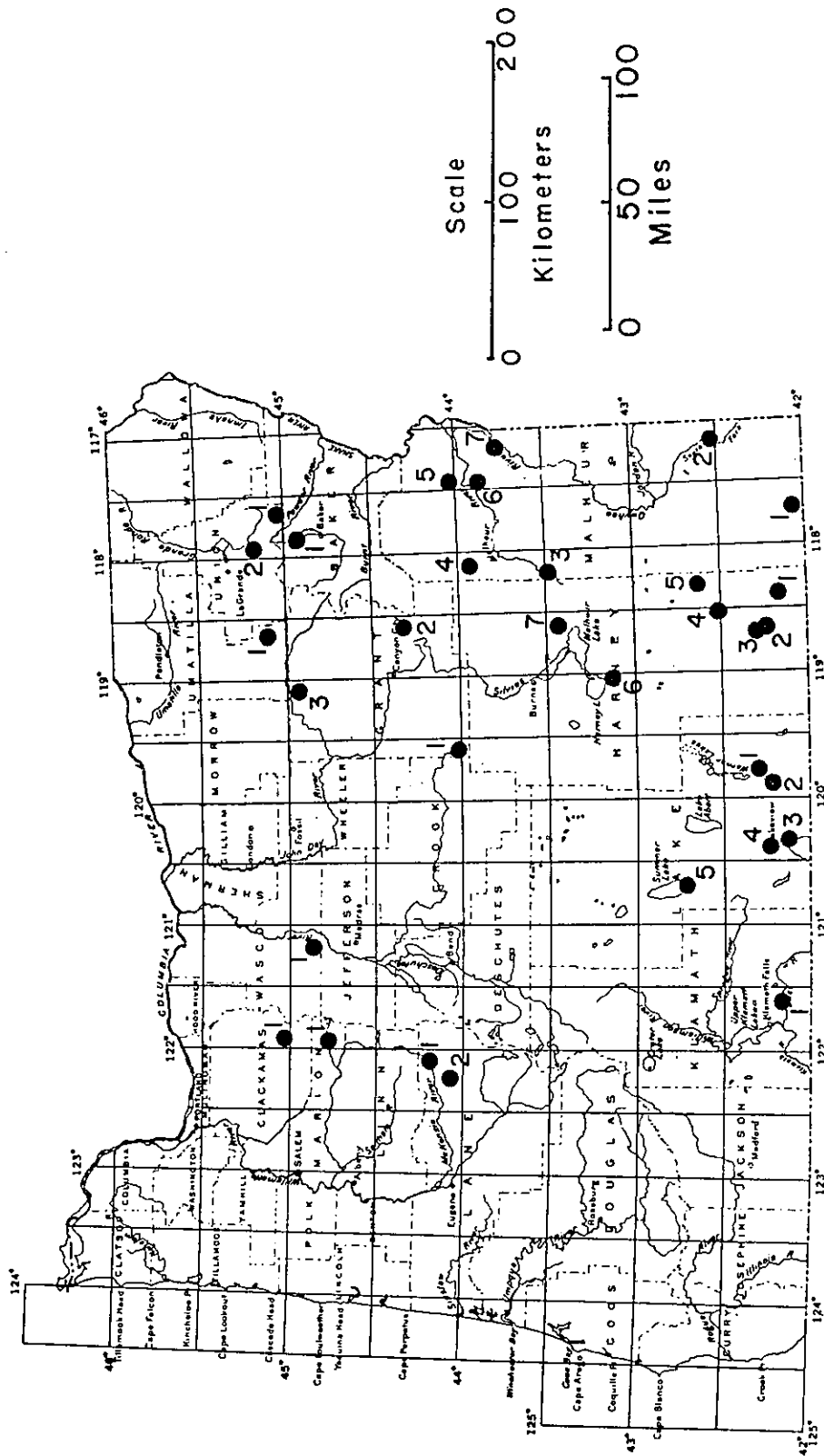


Figure 1. Map of the State of Oregon showing the location of sampled thermal springs and wells. The numbered dots correspond to sampled springs listed by county in tables 1, 2, 3, and 4 of the text.



#### REFERENCES CITED

- Barnes, Ivan, 1964, Field measurement of alkalinity and pH: U.S. Geol. Survey Water-Supply Paper 1535-H, 17 p.
- Bowen, R. G., 1956, Geology of the Beulah area, Malheur County, Oregon: Eugene, Oregon Univ., Master's thesis, 152 p.
- Bowen, R. G., and Peterson, N. V., 1970, Thermal springs and wells, in Oregon: Oregon Dept. of Geology and Mineral Industries, Misc. Paper 14.
- Brown, C. E., and Thayer, T. P., 1966, Geologic map of the Canyon City quadrangle, northeastern Oregon: U.S. Geol. Survey Misc. Geol. Inv. Map I-447.
- Corcoran, R. E., 1969, General geologic history of Oregon in Mineral and Water Resources of Oregon: Oregon Dept. of Geology and Mineral Industries Bull. 64, p. 23-32.
- Corcoran, R. E., Doak, R. A., Porter, P. W., Pritchett, F. I., Jr., and Privrasky, N. C., 1962, Geology of the Mitchell Butte quadrangle, Oregon: Oregon Dept. Geology and Mineral Industries, Geol. Map Series GMS 2.
- Ellis, A. J., 1970, Quantitative interpretation of chemical characteristics of hydrothermal systems in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, part 1: Geothermics Spec. Issue 2, p. 516-528.
- Fournier, R. O., 1973, Silica in thermal waters: laboratory and field investigations in Proceedings of the International Symposium on Hydrogeochemistry and Biogeochemistry, Tokyo, 1970, v. 1: Hydrogeochemistry, p. 122-139.

- Fournier, R. O., and Rowe, J. J., 1962, The solubility of cristobalite along the three-phase curve, gas plus liquid plus cristobalite: Am. Mineralogist, v. 47, p. 897-902.
- \_\_\_\_\_ 1966, Estimation of underground temperatures from the silica content of water from hot springs and wet steam wells: Am. Jour. Sci., v. 264, p. 685-697.
- Fournier, R. O., and Truesdell, A. H., 1970, Chemical indicators of subsurface temperature applied to hot waters of Yellowstone National Park, Wyo., U.S.A. in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, part 1: Geothermics Spec. Issue 2, p. 529-535.
- \_\_\_\_\_ 1973, An empirical Na-K-Ca geothermometer for natural waters: Geochem. Cosmochim. Acta, v. 37, p. 1255-1275.
- \_\_\_\_\_ 1974, Geochemical indicators of subsurface temperatures, Part II: Estimation of temperature and fraction of hot water mixed with cold water: U.S. Geol. Survey Jour. Research, v. 2, no. 3, May-June 1974 (in press).
- Fournier, R. O., White, D. E., and Truesdell, A. H., 1974, Geochemical indicators of subsurface temperature, Part I: Basic assumptions: U.S. Geol. Survey Jour. Research, v. 2, no. 3, May-June 1974 (in press).
- Gilluly, James, 1937, Geology and mineral resources of the Baker quadrangle, Oregon: U.S. Geol. Survey Bull. 879, 119 p.

- Godwin, L. H., Haigler, L. B., Rioux, R. L., White, D. E., Muffler, L. J. P., and Wayland, R. G., 1971, Classification of public lands valuable for geothermal steam and associated geothermal resources: U.S. Geol. Survey Circ. 647, 17 p.
- Greene, R. C., Walker, G. W., and Concoran, R. E., 1972, Reconnaissance geologic map of the Burnes quadrangle, Oregon: U.S. Geol. Survey Misc. Geol. Inv. Map I-680.
- Groh, E. A., 1966, Geothermal energy potential in Oregon: The Ore Bin, v. 28, no. 7. p. 125-136.
- Hampton, E. R., and Brown, S. G., 1964, Geology and ground water resources of the upper Grande Ronde River Basin, Union County, Oregon: U.S. Geol. Survey Water-Supply Paper 1597, 99 p.
- Hodge, E. T., 1941, Geology of the Madras quadrangle: Oregon State College, Studies in Geology, no. 1, geologic map.
- Leonard, A. R., 1970, Ground-water resources in Harney Valley, Oregon: Oregon State Ground Water Report no. 16, 85 p.
- Mahon, W. A. J., 1970, Chemistry in the exploitation of hydrothermal systems in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 2, part 2: Geothermics Spec. Issue 2, p. 1310-1322.
- Peck, D. L., Griggs, A. B., Schlicker, H. G., Wells, F. G., and Dole, H. M., 1964, Geology of the central and northern parts of the Western Cascade Range in Oregon: U.S. Geol. Survey Prof. Paper 449, 56 p.

- Peterson, N. V., 1959, Lake County's new continuous geyser: The Ore Bin, v. 21, no. 9, p. 83-88.
- Peterson, N. V., and Groh, E. A., 1967, Geothermal potential of the Klamath Falls area, Oregon, A preliminary study: The Ore Bin, v. 29, no. 11, p. 209-231.
- Wagner, N. S., 1954, Preliminary report on the geology of the southern half of Umatilla County, Oregon: The Ore Bin, v. 16, no. 3, p. 13-17.
- Walker, G. W., 1963, Reconnaissance geologic map of the eastern half of the Klamath Falls (AMS) quadrangle, Lake and Klamath Counties, Oregon: U.S. Geol. Survey Mineral Inv. Field Studies Map MF-260.
- \_\_\_\_\_ 1973, Preliminary geologic and tectonic map of Oregon east of the 121st meridian: U.S. Geol. Survey Misc. Geol. Inv. Map MF-495.
- Walker, G. W., and Repenning, C. A., 1965, Reconnaissance geologic map of the Adel quadrangle, Lake, Harney, and Malheur Counties, Oregon: U.S. Geol. Survey Misc. Geol. Inv. Map I-446.
- \_\_\_\_\_ 1966, Reconnaissance geologic map of the western half of the Jordan Valley Quadrangle, Owyhee and Malheur Counties, Oregon: U.S. Geol. Survey Misc. Geol. Inv. Map. I-457.
- Walker, G. W., and Swanson, D. A., 1967, Summary report on the geology and mineral resources of the Harney Lake and Malheur Lake areas of the Malheur National Wildlife Refuge, north-central Harney County, Oregon: U.S. Geol. Survey Bull. 1260-L, p. 1-17.

- Waters, A. C., 1968, Reconnaissance geologic map of the Madras quadrangle, Jefferson and Wasco Counties, Oregon: U.S. Geol. Survey Misc. Geol. Inv. Map I-555.
- White, D. E., 1970, Geochemistry applied to the discovery, evaluation, and exploitation of geothermal energy resources in Proceedings United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, 1970, v. 1, part 2: Geothermics Spec. Issue 2 (in press).
- White, D. E., Muffler, L. J. P., and Truesdell, A. H., 1971, Vapor-dominated hydrothermal systems compared with hot-water systems: Econ. Geol., v. 66, no. 1, p. 75-97.

## U T A H

### SUMMARY

Mundorff (1970) conducted a reconnaissance study wherein data were obtained on the thermal, chemical and geologic characteristics of more than 50 thermal springs of Utah. Only three of the springs have temperatures near the boiling point of water; the maximum recorded temperatures of these springs range from 185° to 189° F.

Most thermal springs occur within a north-south zone 110 to 130 Km in width, bordered on the east by the Wasatch-Hurricane normal fault zone, and within the Basin and Range Province. Most of the springs are associated with Late Cenozoic faults. Only the three hottest are associated closely with Late Tertiary or Quaternary volcanic rocks. Most of the hotter springs occur along that portion of the Wasatch fault zone that extends from Provo northward to the Utah-Idaho border.

Temperatures of the thermal springs studied ranged from 68° to 189° F. Nearly all thermal springs in Utah are in or near fault zones.

Dissolved solids contents of the springs range from as low as 214 ppm for a spring having a temperature of 80° F to as high as about 45,000 ppm for a spring having a temperature of 132° F. Calculated dissolved solids average from 5000 to 8000 ppm. Most springs are sodium chloride in type, and all springs that contain more than 3000 ppm of dissolved solids are of the sodium chloride type. Silica concentrations exceed 100 ppm for two of the thermal springs. The pH of springs ranges from 6.2 to 8.8.

Water quality of thermal springs in Utah is, for the most part, unsuitable for industrial and agricultural uses.

UTAH, (cont'd)

Only two springs in the state, Roosevelt and Abraham Hot (Crater) Springs, have been designated as KGRA's by the U.S. Geological Survey.

Roosevelt (McKeans) Hot Springs  
313-405 ppm SiO<sub>2</sub>  
131-185°F (55-85°C)  
Dissolved solids 7800 ppm  
pH 8.3  
--water quality is poor

Abraham (Crater) Hot Springs  
59 ppm SiO<sub>2</sub>  
180°F (82°C)  
Dissolved solids 3630 ppm  
pH 7.6  
Discharge estimate 250 gpm  
--water quality is poor

Thermo Hot Springs is the third hottest in the state.  
100 ppm SiO<sub>2</sub>  
173°F (78°C)  
Dissolved solids 1460 ppm  
pH 8.1  
--water quality is poor

Some thermal springs in Utah have large discharges (up to 900 gpm), low dissolved solids contents, and fairly low temperatures; these springs are valuable as water supplies for irrigation and stock use, but other non-electrical usages may be feasible.

The main undesirable effect of the thermal springs in Utah is that they add significant amounts of water having high dissolved solids contents to some streams and lakes. The inflow of LaVerkin Hot Springs to the Virgin River and of many thermal springs around Utah Lake results in a deterioration of the quality of the surface water supply.

Table 1. Chemical analyses of thermal springs in Utah.

Location	Name of spring	Date of collection	Elev. (ft)	Temp. (°C)	pH	Calcium (Ca)		Magnesium (Mg)		Sulfate (SO <sub>4</sub> )		Chloride (Cl)		Total equivalents per million		Hardness (°C)	Nitrate (NO <sub>3</sub> )	
						ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	Ca	Mg			
(A-6-3) Vicab-S1	Cone Warm Springs	5-19-60	9,100	27	19	109	31	5	8.5	250	-	241	28	9.6	9.66	2.0	0.1	
							2.24	2.55	1.4	21	5.10	-	3.41	26				
(B-1-1) Hadeb-S1	Becks Hot Springs	1881	-	-	37	194	110	3,755	197	-	-	800	8,760	-	-	-	-	
		5-17-22	-	-	32	254	115	3,720	252	200	-	825	8,670	-	-	-	-	
		6-29-52	-	-	35	698	136	4,100	170	10	-	200	2,230	-	-	-	-	
		11-1-51	-	-	36	720	125	2,610	262	227	-	353	10,000	200	-	-	-	
		11-9-66	550	130	32	33	115	4,230	198	200	-	170	430	179.1	8.2	1.2	0.6	
		7-26-67	600	131	32	200	111	4,500	150	211	-	285	4,450	230.9	234.8	1.1	-	
(B-1-1) 25ab-S	Wasatch Hot Springs	3-16-25	-	-	33	1,380	100	2,600	552	300	-	1,330	3,400	-	-	-	-	
		5-2-55	-	-	62	68	96	1,920	163	200	-	1,070	1,490	90.9	106.3	-	-	
		1-2-57	-	-	41	23	36	1,820	190	293	-	1,000	2,580	111.2	111.6	-	-	
		3-19-50	-	-	26	60	63	1,090	62	278	-	700	2,000	90.2	81.3	-	-	
		5-19-52	-	-	22	18	101	2,110	213	269	-	870	1,820	112.5	122.1	1.9	0.9	
		7-26-67	105	16	108	565	109	2,210	113	200	-	1,000	1,170	122.5	121.9	2.8	0.3	
							21.61	2.40	20.22	1.20	2.00	0	17.61	19.55	101.26	109.6		
							5.3	118	2,390	263	255	-	30	5,100	147.0	148.6	2.0	-
(B-5-1) 27c-S	Hooper Hot Springs	9-15-52	-	-	35	57	82	2,520	265	271	-	151	151	127.0	127.0	2	2.0	
		10-27-51	-	-	32	525	110	2,130	292	300	-	100	151	127.0	127.0	12	8.3	
		6-20-57	-	-	35	30	85	2,520	168	207	-	50	151	127.0	127.0	8	-	
		4-13-60	-	-	30	55	90	2,520	230	200	-	50	151	127.0	127.0	12	-	
		11-7-66	-	-	118	106	15	2,130	163	200	-	30	151	127.0	127.0	1.0	1.8	
							25.25	2.44	102.22	6.53	3.67	-	2.79	116.36	121.3	121.0		
(B-5-3) 28d-S	Southwest Hooper Warm Springs	9-15-53	27	90	48	56	558	8,290	201	305	-	210	12,000	443.6	403.9	-	1.6	
(B-6-1) 23cd-S1	Opden Hot Springs	4-27-53	-	-	-	155	11	1,000	-	208	-	102	5,000	-	-	3.0	-	
		11-3-51	137	33	117	8	90	2,720	612	200	-	100	1,000	127.1	127.1	3.2	-	
		11-2-60	75	135	47	16	62	3,190	10	50	1.97	-	2,000	127.1	127.1	3.2	-	
		3-8-67	-	-	-	17	26	30	118,226	8.18	2.13	-	20	139.36	127.1	127.1	3.2	-
		5-18-67	-	-	136	65	170	20	2,690	367	190	-	121	1,000	127.1	127.1	3.2	-
(B-7-2) 14cd-S1	Leach Hot Springs	11-18-11	-	-	-	1,175	28	8,500	-	188	-	203	15,028	-	-	-	-	
(B-10-1) 10hd-S1	Stroking Hot Springs	11-3-51	-	-	41	1,190	46	2,720	1,100	100	-	200	15,000	-	-	-	-	
		1-7-55	-	-	33	835	16	5,550	252	199	-	163	10,000	-	-	-	-	
		2-5-58	-	-	135	1,100	70	2,000	601	190	-	189	11,000	-	-	-	-	
		11-1-66	-	-	137	1,070	39	6,500	935	180	-	201	12,000	-	-	-	-	
		3-8-67	-	-	-	1,000	3.21	26,231	23,291	7.98	-	-	18	136.22	127.2	125.2	3	2
		5-18-67	-	-	135	1,050	79	6,270	932	190	-	201	11,000	-	-	-	-	
							51.90	0.50	266.85	23.83	3.11	-	6.10	375.34	181.1	184.4	1	2.2
							828	179	10,500	-	193	-	20	14,500	-	-	-	-
(B-10-1) 10hd-S1	Stroking Hot Springs	10-27-52	-	-	51	27	21	11,160	55	50	-	51	21,000	227.4	228.2	0	-	
		5-21-52	-	-	124	55	81	29,200	587.20	16.87	8.59	-	1,700	575.48	578.0	585.2	-	-
		6-29-52	-	-	125	-	-	-	-	-	-	-	6,000	-	-	-	-	
		2-6-53	-	-	-	-	-	-	-	-	-	-	11,000	-	-	-	-	
		4-5-58	-	-	118	806	297	12,000	571	127	-	111	21,000	-	-	-	-	
		3-29-66	-	-	113	860	152	10,100	602	169	-	167	17,200	-	-	-	-	
		7-17-67	-	-	117	1,027	28.11	5,990	16.93	1.77	-	-	150	2,000	226.4	508.6	1.8	2.1
							637	2.31	9,280	592	6.36	-	150	17,200	226.4	508.6	1.8	2.1

1/ Calculated Na plus K, reported as Na.  
 2/ Analysis by Salt Lake City Corporation.  
 3/ Analysis by U.S. Bureau of Reclamation - Estimate.

Table extracted from: Mundorff, J. C., Major Thermal Springs of Utah, U.S. Geological Survey Water Resources Bulletin #13, 60 pgs, (1970)



Table 1. (continued).

Boron (B)	Lithium (Li)	Strontium (Sr)	Bromide (Br)	Iodide (I)	Dissolved solids		Specific conductance-micro-mhos at 25 C	pH	Concentration, in micrograms per liter																
					Residue on evaporation at 160°C	Calculated			Aluminum Al	Beryllium Be	Bismuth Bi	Cadmium Cd	Chromium Cr	Cobalt Co	Copper Cu	Gallium Ga	Germanium Ge	Iron Fe	Lead Pb	Manganese Mn	Molybdenum Mo	Nickel Ni	Titanium Ti	Vanadium V	Zinc Zn
0.02	-	-	-	-	672	586	896	7.4	15	0.57	0.29	1.4	1.4	1.4	1.4	5.7	2.9	4.3	1.4	8.6	.29	0.8	0.6	0.29	1.4
-	-	-	-	-	12,585	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	13,700	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	13,100	19,400	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	13,500	20,500	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
2.0	2.1	-	-	0.62	13,800	21,800	7.5	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1.2	-	-	-	-	13,950	20,800	7.4	31	.6	.3	1.4	1.4	6.9	1.4	5.7	20	37	1.4	486	.3	4.3	.6	.7	5.7	-
-	-	-	-	-	12,800	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	8,080	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	7,310	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	5,590	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	7,720	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1.2	.78	-	-	.26	8,680	8,590	13,700	8.0	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
.9	-	-	-	-	6,000	9,540	7.9	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	8,600	14,500	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	9,310	14,900	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	9,980	16,000	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	8,350	14,300	7.8	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1.0	-	-	-	.07	8,620	15,000	7.5	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1.1	7.0	-	-	.14	8,230	14,300	7.6	34	.57	.29	1.4	1.4	1.4	1.4	5.7	46	137	1.4	3,000	.29	5.1	<.57	.29	5.7	-
-	-	-	-	-	22,800	39,400	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	8,650	14,700	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
3.6	6.0	-	-	-	8,820	-	-	7.4	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
2.6	5.4	-	-	.3	8,650	15,000	7.7	40	.57	.29	1.4	1.4	1.4	1.4	5.7	66	149	1.4	3,000	.29	3.7	.57	1.7	5.7	-
-	-	-	-	-	15,400	-	-	43	.57	.29	1.4	1.4	1.4	1.4	5.7	91	200	1.4	800	4.3	2.6	.57	3.1	5.7	-
2.6	3.5	-	15	.39	8,680	14,400	7.7	69	.6	.3	1.4	1.4	1.4	1.4	5.7	56	154	1.4	914	.3	5.4	.6	.3	5.7	-
-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	25,200	37,200	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	18,900	28,300	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
5.1	9.9	28	8.2	.2	22,900	34,300	7.3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
3.1	7.9	-	-	.24	21,600	34,800	7.5	51	.57	.29	1.4	1.4	26	1.4	5.7	80	314	1.4	1,000	.29	7.4	.57	4.9	5.7	-
-	-	-	-	-	34,600	-	-	66	.57	.29	1.4	71	1.4	1.4	5.7	80	214	1.4	860	8.3	5.4	<.57	2.8	5.7	-
3.2	7.5	-	36	.38	22,700	34,100	7.0	63	.6	.3	1.4	1.4	1.4	1.4	5.7	538	608	1.4	1,100	.29	5.48	.6	.3	5.7	-
-	-	-	-	-	10,400	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	33,900	48,800	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	53,000	-	-	6.3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	51,000	-	-	6.4	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	52,200	-	-	6.2	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
3.6	5.9	31	15	1.3	26,600	53,900	6.7	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
3.6	3.86	-	-	-	29,900	46,100	7.3	27	.57	.29	-	-	<1.4	1.4	5.7	.3	-	1.4	286	.29	-	.57	.29	5.7	-
3.2	2.8	-	35	.76	29,000	42,500	7.1	43	.6	.3	1.4	1.4	1.4	1.4	5.7	21	24	1.4	237	.3	3.7	.6	.3	5.7	-

Utah Geological and Mineralogical Survey Water-Resources Bulletin 13, 1970

Table 1. (continued).

Location	Name of spring	Date of collection	Estimated discharge (gpm)	Temperature (°F)	Salinity (SAS)	Calcium (Ca)		Magnesium (Mg)		Sodium (Na)		Potassium (K)		Bicarbonate (HCO <sub>3</sub> )	Carbonate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Total equivalents per million		Fluoride (F)	Nitrate (NO <sub>3</sub> )	
						ppm eq	ppm eq	ppm eq	ppm eq	ppm eq	ppm eq	ppm eq	ppm eq					Major cations	Minor anions			
(B-11-2) 20da-S	Crystal (Hades?) Hot Springs	11-18-51	-	-	-	601	218	16,600	-	-	254	-	599	27,300	-	10.39	62.59	285.0	282.3	-	-	
		10-27-51	-	132	31	801	219	15,700	264	503	-	566	9,500	25,000	-	9.50	205.25	117.0	222.6	0	-	
		6-20-52	-	115	25	50,002	18,011	6,499.55	19,354	3,655	-	-	-	27,700	-	-	650.36	-	-	-	-	
		2-6-53	-	-	-	-	-	582,900	-	-	-	-	26.8	-	480	25,000	-	-	-	-	-	
		11-2-66	1,800	127	26	782	186	13,600	65	171	-	553	25,600	-	9.49	713.11	-	-	662.7	653.0	1.9	0.9
(B-11-19) 11dda-S1	Unnamed hot spring	5-10-60	75	10	23	64	14	1317	-	-	-	184	0	29	-	9.1	-	-	-	-		
(B-11-15) 11ada-S1	Warm Springs	8-12-60	530	80	14	6	8.0	27	1.7	10*	-	1.77	0	15	-	57	1.61	3.67	1.20	2	.1	
(B-11-3) 230aa-S1	Lady Hot Springs	11-1-66	-	109	26	212	53	2,690	118	366	-	90	-	1.87	4,470	126.10	135.1	134.0	1.5	2.6		
(B-11-5) 29-S	Blue Warm Springs 1/2	9-10-64	3,600	80	-	83	25	540	12	268	-	68	-	1.41	866	25.0	30.4	30.8	-	-		
		5-1-65	-	-	-	4.1	2.0	23.5	.8	4.0	-	-	-	-	-	-	-	-	-	-		
(C-1-7) 59-S	Big Warm Springs	9-28-59	3,600	-	7.9	102	97	2,821	-	-	133	93	352	-	7.33	4,550	128.25	-	-	-	9.7	
		9-29-59	-	-	10	16	29	2,710	92	215	0	327	4,360	-	5.81	122.99	134.5	131.3	2	1.1		
		1-5-60	2,750	-	9.5	138	81	2,810	106	223	-	355	4,550	-	7.18	128.35	136.5	139.2	-	10		
		2-12-60	2,100	-	7.1	135	80	2,850	103	193	-	351	4,670	-	7.11	131.25	139.9	141.9	-	12		
		7-8-60	3,200	-	7.0	135	73	2,790	99	266	-	342	4,560	-	6.91	126.96	137.1	137.2	-	12		
		10-1-60	2,700	-	15	136	73	2,830	98	193	-	349	4,560	-	7.26	128.48	138.9	139.9	-	9.7		
		10-11-60	1,000	-	9.0	149	77	2,800	94	203	-	345	4,510	-	7.16	127.22	137.4	137.8	-	10		
		1-10-61	2,900	-	11	152	81	2,840	100	220	-	354	4,600	-	7.15	129.26	139.8	140.5	-	10		
		2-2-61	-	-	9.5	145	88	2,890	101	218	-	352	4,720	-	7.31	131.15	142.8	144.0	-	13		
		5-27-66	-	65	2.9	140	85	2,750	112	202	-	318	4,420	-	6.62	124.69	136.5	134.8	-	1.0	1.4	
(C-1-6) 10ada-S1	Grantville Warm Springs	3-15-66	-	76	27	585	188	8,910	232	233	-	662	15,000	-	13.28	423.15	438.2	460.7	1.7	1.3		
(C-1-11) 6 12b-S	Crystal Hot Springs	10-2	-	-	28	141	28	465	55	216	-	378	337	-	-	-	-	-	-	-		
		5-22-51	-	137	60	176	25	3041/2	-	263	-	97	598	-	-	-	-	-	-	-		
		7-25-50	-	-	73	102	24	2251/2	-	156	-	72	560	-	-	-	-	-	-	-		
		9-10-52	-	-	66	102	24	2251/2	-	156	-	72	560	-	-	-	-	-	-	-		
		5-27-58	-	-	50	152	31	3301/2	-	460	-	72	595	-	-	-	-	-	-	0	9.6	
(C-5-1325-S	Crater Hot Springs	5-27-58	-	107	27	180	59	2121/2	-	170	-	425	318	-	-	-	-	-	-	-		
		6-6-58	-	110	28	191	52	2151/2	-	170	-	425	318	-	-	-	-	-	-	-	2.5	
(C-5-1) 25da-S	Saratoga Hot Springs	1931	-	-	-	158	36	231	-	283	-	188	161	-	-	-	-	-	-	-		
		3-26-66	-	111	25	194	66	2151/2	-	170	-	425	318	-	-	-	-	-	-	-		
(C-5-5) 9cha-S1	Horseshoe Warm Springs	9-22-64	1,000	75	20	58	25	110	11	174	-	90	179	-	-	-	-	-	-	1.0	0.4	
		5-27-66	-	77	18	16	25	109	11	172	-	89	182	-	-	-	-	-	-	-	1.2	.3
		7-18-67	-	80	19	279	1.97	4.76	.13	2.82	-	1.85	5.13	9.83	9.60	-	-	-	-	-	1.4	.0
(C-5-5) 11aa-S1	Russells Warm Springs	4-29-66	450	71	17	51	21	73	11	170	-	55	125	-	-	-	-	-	-	1.6	.2	
		5-27-66	-	72	17	53	19	76	12	175	-	55	122	-	-	-	-	-	-	-	1.5	.1
		7-18-67	-	-	19	2.64	1.56	3.22	.31	2.87	-	1.15	3.44	7.72	7.46	-	-	-	-	-	1.5	.0

1/ Calculated Na plus K, reported as Na.  
2/ Analysis from Milligan and others (1966, table 1).



Utah Geological and Mineralogical Survey Water-Resources Bulletin 13, 1970

Table 1, (continued).

Location	Name of spring	Date of collection	Estimated discharge (gpm)	Temperature (°F)	Silica (SiO <sub>2</sub> )	Calcium (Ca)		Magnesium (Mg)		Sodium (Na)		Potassium (K)		Bicarbonate (HCO <sub>3</sub> )		Carbonylate (CO <sub>3</sub> )		Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Total equivalents per million		Fluoride (F)	Nitrate (NO <sub>3</sub> )					
						ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm	ppm			ppm	ppm			ppm				
(C-10-14) 31-S	Wilson Hot Springs	7-12-67	100	51	33	751	224	7,000	48	178	-	1,560	11,900	-	-	1,560	11,900	-	-	365.3	371.1	4.0	0					
						36.97	18.43	308.43	7.58	7.92	-	32.48	335.70	-	-	-	-	-	-	-	-	-	-	-	-	-		
(C-11-14) 23c-S (C-11-14) 23db-S (C-11-14) 23de-S (C-11-14) 23dd-S	Fish Springs	7-12-67	75	82	20	136	26	920	36	312	-	150	630	-	-	150	630	-	-	17.72	30.3	29.92	2.0	0				
						6.79	2.15	20.25	0.92	5.11	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
						-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
						-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
(C-15-8) 10-S	Abraham (Crater) Hot Springs	11-12-27	-	-	-	150	-	6.12	-	-	-	136	-	-	560	1,100	-	-	-	-	-	-	-					
						661	56	812	67	153	-	731	1,550	-	-	-	-	-	-	-	-	-	-	-	-			
						12.01	5.60	35.42	1.21	2.51	-	13.26	23.73	50.6	61.5	3.0	2											
(C-15-19) 31bc-S	Gandy Warm Springs	3-3-66	4,500	81	20	56	21	26	7.8	278	-	21	20	-	-	21	20	-	-	5.09	3.55	7	3.6					
						2.50	1.73	1.22	0.02	5.56	-	44	56	-	-	-	-	-	-	-	-	-	-	-	-			
(C-2-6) 21ccc-S1 (C-2-6) 21ddd-S1	Meadow Hot Springs	5-12-66	-	106	44	619	97	1,020	157	314	-	1,020	1,750	-	-	1,020	1,750	-	-	28.1	75.2	5.1	0					
						20.91	7.94	55.25	3.01	5.18	-	21.24	32.37	78.1	75.2	-	-	-	-	-	-	-	-	-	-			
(C-22-6) 35ddb-S1	Hutton Hot Springs	6-19-57	-	100	44	665	89	1,090 1/2	-	422	-	985	1,260	-	-	985	1,260	-	-	60.21	-	-	2.4					
						23.20	7.13	-	-	6.69	-	-	50.21	-	-	-	-	-	-	-	-	-	-	-	-	-		
						8.27-58	25	-	-	22.40	6.16	-	-	7.93	-	-	51.06	-	-	-	-	-	-	-	-	-	-	
						45	38	12	4.0	296	-	27	20	-	-	27	20	-	-	27	20	-	-	6.00	6.00	3	1	
						2.25	3.13	1.52	10	4.88	-	56	56	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
						6-6-66	700	72	11	51	35	15	1.2	280	3	29	20	6.13	5.85	-	-	-	-	-	-	-	-	
(C-25-3) 10dda-S1	Nantux (Cooper) Hot Springs	7-23-57	-	-	54	282	34	562	63	354	-	898	630	-	-	898	630	-	-	42.9	42.3	2.6	0					
						16.07	2.79	24.54	1.61	5.80	-	18.70	17.77	62.9	62.3	3.0	0											
						288	33	555	67	416	-	831	600	-	-	831	600	-	-	831	600	-	-	17.15	18.61	42.9	42.8	2.8
(C-25-3) 15a-S	Nantux (Cooper) Hot Springs	5-3-67	-	97	51	281	49	553	69	386	-	924	600	-	-	924	600	-	-	43.4	42.5	1.8	2					
						15.02	4.03	26.06	1.25	6.33	-	19.24	16.93	6.66	6.65	1.8	2											
(C-25-4) 23-S	Joseph Hot Springs	7-23-57	-	130	85	287	36	1,440	68	426	-	1,270	1,750	-	-	1,270	1,750	-	-	81.4	82.8	2.7	-					
						16.07	2.96	32.64	1.73	6.98	-	20.25	24.36	81.4	82.8	6.0	-											
(C-26-9) 34-icb-S1	Roosevelt (McKeans) Hot Springs	11-4-50	-	185	405	19	3.3	2,080	472	158	-	65	3,810	-	-	65	3,810	-	-	103.48	103.8	111.7	7.5	11				
						0.95	0.27	98.48	12.02	5.52	-	1.35	4.240	103.8	111.7	7.5	11											
						22	0	2,500	488	156	-	73	4,240	-	-	73	4,240	-	-	73	4,240	-	-	119.61	122.3	123.7	-	
						3.10	0	108.75	12.48	2.56	-	1.52	119.61	122.3	123.7	-	-	-	-	-	-	-	-	-	-	-		
(C-30-9) 7aca-S1	Kadium (Doctors) Warm Springs	8-21-63	-	91	31	110	24	172	18	220	3	480	65	-	-	480	65	-	-	15.2	15.4	9.8	2					
						5.48	1.97	7.68	0.66	3.60	-	9.99	1.83	15.2	15.4	4.3	1											
						104	26	165	18	222	-	458	65	-	-	458	65	-	-	458	65	-	-	14.9	14.9	4.3	1	
(C-30-9) 7aca-S1	Kadium (Doctors) Warm Springs	7-11-67	100	89	32	88	15	169	17	228	-	435	63	-	-	435	63	-	-	15.0	14.6	4.5	0					
						4.39	2.87	7.35	0.43	3.74	-	9.06	1.78	15.0	14.6	4.5	0											

1/ Calculated Na plus K, reported as Na.

Mundorff - Major Thermal Springs of Utah

Table 1. (continued).

Boron (B)	Lithium (Li)	Strontium (Sr)	Bromide (Br)	Iodide (I)	Dissolved solids		Specific conductance (micro-mhos at 25°C)	pH	Concentration, in micrograms per liter																	
					Residue on evaporation at 180°C	Calculated			Aluminum (Al)	Beryllium (Be)	Bismuth (Bi)	Cadmium (Cd)	Chromium (Cr)	Cobalt (Co)	Copper (Cu)	Gallium (Ga)	Germanium (Ge)	Iron (Fe)	Lead (Pb)	Manganese (Mn)	Molybdenum (Mo)	Nickel (Ni)	Titanium (Ti)	Vanadium (V)	Zinc (Zn)	
2.6	2.1	-	23	.36	21,800	11,200	7.4	74	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	20	43	<1.4	<1.4	8.0	4.0	<.57	<.29	<5.7		
.74	.33	-	1.2	.02	1,820	3,050	7.7	25	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	<.29	110	<1.4	<1.4	9.4	2.5	<.57	1.0	110		
-	-	-	-	-	-	3,160	7.2	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	-	-	-	-	-	3,130	7.2	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	-	-	-	-	-	3,100	7.3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	-	-	-	-	3,170	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
.70	.28	-	-	-	3,210	-	5.910	7.8	11	<.57	<.29	-	-	<1.4	<1.4	<5.7	<.29	-	<1.4	100	<.29	-	<.57	<.29	<5.7	
.83	.62	-	1.8	.09	3,630	5,570	7.3	4.3	<.57	<.29	1.4	<1.4	<1.4	<1.4	<5.7	<.29	86	5.1	1.10	<.29	2.7	1.4	<.29	<5.7		
.09	-	-	-	-	305	498	7.6	4.3	<.57	<.29	1.4	<1.4	<1.4	<1.4	<5.7	4.6	3.4	<1.4	<1.4	1.5	1.6	<.57	1.6	197		
.09	.02	-	.1	.00	301	485	7.6	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
4.2	2.5	-	1.2	.02	4,690	7,350	7.0	629	<.57	<.29	1.4	<1.4	<1.4	<1.4	<5.7	57	7.4	<1.4	<1.4	<.29	2.7	<.57	<.29	140		
-	-	-	-	-	4,810	7,350	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	-	-	-	-	4,690	7,360	7.0	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	-	-	-	-	4,850	7,370	7.2	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
6.1	1.2	-	1.3	.09	4,870	7,440	7.7	106	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	571	13	<1.4	<1.4	<.29	2.5	<.57	<.29	<109		
-	1.2	-	4.0	.45	4,900	7,130	7.5	61	<.57	<.29	<1.4	<1.4	<20	<1.4	<5.7	55	40	<1.4	<1.4	<.3	4.0	<.6	<.3	<5.7		
-	-	-	-	-	4,670	7,370	6.7	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	-	-	-	-	4,850	7,380	6.5	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	.5	-	-	-	310	548	7.9	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
.04	.04	-	.1	.01	307	551	8.3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	4.8	-	-	-	2,700	4,100	7.6	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
1.9	1.1	-	-	-	2,860	4,020	6.4	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
2.6	.51	-	.4	.03	2,680	4,000	7.6	4.0	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	19	5.1	<1.4	<1.4	1.2	.9	<.57	<.29	<5.7		
1.4	.9	-	-	-	4,070	7.5	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
2.7	.57	-	.3	.06	2,630	4,100	7.8	6.0	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	16	11	<1.4	<1.4	1.2	1.1	<.57	<.29	<5.7		
2.3	.4	-	1.6	.04	2,700	3,900	7.9	571	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	16	29	<1.4	166	<.29	4.1	57	<.29	<5.7		
.08	.01	-	.1	.01	428	623	7.4	11	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	<.29	14	<1.4	<1.4	18	1.7	<.57	1.5	<5.7		
-	-	-	-	-	5,150	7,290	6.9	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
4.8	1.5	-	-	-	4,970	7,520	6.6	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
4.0	1.5	-	.9	.10	5,090	7,480	7.6	716	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	129	16	37	127	19	5.4	<.57	<.29	<5.7		
1.7	1.9	-	1.0	.12	5,180	7,530	7.8	19	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	57	270	<1.4	163	25	4.6	<.6	<.1	<5.7		
-	-	-	-	-	7,040	11,500	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
.18	.71	-	1.3	.1	7,800	12,700	7.9	.04	-	-	-	-	-	-	00	-	-	00	0	-	-	0	-	0		
.31	.4	1.2	.0	.01	1,020	1,420	8.3	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		
-	.13	-	.0	.01	927	1,410	8.2	31	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	19	4.3	<1.4	<1.4	2.1	1.4	<.57	<.29	41		
.47	.11	-	.2	.00	956	1,390	7.4	29	<.57	<.29	<1.4	<1.4	<1.4	<1.4	<5.7	27	110	<1.4	<1.4	<.29	2.3	<.57	<.29	<5.7		

Utah Geological and Mineralogical Survey Water-Resources Bulletin 13, 1970

Table 1. (continued).

Loca-tion	Name of spring	Date of collection	Esti-mated dis-charge (gpm)	Temp-erature (°F)	Silica (SiO <sub>2</sub> )	Calcium (Ca)	Magnes-ium (Mg)	Sodium (Na)	Potassium (K)	Bicarbon-ate (HCO <sub>3</sub> )	Car-bon-ate (CO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Total equivalents per million		Fluor-ide (F)	Ni-trate (NO <sub>3</sub> )	
						Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Ppm epm	Major cations
(C-30-12) 21-S	Thermo Hot Springs	10-23-39	-	-	-	-	-	-	-	370	-	460	217	-	-	-	-	
		8-21-63	-	-	108	83	9.7	358	49	6.06	-	9.58	6.12	-	-	14	1.0	
(C-30-12) 28-S	do	8-21-63	-	-	103	4.14	.79	15.57	1.25	6.29	-	10.04	5.42	21.78	22.2	4.7	1.1	
		5-25-66	10	173	100	3.54	.75	15.96	1.15	5.15	.53	9.76	6.07	21.4	21.5	6.7	.1	
		7-11-67	-	170	100	75	9.7	361	52	359	-	460	208	5.87	21.6	21.3	14	.0
						3.74	.80	15.70	1.33	5.88	-	9.58	5.87	21.6	21.3	14	.0	
						76	12	364	47	374	-	460	212	5.98	21.8	21.7	14	.0
						3.79	.98	15.83	1.20	6.13	-	9.58	5.98	21.8	21.7	14	.0	
(C-40-16) 7ac6-S1	Vuyo Hot Spring	3-30-66	90	90	30	59	29	32	4.4	220	-	100	30	6.82	6.52	.6	8.3	
		4-20-67	-	90	32	2.94	2.39	1.39	.11	3.61	-	2.08	.85	6.82	6.52	.7	6.9	
						53	28	32	3.6	230	-	90	30	6.42	6.47	.7	6.9	
						2.64	2.30	1.39	.09	3.77	-	1.67	.85	6.42	6.47	.7	6.9	
(C-41-13) 25-S	LaVeckin (Dixie) Hot Springs	6-3-40	-	-	10	787	165	2,300	167	1,270	-	1,960	3,440	157.0	158.7	-	.2	
		8-15-50	-	-	9.3	39.27	13.57	100.05	4.27	20.82	-	40.81	97.04	157.0	158.7	-	.2	
		2-5-51	-	-	30	816	197	2,430	175	1,290	-	1,940	3,580	100.99	101.56	3.1	3.2	
		8-11-60	4,800	100	28	40.71	16.20	101.79	4.47	21.14	-	40.41	100.99	161.4	165.6	2.1	3.2	
		3-25-66	4,500	108	28	825	169	2,340	175	1,300	-	1,970	3,600	154.6	154.1	2.6	.9	
						41.16	13.90	101.79	4.47	21.30	-	41.03	101.56	154.6	154.1	2.6	.9	
						590	148	2,490	177	583	-	3,050	3,610	154.6	154.1	2.6	.9	
						29.44	12.17	108.31	4.52	9.55	-	42.70	101.83	154.6	154.1	2.6	.9	
						643	128	2,530	220	721	-	1,990	3,620	158.3	155.4	2.6	.9	
						32.09	10.53	110.06	5.63	11.82	-	41.43	102.12	158.3	155.4	2.6	.9	
(D-3-4) 26bra-S1	Midway Hot Springs	7-1-52	-	-	5.6	358	16	257	17	754	-	698	121	27.7	27.9	2.2	.1	
(D-3-4) 26cca-S1	do	9-28-66	50	103	23	17.86	1.31	114	25	12.35	-	34.53	108	27.7	27.9	2.2	.1	
(D-3-4) 27bab-S1	do	9-13-67	-	113	27	331	68	114	25	674	-	661	108	30.5	30.9	2.5	.4	
(D-3-4) 27bac-S1	do	5-10-67	10	104	28	16.51	5.59	4.96	.64	11.05	-	13.76	132	32.7	33.1	3.1	.0	
(D-3-4) 27bad-S1	do	9-28-66	150	103	28	17.22	6.83	6.44	.41	10.56	-	15.45	140	32.7	33.1	2.5	.1	
(D-3-4) 27cbd-S1	do	9-28-66	0	84	21	361	88	152	32	646	-	853	138	32.8	32.9	2.1	.1	
(D-3-4) 27cbd-S2	do	5-15-67	0	84	22	18.01	7.23	6.61	.82	11.41	-	17.76	115	32.8	32.9	2.3	.1	
		9-28-66	0	85	19	389	73	151	31	728	-	820	138	32.8	32.9	2.3	.1	
		5-15-67	0	90	17	19.41	6.00	6.57	.79	11.93	-	17.07	138	32.8	32.9	2.3	.1	
		5-16-67	0	83	21	353	72	125	28	716	-	702	115	32.8	32.9	2.3	.1	
						17.61	5.92	5.44	.72	11.74	-	15.62	115	32.8	32.9	2.3	.1	
						228	95	130	28	476	-	719	115	32.8	32.9	2.3	.1	
						11.37	7.81	5.66	.72	7.80	-	14.07	115	32.8	32.9	2.3	.1	
						329	70	111	25	646	-	643	103	27.6	27.5	2.4	.0	
						16.41	5.75	4.83	.64	11.24	-	13.39	103	27.6	27.5	2.4	.0	
						279	74	114	26	572	-	611	105	25.5	25.1	2.7	.0	
						13.92	6.08	4.96	.66	9.38	-	12.72	105	25.5	25.1	2.7	.0	
						329	88	163	33	584	-	805	150	31.6	29.5	2.7	.0	
						16.41	7.23	7.09	.84	9.57	-	15.66	4.23	31.6	29.5	2.7	.0	
(D-4-24) 16cdd-S1	Split Mountain Warm Spring	9-19-48	5,400	86	18	97	32	193	193	198	-	212	291	-	-	-	1.2	
						4.84	2.63	-	-	3.24	-	4.41	8.20	-	-	-	1.2	
(D-7-1) 8cc1-S1	Warm spring west shore of Utah Lake	9-11-58	-	77	15	144	58	342	-	348	-	325	510	-	-	-	.8	
						7.18	4.77	-	-	5.70	-	6.76	14.38	-	-	-	.8	
(D-7-1) 8bbc-S1	do	9-11-58	-	75	16	88	59	342	-	196	-	314	510	-	-	-	.8	
						4.39	4.85	-	-	3.21	-	6.54	14.38	-	-	-	.8	
(D-8-1) 3dda-S1	Lincoln Point Warm Springs	5-27-64	-	87	.22	457	114	1,820	-	756	-	953	2,800	-	-	-	-	
		6-16-66	-	89	21	27.80	9.37	1,510	159	11.39	-	19.84	78.99	-	-	-	-	
						4.51	136	65.68	4.02	751	-	940	2,530	103.4	103.2	2.8	2.4	
						22.50	11.19	65.68	4.02	12.31	-	19.57	71.37	103.4	103.2	2.8	2.4	
(D-8-5) 14d-S	Diamond Fork Warm Springs	10-20-67	450	68	17	104	32	117	8.3	264	0	340	36	13.1	13.5	1.6	.5	
						5.20	2.64	5.09	.21	4.33	-	8.12	1.02	13.1	13.5	1.6	.5	
(D-9-4) 18ba-S	Castilla Hot Springs	10-20-67	20	104	30	469	80	1,680	10	562	0	1,400	2,320	105.6	103.6	3.6	4.8	
						23.40	6.60	73.08	2.56	8.88	-	29.15	65.45	105.6	103.6	3.6	4.8	
(D-10-1) 8c-S	Goahen Warm Springs	5-27-64	-	70	18	84	41	356	-	316	-	102	558	-	-	-	2.2	
		6-15-66	-	70	17	4.19	3.40	-	-	5.18	-	2.12	15.74	-	-	-	2.2	
						87	40	363	19	314	-	115	560	-	-	-	2.1	
						4.34	3.29	14.92	.49	5.15	-	2.39	15.23	23.0	22.8	1.2	2.1	
(D-18-2) 13cad-S1	Livingston(Crystal) Warm Spring	2-6-61	-	-	5.6	26	15	129	-	421	-	86	55	-	-	.7	3.0	
						1.29	1.23	-	-	6.90	-	-	-	-	-	-	3.0	
(D-19-2) 4dca-S1	Sterling Warm Spring	8-27-57	-	67	13	38	19	94	3.8	310	-	71	34	7.62	7.50	1.1	.1	
						1.89	1.56	4.08	.09	5.08	-	1.47	.95	7.62	7.50	1.1	.1	

1/ Calculated Na plus K, reported as Na.



Utah Geological and Mineralogical Survey Water-Resources Bulletin 13, 1970

Table 2. Ratios, by weight, of selected constituents in thermal springs in Utah.

Location	Name of spring	Date	Dis-solved solids (ppm)	Calcium Magnesium Ca/Mg	Sodium Calcium Na/Ca	Sodium Potassium Na/K	Sodium Lithium Na/Li	Chloride Sulfate Cl/SO <sub>4</sub>	Chloride Fluoride Cl/F	Chloride Bicarbonate Cl/HCO <sub>3</sub>	Chloride Bromine Cl/Br	Bromine Iodine Br/I		
				Magnesium Calcium Mg/Ca	Calcium Sodium Ca/Na	Potassium Sodium K/Na	Lithium Sodium Li/Na	Sulfate Chloride SO <sub>4</sub> /Cl	Fluoride Chloride F/Cl	Bicarbonate chloride HCO <sub>3</sub> /Cl	Bromine Chloride Br/Cl	Iodine Bromine I/Br		
(A-4-3)31cab-S1	Como	5-18-66	576	3.5	0.11	4.0	-	0.12	-	-	-	-		
				.29	3.2	.75	-	8.3	-	-	-	-		
(B-1-1)14deb-S1	Becks	1881	12,580	6.3	5.4	19.0	-	8.0	-	-	-	-		
		5-19-42	13,900	4.9	6.2	9.1	-	8.8	-	32.1	-	-		
		8-29-47	13,100	5.1	6.0	-	-	9.0	-	70.7	-	-		
		11-3-51	13,500	5.8	5.6	15.5	-	8.3	3,160	32.0	-	-		
		11-9-66	13,600	5.5	5.1	20.8	1,790	7.9	2,280	31.9	-	-		
		7-26-67	13,900	5.3	5.7	26.8	-	7.8	2,260	33.8	-	-		
						.18	.18	.06		.13	.0006	.03		
(B-1-1)25db-S	Wasatch	3-16-34	12,800	10.1	1.9	4.2	-	4.4	-	18.9	-	-		
		May 1935	8,080	4.8	4.1	11.4	-	3.3	-	11.6	-	-		
		Jan. 1937	7,310	7.1	4.5	9.4	-	3.4	-	9.5	-	-		
		3-16-40	5,590	7.1	3.0	22.1	-	2.1	-	7.4	-	-		
		5-19-42	7,770	5.2	3.9	10.0	-	4.0	-	14.2	-	-		
		11-3-66	8,590	5.2	4.3	21.2	1,090	3.5	2,290	19.0	-	-		
		7-26-67	6,000	4.8	3.7	23.2	-	3.4	1,010	11.6	-	-		
						.21	.27	.04		.29	.001	.08		
						.23	.22	.12		.007	.0001	.05		
(B-5-3)27c-S	Hooper	9-15-53	9,310	5.8	4.7	8.8	-	121	6,250	18.7	-	-		
		10-7-53	9,980	4.8	4.5	8.1	-	140	12,950	19.5	-	-		
		6-20-57	8,350	6.1	4.4	14.4	-	126	-	5.5	-	-		
		4-13-60	8,620	5.3	4.7	10.3	-	86.1	-	23.1	-	-		
		11-7-66	8,230	5.6	4.6	9.2	1,125	127	4,850	21.6	-	-		
						.18	.21	.11		.009	.0002	.05		
(B-5-3)28d-S	Southwest Hooper	9-15-53	27,800	1.2	15.5	10.3	-	65.8	-	47.4	-	-		
						.83	.06	.10		.02	.02			
(B-6-1)23cc-S1	Ogden	4-27-43	8,650	32.3	8.5	-	-	49.8	1,890	24.4	-	-		
		11-3-51	8,820	52.1	8.1	6.7	456	20.6	1,490	25.3	-	-		
		11-2-66	8,650	7.1	7.7	7.6	66.5	46.6	1,320	75.7	-	-		
		5-18-67	8,480	6.5	8.4	7.7	99	41.2	1,310	26.0	333	38		
						.15	.12	.13		.01	.02	.0008	.04	.003
(B-7-2)14cca-S1	Utah	11-18-11	-	41.9	-	-	-	15.3	-	80.2	-	-		
		11-3-51	25,200	25.9	6.5	7.0	-	74.0	-	74.4	-	-		
		3-2-54	16,900	22.0	8.5	7.5	-	59.7	4,150	56.8	-	-		
		4-5-58	22,900	16.3	6.7	7.8	710	70.4	4,110	65.2	1,620	41		
		11-3-66	21,600	26.2	6.5	7.0	833	61.2	2,950	69.8	-	-		
		5-18-67	22,700	13.1	6.6	7.4	416	45.3	3,240	70.0	3,910	89		
						.08	.15	.14		.001	.02	.0003	.02	.0003
(B-10-3)30bb-S1	Stinking	11-18-11	30,400	2.3	11.8	-	-	925	-	47.1	-	-		
		10-27-51	33,900	2.5	12.5	17.0	-	352	-	34.4	-	-		
		2-6-53	-	-	-	-	-	164	-	44.3	-	-		
		4-5-58	36,600	3.2	13.3	22.1	1,830	194	11,370	66.7	1,440	9.4		
		5-24-66	29,900	2.5	12.0	15.3	2,620	109	11,100	105	-	-		
		5-17-67	29,000	4.0	10.8	07	-	109	-	113	491	48		
				.87	.07	.05		.0003	.009	.0002				
(B-11-2)29da-S	Crystal	11-18-11	45,500	4.1	18.4	-	-	56.3	-	59.7	-	-		
		10-27-51	42,200	3.7	18.3	-	-	53.6	-	54.0	-	-		
		2-6-53	-	-	-	-	-	52.7	-	56.0	-	-		
		11-2-66	38,500	4.2	17.3	20.7	2,830	50.9	11,500	60.9	-	-		
				.24	.06	.05		.02	.0009	.02				
(B-11-19)11dda-S1	Unnamed hot spring	5-16-6d	248	3.1	-	-	-	3.1	-	.05	-	-		
				.32	-	-	-	3.2	-	20.2	-	-		
(B-12-15)19aab-S1	Warm	8-12-66	214	4.5	.75	15.9	1,350	3.8	285	.53	285	-		
				.22	1.33	.06	-	.26	.004	1.9	.003	-		
(B-13-3)23baa-S1	Uddy	11-1-66	7,850	3.9	12.2	22.8	3,240	49.7	2,910	12.2	-	-		
				.26	.08	.02		.02	.0003	.08				



Mundorff - Major Thermal Springs of Utah

Table 2. (continued).

Location	Name of spring	Date	Dis-solved solids (ppm)	Calcium	Sodium	Sodium	Sodium	Chloride	Chloride	Chloride	Chloride	Bromine		
				Magnesium	Calcium	Potassium	Lithium	Sulfate	Fluoride	Bicarbonate	Bromine			
				Ca/Mg	Na/Ca	Na/K	Na/Li	Cl/SO <sub>4</sub>	Cl/F	Cl/HCO <sub>3</sub>	Cl/Br	Br/I		
				Ca/Mg	Ca/Na	K/Na	Li/Na	SO <sub>4</sub> Cl	F/Cl	HCO <sub>3</sub> Cl	Br/Cl	I/Br		
(C-1-7)869-S	Big	9-28-59	8,100	1.05	27.5	-	-	-	12.9	-	26.5	-	-	
				.95	.06	-	-	-	.08	-	.67	-	-	
		9-29-59	7,860	1.70	20.1	28.1	1,520	-	13.3	21,800	20.0	-	-	
				.59	.05	.04	.0007	-	.06	-	.0005	.05	-	-
		1- 5-60	8,160	1.70	20.4	26.5	2,810	-	13.2	-	20.4	-	-	
				.52	.05	.04	.0004	-	.08	-	-	.05	-	-
		4-12-60	8,300	1.70	21.3	27.7	-	-	13.3	-	28.2	-	-	
				.50	.05	.04	-	-	.06	-	.05	-	-	-
		7- 8-60	8,060	1.45	19.4	27.2	-	-	13.6	-	21.8	-	-	
				.57	.05	.04	-	-	.07	-	.05	-	-	-
		10- 1-60	8,210	2.00	19.6	28.9	-	-	11.2	-	23.7	-	-	
				.50	.05	.04	-	-	.08	-	.04	-	-	-
		10-11-60	8,040	1.80	20.1	30.1	-	-	13.1	-	25.0	-	-	
				.56	.05	.03	-	-	.05	-	.05	-	-	-
1-10-61	8,250	1.75	20.0	28.2	-	-	13.0	-	20.9	-	-			
		.57	.05	.04	-	-	.08	-	.05	-	-	-		
4- 4-61	8,430	1.65	20.1	28.1	-	-	13.4	-	21.7	-	-			
		.51	.05	.04	-	-	.08	-	.05	-	-	-		
5-27-66	7,940	1.6	19.6	24.1	4,100	.0002	13.9	4,425	21.4	-	-			
		.62	.05	.04	-	-	.07	.0002	.05	-	-	-		
(C-2-6)16aad-S1	Grantville	3-15-66	25,800	3.1	15.3	37.6	-	22.7	8,520	64.4	-	-		
				.32	.07	.73	-	.04	.0001	.02	-	-		
(C-4-1)12b-S	Crystal	1882	1,580	5.0	2.9	2.4	-	.89	-	1.6	-	-		
				.20	.35	.15	-	1.1	-	.62	-	-		
5-22-34		1,665	4.2	2.9	-	-	6.2	-	2.1	-	-	-		
				.24	.35	-	-	.16	-	.47	-	-		
7-25-50		1,300	4.2	2.2	-	-	2.4	-	3.8	-	-	-		
				.25	.45	-	-	.13	-	.26	-	-		
9-10-52	1,550	2.5	-	-	-	-	7.1	-	1.9	-	-			
			.40	-	-	-	.14	-	.53	-	-			
5-27-58	1,390	4.6	-	-	-	-	8.1	-	1.8	-	-			
			.22	-	-	-	.12	-	.56	-	-			
(C-5-1)25-S	Crater	5-27-58	1,390	3.7	-	-	-	.75	-	1.0	-	-		
				.27	-	-	-	1.4	-	1.0	-	-		
		6- 4-58	1,440	3.7	-	-	-	.77	-	1.1	-	-		
				.27	-	-	-	1.3	-	.91	-	-		
(C-5-1)25cd-S	Saratoga	1933	1,500	4.4	1.8	-	-	.93	-	1.3	-	-		
				.23	.56	-	-	1.1	-	.77	-	-		
		3-24-66	1,450	2.8	-	-	-	.78	-	1.1	-	-		
				.36	-	-	-	1.3	-	.91	-	-		
(C-5-5)9cba-S1	Morgans	9-22-64	583	2.4	1.9	10.0	-	2.0	179	1.0	-	-		
				.42	.53	.10	-	.50	.005	1.0	-	-		
5-27-66		578	2.1	1.9	8.4	2,720	2.0	151	1.1	-	-			
			.43	.53	.12	.0004	.50	.007	.91	-	-			
7-18-67	586	1.7	2.3	10.2	3,730	1.9	132	1.2	627	.0015	30			
			.59	.43	.09	.0003	.53	.067	.83	.0015	-			
(C-5-5)17aaa-S1	Russells	4-29-66	438	2.4	1.4	6.6	-	2.3	-	.72	-	-		
				.42	.71	.15	-	.53	-	1.5	-	-		
5-27-66		440	2.8	1.4	6.2	3,750	2.2	81	.70	-	-			
			.36	.71	.16	.0003	.45	.012	1.4	-	-			
7-18-67	445	3.2	1.3	7.1	3,550	1.9	81	.72	610	.0010	10			
			.31	.77	.14	.0001	.53	.012	1.4	.0010	-			
(C-10-14)33-S	Wilson	7-12-67	21,800	3.3	9.6	39.4	3,380	7.6	2,980	66.9	517	63.9		
				.30	.10	.02	.0003	.13	.0003	.01	.0019	-		
(C-11-14)23c-S	Fish	7-12-67	1,820	5.2	3.5	13.1	1,420	1.9	315	2.0	485	65		
				.19	.28	.08	.0007	.53	.003	.50	.0020	-		
(C-14-8)10c-S	Abraham	11-12-27	3,170	-	-	-	-	2.5	-	9.0	-	-		
					-	-	-	.50	-	.11	-	-		
5-31-66		1,710	6.4	2.2	12.1	1,040	2.1	117	10.1	-	-			
			.15	.45	.08	.0004	.48	.019	.10	-	-			
7-13-67	3,630	5.1	2.4	17.0	1,300	1.9	354	9.1	806	.0012	20			
			.70	.42	.06	.0008	.53	.028	.11	.0012	-			
(C-15-19)31bc-S	Gandy	3- 3-66	305	2.4	.56	10	-	.95	-	.07	-	-		
				.42	1.8	.10	-	1.0	-	1.4	-	-		
7-12-67	303	2.8	.58	7.8	1,450	1.9	37	.10	260	-	-			
			.36	1.7	.13	.0006	1.1	.025	1.0	.038	-			
(C-22-6)26ccc-S1	Headow	5-12-66	4,690	4.1	2.5	6.6	416	1.7	370	5.6	3,350	43		
				.43	.40	.15	.002	.59	.003	.18	.0007	-		
(C-22-6)27ddd-S1	do	4- 8-43	4,810	4.9	-	-	-	1.8	-	4.0	-	-		
				.30	-	-	-	.56	-	.25	-	-		
6-18-57		4,690	3.3	-	-	-	-	1.8	-	4.0	-	-		
				.19	-	-	-	.50	-	.25	-	-		
8-27-58		4,850	7.1	-	-	-	-	1.7	-	4.2	-	-		
				.14	-	-	-	.59	-	.24	-	-		
5-12-66	4,870	4.4	2.4	6.8	707	1.7	156	4.1	370	.0007	16.2			
			.23	.42	.15	.001	.59	.003	.24	.0007	-			
5-22-67	4,900	3.8	2.4	73.9	319	1.6	327	4.4	450	.022	8.9			
			.26	.42	.01	.003	.62	.003	.23	.022	-			
(C-22-6)35ddb-S1	Hutton	6-19-57	4,670	5.2	-	-	-	1.8	-	4.2	-	-		
				.19	-	-	-	.56	-	.24	-	-		
8-27-58	4,850	6.5	-	-	-	-	1.7	-	3.7	-	-			
			.15	-	-	-	.59	-	.27	-	-			
(C-23-3)26aca-S1	Richfield	7-30-57	310	1.2	.26	3.0	24	.74	100	.07	-	-		
				.83	3.8	.33	.042	1.3	.010	14.3	-	-		
6- 6-66	307	1.5	.29	5.2	375	.69	67	.34	200	.050	10			
			.67	3.4	.21	.003	1.4	.015	7.1	.050	-			

Utah Geological and Mineralogical Survey Water-Resources Bulletin 13, 1970

Table 2. (continued).

Location	Name of spring	Date	Dis-solved solids (ppm)	Calcium	Sodium	Sodium	Sodium	Chloride	Chloride	Chloride	Chloride	Bromine	Bromine
				Magnesium	Calcium	Potassium	Lithium	Sulfate	Fluoride	Bicarbonate	Bromine	Iodine	
				Mg/Mg	Na/Ca	Na/K	Na/Li	Cl/50 <sub>2</sub>	Cl/FF	Cl/CO <sub>3</sub>	Cl/Br	Br/I	Br/I
				Mg/Ca	Ca/Na	K/Na	Li/Na	Cl/Cl	Cl/Cl	Bicarbonate	Bromine	Iodine	Iodine
				Mg/Ca	Ca/Na	Na/K	Li/Na	Sulfate	Fluoride	Cl/CO <sub>3</sub>	Cl/Br	Br/I	Iodine
				Mg/Ca	Ca/Na	Na/K	Li/Na	Cl/Cl	Cl/Cl	Cl/CO <sub>3</sub>	Cl/Br	Br/I	Iodine
(C-25-3)10da-S1	Nonrow	7-23-57	2,700	8.3	2.0	8.9	.117	0.70	242	1.8	-	-	-
		9-10-57	2,850	8.9	2.1	12.5	1,060	0.8	-	1.4	55	-	-
		5-17-66	2,680	15.1	2.2	9.7	1,180	0.67	222	2.0	1,500	13.7	-
(C-25-3)11ca-S1	Red Hill	5-2-66	2,630	5.9	3.0	9.0	1,050	.72	274	4.2	2,220	5.0	-
(C-25-3)15a-S	Munroe	5-3-67	2,700	5.7	2.0	11.1	1,380	.65	333	1.6	375	40.0	-
(C-25-3)27a-S	Johnson	4-19-67	628	4.7	1.6	24.3	4,600	.09	7.8	.08	140	10.0	-
(C-25-4)27-S	Joseph	7-23-57	5,150	7.8	5.1	21.3	180	1.4	448	4.1	-	-	-
		10-11-57	4,970	6.7	5.2	30.1	-	1.4	282	5.1	-	-	-
		5-3-66	5,090	5.7	3.0	18.1	98.0	1.3	367	4.1	1,400	4.0	-
		5-15-67	5,180	5.4	3.0	11.7	746	1.2	370	4.3	570	75	-
(C-26-9)34deb-S1	Roosevelt	11-4-50	7,040	5.6	100	6.5	-	58.6	-	24.1	-	-	-
		9-11-57	7,800	18	114	5.1	-	48.1	-	27.2	-	-	-
(C-30-9)7aca-S1	Radium	8-21-63	1,020	4.6	1.6	9.6	800	.15	6.6	.68	-	-	-
		3-12-66	977	4.0	1.6	9.1	1,200	.16	15.9	.40	-	-	-
		7-11-67	956	2.5	1.9	10.0	1,540	.15	15.0	.25	315	-	-
(C-30-12)21-S	Thermo	10-23-39	-	-	-	-	-	.47	-	.59	-	-	
(C-30-12)28-S	do	8-21-63	1,500	8.6	4.3	7.3	275	2.1	44	15.0	1.7	-	-
		5-25-66	1,450	7.7	4.8	8.9	398	1.5	31.0	.58	1,050	6.7	-
		7-11-67	1,470	6.3	4.8	7.7	428	1.2	35.1	.57	530	10	-
(C-40-16)6cda-S1	Veyo	3-30-66	402	2.0	.54	7.3	640	.30	50.0	.14	150	0.2	-
		4-20-67	389	1.9	.60	8.9	1,600	.33	42.9	.13	300	10	-
(C-41-13)25-S	LaVerkin	6-3-60	9,460	4.8	2.9	17.8	-	1.8	-	2.3	-	-	-
		2-5-51	9,760	4.8	2.8	13.4	-	1.9	-	2.8	-	-	-
		3-25-66	9,530	5.0	3.9	11.5	1,240	1.8	1,390	5.9	2,410	4.7	-
(D-3-4)26bca-S1	Midway	7-1-52	1,890	22.4	-	-	-	.17	-	.16	-	-	
(D-3-4)26cca-S1	do	9-28-66	1,670	4.9	2.9	4.6	407	.16	44.1	.16	120	90	
(D-3-4)27bac-S1	do	5-16-67	2,000	4.1	4.2	4.8	507	.16	45.1	.20	280	35	
(D-3-4)27had-S1	do	9-28-66	1,990	5.3	2.39	4.9	414	.17	55.2	.19	197	70	
(D-3-4)27bcd-S1	do	9-28-66	1,770	4.9	2.8	4.5	403	.16	54.8	.16	192	60	
		5-15-67	1,570	2.4	1.7	4.2	411	.16	50.0	.22	230	50	
(D-3-4)27cbd-S2	do	9-28-66	1,640	4.7	3.3	4.4	411	.16	46.8	.15	206	-	
		5-15-67	1,510	3.8	2.1	4.4	570	.17	43.8	.18	210	50	
(D-3-4)27cbd-S3	do	5-16-67	1,880	3.7	2.0	4.9	543	.19	55.7	.28	167	45	
(D-4-24)16cdd-S1	Split Mountain	9-19-68	962	3.0	-	-	-	1.4	242	1.5	-	-	
(D-7-1)5ecb-S1	Warm spring, west shore of Utah Lake	9-11-58	1,570	2.5	-	-	-	1.6	-	1.5	-	-	
(D-7-1)8bbr-S1	do	9-11-58	1,630	1.5	-	-	-	1.6	-	2.6	-	-	
(D-8-1)30da-S1	Lincoln Point	5-27-64	6,550	4.0	-	-	-	2.9	-	3.7	-	-	
		6-16-66	6,740	3.3	3.0	9.4	-	8.2	-	3.4	-	-	
(D-8-5)14d-S	Diamond Fork	10-20-67	877	3.2	1.1	14.1	1,170	.09	22.5	.16	-	-	
(D-9-4)11ba-S	Cistern	10-20-67	6,360	5.9	3.6	15.8	1,290	1.7	654	4.3	-	-	
(D-10-1)8c-S	Goshen	5-27-64	1,320	2.0	-	-	-	5.5	-	1.8	-	-	
		6-15-66	1,320	2.2	3.9	18.1	-	4.7	-	1.7	-	-	
(D-18-2)13cad-S1	Livingston	2-6-61	635	1.7	5.0	-	-	6.4	-	1.1	-	-	
(D-19-2)4dca-S1	Sterling	8-27-57	429	2.0	-	-	-	4.8	-	.11	-	-	

## WASHINGTON

Because of the basaltic and andesitic nature of the extensive Tertiary and Quaternary volcanism in Washington; hot spring activity is relatively minor. There are approximately 20 hot springs with temperatures ranging up to 190°F (88°C) which tend to be directly associated with the volcanism.

Average temperature 95-100°F (35-38°C)

Flows are up to 7,700 gpm

North slope of Mt. St. Helens - temperature 142-190°F (61-88°C)

Some fumarolic activity

Hottest springs:

### So1 Duc Hot Springs

50°C temperature

Flow - 50 gpm at 56°C

SiO<sub>2</sub> - 120 ppm

### Kennedy (Byrne) Hot Springs

Temperature - 34°C

Flow - 30 gpm at 43°C

SiO<sub>2</sub> - 380 ppm

### Gamma Hot Springs

Temperature - 60°C

Flow - 3-4 gpm

SiO<sub>2</sub> - 150 ppm

Blue Mountain Region - Walla Walla, Washington to Milton Freewater, Oregon

High temperature found in water wells

Thermal spring

Wallowa-Olympic Lineament (major structural discontinuity)

Table on page 178 and map on page 179 extracted from: Thorsen, G. W., Prospects for Geothermal Energy in Washington, First Northwest Conference on Geothermal Power, Washington State Department of Natural Resources, 18 pages. (1971).

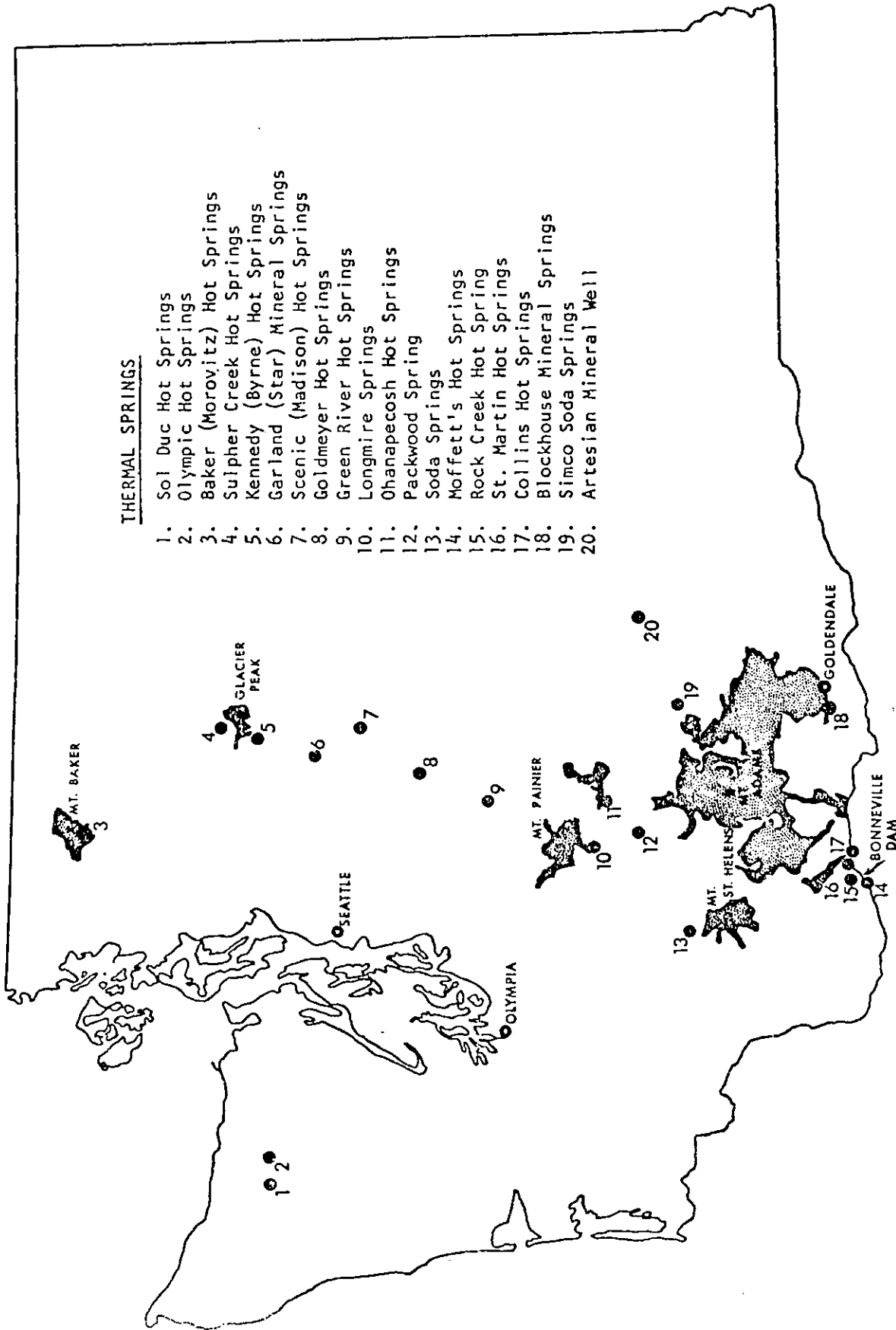
SOME THERMAL SPRINGS OF WASHINGTON

NAME	LOCATION	TEMPERATURE		FLOW (gpm)	pH	xxx SiO <sub>2</sub>	Mg	Li	Na	K	Ca	Cl	GEOLOGY and REMARKS
		Cent.	Fah.										
Olympic (1)	Clallam County SW 1/4 27 (29-8W) 7 more springs max.	26 30 48	79 86 118	135 at 52°C (2)	6-7 7 7.5	<20 30 80	<0.005 <0.005 <0.005	<0.1 <0.1 <0.1	39 51 78	0.7 0.9 1.3	3.6 2.7 1.4	<0.25 0.43 0.74	Mesozoic - Tertiary volcanics, springs along a fault zone, faint H <sub>2</sub> S, minor ppt.
Sal Duc (1)	Clallam County 1/4 W 1/4 32 (29-9W) 3 springs	50	122	50 at 56°C (2)	7.5	120	<0.005	<0.1	84	1.6	1.6	1.7	Mesozoic - Tertiary volcanics, very faint H <sub>2</sub> S
Garland (Star) (1)	Snohomish County 25 (28-11E) 2 springs	21	70	25 at 38°C (2)	6	120	74.8	75	1592	130	336	26.71	Tertiary granitics
Baker (Morovitz) (1)	Whatcom County NE 1/4 30 (38-9E)	42	108	7	8	140	<0.005	0.4	165	10.0	7.9	108	Granite overlain by Tertiary flows associated with Mt. Baker flows very faint H <sub>2</sub> S, minor ppt
Longmire (1)	Pierce County SE cor. 29 (15-8E)	21	70	6	6	170	151.2	1.8	402	37.2	298	615	Tertiary volcanics within Mt. Rainier flows.
Bonneville (Moffetts) (1)	Skamania County 16 (2-7E)	32	90	20 at 35°C (2)	9.5	<20	<0.005	<0.1	126	1.5	42	151	Tertiary andesite flows, 50 mi. from Mt. St. Helens
St. Martin's (1)	Skamania County SE cor. 21 (3-8E)	49	120	7	7	<20	<0.005	0.2	291	6.2	104	636	Tertiary volcanics, Recent vol- canics
Kennedy (Byrne) (1)	Snohomish County 1 (30-12E)	34	93	30 at 43°C (2)	7	380	60.4	37	808	67.8	228	612	Granitics overlain by Recent volcanics associated with Glacier Peak flows strong H <sub>2</sub> S, extensive ppt.
Sulphur Creek (1)	Snohomish County NE 1/4 30 (32-12E)	37	99	4 at 37°C (2)	8	120	<0.005	0.1	108	2.4	<0.2	52	Tertiary granitics, 10 miles from Glacier Peak. Strong H <sub>2</sub> S, minor ppt.
Ohanapecosh (1)	Lewis County	40	104	60 at 49°C (2)	7	80	7.5	3.3	981	50.9	85	869	Tertiary volcanics and sand- stone 10 miles from Mt. Rainier, extensive ppt.
Malotte (3)	Yakima County SE 1/4 32 (16-17E)	17	62	7.7	7.7	53	6.6	17	43	4.3	12	1.8	Tertiary volcanics (Yakima Basalt) in Wenas Syncline.
Gamma (4)	Snohomish County	60	140	3-4	7.9	150	2.6	2.6	491	77	47	86	5000' elevation on NE flank of Glacier Peak
Spring (3)	Franklin County NW 1/4 24 (12-28E)	16	60	7.9	7.9	32	43	68	11	110	110	86	Tertiary volcanics (Yakima Basalt) in Antanum-Moxee Syncline.
Worm Spring (5)	Walla Walla County NW 1/4 14 (4-37E)	22	72	50	7.9	32	43	68	11	110	110	86	Tertiary volcanics (Yakima Basalt) concealed fault?

178

i) Campbell and others, 1970  
 x Other springs listed only as "hot" or "warm" in Valentine (1960) not included.  
 xx Based on the criterion of a temperature greater than 5°C above mean annual.  
 xxx Quantities listed in parts per million.

NOTE: all data is from reference given in "Name" column unless otherwise indicated.



THERMAL SPRINGS

1. Sol Duc Hot Springs
2. Olympic Hot Springs
3. Baker (Morovitz) Hot Springs
4. Sulpher Creek Hot Springs
5. Kennedy (Byrne) Hot Springs
6. Garland (Star) Mineral Springs
7. Scenic (Madison) Hot Springs
8. Goldmeyer Hot Springs
9. Green River Hot Springs
10. Longmire Springs
11. Ohanapecosh Hot Springs
12. Packwood Spring
13. Soda Springs
14. Moffett's Hot Springs
15. Rock Creek Hot Spring
16. St. Martin Hot Springs
17. Collins Hot Springs
18. Blockhouse Mineral Springs
19. Simco Soda Springs
20. Artesian Mineral Well

Map showing areas covered by young volcanic rocks, which are considered to be favorable places to prospect for geothermal steam. The circles show locations of thermal springs.

to block the downward flow. The circles show locations of primary deposits. The arrows show secondary deposits. The circles and arrows are connected by lines to show the flow of the material.

TABLE 1.—Thermal springs in Washington

[Modified from Tables 1 and 2, Campbell and others, 1970]

Name	Location	Geology	Temp. °C	pH	Analyses <sup>1/</sup>											Remarks (presence of H <sub>2</sub> S vapor and/or ppt. at site)
					Li	Na	K	Mg	Ca	Sr	Mn	Si	Cl	CO <sub>2</sub>	SO <sub>4</sub>	
Olympic	Clallam County Sec. 27, T. 29 N., R. 8 W.	Springs located along fault zone in early Tertiary volcanic rocks.	46	7.5	<0.1	74	1.3	<0.1	1.4	<0.1	<0.1	120	0.49	X	X	Very faint H <sub>2</sub> S, minor ppt.
Gorland	Snohomish County Sec. 25, T. 28 N., R. 11 E.	Tertiary granites; 13 mi. S. of Glacier Peak.	21	6	1.4	358	27.9	21.6	90	0.3	0.25	<0.1	461	X	X	Extensive ppt.
Onanopcech	Lewis County Sec. 4, T. 14 N., R. 10 E.	Tertiary volcanic breccia, sandstone and pyroclastic; 10 mi. E. of Mt. Rainier.	40	7	3.3	981	50.9	7.5	85	0.3	<0.1	80	869	X	X	Extensive ppt.
Mt. Baker	Whatcom County Sec. 20, T. 38 N., R. 9 E.	Granite overlain by Tertiary vol- canics, assoc. with Mt. Baker flows.	42	8	0.4	165	10.0	<0.1	7.9	<0.1	<0.1	140	108	X	X	Very faint H <sub>2</sub> S, minor ppt.
Sol Duc	Clallam County Sec. 32, T. 29 N., R. 9 E.	Early Tertiary basalt, pillow basalt, and flow breccia.	50	7.5	<0.1	84	1.6	<0.1	1.6	<0.1	<0.1	120	1.7	X	X	Very faint H <sub>2</sub> S.
Longmire	Pierce County Sec. 29, T. 15 N., R. 8 E.	Tertiary volcanic breccia and pyro- clastic, within Mt. Rainier flows.	21	6	1.8	402	37.2	151.2	298	2.1	0.07	170	615	X	X	Faint H <sub>2</sub> S, some ppt.
Bonneville	Skamania County Sec. 16, T. 2 N., R. 7 E.	Tertiary andesite flows, congl., and tuff; 30 mi. S. of Mt. St. Helens.	32	9.5	<0.1	126	1.5	<0.1	42	<0.1	<0.1	<0.1	151	X	X	
Kennedy	Snohomish County Sec. 1, T. 30 N., R. 12 E.	Granites overlain by Glacier Peak flows.	34	7	3.7	808	67.8	60.4	228	0.5	0.27	380	612	X	X	Strong H <sub>2</sub> S, extensive ppt.
Sulfur Creek	Snohomish County Sec. 30, T. 32 N., R. 12 E.	Tertiary granites; 10 mi. N. of Glacier Peak on Downey Mt.	37	8	<0.1	108	2.4	<0.1	<0.1	<0.1	<0.1	120	52	X	X	Strong H <sub>2</sub> S, minor ppt.
Gamma	Snohomish County NE flank of Glacier Peak on Gamma Creek	Tertiary volcanic rocks, assoc. with Glacier Peak flows.	60	7.9	2.6	491	77	2.6	47	ND	ND	150	728	X	X	Strong H <sub>2</sub> S, some ppt.
St. Martin	Skamania County Sec. 21, T. 3 N., R. 8 E.	Tertiary volcanic rocks, within Recent flows.	49	7	0.2	291	6.2	<0.1	104	<0.1	<0.1	<0.1	636	X	X	

<sup>1/</sup> Concentrations in ppm. X = present, ND = not determined.

TABLE 2.—Mineral springs in Washington  
Modified from Tables 1 and 2, Campbell and others, 1970]

Name	Location	Geology	Temp. °C	pH	Analyses <sup>1/</sup>											Remarks (presence of H <sub>2</sub> S vapor and (or) ppt. at site)
					Li	Na	K	Mg	Ca	Sr	Mn	Si	Cl	CO <sub>2</sub>	SO <sub>4</sub>	
Diamond	King County Sec. 21, T. 21 N., R. 6 E.	Tertiary volcanic rocks; 25 mi. NW. of Mt. Rainier.	11	8	<0.1	1280	5.5	55.3	118	1.4	<0.1	<0.1	1574		X	Minor ppt.
Summit Creek	Lewis County Sec. 28, T. 14 N., R. 11 E.	Tertiary volcanic breccia and carb. shale; 14 mi. E. of Mt. Rainier.	13	6	5.9	1790	86.7	87.5	278	1.9	0.35	170	1552	X	X	Extensive ppt.
Iron Mike	Skamania County Sec. 31, T. 5 N., R. 7 E.	Tertiary volcanic rocks, within Recent flows.	10	7	0.4	211	6.2	50.1	192	0.5	0.83	40	318	X	X	Ppt. present.
Bubbling Mike	Skamania County Sec. 31, T. 5 N., R. 7 E.	Tertiary volcanic rocks, within Recent flows.	8.5	6.5	0.3	176	5.1	42.8	154	0.4	0.88	50	276	X	X	Ppt. present.
Little iron Mike	Skamania County Sec. 31, T. 5 N., R. 7 E.	Tertiary volcanic rocks, within Recent flows.	10	6.5	0.8	404	9.6	82.4	309	0.6	2.11	<0.1	561	X	X	Ppt. present.

<sup>1/</sup> Concentrations in ppm. X = present.

## WYOMING

Yellowstone National Park is the only KGRA in Wyoming. The available data on thermal areas in Wyoming implies that reservoir temperatures are not high enough for development. Within the borders of Wyoming, the occurrence of geothermal phenomena can be divided into two distinct geological provinces: 1) Yellowstone National Park, a high volcanic plateau with nearly 100 separate localities of various types of intense, high-temperature hot spring and geyser activity; and 2) the rest of the state, which is mostly non-volcanic Laramide Rocky Mountain ranges and basins has only about 20 scattered low-temperature springs. The few thermal springs known elsewhere in Wyoming have temperatures less than 60°C. Heat flow in the region is about average for the rest of western U.S., 1.5 HFU.



Table 1. Thermal Springs and Wells in Wyoming, excluding Yellowstone Park

No. on Fig. 2	Name or Location	Temp. °C	Flow gal/min	Geology	Comments and References
1	DeMaris Hot Springs. 4 miles southwest of Cody, Wyo.	24-38	50-100	Dinwoody and Park City Fm. (Permian).	Several Springs. Resort and Sanitarium. Deposits of sulfur and Travertine. Refs. 1, 14, 15, 16, 17, 18, 36, 42, 43, 54, 58.
2	T. 55 N., R. 94W. In Sheep Canyon of the Big Horn River, near mouth of Five Springs Creek.	Warm	?	Cretaceous sediments.	Several Springs. Water used locally. Refs. 16, 54.
3	T. 53 N., R. 94W. Near Upper end of Black Canyon (Wyoming) of the Big Horn River.	Warm	Small	Paleozoic sediments.	Refs. 16, 54.
4	Flagg Ranch Hot Springs (Later called Huckleberry Hot Springs). Sec. 8, T. 48N., R. 115 W.	Many boiling	100-800	Pleistocene rhyolite.	Refs. 1, 36, 42, 48, 54.
5	Astoria Springs. T. 39 N., R. 116 W.	34-40 (Main pool)	100	Permian sediments.	Hot Baths. Refs. 36, 54.
6	Granite Hot Springs. Sec. 6, T. 39 N., R. 113 W.	41-43	360 or large	Flathead sandstone (Cambrian), near granite.	Several Springs. Refs. 36, 54.

Table 1 (cont'd)

No. on Fig. 2	Name or Location	Temp. °C	Flow gal/min	Geology	Comments and References
7	Auburn Hot Springs, 2.5 miles north of Auburn, Wyo.	20-62	38	Faulted Phosphoria Fm. (Permian)	Many springs. H <sub>2</sub> S smell, salty taste. Bubbling at some springs. Sulfur and salt deposits nearby, and in neighboring subsurface section. Travertine deposits. Refs. 39, 42, 46.
8	Kendall Warm Springs, Sec. 2, T. 38 N., R. 110 W., on Green River near Wells, Wyo.	Warm	> 850	Phosphoria Fm. (Permian)	Six Springs. Ref. 36, 54.
9	Steele Hot Springs, T. 32 N., R. 107 W., Near Fremont Butte.	Hot	Small	Precambrian granite	Water used for bathing. Refs. 27, 33, 54.
10	Near Warm Spring Creek, 4 miles north- west of Dubois, Wyo.	29 (Max)	?	Phosphoria Fm. (Permian)	Several springs. Deposits of Travertine. Refs. 19, 28, 42, 45, 52, 54.
11	Near mouth of Little Warm Spring Creek, 3 miles southwest of Dubois.	20	?	Phosphoria Fm. (Permian)	Several springs. Deposits of Travertine. Refs. 19, 42, 52, 54.
12	Fort Washakie Hot Springs, Sec. 2, T. 15, R. 1 W., 24 miles west of Riverton	43+	2,000	Cnugwater Fm., Red Peak member (Triassic)	Several Springs, rising in deep pools. Refs. 1, 7, 14, 15, 18, 29, 30, 42, 49, 54.

Table 1 (cont'd)

No. on FIG. 2	Name or Location	Temp. °C	Flow gal/min	Geology	Comments and References
13	T, 30 N., R. 97 W., 4 miles southwest of Hailey.	38-49	100	Undivided Triassic sediments, near Phosphoria Fm. (Permian)	Several springs, water smells of H <sub>2</sub> S. Refs. 1, 15, 22, 42, 54.
14	T, 29 N., R. 20 W., Near Sweetwater River, 12 miles southwest of Myersville.	Warm	?	Sandstone (Oligocene)	Several springs, water used locally. Refs. 22, 54.
15	Big Horn (Thermopolis) Hot Springs on the Big Horn River at Thermopolis.	57+	12,600	Red Beds (Triassic), overlying Embar limestone.	One large spring, several small. Large deposits of Travertine. Refs. 1, 7, 8, 10, 11, 12, 14, 16, 17, 18, 24, 29, 42, 44, 51, 54, 59.
16	3.5 miles northwest of Thermopolis, near sulfur deposits.	Hot	Small	Red Beds (Triassic)	Deposits of Travertine and sulfur. Flow formerly much greater. Refs. 54, 59.
17	Sec. 35, T. 32 N., R. 36 W., on Horse Creek near Independence	Warm	Large	Oligocene strata near Chugwater Fm. (Triassic and Permian)	Several springs, water used locally. Refs. 30, 42, 54.
18	Alcova Hot Springs, T. 30 N., R. 82 W., in Fremont Canyon of the North Platte River.	50+	75	Faulted Tensleep and Amsden Fm. (Pennsylvanian)	Several springs, resort. Refs. 1, 6, 22, 42, 50, 54. Alcova Dam covers springs.

Table 1 (cont'd)

No. on Fig. 2	Name or Location	Temp. °C	Flow gal/min	Geology	Comments and References
19	Sec. 8, T. 31 N., R. 71 W., Near the North Platte River, 9 miles south of Douglas.	Warm	?	Triassic and Permian sediments, undivided (Includes Chugwater Fm.)	Water used for bathing and irrigation. Refs. 1, 54.
20	Saratoga Hot Springs. T. 17 N., R. 84 W.	48-53	10-120	North Park Formation (Late Miocene)	Resort. Several Springs. Refs. 1, 28, 41, 42.
21	Guernsey Hot Springs(?)	23+	?	Tertiary sediments (?)	Possibly mislocated by Waring (1965) (?). Refs. 4, 54.
22	Jackson Lake Hot Springs	50 (Max)	?	Madison limestone (Mississippian)	13 orifices visible above lake level at low water. Ref. 36.
23	Kelley Warm Spring	30 (Max)	> 3400	Pliocene tuff and lake sediments	Ref. 36.
24	Abercrombie Warm Spring	?	?	Fault between Ordovician and Pliocene	Ref. 36.
25	Teton Valley Warm Spring. T. 42 N., R. 115 W.	23- (Max)	> 6900	Madison limestone (Mississippian)	Ref. 36.
26	School Section Springs. T. 41 N., R. 117 W.	31+	?	Thrust fault between Cambrian on upper Crataceous	Flows 6" pipe of water. Ref. 36.
27	Bradco-Gilcrease Well. T. 39 N., R. 116 W.	48-		Triassic sediments	6" pipe flows 48-°C water from depth of 3065 ft. Ref. 36.

Table 1 (cont'd)

No. on Fig. 2	Name of Location	Temp. °C	Flow gal/min	Geology	Comments and References
28	Dallas Well T. 33 N., R. 29 W.	38-43		Madison limestone (Mississippian)	Flows 8" pipe of 38-43°C water. Ref. 36.
29	Carter Cole Well (Carter Oil Co.) T. 33 N., R. 68 W.	135		Lower Cretaceous sediments (50% shale, 50% sandstone)	Bottom-hole temperature at 11,000 ft. Ref. 36.
30	Fremont Petroleum Co. Well. T. 52 N., R. 101 W.	69-		Tensleep Sandstone (Pennsylvanian)	Bottom-hole temperature at 1476 ft (flowing water). Ref. 36.
31	SW $\frac{1}{4}$ , Sec. 20, T. 28 N., R. 118 W.	Warm(?)	8000(?)	Mesozoic and Paleozoic strata.	Several cold springs, one warm. Several travertine deposits. Ref. 23.

Table 2. Chemical analyses of thermal springs in Wyoming, exclusive of Yellowstone Park

References: (a) Love (personal communication, 1972); (b) open file, Wyoming Geological Survey; (c) White et al. (1963); (d) White (written communication, 1972).

Locality	DeMaris Hot Springs		Saratoga		Thermopolis		Auburn		Flagg Ranch Hot Springs (d)
	#WR-429(a)	#WR-430(a)	Hot Springs Hobo Pool (b)	Hot Springs #1(b)	Hot Springs #2(c)	Hot Springs #1(d)	Hot Springs #2(d)		
Flow (gal./min)	50	50	120	12,600(?)	12,600(?)	?	3 <sup>†</sup>	?	?
T, °C	~28	~28	48	57.2(?)	57.2(?)	62	15.5	71	71
pH	6.9	7.1	7.3	-	6.2	8.2	8.2	7.1	7.1
			ppm	ppm	ppm	ppm	ppm	ppm	ppm
SiO <sub>2</sub>	18	18	62	82	36	110	110	124	
Fe	.06	.2	-	.08	.04	-	-	.03	
Ca	354	369	125	385	374	134	252	12	
Mg	72	63	9	76	74	74	70	1.1	
Na	33	33	453	262	271	1410	1500	201	
K	16	16	29	49	44	156	180	7.8	
HCO <sub>3</sub>	952	993	77	766	756	94	70	372	
SO <sub>4</sub>	422	418	558	769	726	1100	1430	12	
Cl	21	20	511	328	320	1890	2000	102	
F	2	2	6.5	3.7	3.5	3.5	3.5	10	
NO <sub>3</sub>	.1	.1	6.5	.1	-	<10	6.6	.1	
B	-	-	.1	-	7.8	1.8	3.0	-	

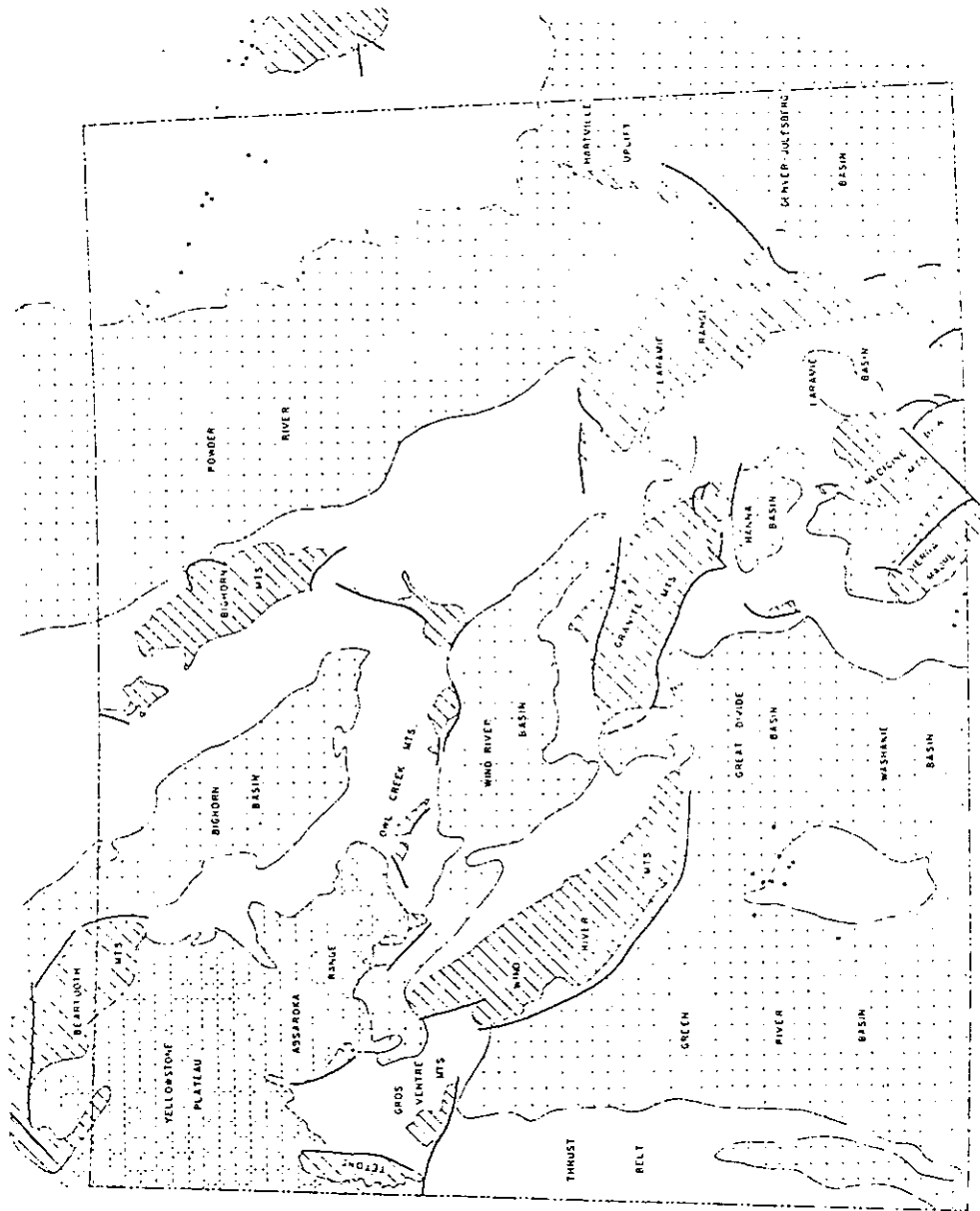


Figure 1. Generalized geology of Wyoming (after Houston (1969)).  
 Diagonal lines = Precambrian; no pattern = Paleozoic-Mesozoic;  
 dots = Tertiary; cross-hatch = Tertiary volcanic sedimentary rocks,  
 flows and tuffs; solid dots = Tertiary intrusive centers.

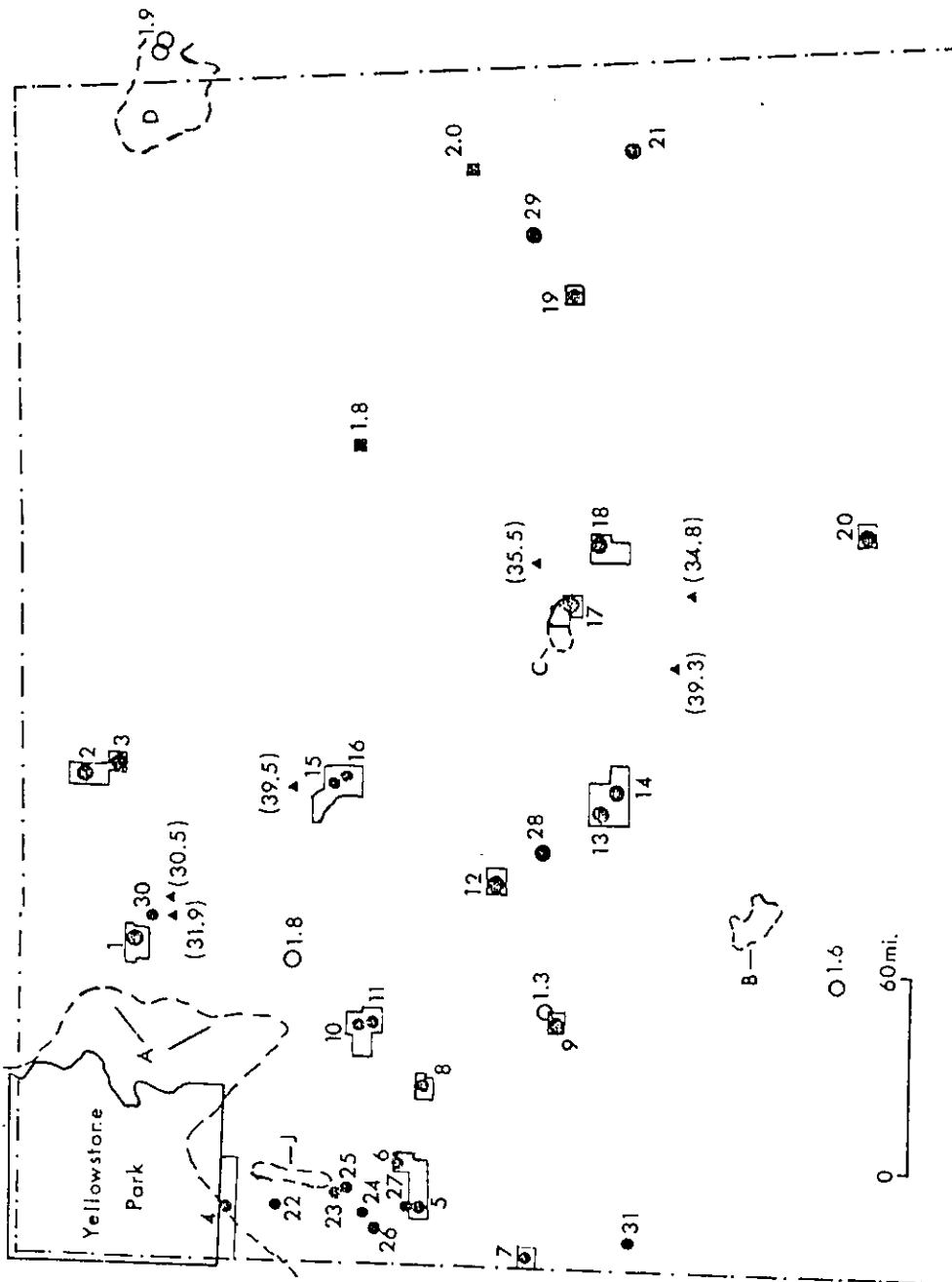


Figure 2. Thermal springs and wells, and other geothermal data in Wyoming. Solid circles represent locations of thermal springs or wells. Open circles represent locations of published heat flows (Blackwell, 1969; Sass et al., 1971); solid squares represent locations of estimated heat flows (Blackwell, 1969). Solid triangles represent locations of gradients, in brackets, determined by Blackwell (1969); Heat flows in microcal/cm<sup>2</sup>sec, gradients in °C/km. Tertiary volcanic fields (after Houston (1963)): A = Yellowstone-Absaroka; B = Leucite Hills; C = Rattlesnake Hills; D = Black Hills; J = Jackson Source Area. See Table 1 for locations of springs and wells.



## BIBLIOGRAPHY

- Bowen, R. G., 1972, Geothermal activity in 1972, State of Oregon Dept. of Geology and Mineral Industries' The Ore Bin, vol. 35, no. 1, p. 4 - 7.
- Garside, L. J., 1974, Geothermal exploration and development in Nevada through 1973, Mackay School of Mines at Univ. of Nevada, report 21, 12 p.
- Geothermal Resources Council, 1972, Geothermal overviews of the Western United States, El Centro Conference, 201 p.
- Grose, L. T., Geothermal energy: geology, exploration, and developments, Colorado School of Mines Research Institute Mineral Industries Bulletin, vol. 14, no. 1, pt. 1, 14 p.
- \_\_\_\_\_, 1972, Geothermal energy: geology, exploration, and developments, Colorado School of Mines Research Institute Mineral Industries Bulletin, vol. 15, no. 1, pt. 2, 16 p.
- Jones, P. H., 1970, Geothermal resources of the Northern Gulf of Mexico Basin, Geothermics: U. N. Symposium on the Development and Utilization of Geothermal Resources, vol. 2, pt. 1, p. 14 - 26.
- Koenig, J. B., 1970, Geothermal exploration in the Western United States, Geothermics: Proceedings of the U. N. Symposium on the Development and Utilization of Geothermal Resources, vol. 2, pt. 1, p. 1 - 13.
- Miller, T. P., 1973, Distribution and chemical analyses of thermal springs in Alaska, Open File Map.
- Mundorff, J. C., 1970, Major thermal springs of Utah, U. S. Geol. Survey Water-Resources Bull. 13, 60 p.
- Ross, S. H., 1971, Geothermal potential of Idaho, Idaho Bureau of Mines and Geology, pamphlet 150, 72 p.
- Summers, W. K., 1965, A preliminary report on New Mexico's geothermal energy resources, State Bureau of Mines and Mineral Resources, New Mexico Institute of Mining and Technology, circular 80, 41 p.
- Thorsen, G. W., 1971, Prospects for geothermal energy in Washington, First Northwest Conference on Geothermal Power, Wash. State Dept. of Natural Resources, 18 p.
- Waring, G. A., 1965, Thermal springs of the United States and other countries of the world; a summary, U. S. Geol. Survey Prof. Paper 492, 383 p.

- White, D. E., 1968, Hydrology, activity, and heat flow of the Steamboat Springs thermal system, Washoe County, Nevada, U.S. Geol. Survey Prof. Paper 458-C, 109 p.
- Young, H. W. and Mitchell, J. C., 1973, Geothermal investigations in Idaho, Idaho Department of Water Administration, Bull. no. 30, pt. 1, 43 p.
- Pacific Northwest Regional Commission, 1974, Regional Geothermal Energy Program, 38 p.
- Blackwell, D. D. and Baag, Czang-Go, 1973, Heat flow in a "blind" geothermal area near Marysville, Montana, Geophysics, vol. 38, no. 5, p. 941-956.
- White, D. E., Ivan Barnes, and James R. O'Neil, 1973, Thermal and mineral waters of nonmeteoric origin, California coast ranges, G.S.A. Bull, V. 84, p. 547-560.

## ALASKA

### Seward Peninsula

Surface temperatures average about 65°C. Silica Predicted temperatures average 100°C. Water Quality is marginal for most agricultural purposes (Total dissolved solids for warmer springs range from 2,000 to 5,000ppm.

### Interior

Surface temperatures average 55°C. Silica predicted temperatures average 70°C. Water quality is marginal for vegetable crops - about 200-1,000ppm

### Aleutian Islands

Surface temperatures are hot, - boiling point.  
Silica Predicted temperatures - 100-200°C.  
Water quality marginal for agricultural purposes - generally 1,000-2,000ppm total dissolved solids.

### Southeastern Alaska

Surface temperatures average 58°C.  
Silica predicted temperatures range up to 150°C, but average 100°C.  
Water quality is variable, ranging from 200 to 5,000ppm total dissolved solids. The quality of the hottest spring is excellent, 422ppm.

The U.S., east of Montana, Wyoming, Colorado and New Mexico, has very few significant geothermal areas. With the exceptions of Arkansas, Georgia, West Virginia and Virginia, the central and eastern portions of the U.S. are virtually devoid of thermal waters.

#### Arkansas

There are approximately 60 thermal springs in Arkansas, with 46 of these located in Hot Springs National Park. Water temperature ranges up to 147°F and flow rates up to 185 gpm.

#### Florida

Florida has one major hot spring with a temperature of 86°F and a flow rate of 4,900 gpm. The water is used for bathing.

#### Massachusetts

There is one major hot spring in Massachusetts with a maximum temperature of 76°F and a flow rate of 400 gpm. Water is bottled for table use and in manufacture of soft drinks.

#### New York

New York has one significant hot spring with a temperature of 76°F and a flow rate of 500 gpm. Water is bottled and marketed.

#### North Carolina

North Carolina has one hot spring area with water temperature up to 117°F and flow rate up to 30 gpm.

#### Pennsylvania

Pennsylvania has one hot spring with water temperature of 72°F and a flow rate of 90 gpm. This hot spring was used as a resort at one time.

#### South Dakota

Four springs with temperatures up to 90°F and flow rates up to 7,000 gpm. Water is used mainly for irrigation.

#### Texas

Texas has three main hot springs with temperatures up to 118°F and flow rates up to 45 gpm. Water is used for bathing.

#### Virginia

Virginia has approximately 20 hot spring areas with temperatures up to 106°F, averaging 70°F, with flow rates up to 2,000 gpm. Some water is bottled, but most is used locally. Springs occur in folded or faulted Paleozoic strata.

#### West Virginia

There are 30 major thermal areas in West Virginia with temperatures ranging up to 73°F and averaging 65°F. Flow rates are up to 1,630 gpm, and water is used for bathing and drinking.

# Classification of Public Lands Valuable for Geothermal Steam and Associated Geothermal Resources

By L. H. Godwin, L. B. Haigler, R. L. Rioux, D. E. White, L. J. P. Muffler, and R. G. Wayland

## ABSTRACT

The Organic Act of 1879 (43 U.S.C. 31) that established the U.S. Geological Survey provided, among other things, for the classification of the public lands and for the examination of the geological structure, mineral resources, and products of the national domain. In order to provide uniform executive action in classifying public lands, standards for determining which lands are valuable for mineral resources, for example, leasable mineral lands, or for other products are prepared by the U.S. Geological Survey. This report presents the classification standards for determining which Federal lands are classifiable as geothermal steam and associated geothermal resources lands under the Geothermal Steam Act of 1970 (84 Stat. 1566).

The concept of a geothermal resources province is established for classification of lands for the purpose of retention in Federal ownership of rights to geothermal resources upon disposal of Federal lands. A geothermal resources province is defined as an area in which higher than normal temperatures are likely to occur with depth and in which there is a reasonable possibility of finding reservoir rocks that will yield steam or heated fluids to wells.

The determination of a "known geothermal resources area" is made after careful evaluation of the available geologic, geochemical, and geophysical data and any evidence derived from nearby discoveries, competitive interests, and other indicia. The initial classification required by the Geothermal Steam Act of 1970 is presented.

## INTRODUCTION

The Organic Act of 1879 (43 U.S.C. 31) that established the U.S. Geological Survey provided, among other things, for the classification of the public lands and for the examination of the geological structure, mineral resources, and other products of the national domain. With the enactment of the Geothermal Steam Act of 1970, that authority and responsibility now includes, without limitation,

to the same extent as in classifying lands under the mineral leasing laws, the authority and responsibility to classify lands as valuable for geothermal steam and associated geothermal resources. Land is so classified in order to reserve or retain those substances in Federal ownership and to determine for the Department of the Interior which lands are included within "known geothermal resources areas" and thus subject to the competitive leasing provisions of the Geothermal Steam Act of 1970 (84 Stat. 1566).

The Geothermal Steam Act of 1970 (see "Appendix"), effective December 24, 1970, includes the following provisions:

SEC. 2(e). "known geothermal resources area" means an area in which the geology, nearby discoveries, competitive interests, or other indicia would, in the opinion of the Secretary, engender a belief in men who are experienced in the subject matter that the prospects for extraction of geothermal steam or associated geothermal resources are good enough to warrant expenditures of money for that purpose.

SEC. 4. If lands to be leased under this Act are within any known geothermal resources area, they shall be leased to the highest responsible qualified bidder by competitive bidding under regulations formulated by the Secretary. If the lands to be leased are not within any known geothermal resources area, the qualified person first making application for the lease shall be entitled to a lease of such lands without competitive bidding. \* \* \*

SEC. 21(a). Within one hundred and twenty days after the effective date of this Act, the Secretary shall cause to be published in the Federal Register a determination of all lands which were included within any known geothermal resources area on the effective date of the Act. He shall likewise publish in the Federal Register from time to time his determination of other known geothermal resources areas specifying in each case the date the lands were included in such area;

SEC. 25. As to any land subject to geothermal leasing under section 3 of this Act, all laws which either (a) provide for the disposal of land by patent or other form of conveyance or by grant or by operation of law subject to a reservation of any mineral or (b) prevent or restrict the disposal of such land because of the mineral character of the land, shall hereafter be deemed to embrace geothermal steam and associated geothermal resources as a substance which either *must be reserved* or *must prevent or restrict the disposal of such land*, as the case may be. [Italics added.] This section shall not be construed to affect grants, patents, or other forms of conveyances made prior to the date of enactment of this Act.

In order to assure uniform executive action in the classification of leasable mineral lands in the public domain, standards for determining which lands are mineral lands have been prepared from time to time by the U.S. Geological Survey. It is the duty of the Geological Survey to use geologic expertise to identify those Federal lands that are underlain by or have a reasonable expectation of containing mineral deposits or other products that meet

or exceed the minimum limits set by the classification standards. Field examination and a study of subsurface, geophysical, and geochemical data may precede classification. Similarly, all known pertinent geologic facts are considered in determining which legal subdivisions of lands are classified as geothermal steam and associated resources lands.

The purpose of this report is to present the classification standards that have been established to implement the Geothermal Steam Act of 1970. Figures 1 and 2 show those areas which include, in part, Federal lands classified as known geothermal resources areas that may be leased only competitively to the highest qualified bidder and those lands classified as valuable prospectively for the purpose of retention of geothermal rights in Federal ownership upon disposal of the lands. Table 1 lists the known geothermal resources areas effective December 24, 1970.

TABLE 1.—Known geothermal resources areas

[Number corresponds to location shown in fig. 1 or 2. Detailed land descriptions of these areas have been published in the "Federal Register," v. 36, p. 5626, March 25, 1971; v. 36, p. 6118, April 2, 1971; v. 36, p. 6441, April 3, 1971; v. 36, p. 7319, April 17, 1971; and v. 36, p. 7759, April 24, 1971]

Locality	Name	Locality	Name
	<i>Alaska</i>		<i>Nevada—Continued</i>
1.....	Pilgrim Springs	4.....	Steamboat Springs
2.....	Geyser Spring Basin and Okmok Caldera	5.....	Brady Hot Springs
	<i>California</i>	6.....	Stillwater-Soda Lake
1.....	The Geysers	7.....	Darrrough Hot Springs
2.....	Salton Sea	8.....	Gerlach
3.....	Mono-Long Valley	9.....	Moana Springs
4.....	Calistoga	10.....	Double Hot Springs
5.....	Lake City	11.....	Wabuska
6.....	Wendel-Amedee	12.....	Monte Neva
7.....	Coso Hot Springs	13.....	Elko Hot Springs
8.....	Lassen		<i>New Mexico</i>
9.....	Glass Mountain	1.....	Baca Location No. 1
10.....	Sespe Hot Springs		<i>Oregon</i>
11.....	Heber	1.....	Breitenbush Hot Springs
12.....	Brawley	2.....	Crump Geyser
13.....	Dunes	3.....	Vale Hot Springs
14.....	Glamis	4.....	Mount Hood
	<i>Idaho</i>	5.....	Lakeview
1.....	Yellowstone	6.....	Carey Hot Springs
2.....	Frazier	7.....	Klamath Falls
	<i>Montana</i>		<i>Utah</i>
1.....	Yellowstone	1.....	Crater Springs
	<i>Nevada</i>	2.....	Roosevelt
1.....	Beowawe		<i>Washington</i>
2.....	Fly Ranch	1.....	Mount St. Helens
3.....	Leach Hot Springs		

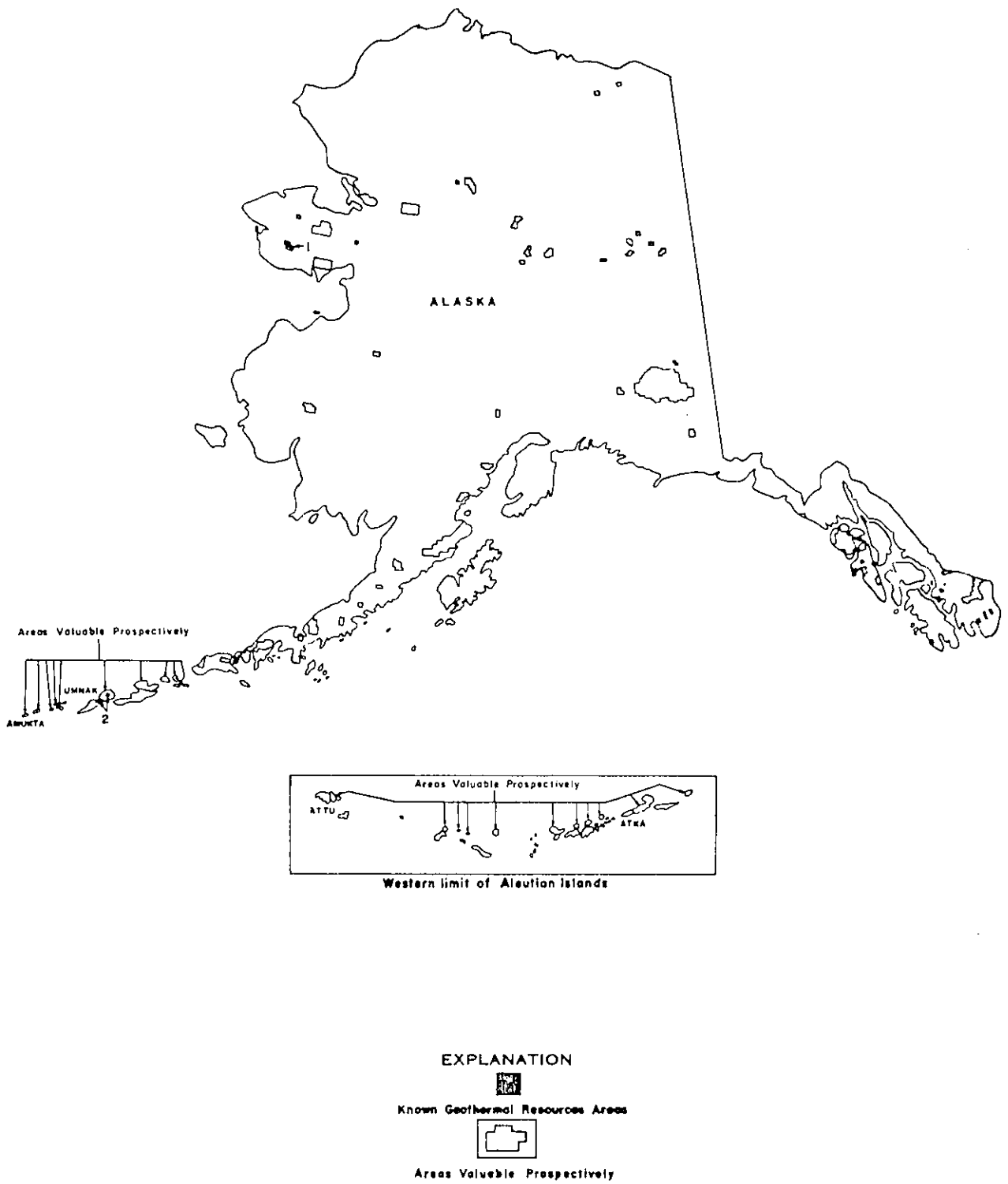


FIGURE 1.—Map of Alaska showing lands classified for geothermal resources effective December 24, 1970. Numbers correspond to areas listed in table 1.

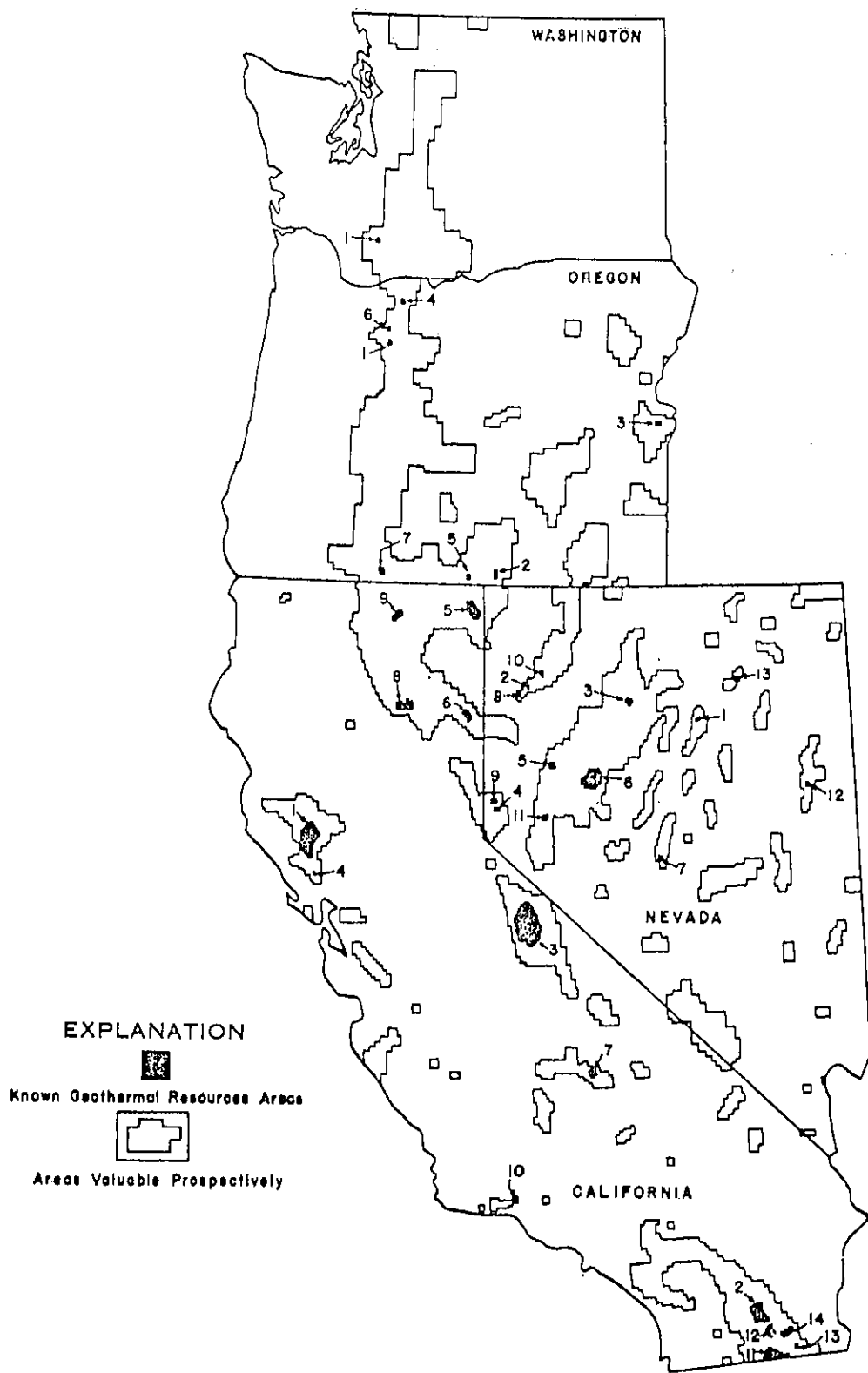
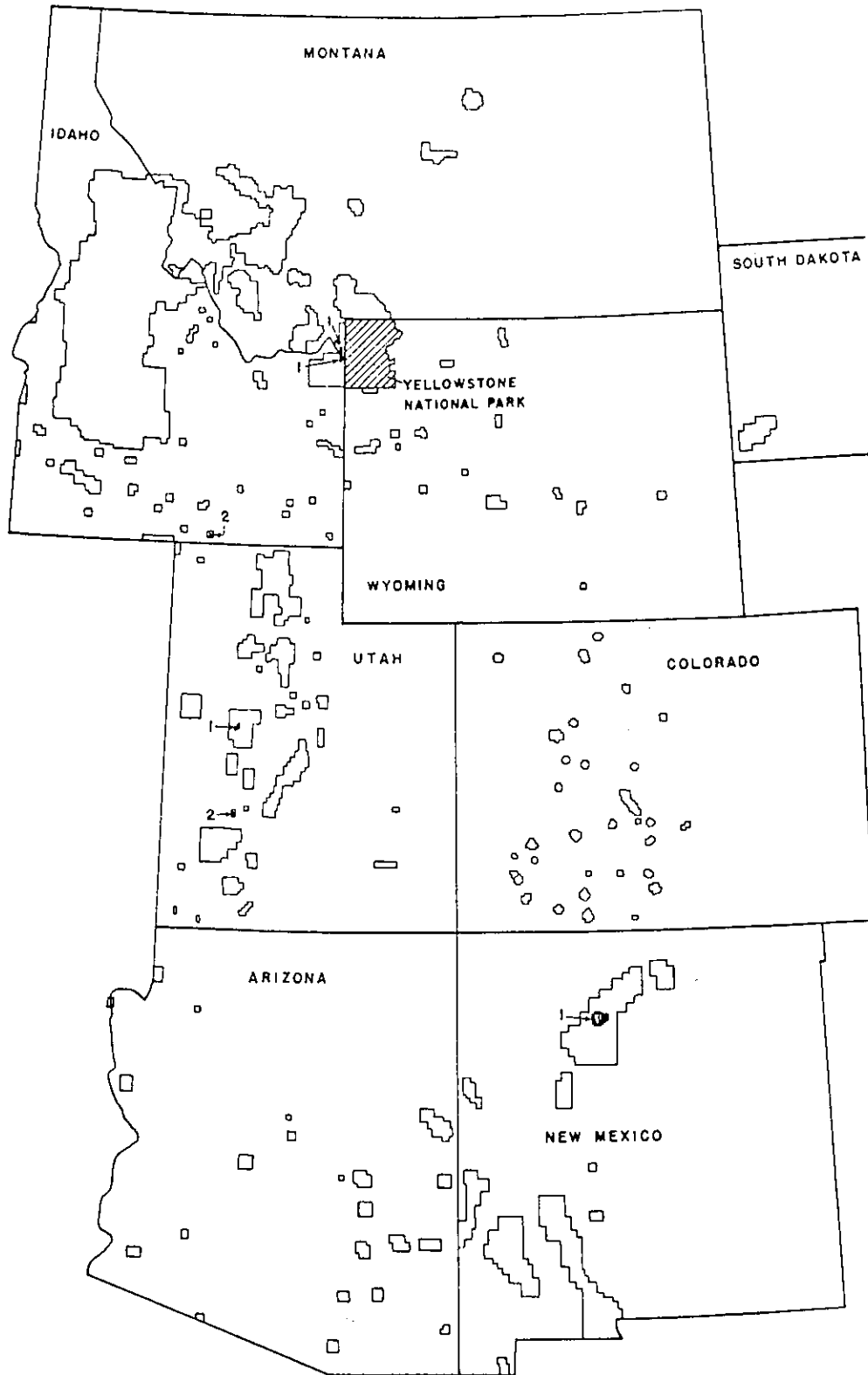


FIGURE 2.—Map of the Western United States showing lands classified for geothermal





resources effective December 24, 1970. Numbers correspond to areas listed in table 1.

Several Geological Survey personnel contributed helpful suggestions in critical reviews during the preparation of this report, and their assistance is gratefully acknowledged. The review and suggestions made by John F. Hughes, Office of the Solicitor, concerning the classification standards were especially useful.

#### NATURE OF GEOTHERMAL RESOURCES

The earth is an immense reservoir of energy, but most of this energy is contained in the earth's core and mantle at depths unlikely ever to be tapped by any foreseeable drilling technology. Within the earth at depths potentially accessible to drilling (about 6 miles) are stored approximately  $10^{24}$  British thermal units of heat (White, 1965, p. 2), but most of this heat is far too diffuse to be considered as a potential resource. However, economically significant concentrations of geothermal energy do occur in local "hot spots" where high temperatures ( $150^{\circ}$  to  $650^{\circ}$ F) are found in porous rocks containing liquid water and (or) steam; such concentrations of extractable heat are known as "geothermal reservoirs." The reservoirs are found in regions of recent volcanism and mountain-building and in the deep parts of many sedimentary basins.

The energy in a geothermal reservoir consists of heat, largely stored in rocks and to a lesser extent in liquid water and (or) steam-filling pores and fractures. The water and steam provide the means by which heat from deep sources is transferred by convection to depths shallow enough to be tapped by drilling. Water and steam also serve as the agents by which geothermal heat escapes at the surface in hot springs and fumaroles and through which geothermal heat can be tapped commercially by wells.

The fluid in most geothermal reservoirs is liquid water (White and others, 1971) that is held at temperatures above surface boiling by the confining pressure. Decrease in pressure upon withdrawal of the liquid water causes steam to form by boiling, and a mixture of steam and water is produced at the surface. A few reservoirs contain primarily steam, and the wells produce dry or superheated steam with no water. These dry steam reservoirs are

positively known only in the Larderello-Mt. Amiata region of Italy and at The Geysers, Calif.

For a geothermal reservoir to have appreciable potential for exploitation, it must meet the following requirements: (1) relatively high temperature (greater than  $150^{\circ}$  to  $400^{\circ}$ F, depending on processing technology), (2) a depth shallow enough to permit drilling (currently 10,000 ft or less), (3) sufficient rock permeability to allow the heat transfer agent (water and (or) steam) to flow continuously at a high rate, and (4) sufficient water recharge to maintain production over many years.

Limited exploitation of geothermal resources has occurred since the turn of the century, primarily to generate electric power. Geothermal resources also have been used for space heating, product processing, and agricultural heating, and in addition some geothermal fluids contain chemicals and metals that are potentially valuable byproducts. Furthermore, geothermal energy appears to have an important potential use in desalination, either of the geothermal fluid itself or of other saline waters that may occur near a source of geothermal energy.

#### PRESENT STATE OF GEOTHERMAL KNOWLEDGE

Geothermal areas exist throughout the world, primarily along the belts of young volcanism that ring the Pacific Ocean and that follow the midoceanic ridges. Geothermal areas of the United States are found primarily in the Western States, along the circum-Pacific belt of young volcanism and mountain-building and where the Pacific ridge system (a locus of high heat flow) intersects the North American continent along the Gulf of California and the Imperial-Coachella Valley of California. In the Eastern United States, potentially economic reservoirs of geothermal heat have been identified in the deep parts of the Gulf of Mexico sedimentary basin.

The distribution, extent, and magnitude of the geothermal resources of the United States are, at present, poorly known. The general extent of the resource in the Western States can be inferred from the distribution of hot springs and in a more general way from the distribu-

tion of young volcanic rocks. In the past, geothermal exploration was primarily on sites identified by hot springs (an exploration method analogous to the primitive oil exploration methods of the turn of the century when oil fields could be located only by finding surface oil seeps). Available geologic and geochemical techniques have not been used adequately in discovering and evaluating new fields, and geophysical principles and techniques are only now beginning to be adapted to geothermal exploration.

Knowledge of the characteristics and parameters of individual geothermal systems in the United States has come from exploratory drilling in about 40 hot-spring areas (J. B. Koenig, "Geothermal Exploration in the Western United States": Paper II/19, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September 1970), from developmental drilling (mainly at The Geysers of northern California and the Salton Sea in southern California), and from scientific studies of a few major thermal-spring areas, supplemented by shallow research drilling. Extrapolation of knowledge from other countries, mainly New Zealand, Italy, and Iceland, has proven useful, but techniques for estimating the size and power potential of geothermal sites prior to drilling are only beginning to be developed. Techniques for efficiently utilizing all the energy produced from geothermal wells (that is, by desalination or by two-phase generation) are only beginning to be investigated and developed.

#### CLASSIFICATION FACTORS FOR RETENTION

The petroleum province is an established concept in classification of oil and gas lands for purposes of retention of Federal mineral rights upon disposal of public lands. A geothermal resources province (GRP) similar in concept to a petroleum province is necessary to define those areas valuable prospectively for geothermal steam and associated geothermal resources.

A geothermal resources province is an area in which higher than normal temperatures are likely to occur with depth and there is a reasonable possibility of finding reservoir rocks that will yield steam or heated fluids to wells.

In most prospective areas, data on geothermal gradients and conductive heat flow are scarce. Adequate temperature-depth data exist only in sedimentary basins that have been extensively explored for oil and gas. Most of these basins are characterized by nearly "normal" geothermal gradients rather than the abnormally high rates needed for development of geothermal energy.

The present-day use of geothermal energy includes generation of electricity, manufacturing, agriculture, and space heating. The minimum present-day use for geothermal resources is the exploitation of stored heat energy for space heating. Geothermal fluids used for space heating are generally delivered at above 100°F and are available at the surface or at shallow depths below the surface. S. S. Einarsson ("Utilization of Low Enthalpy Water for Space Heating, Industrial, Agricultural and Other Uses": Rapporteur rept., Sec. X, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September 1970) reported use of 118°F water in space heating in Olafsfjordhur, Iceland, and 104°F water in Japan. S. H. Ross ("Geothermal Potential of Idaho": Paper II/1, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September 1970) reported use of 104°F water for agricultural purposes in Idaho. Because a heat exchanger can extract the heat from the geothermal fluid, the chemical composition or the corrosiveness of the fluid is not necessarily a controlling factor in classification.

The classification of geothermal resources provinces is based on geologic inference similar to that used by the Geological Survey in classifying lands for retention of oil and gas mineral rights and should provide adequate protection against alienation of leasable geothermal resources. One or more of the following indicia are necessary for the retention classification of lands in geothermal resources provinces:

1. Volcanism of late Tertiary or Quaternary age—especially caldera structures, cones, and volcanic vents.
2. Geysers, fumaroles, mud volcanoes, or thermal springs at least 40°F higher than

average ambient temperature.

3. Subsurface geothermal gradients generally in excess of two times normal, as reflected in deep water wells, oil well tests, and other test holes.

#### CLASSIFICATION FACTORS FOR A KNOWN GEOTHERMAL RESOURCES AREA

Lands shall be classified as a "known geothermal resources area" (KGRA) when "the prospects for extraction of geothermal steam or associated geothermal resources from an area are good enough to warrant expenditures of money for that purpose." The accumulation of geothermal resources is in some ways similar to the accumulation of oil and gas resources, and only a test hole can establish with certainty the existence of adequate temperatures, pressures, and production capacity of an area. However, the definition of a "known geothermal resources area" departs from the concept of a "known geologic structure of a producing oil or gas field" (Finley, 1959) in that it does *not* require a producible well. Thus, any relevant data and information pertaining to the criteria enumerated in sec. 2(e) of the Geothermal Steam Act can be considered in determining whether lands are included within any KGRA.

The extent of a KGRA is influenced by such geologic factors as the pattern of temperature gradient, structure, stratigraphy, porosity, conductivity, permeability, heat source, and rate of recharge of fluids. The determination of a KGRA is made after evaluating the net effect of all geologic, geochemical, and geophysical data and any evidence derived from nearby discoveries, competitive interests, and other indicia.

#### GEOLOGY (INCLUDING GEOPHYSICAL AND GEOCHEMICAL DATA)

The following kinds of data, considered together, indicate that the prospects for extraction of geothermal steam or associated geothermal resources are good enough to warrant expenditures of money for that purpose:

1. Siliceous sinter and natural geysers both imply high subsurface temperatures, generally 350°F or greater (D. E. White, "Geochemistry Applied to the Discovery,

Evaluation and Exploitation of Geothermal Energy Resources": Rapporteur rept., Sec. V, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September 1970) in hot-water systems, because of relationships generally existing between temperature and SiO<sub>2</sub> content of liquid water.

2. The temperatures of fumaroles, thermal springs, and mud volcanoes provide minimum subsurface temperatures.
3. The SiO<sub>2</sub> content of spring water is a very useful chemical geothermometer for indicating the reservoir temperatures of many hot-water systems (R. O. Fournier and A. H. Truesdell, "Chemical Indicators of Subsurface Temperature Applied to Hot Spring Waters of Yellowstone National Park, Wyoming, U.S.A.": Paper V/2, United Nations Symposium on the Development and Utilization of Geothermal Resources, Pisa, Italy, September 1970; D. E. White, see above).
4. The Na/K ratio in spring waters of many hot-water systems is also a useful chemical geothermometer when there is adequate knowledge of competing influences (D. E. White, see above).
5. Most known potential geothermal systems occur in or near volcanoes and calderas of late Tertiary or Quaternary age.
6. Abnormally high conductive heat flow and the geothermal gradient are the best indicators of deep, concealed geothermal reservoirs. Although specific limits have not yet been established, two to 10 times the world-wide average (heat flow of 1.5 microcalories per cm<sup>2</sup> per second; temperature gradient of 1°F per 100 ft) extended consistently over hundreds of feet of depth appears favorable.
7. The porosity and the permeability of a potential reservoir are important parameters but can be established only by drilling and testing. Where stratigraphic control of the reservoir fluid or steam by a caprock is expected, near-surface characteristics of the rocks may provide preliminary evaluations.
8. Electrical resistivity surveys are probably

the best geophysical means to geothermal evaluation available at this time, especially for the hot-water systems.

9. Magnetic, gravity, and airborne infrared geophysical surveys may provide useful supplemental data.
10. Other geophysical methods such as microseismic, seismic ground noise, electromagnetic, and telluric surveys may have significant future use in evaluation.

#### NEARBY DISCOVERIES

In classifying land as a KGRA, the discovery of a deposit of geothermal steam or associated geothermal resources in the vicinity of such lands is evaluated together with the available data concerning the other criteria enumerated in sec. 2(e) of the Act which are to be considered in classification action.

#### COMPETITIVE INTERESTS

Competitive interest is considered together with the available data concerning the other criteria enumerated in sec. 2(e) of the Act in classifying lands as being within a KGRA.

Available information which could be considered in determining the existence of competitive interest in connection with an application for a geothermal lease would include information concerning the existence of bona fide, allowable applications for geothermal leases which have been filed for all or any part of the lands sought under the application being considered, or for lands in the vicinity of the lands being sought. The circumstance that two or more companies are exploring, applying for, or actually leasing available State or fee lands for geothermal resources in the same general area might constitute competitive interest that

would affect Federal lands considered valuable prospectively for geothermal resources and could warrant its classification as a KGRA. The absence of indicated competitive interest, however, is, in and of itself, no bar to classification of lands for inclusion in a KGRA and would not be sufficient to warrant revocation of a KGRA.

#### OTHER INDICIA

Any pertinent engineering and (or) economic data may be considered together with other available data relating to the criteria enumerated in sec. 2(e) of the Geothermal Steam Act in classifying land for inclusion in a KGRA.

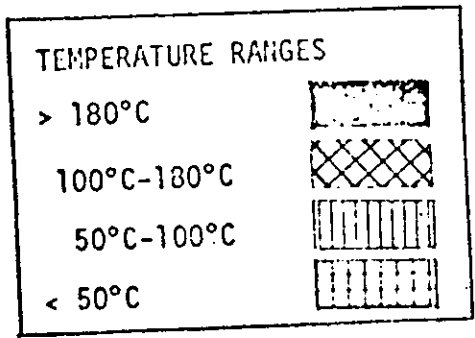
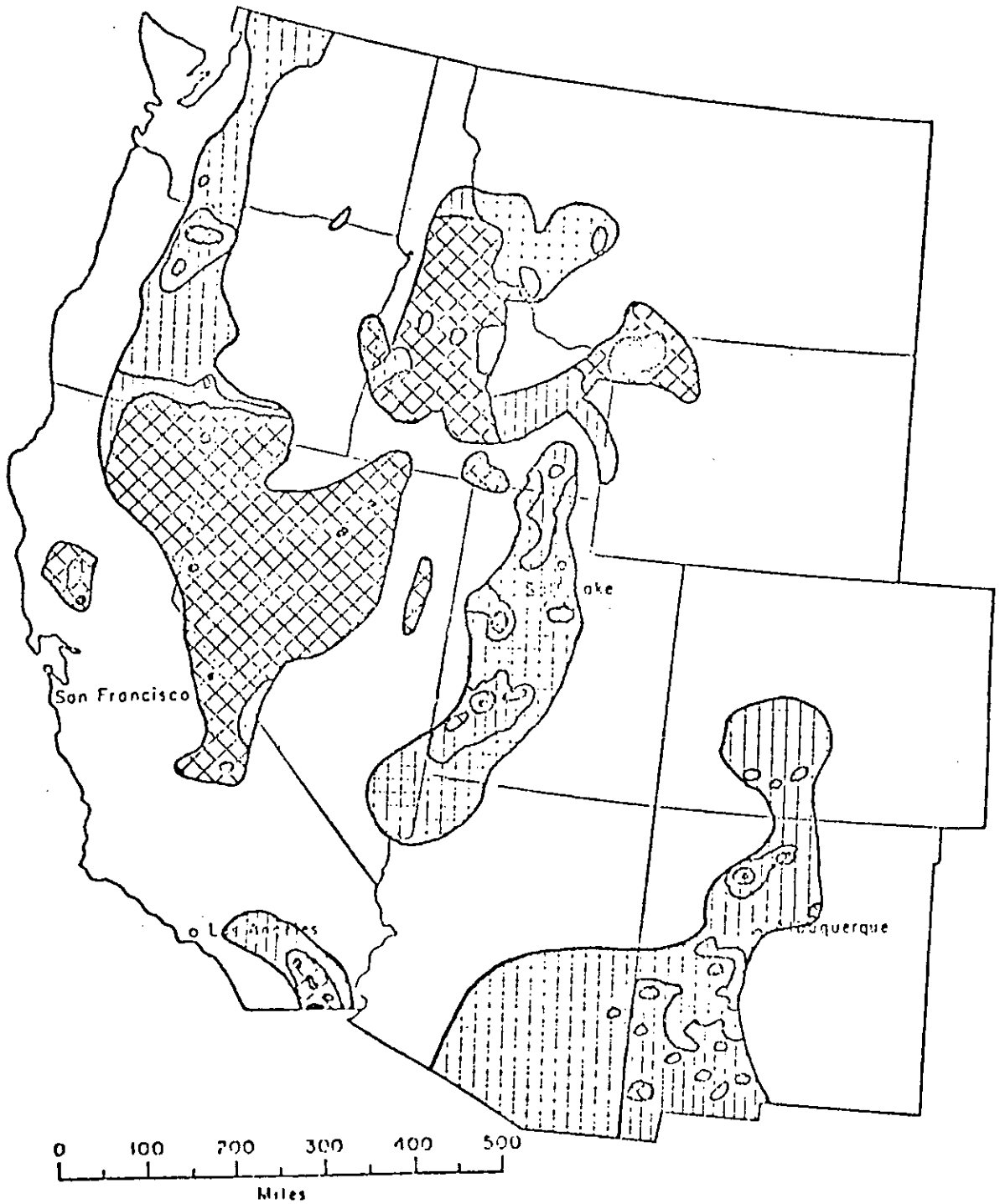
In defining the lands valuable for geothermal resource development under the proposed withdrawal published in the "Federal Register," v. 32, p. 4506, March 24, 1967, the Director of the Geological Survey used primarily a combination of the then known geologic and geophysical data as well as temperature and chemical data, in part supplied by industry, from areas that had been drilled in exploration for geothermal steam. Future developments will provide substantially more geologic and geophysical information as well as engineering and economic data. These factors will be considered in future determinations of KGRA's.

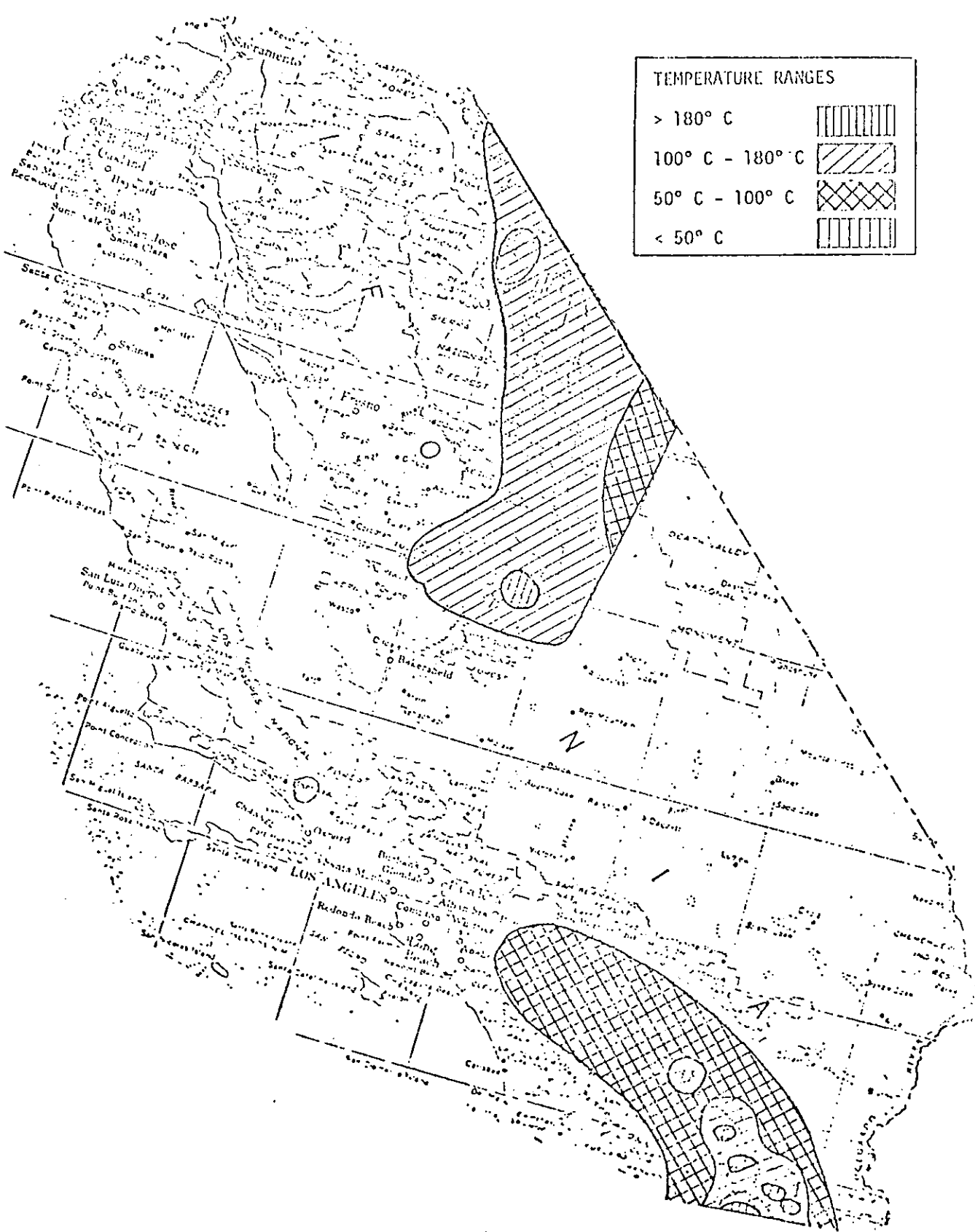
#### REFERENCES CITED

- Finley, E. A., 1959, The definition of known geologic structures of producing oil and gas fields: U.S. Geol. Survey Circ. 419, 6 p.
- White, D. E., 1965, Geothermal energy: U.S. Geol. Survey Circ. 519, 17 p.
- White, D. E., Muffler, L. J. P., and Truesdell, A. H., 1971, Vapor-dominated hydrothermal systems, compared with hot-water systems: Econ. Geology, v. 66, no. 1, p. 75-97.

Appendix II.

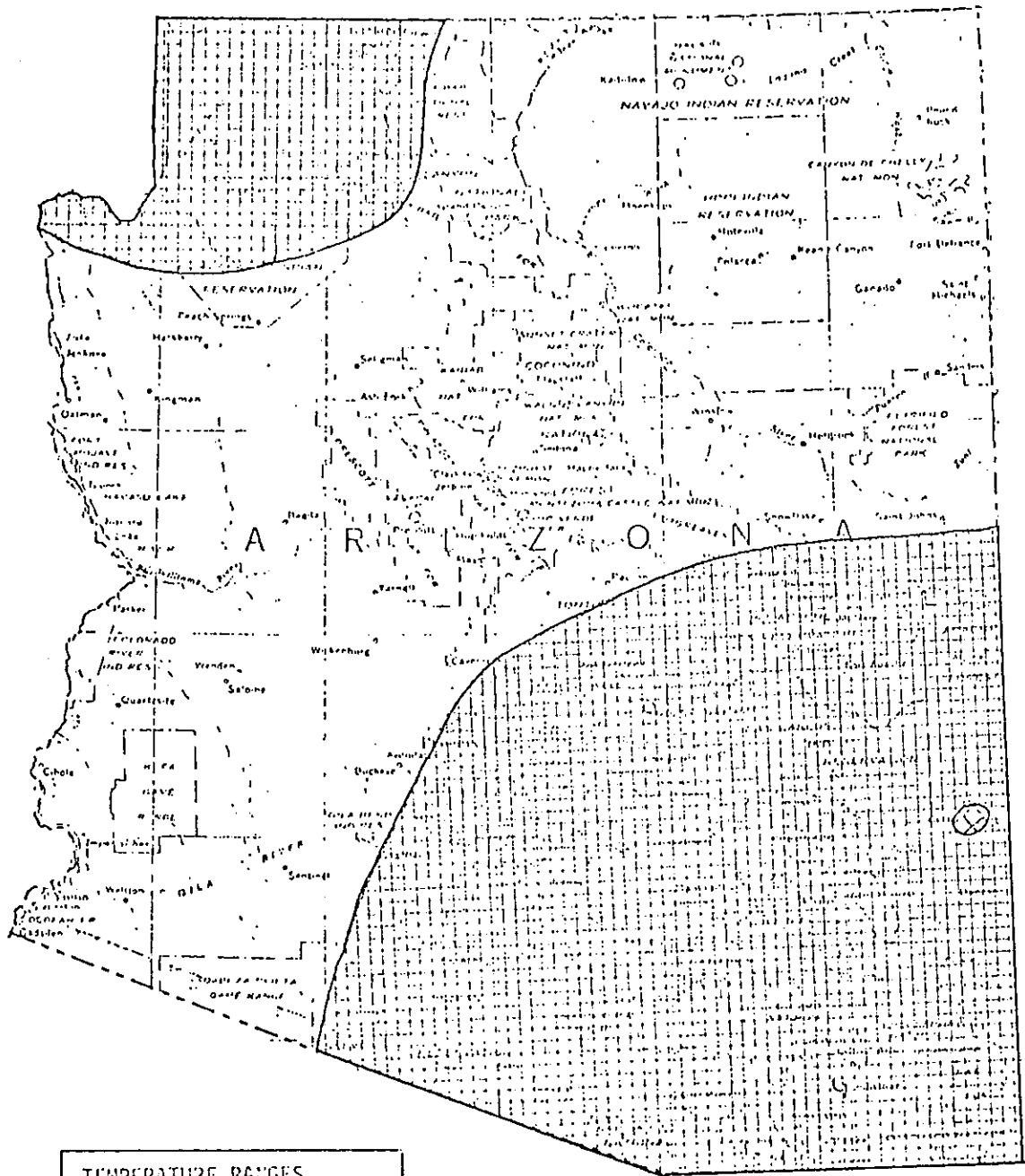
MAP CLASSIFICATION OF AREAS BY TEMPERATURE RANGE, WATER QUALITY  
AND BORON CONTENT





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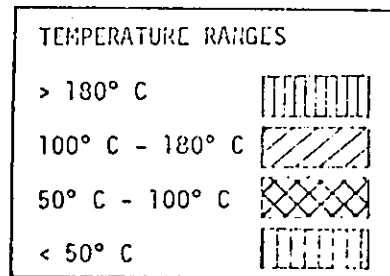
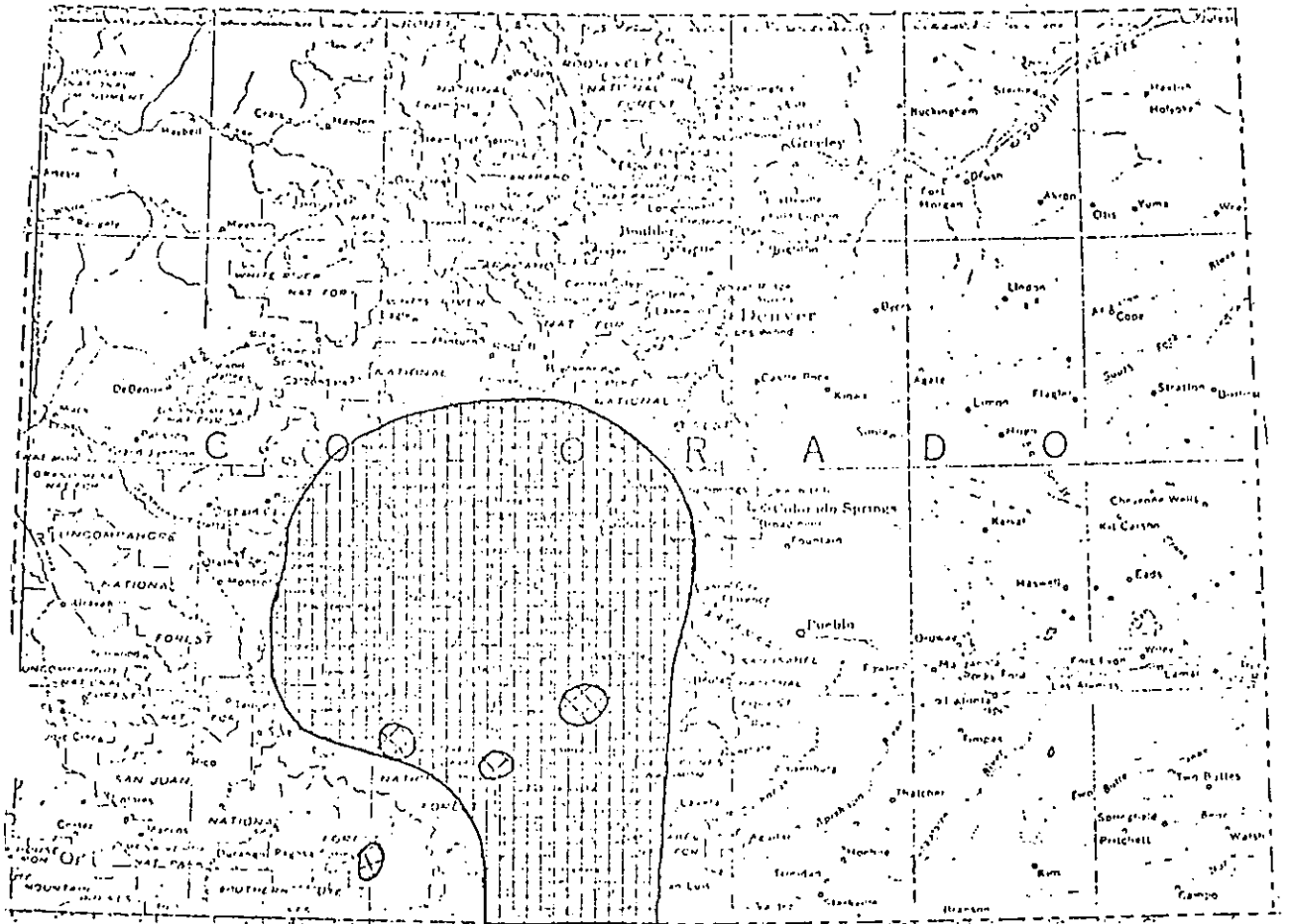


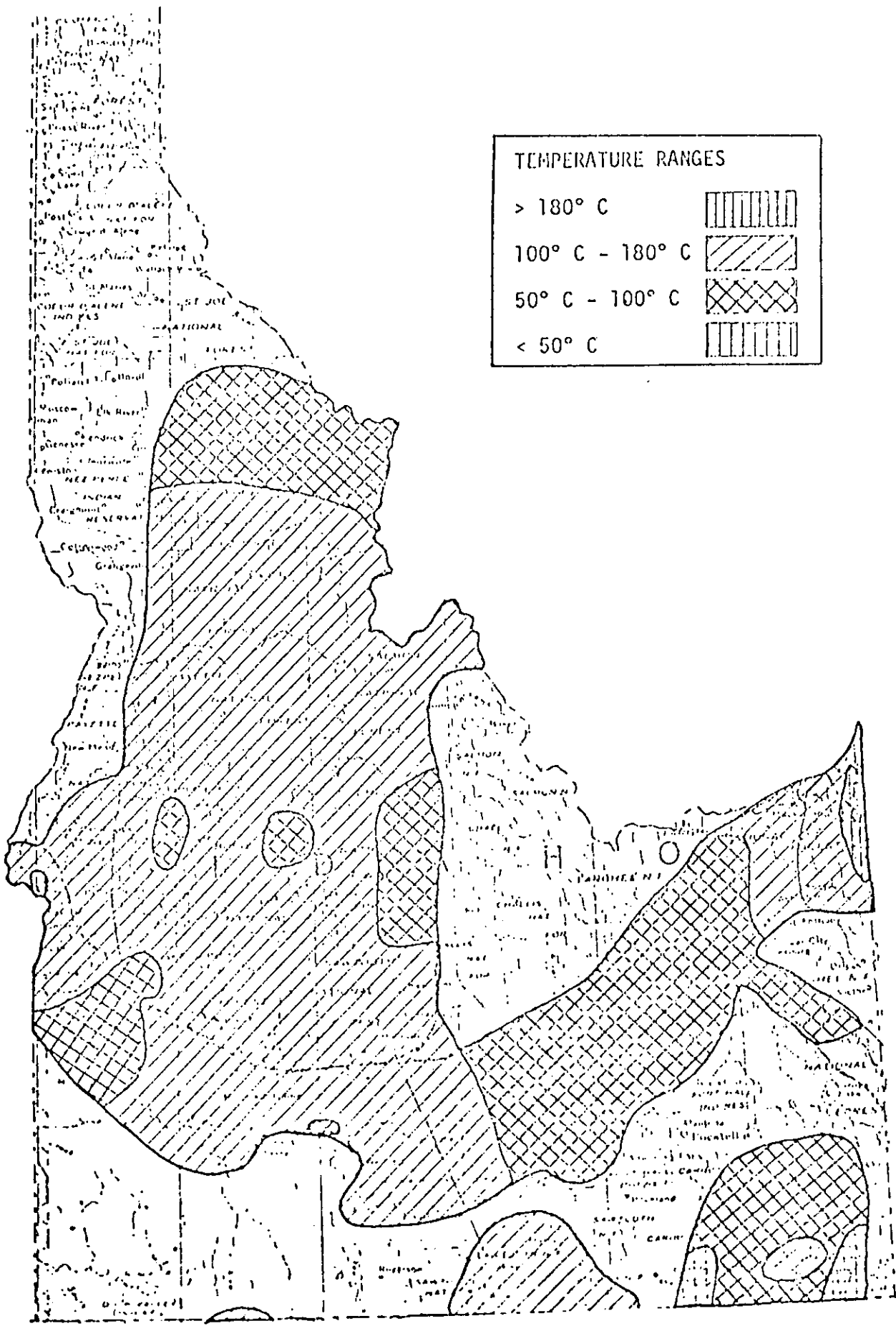


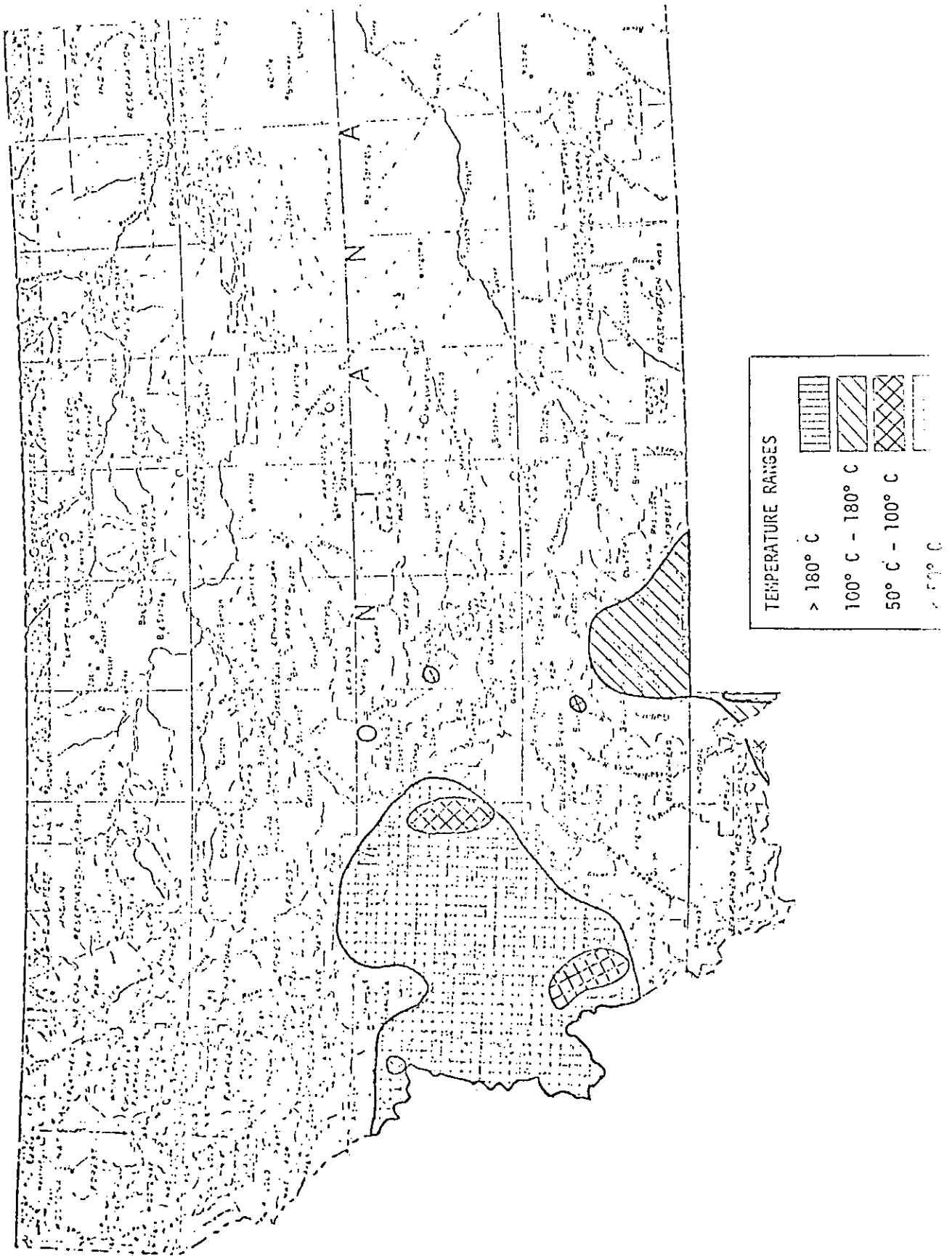
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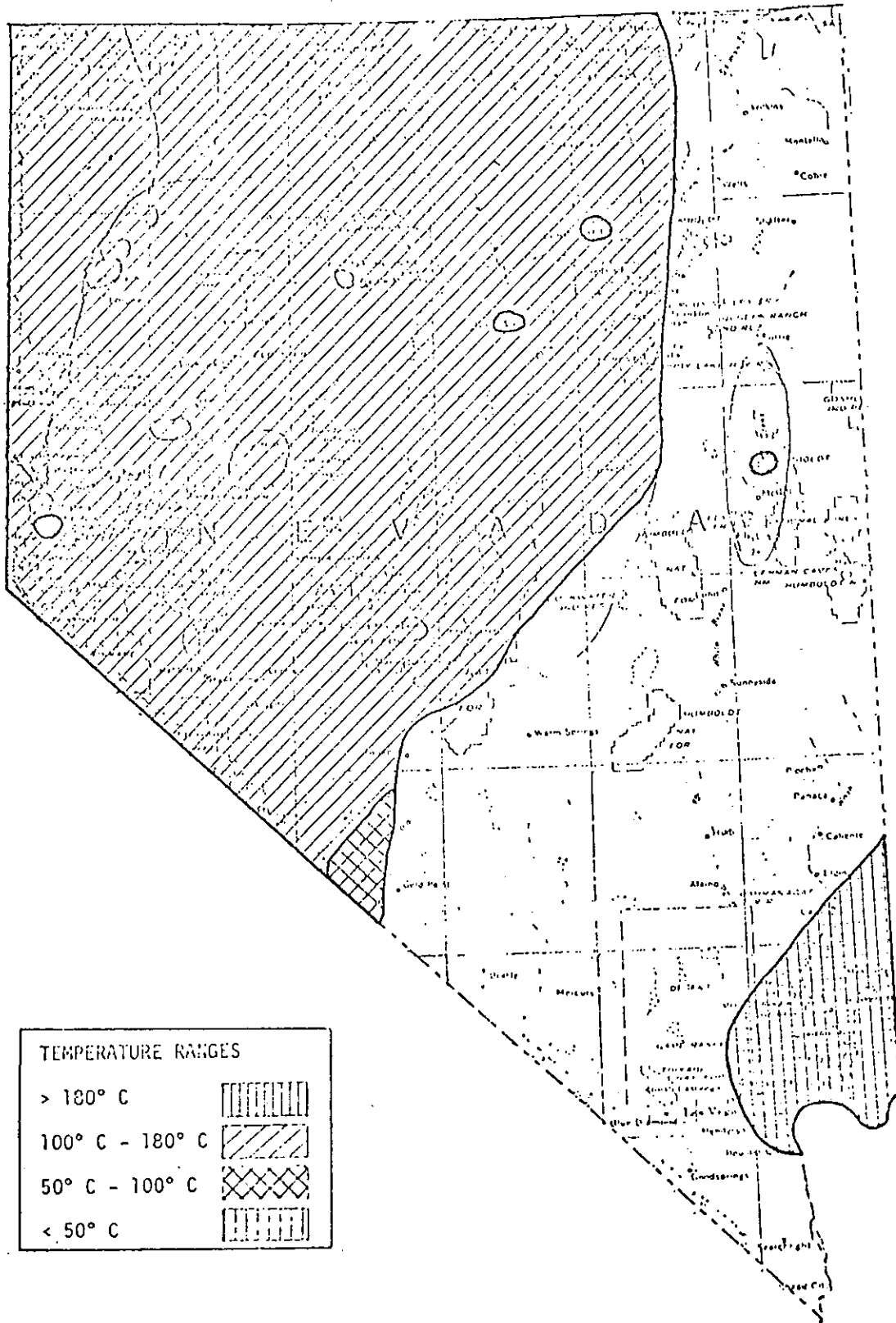


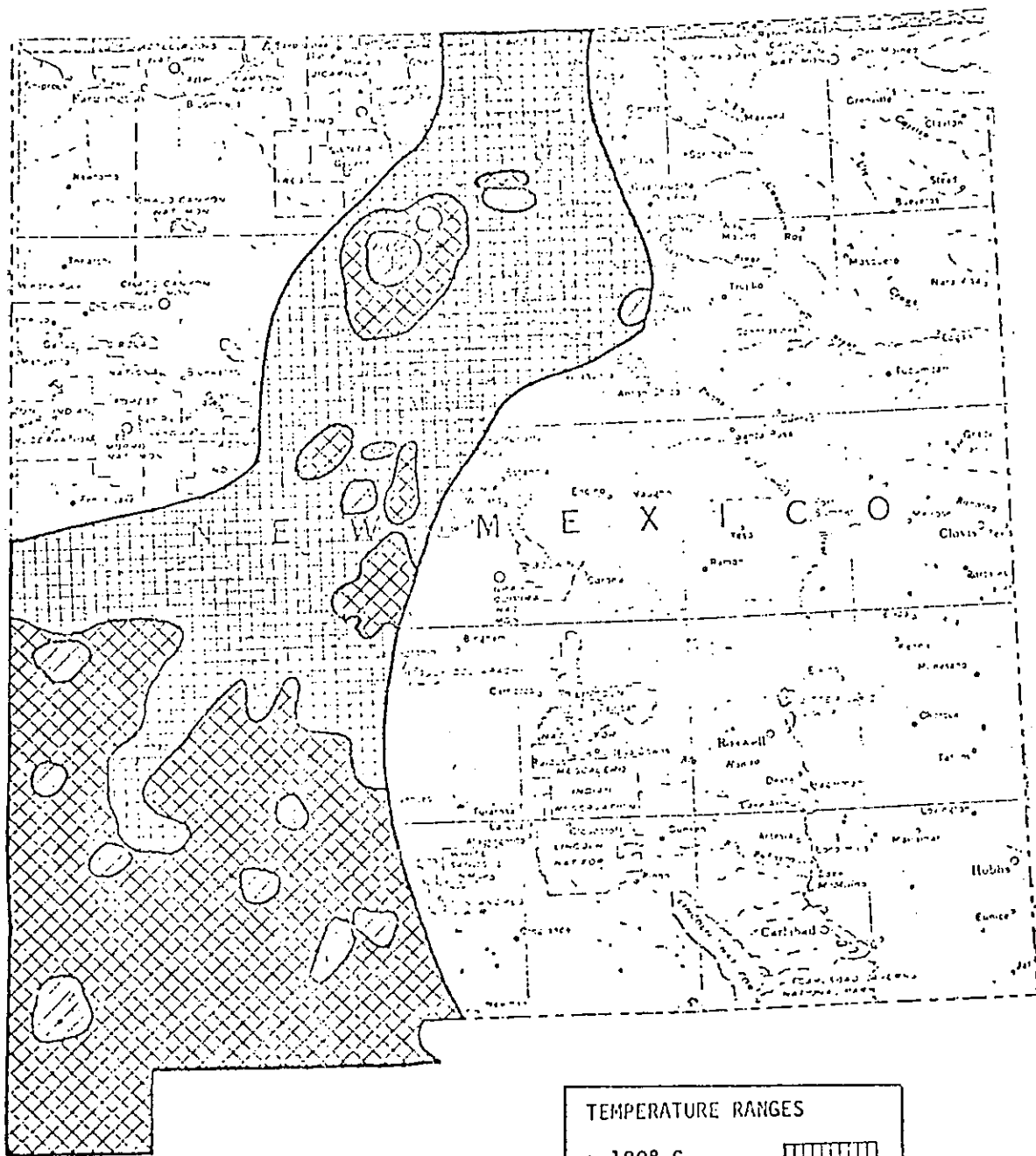
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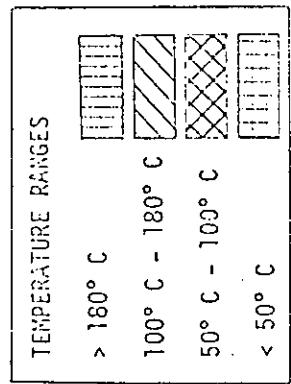
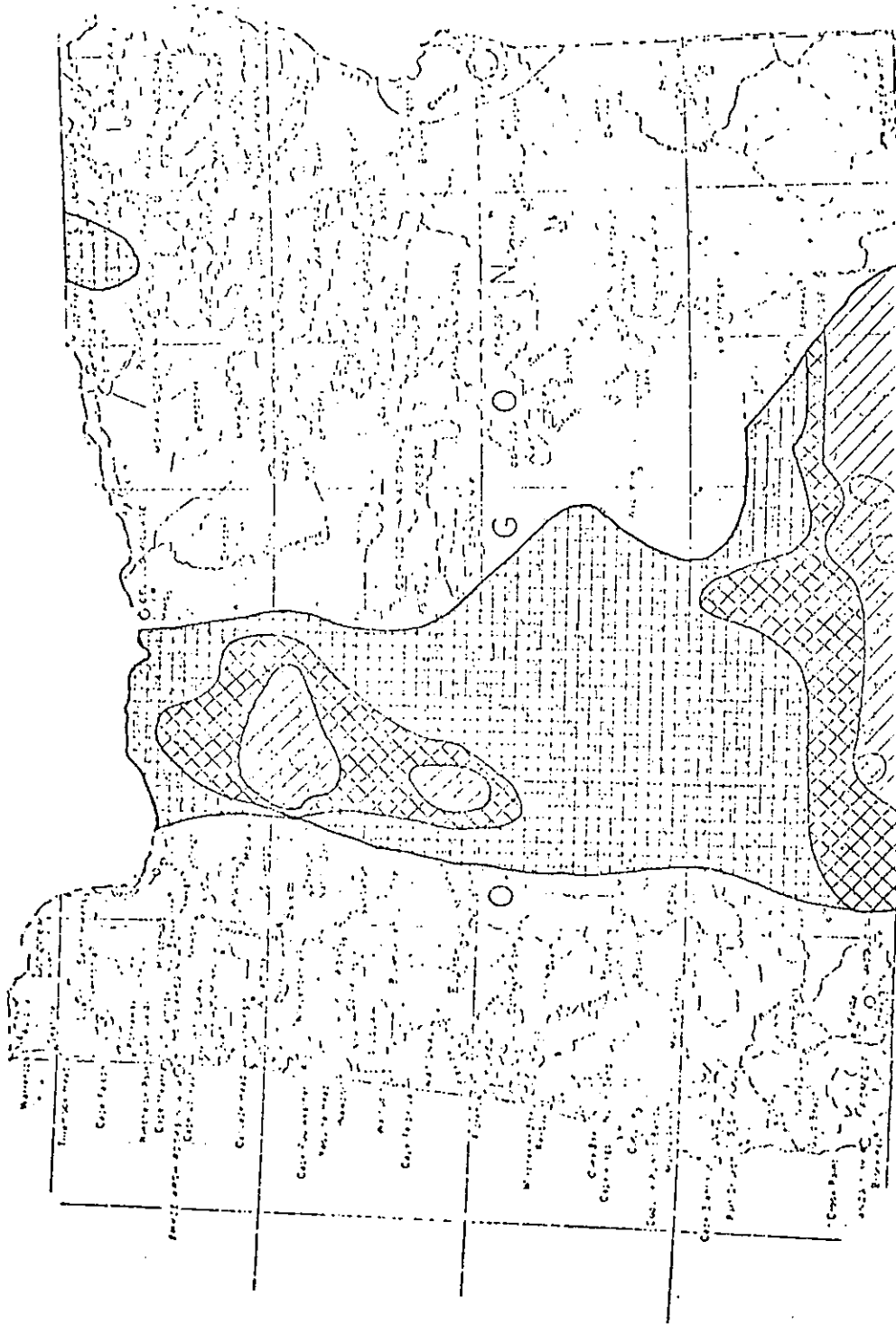












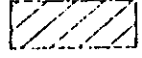


TEMPERATURE RANGES

> 180° C



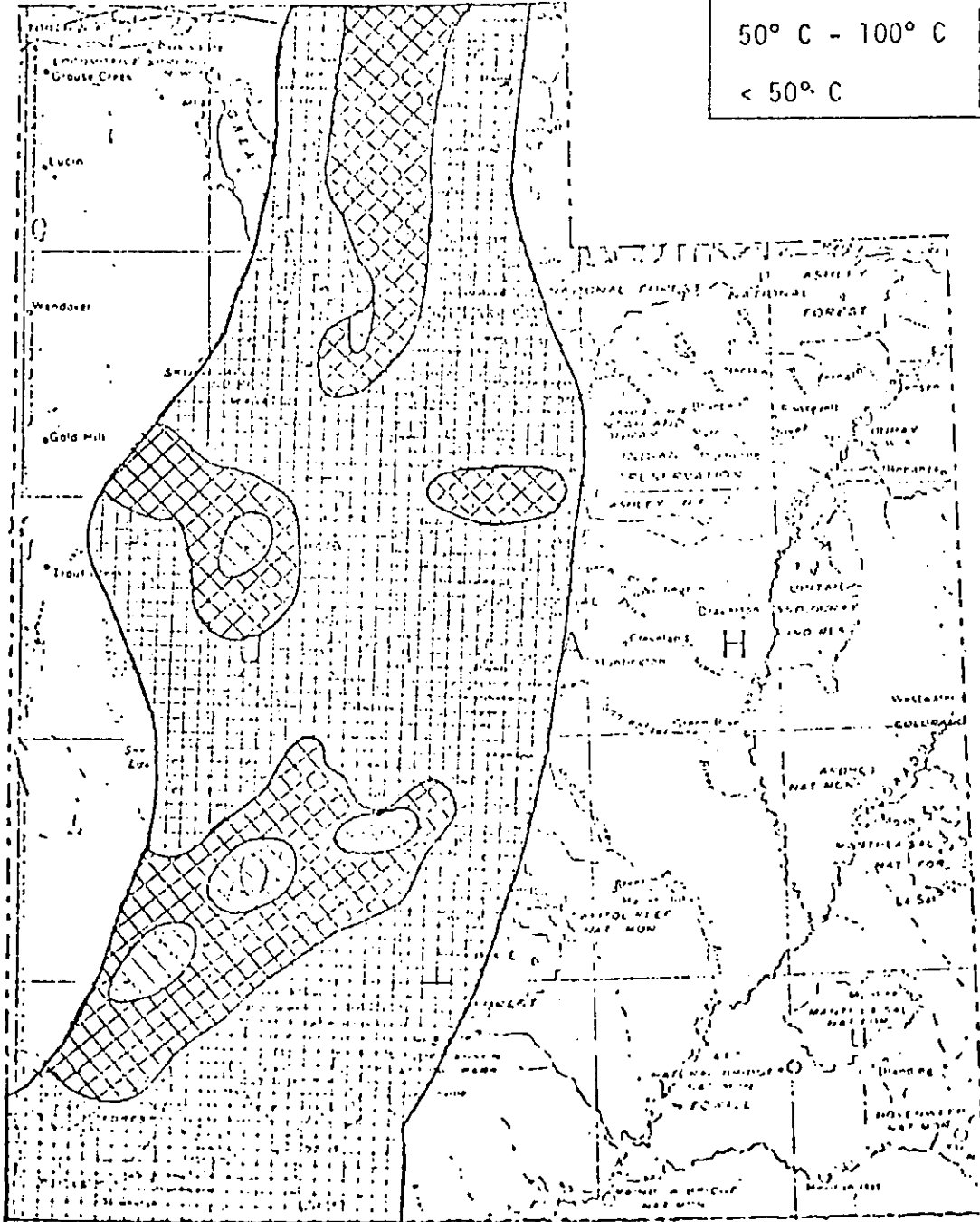
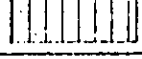
100° C - 180° C

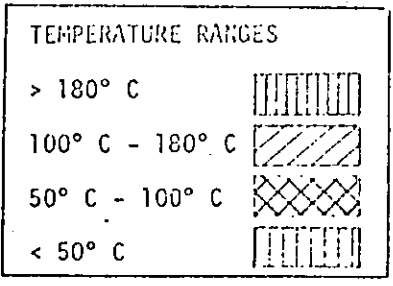
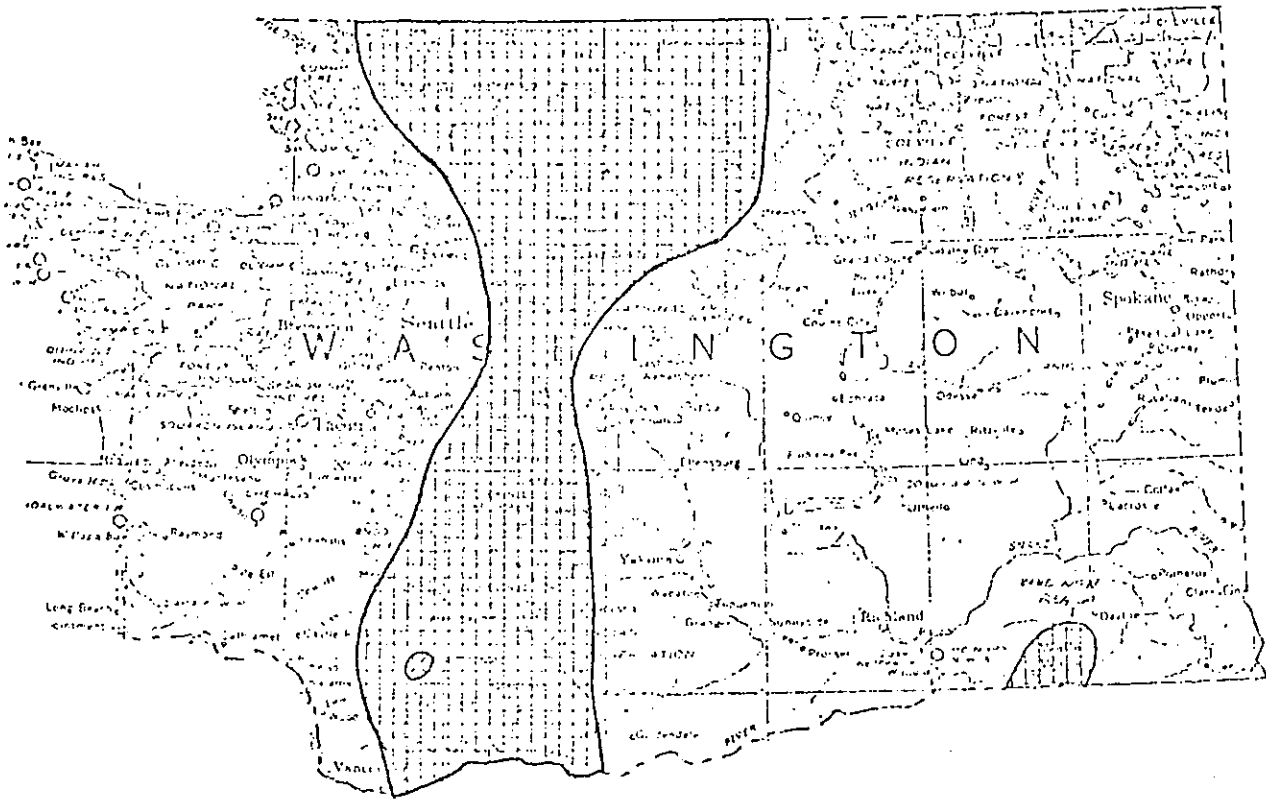


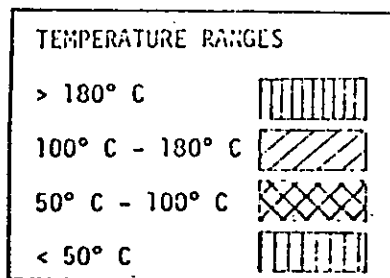
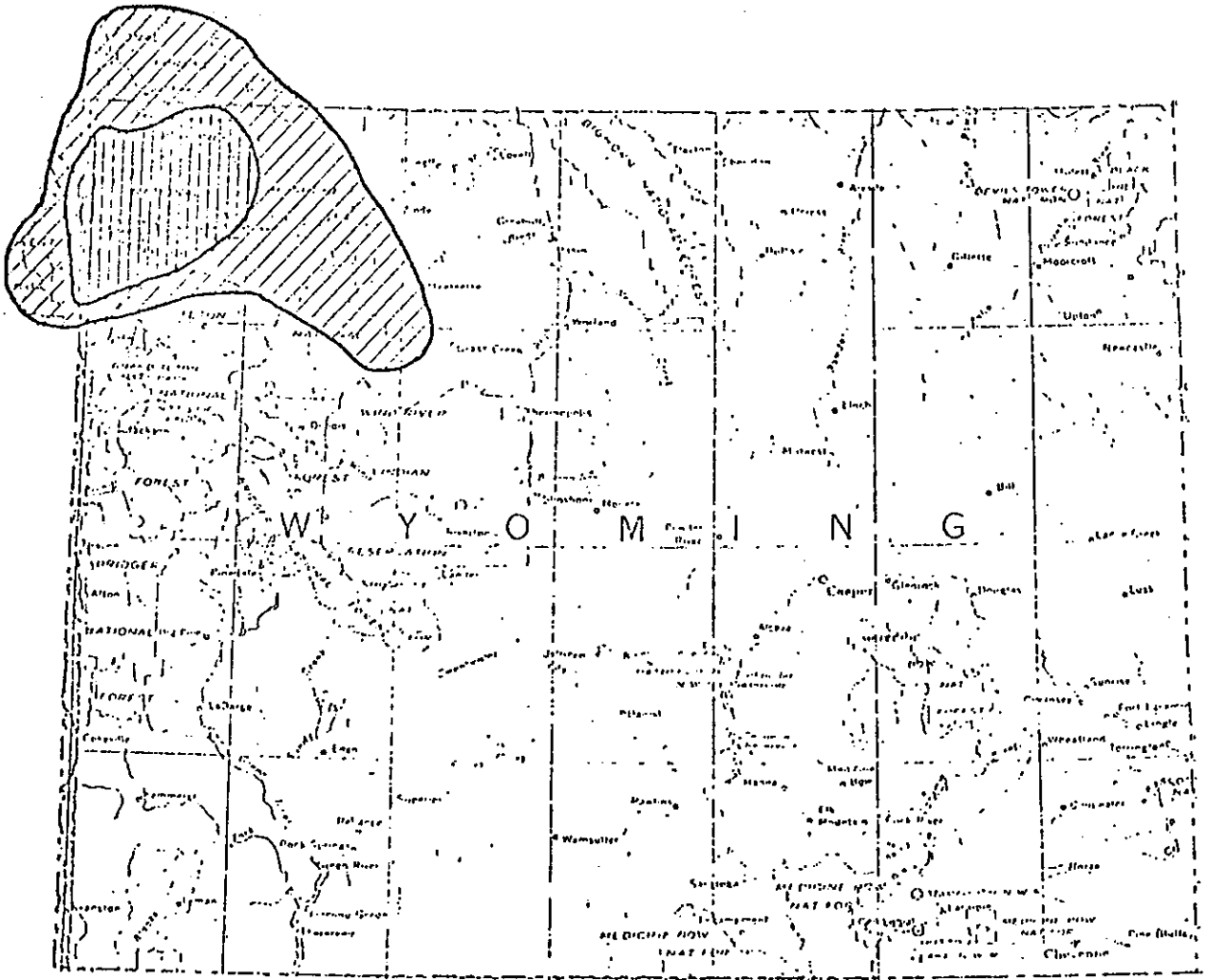
50° C - 100° C

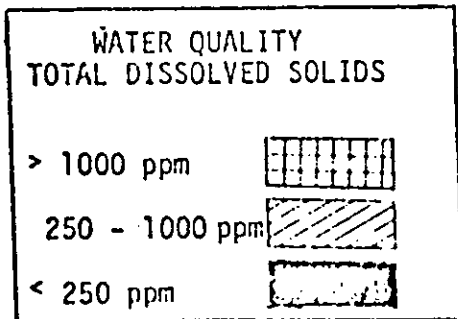
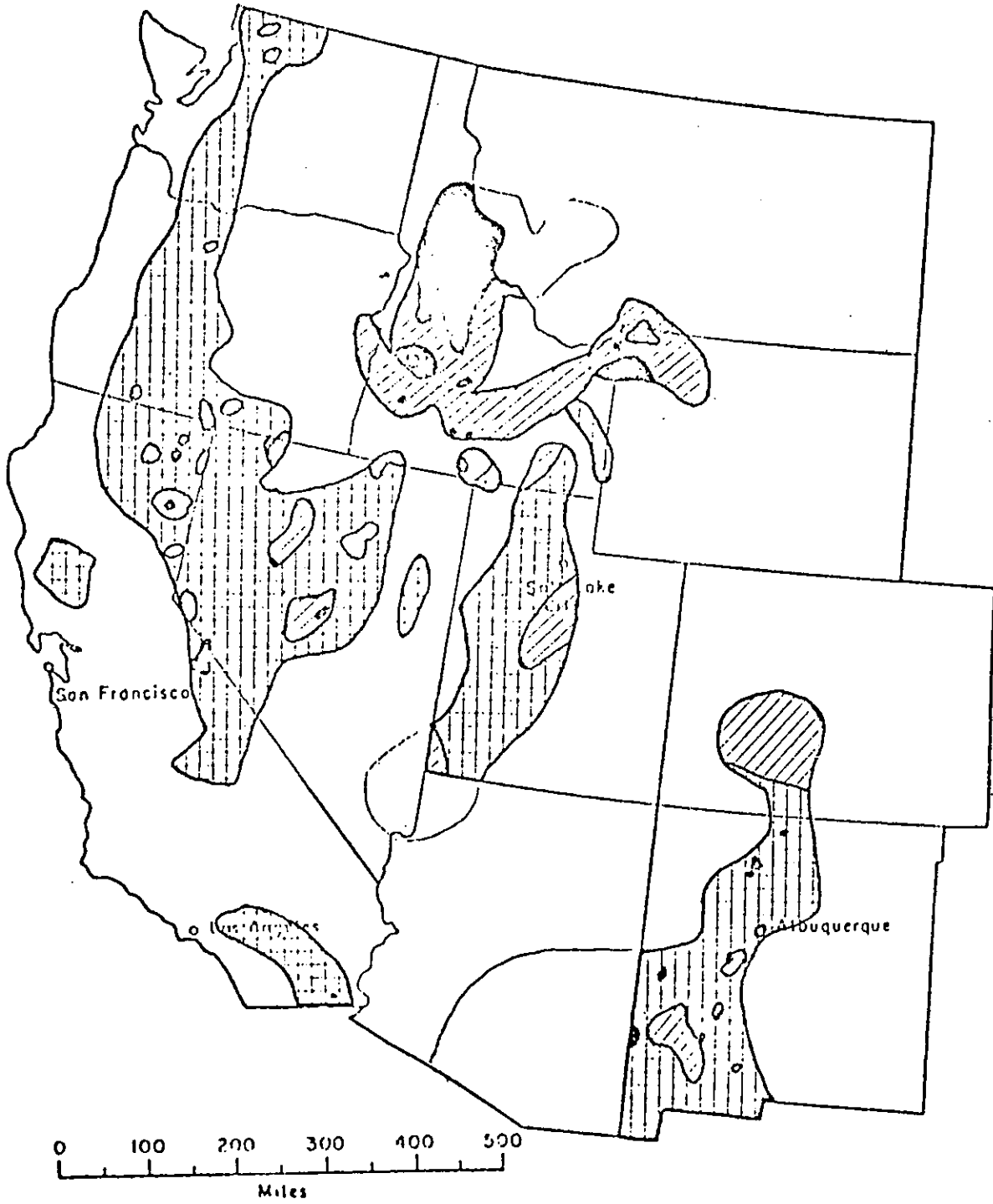


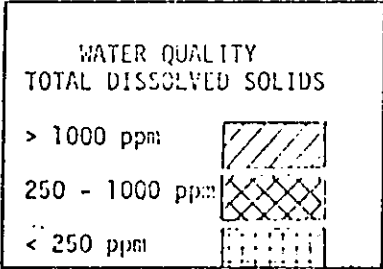
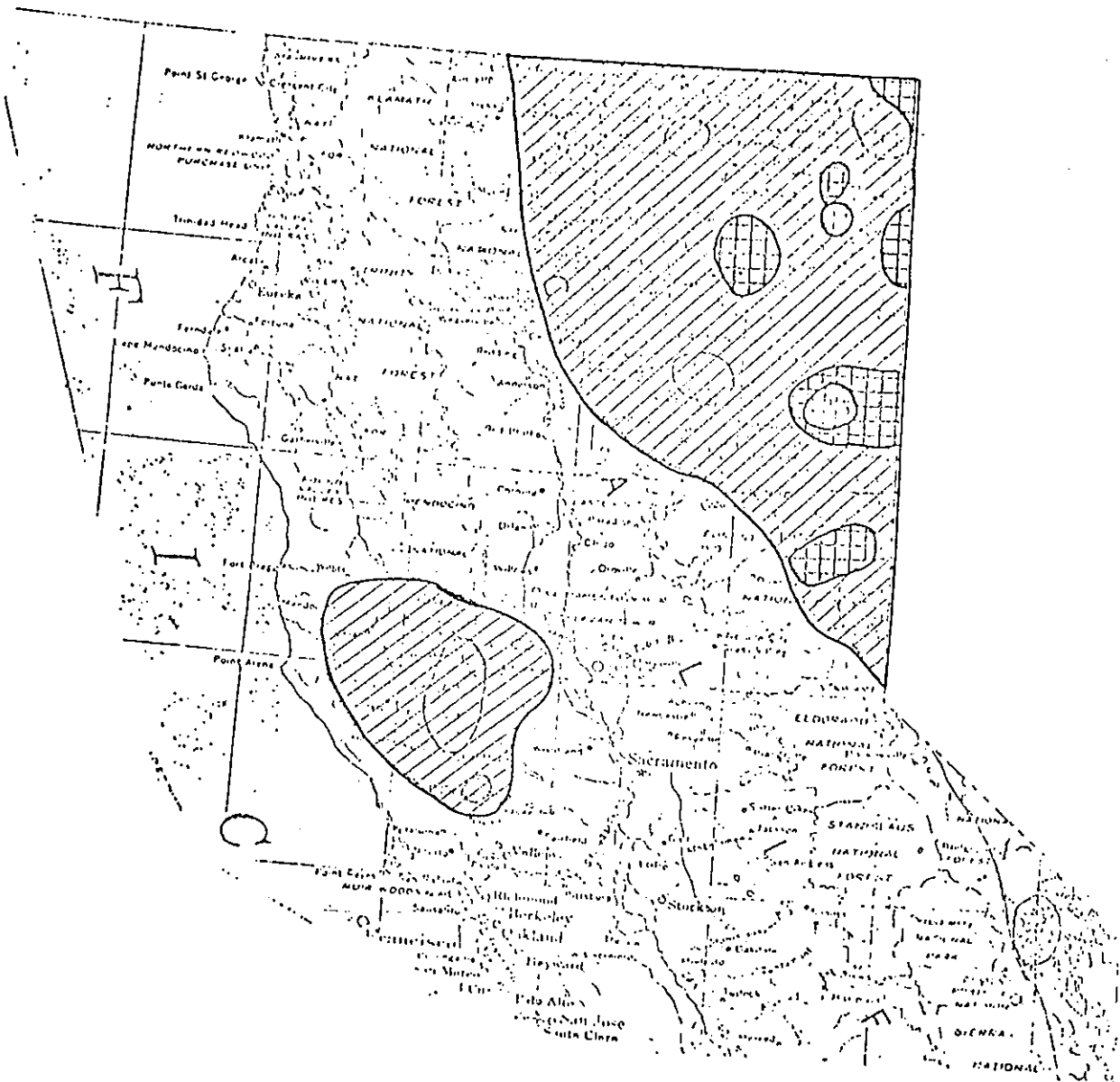
< 50° C



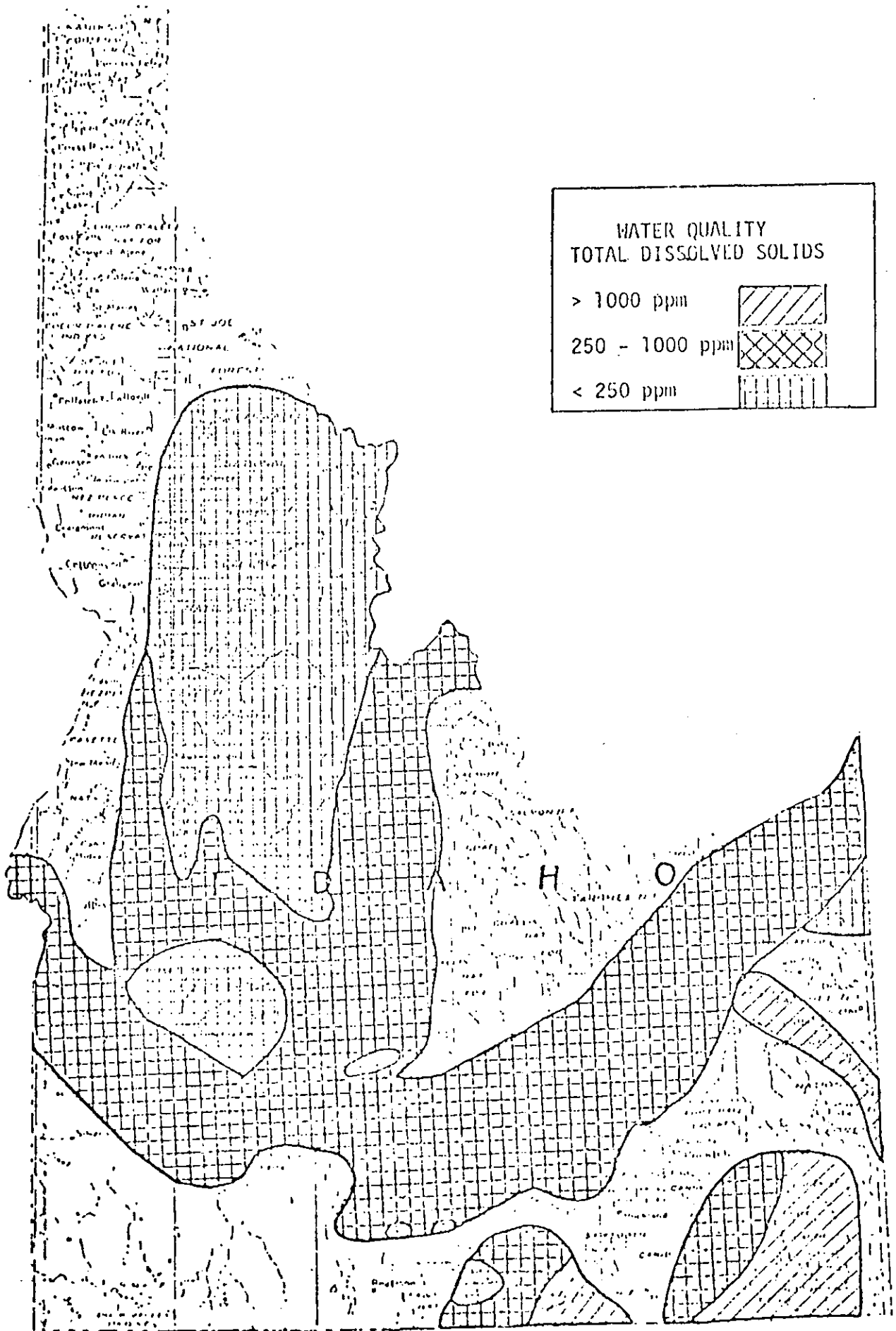






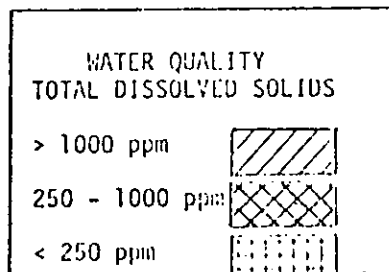
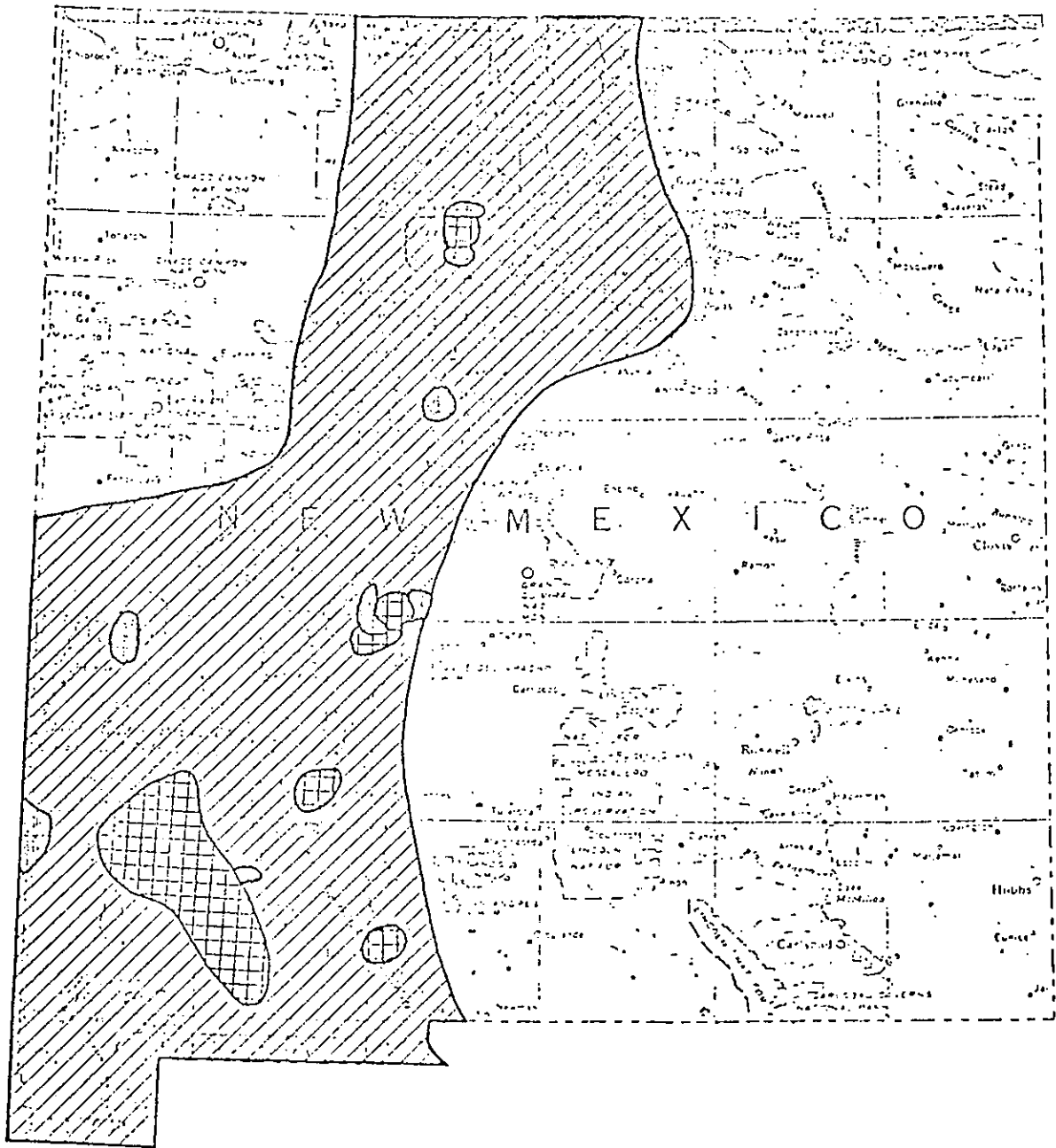


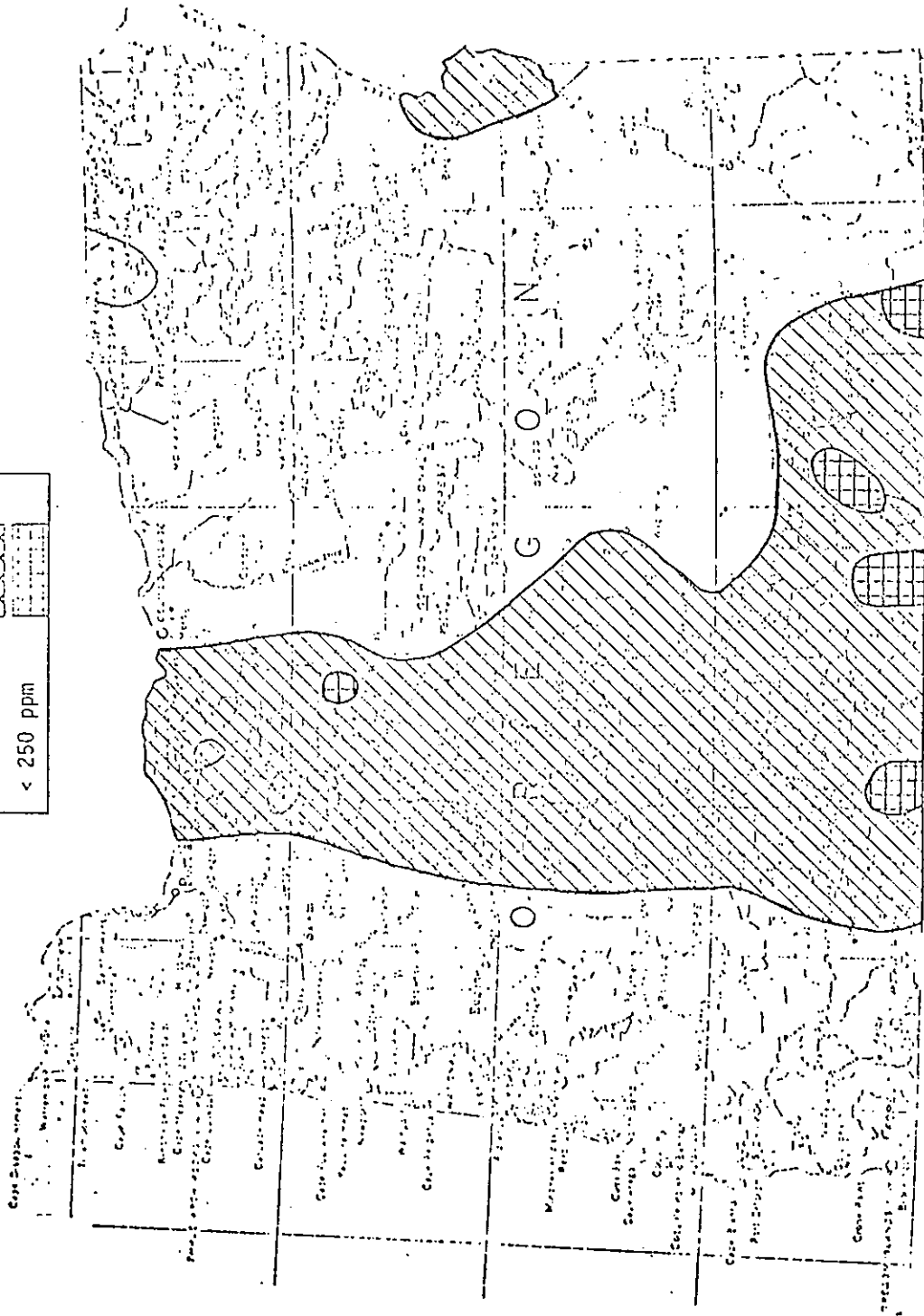
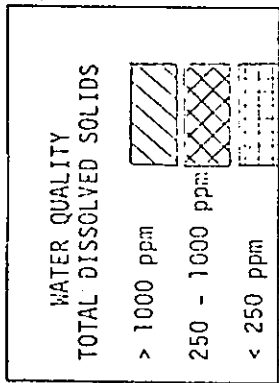






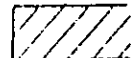




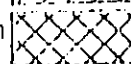


WATER QUALITY  
TOTAL DISSOLVED SOLIDS

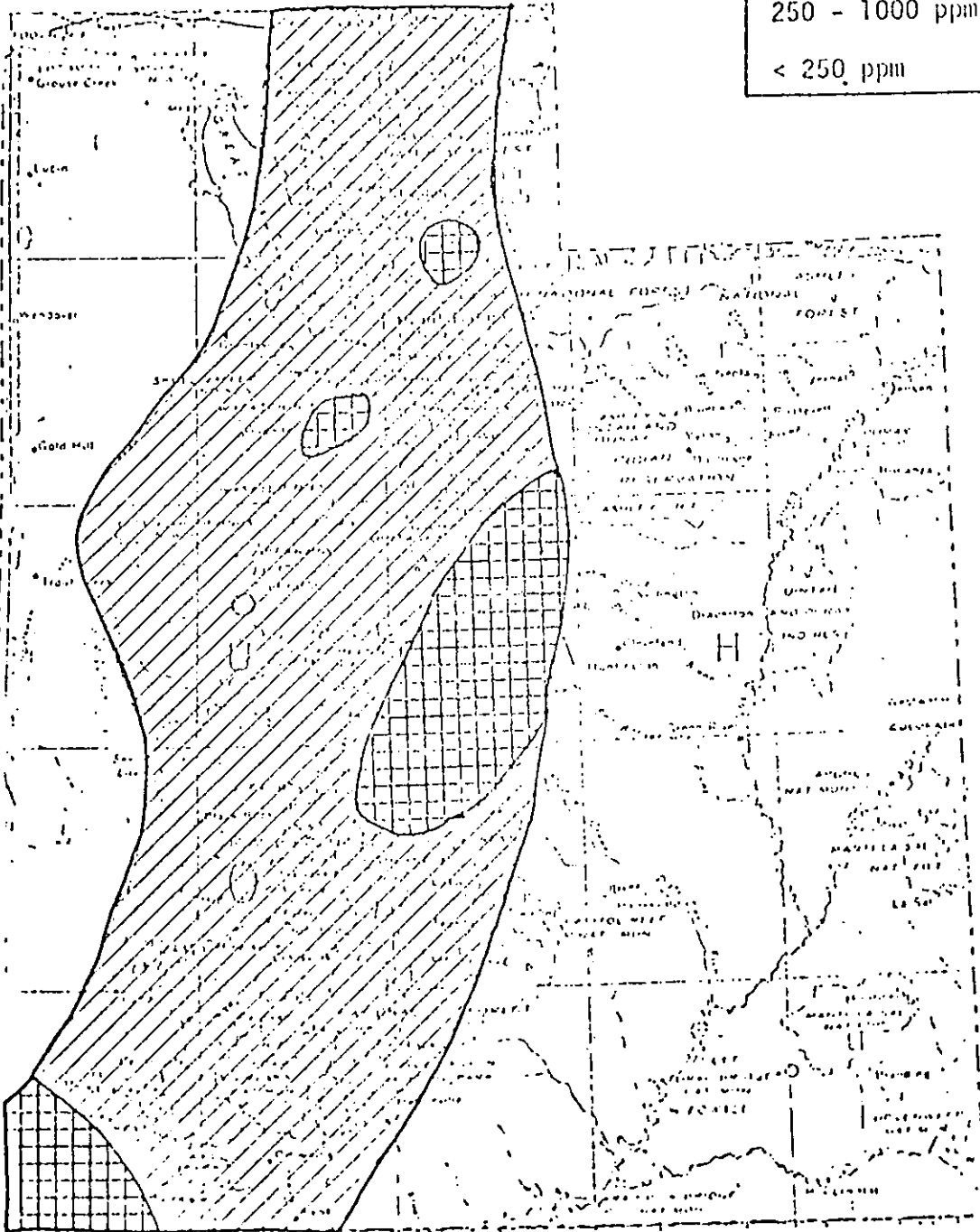
> 1000 ppm

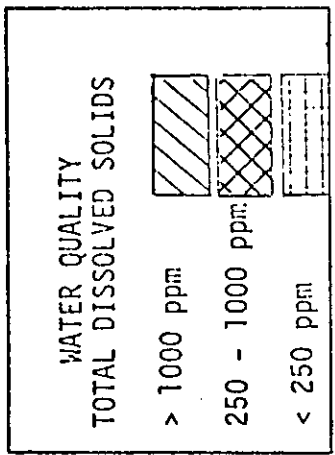
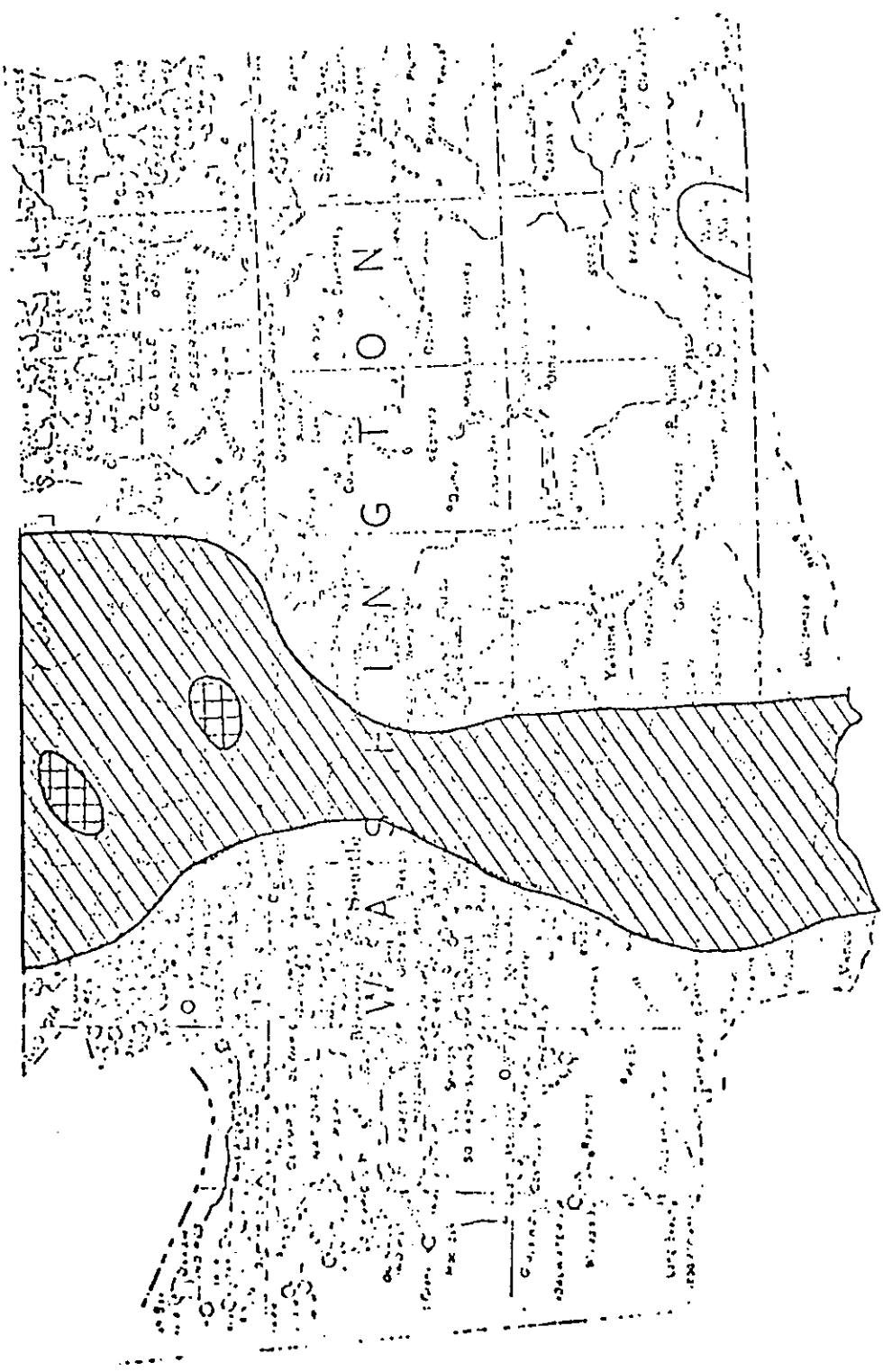


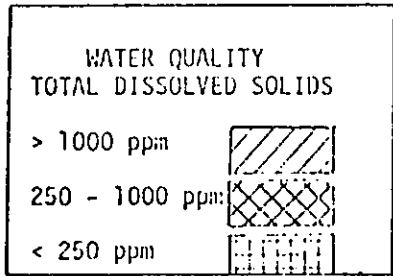
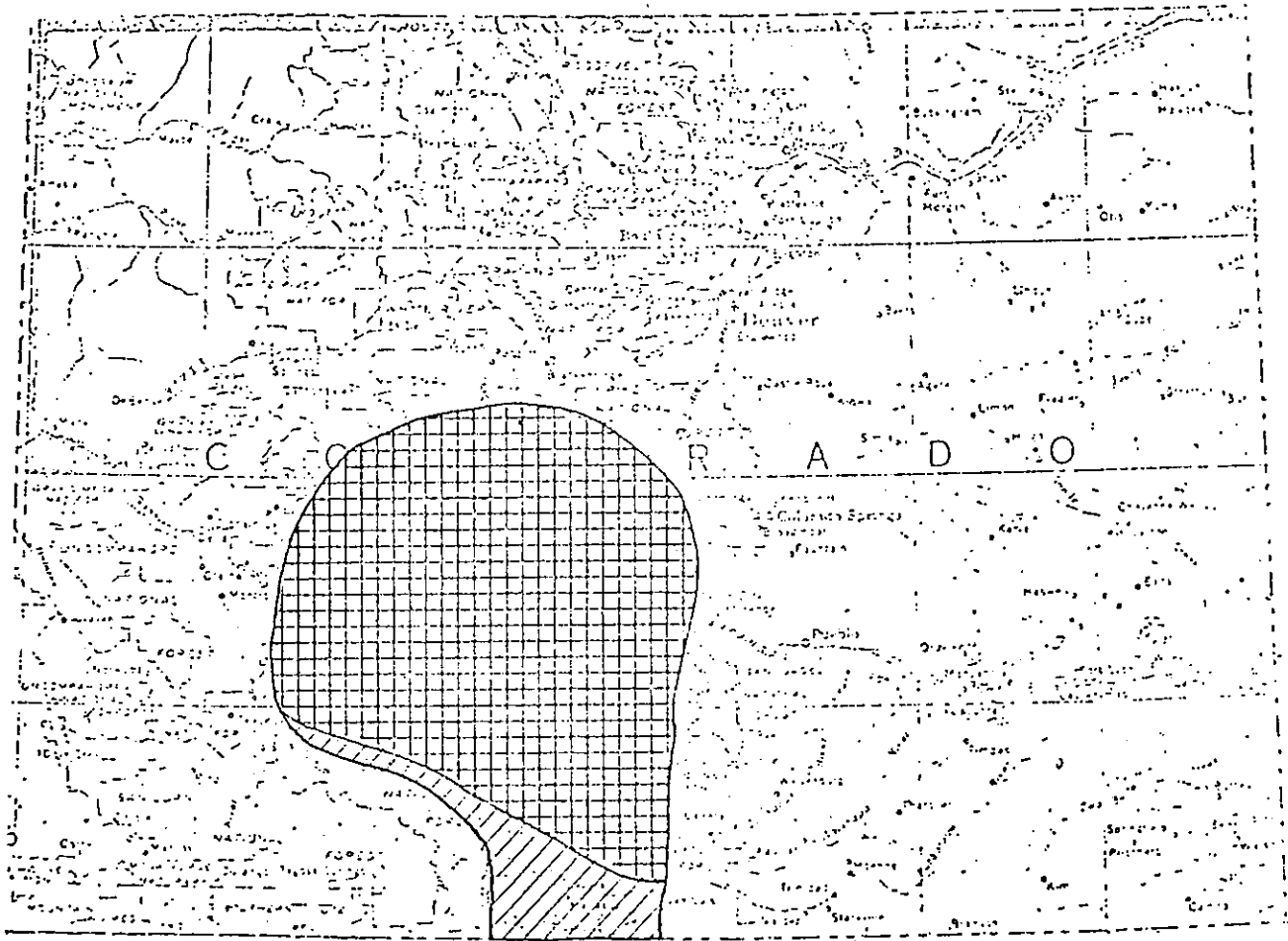
250 - 1000 ppm

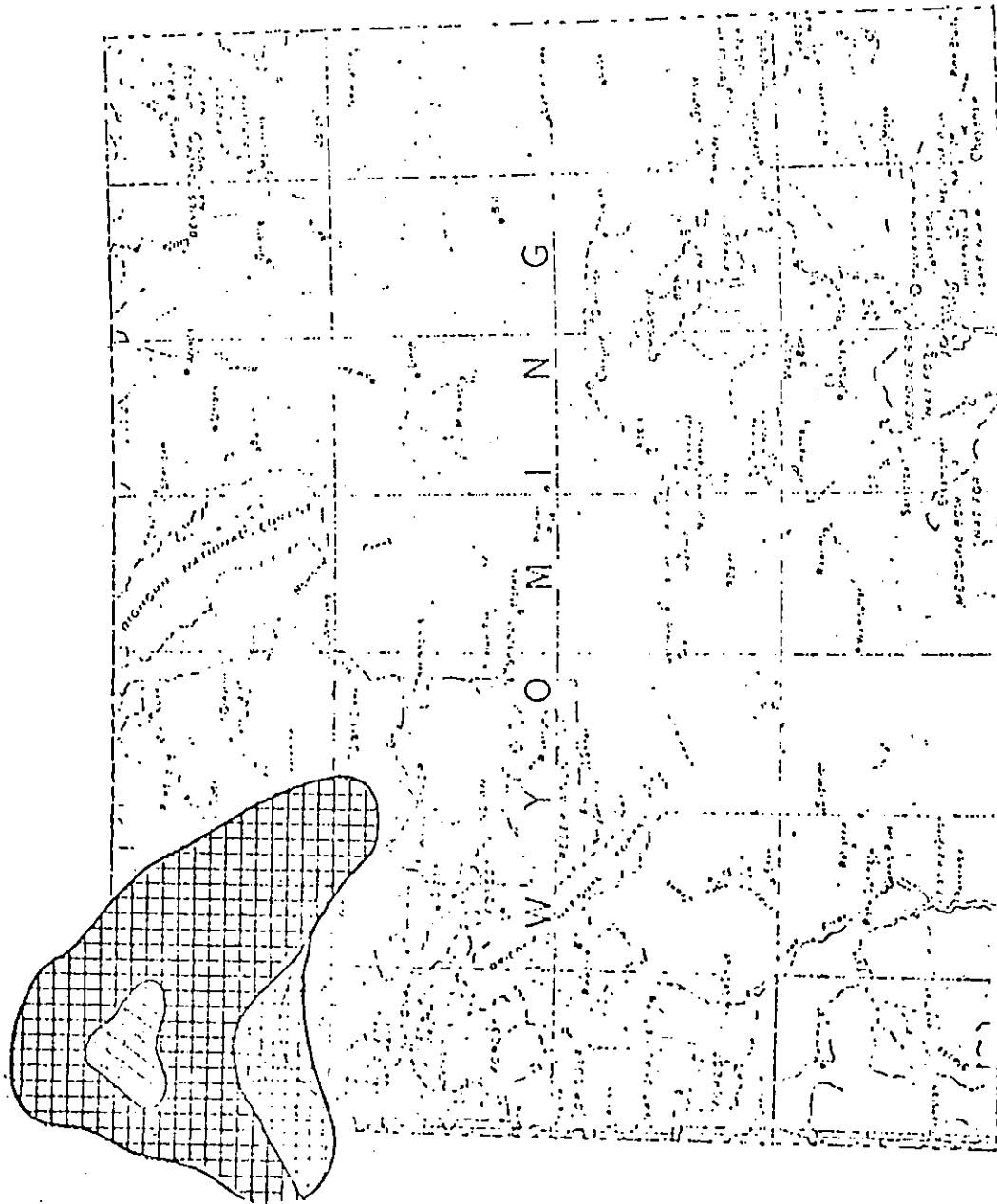
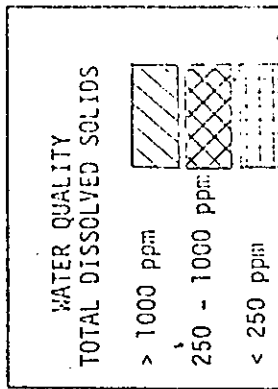


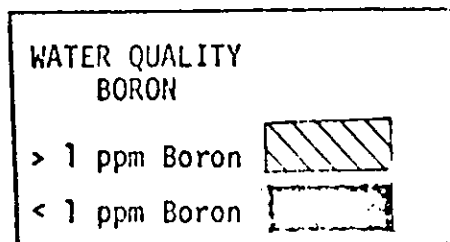
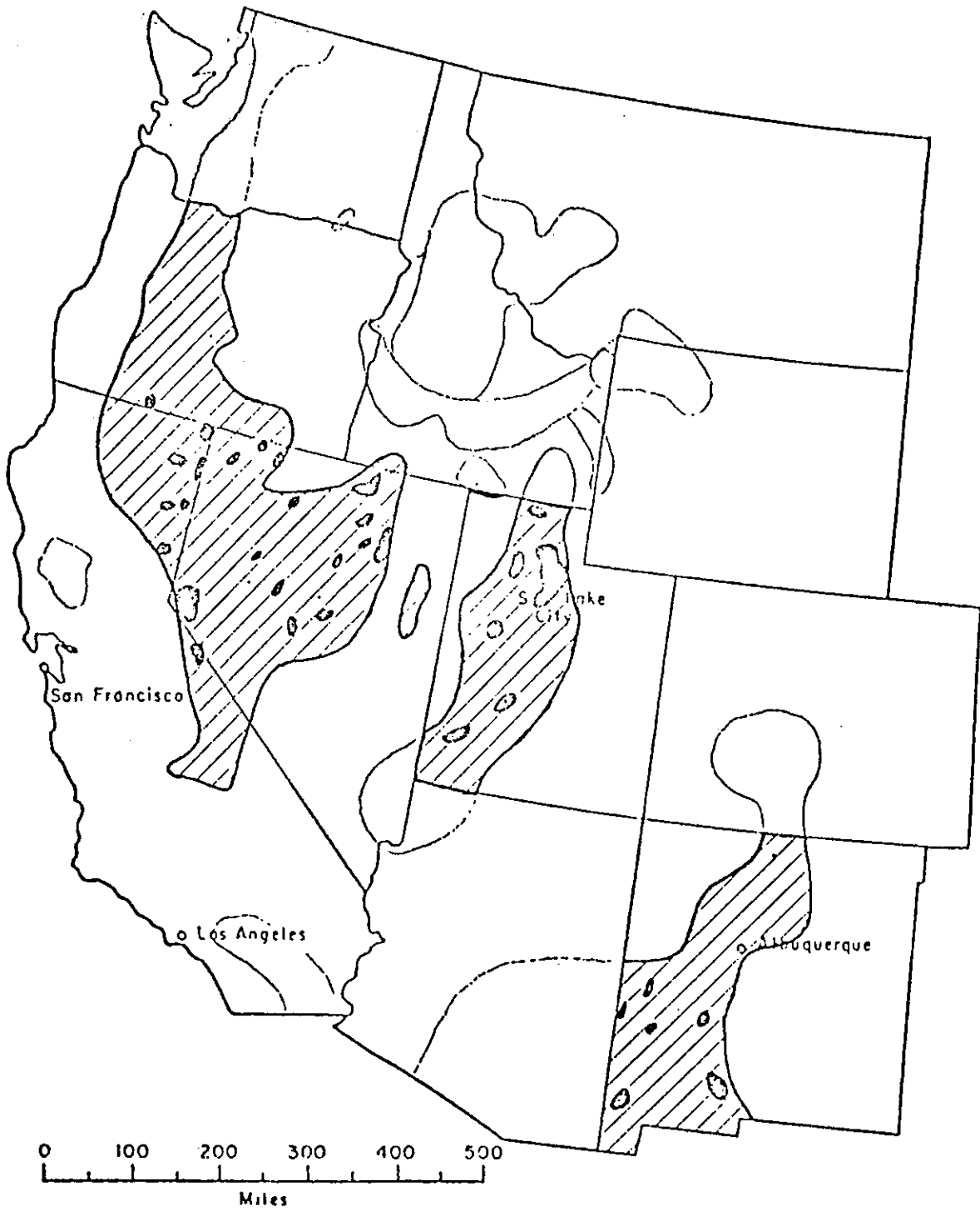
< 250 ppm

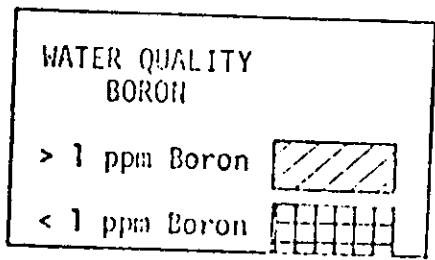




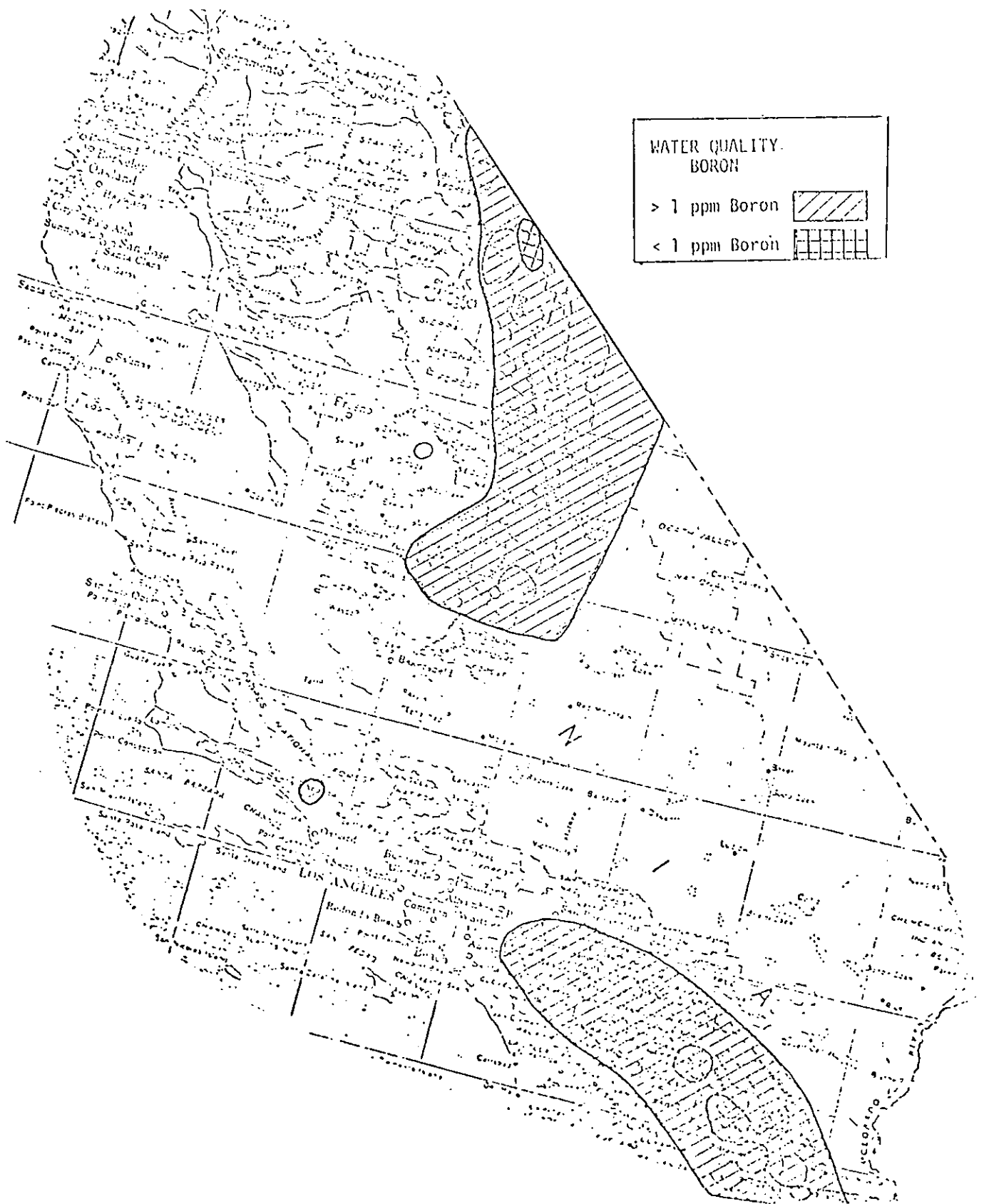


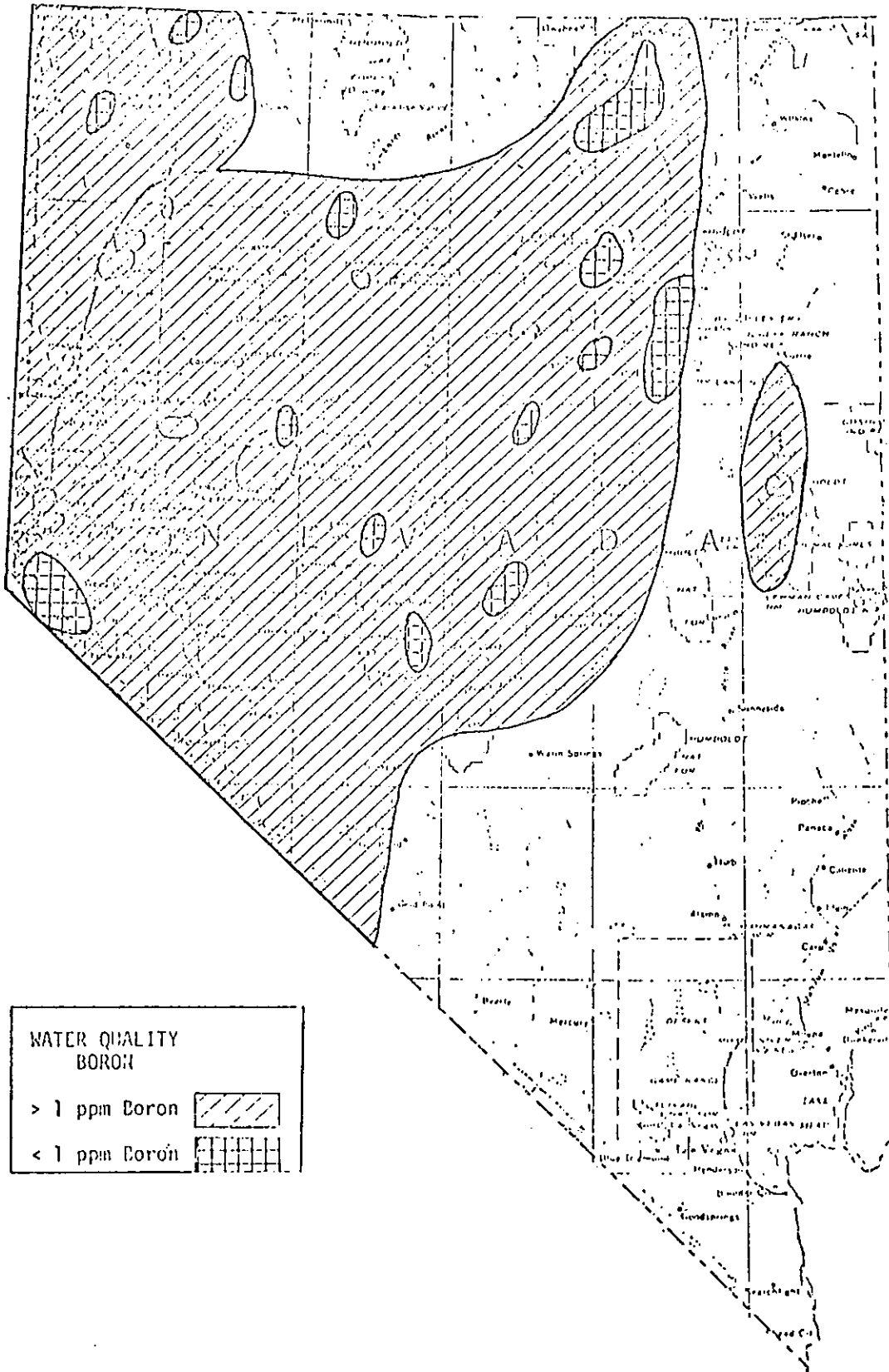


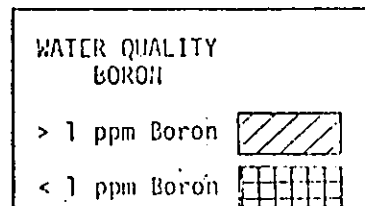
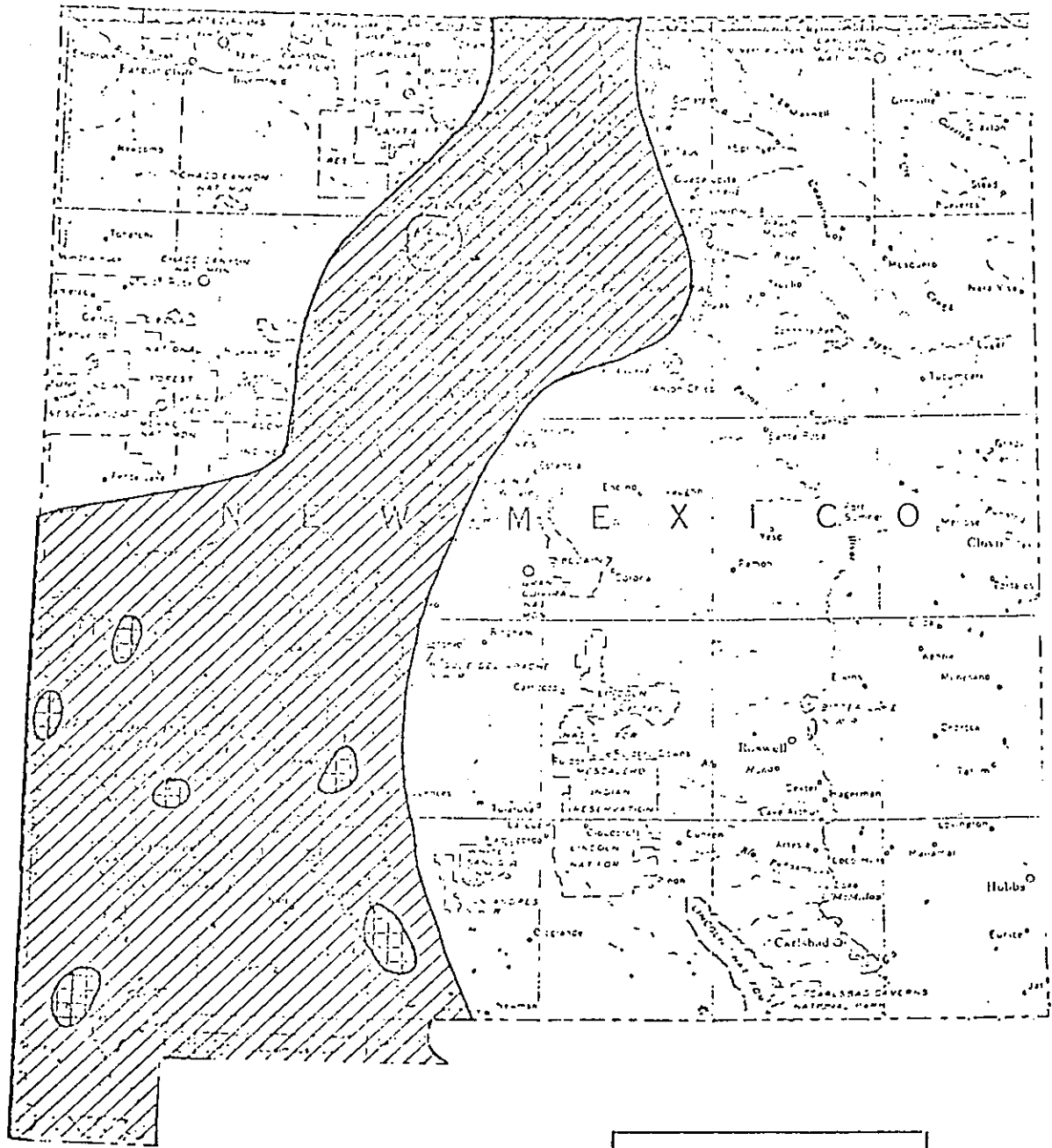


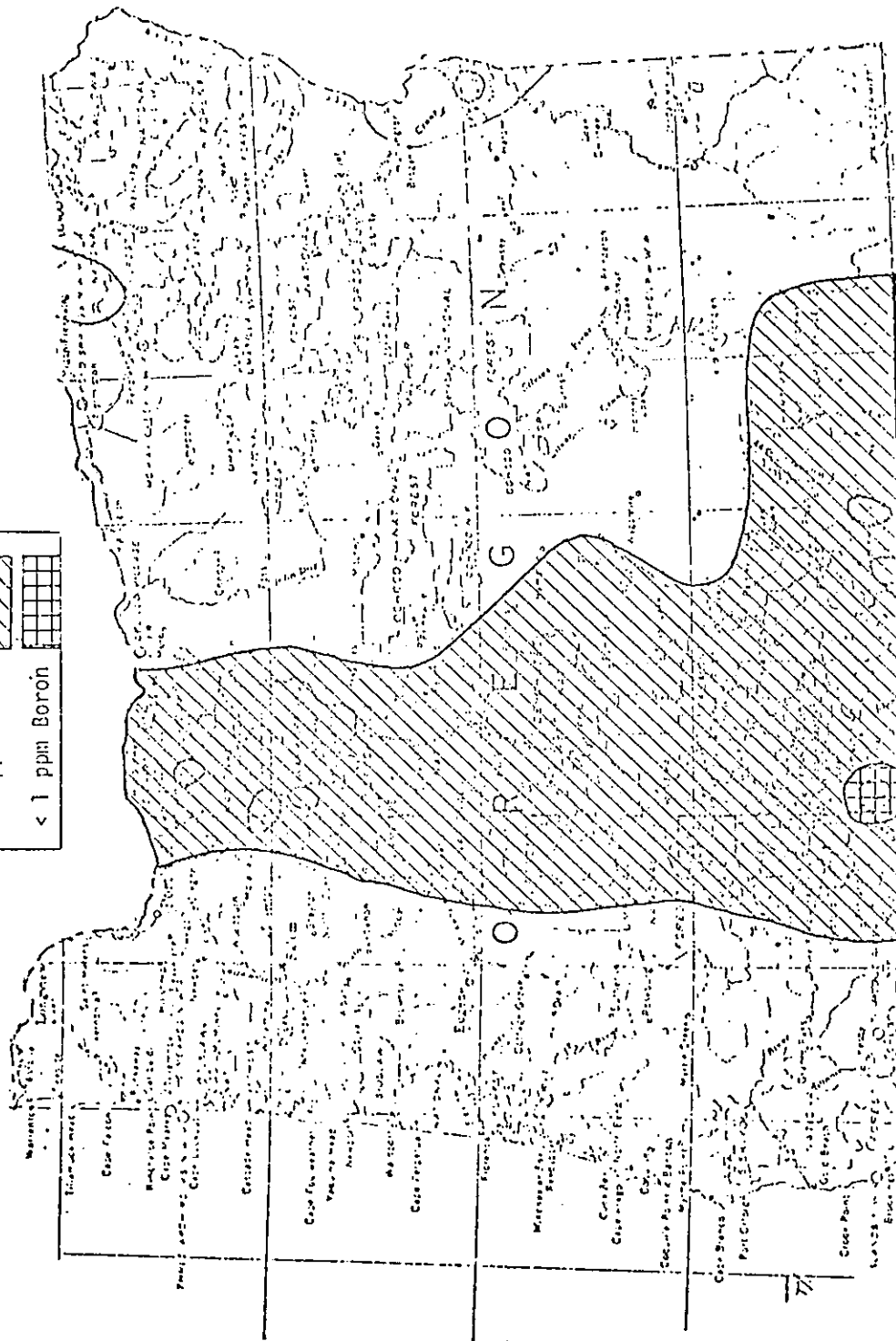
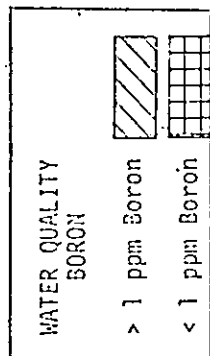


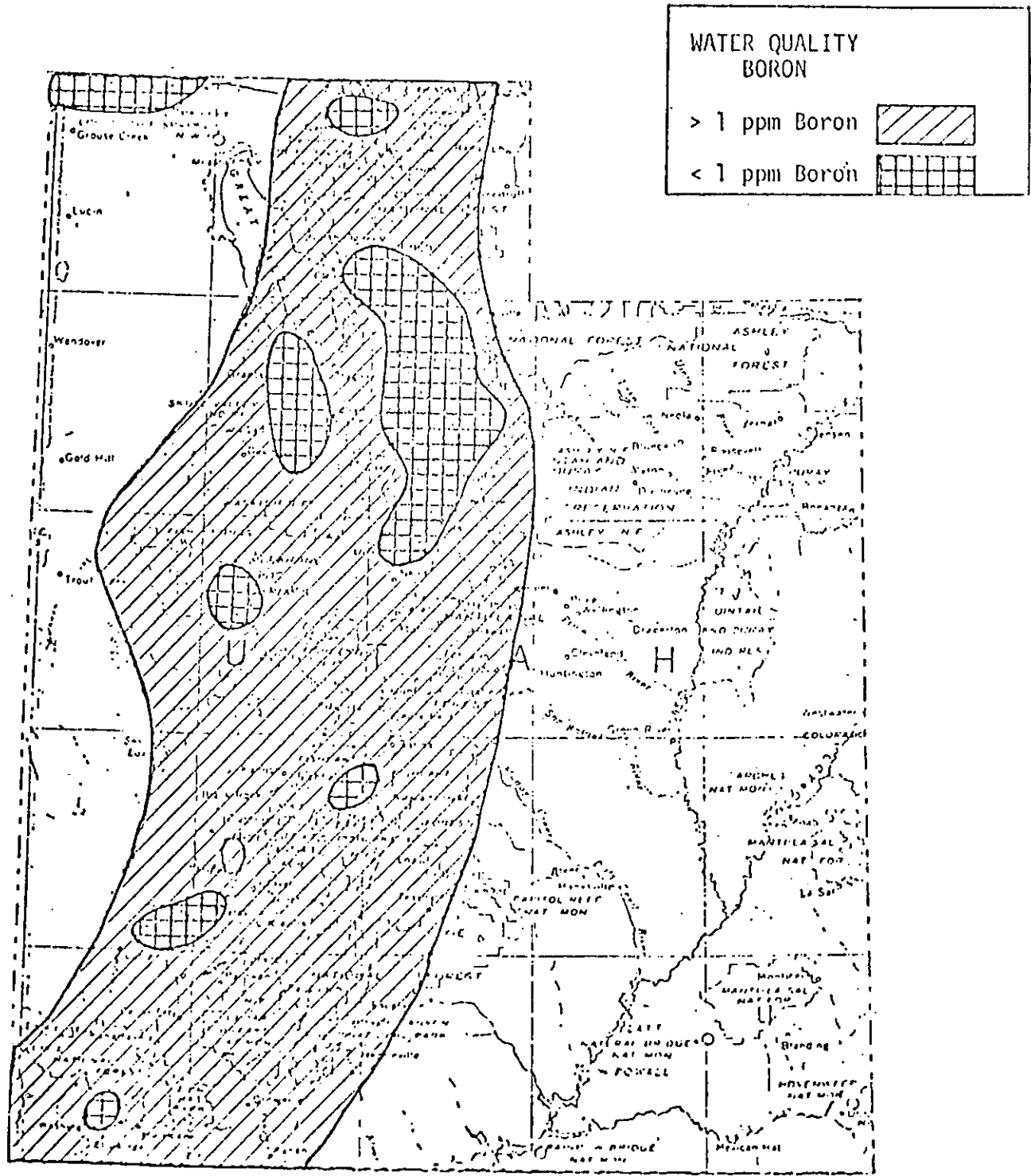


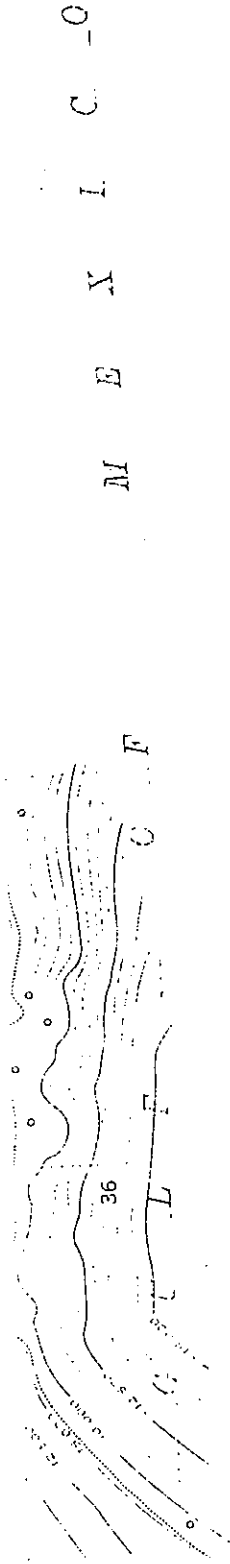












M E X I C O

# TECTONIC MAP OF THE UNITED STATES

*Exclusive of Alaska and Hawaii*

BY THE

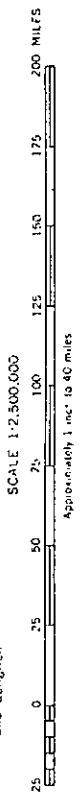
UNITED STATES GEOLOGICAL SURVEY

AND

THE AMERICAN ASSOCIATION OF PETROLEUM GEOLOGISTS

Prepared by a Committee composed of: George V. Cobec, Chairman  
 Paul L. Appin, Marland P. Billings, J. R. Dornell, Harold W. Hoots, Paul L. Lyons  
 N. W. Bass, Ronald K. DeFord, James D. Forrester, Philip B. King, Grover E. Murray  
 Alfred H. Bell, Carroll E. Dolbin, James Gilluly, Chester R. Longwell, Aaron C. Waters

This map is a revision of the Tectonic Map published in 1944 by the American Association of Petroleum Geologists, which was prepared under the direction of the Committee on Tectonics (Chester R. Longwell, Chairman), Division of Geology and Geography, National Research Council. Coordination between the 1944 map and the present edition was provided by King and Longwell.



1962

87°

88°

89°

## EXPLANATION

### METAMORPHIC ROCKS

*Of sedimentary and igneous origin, forming part of outcrop area of basement rocks*



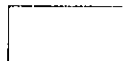
#### Paleozoic and Mesozoic metamorphic rocks

*In western United States. Deformed and metamorphosed during Mesozoic and older orogenies; may be little metamorphosed in some areas; in California includes meta-sedimentary and meta-volcanic rocks of Sierra Nevada and Klamath Mountains. Mesozoic in Arizona*



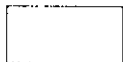
#### Paleozoic low-grade metamorphic rocks (Rocks of Carolina slate belt)

*Slate, graywacke, tuff, and lava, mostly of low-grade metamorphism and moderate deformation. Forms synclinoria in southeast part of Piedmont area*



#### Paleozoic metamorphic rocks

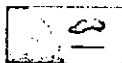
*In places known to be of various Paleozoic ages, but elsewhere of poorly determined age; may include some areas of earlier Precambrian basement rocks, and of late Precambrian stratified rocks. Deformed and metamorphosed during Paleozoic orogenies. In New England represent metamorphosed Paleozoic sedimentary and volcanic rocks of the garnet, staurolite and sillimanite zones*



#### Precambrian metamorphic rocks

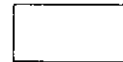
*Undifferentiated as to age, including several subdivisions in some areas; deformed and metamorphosed during one or more orogenies. In places includes minor Precambrian intrusive bodies*

### INTRUSIVE IGNEOUS ROCKS



#### Tertiary intrusive bodies

*Dikes, plugs, stocks, laccoliths, and small batholiths; may include some granitic bodies of Mesozoic age in Oregon and Washington. In Arizona includes Tertiary and Mesozoic intrusives*



#### Triassic basin deposits

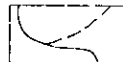
*Continental sedimentary rocks with associated basaltic and pyroclastic rocks; in structurally negative areas in eastern United States*



#### Precambrian sedimentary and volcanic rocks

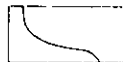
*Includes Belt series of northern Rocky Mountains, Keweenaw series of Lake Superior region, and sedimentary and volcanic rocks of southern Appalachians*

### GEOLOGIC BOUNDARIES



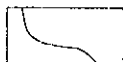
#### Base of upper Tertiary sedimentary rocks

*Base of Miocene (Ogallala and Arikaree formations) in High Plains of west-central United States; base of Oligocene (White River formation) in Nebraska, South Dakota and North Dakota shown dashed*



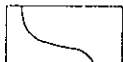
#### Base of lower Tertiary sedimentary rocks

*Base of Eocene series in Atlantic and Gulf Coastal Plains; base of Wasatch formation and equivalents in Rocky Mountains and in North Dakota; commonly underlain by Paleocene rocks*



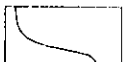
#### Base of Cretaceous sedimentary rocks

*Atlantic and Gulf Coastal Plains; Central Interior region*



#### Base of Pennsylvanian sedimentary rocks

*Central Interior region*



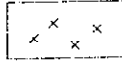
*Precambrian boundary in Canada  
Along north shore of Lake Superior eastward to Frontenac axis*



**Mesozoic felsic intrusive bodies**  
*Granitic bodies of Mesozoic age in western United States. May include some bodies of early Tertiary age*



**Mesozoic dikes, sills and stocks**  
*Diabase dikes and shallow intrusives of Triassic age in Appalachian area; syenite and peridotite of Cretaceous (?) age in Arkansas*



**Small dikes and other intrusive bodies**  
*In central and eastern United States; of Paleozoic and younger ages*



**Ultramafic bodies**  
*Peridotite, dunite, serpentine, and related rocks, forming small bodies in Piedmont and Blue Ridge areas. In Appalachian area mostly of Paleozoic age, some may be of Precambrian age; in western United States mostly of Mesozoic age*



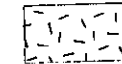
**Paleozoic felsic intrusive bodies**  
*In New England of Late Ordovician, Middle Devonian and Late Pennsylvanian age; some deformed. In eastern Massachusetts may be Precambrian. In the Piedmont of middle Paleozoic and earlier ages; some are little deformed; others have gneissic structure, as suggested by trend lines*



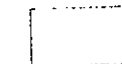
**Paleozoic mafic intrusive bodies**  
*In Piedmont area and New England; in eastern Massachusetts may be Precambrian*



**Paleozoic and Mesozoic alkalic rocks**  
*Paleozoic and Mesozoic rocks with alkalic affinities in New England; in Massachusetts and Rhode Island of Mississippian (?) age; in New Hampshire, Maine, and Vermont of Triassic age. In Piedmont area of North Carolina syenite of late Paleozoic (?) age*

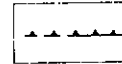


**Precambrian mafic intrusive bodies**  
*Duluth gabbro topolith and related bodies of Lake Superior region, and similar smaller bodies elsewhere; anorthosite bodies in Adirondack region and western United States. Diabase in Arizona*

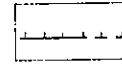


**Precambrian felsic intrusive bodies**  
*Granitic bodies of various ages; principally in Lake Superior region and western United States*

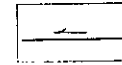
**STRUCTURAL FEATURES**



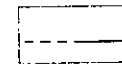
**Thrust fault**  
*Saw teeth on upthrown side. Includes low-angle faults, but some may dip at steep angle at surface. Broken line indicates hypothetical faults*



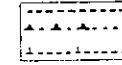
**Normal fault**  
*Hachures on downthrown side. Includes some high-angle thrust faults whose dip is indicated by arrows. Broken line indicates hypothetical faults*



**Lateral fault**  
*With dominant strike-slip movement, whose direction is indicated by arrow on one side. Broken line indicates hypothetical faults*



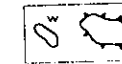
**Unclassified faults**  
*Nature of displacement unknown on many*



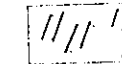
**Buried faults**  
*Includes faults of all classes shown above. Direction of displacement shown only in part.*



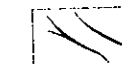
**Klippe**  
*Thrust mass outlier*



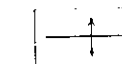
**Window**  
*Fenster in thrust mass*



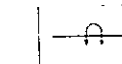
**En echelon fault system**  
*Direction of displacement not shown on all faults of system*



**Elongate, closely compressed anticline**  
*Width of line suggests height, steepness, or size of fold*



**Anticlinal axis**  
*Includes axes of broadly arched uplifts and minor folds. Plunge of some anticlines shown by arrows*



**Axis of overturned anticline**



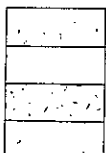
**VOLCANIC ROCKS AND FEATURES**



**Volcanoes and volcanic cones**  
Of Quaternary and late Tertiary age; western United States



**Tertiary and Quaternary volcanic rocks**  
In western United States, where only larger areas or areas of tectonic significance are shown. Mainly of various Tertiary ages. Volcanic rocks of Quaternary age distinguished in places by v pattern. Cretaceous volcanic rocks in southern Arizona are included

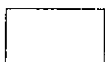


**Tertiary and Quaternary volcanic rocks of Washington and Oregon**

Includes Cascade olivine basalt flows and volcanoes of Quaternary and Tertiary age (v pattern); Columbia River basalt of Miocene and Pliocene age (white hatchures); altered andesitic and dacitic volcanic rocks of western Cascade (red hatchures); basalt flows and pillow complexes of Coast Ranges, chiefly of Eocene age (blue hatchures)

**SEDIMENTARY ROCKS**

Only selected units shown; many units younger than any major orogeny or epirogeny of area; patterns generally omitted in regions of complex structure



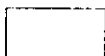
**Quaternary and upper Tertiary sedimentary rocks**

Thick deposits in structurally negative areas, mainly non-marine; includes bolson deposits of Basin and Range province



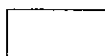
**Tertiary sedimentary rocks**

Chiefly marine; in the Coast Ranges of Washington, Oregon and southern California; includes some small areas of Mesozoic rocks



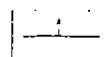
**Upper Mesozoic sedimentary and volcanic rocks in California and in Olympic Mountains, Washington**

Includes Knoxville formation and younger rocks, exclusive of Franciscan, and Tertiary rocks in some areas



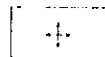
**Rocks of Franciscan type**

Mainly clastic sedimentary rocks, but including bodies of chert, limestone, schist and basalt; of Jurassic and Cretaceous age, depending on locality. Dothan formation in southwest Oregon included



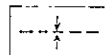
**Monoclinal flexure**

Symbol used where feature is not shown by contour lines

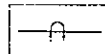


**Dome**

Symbol used where feature is not shown by contour lines



**Synclinal axis**

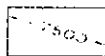


**Axis of overturned syncline**



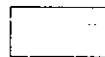
**Ouachita tectonic belt**

Northern edge of stippled band indicates edge of tectonic belt; southern limits undetermined



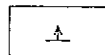
**Structure contours**

Interval of contours is indicated in each area; datum is sea level; where contours are superimposed in the upper part of the Mississippi Embayment and in Georgia and Florida the contours on the lower horizons are indicated by short dashes; for horizons contoured see list and index map. Boundary between areas contoured shown by dotted line

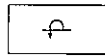


**Form lines**

Similar to structure contours but not drawn on any definite horizon



**Strike and dip of beds**

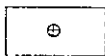


**Strike and dip of overturned beds**

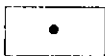


**Trend lines in metamorphic rocks**

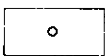
Strike of bedding or schistosity; in places include contacts between lithologic units



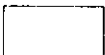
**Intensely disturbed, localized uplift**



**Salt dome**

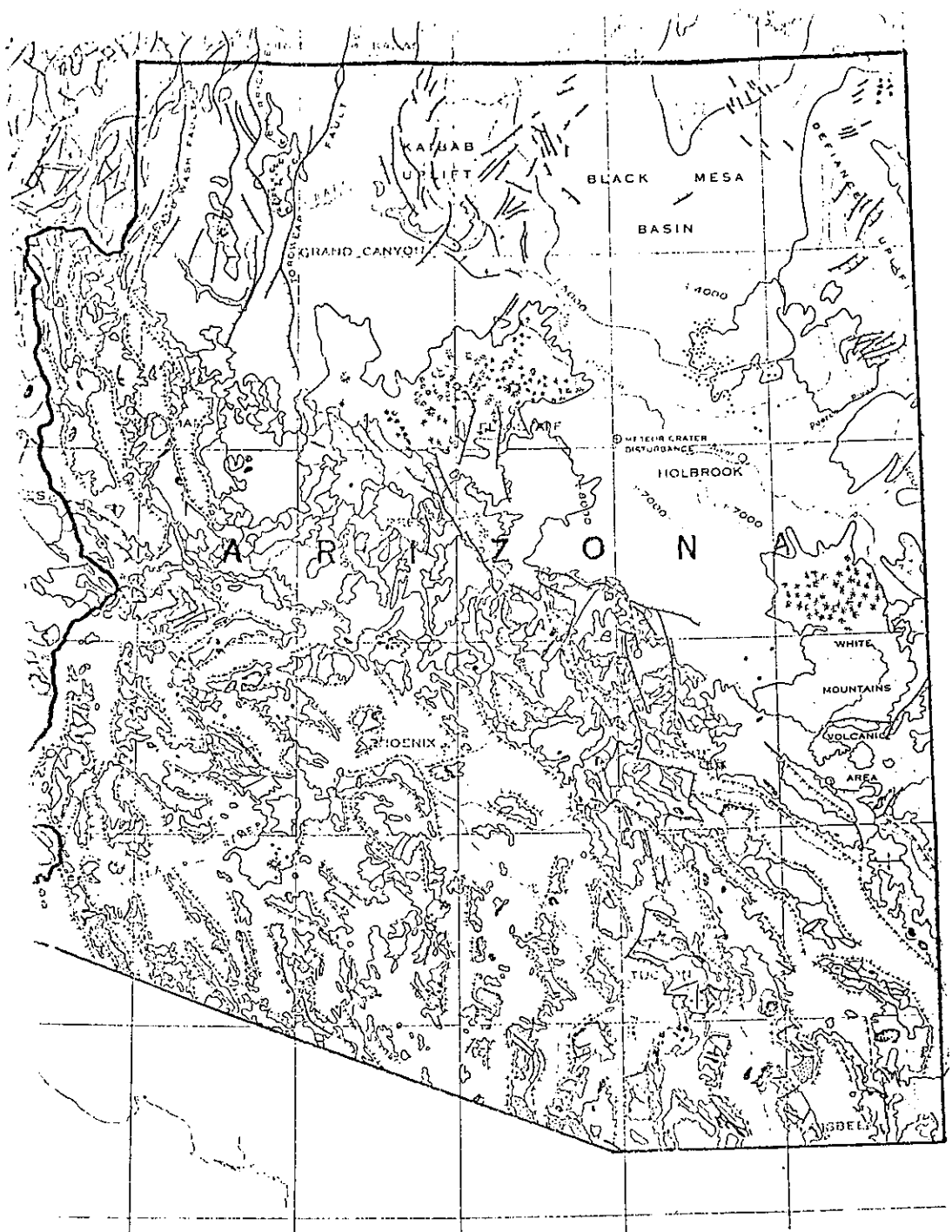


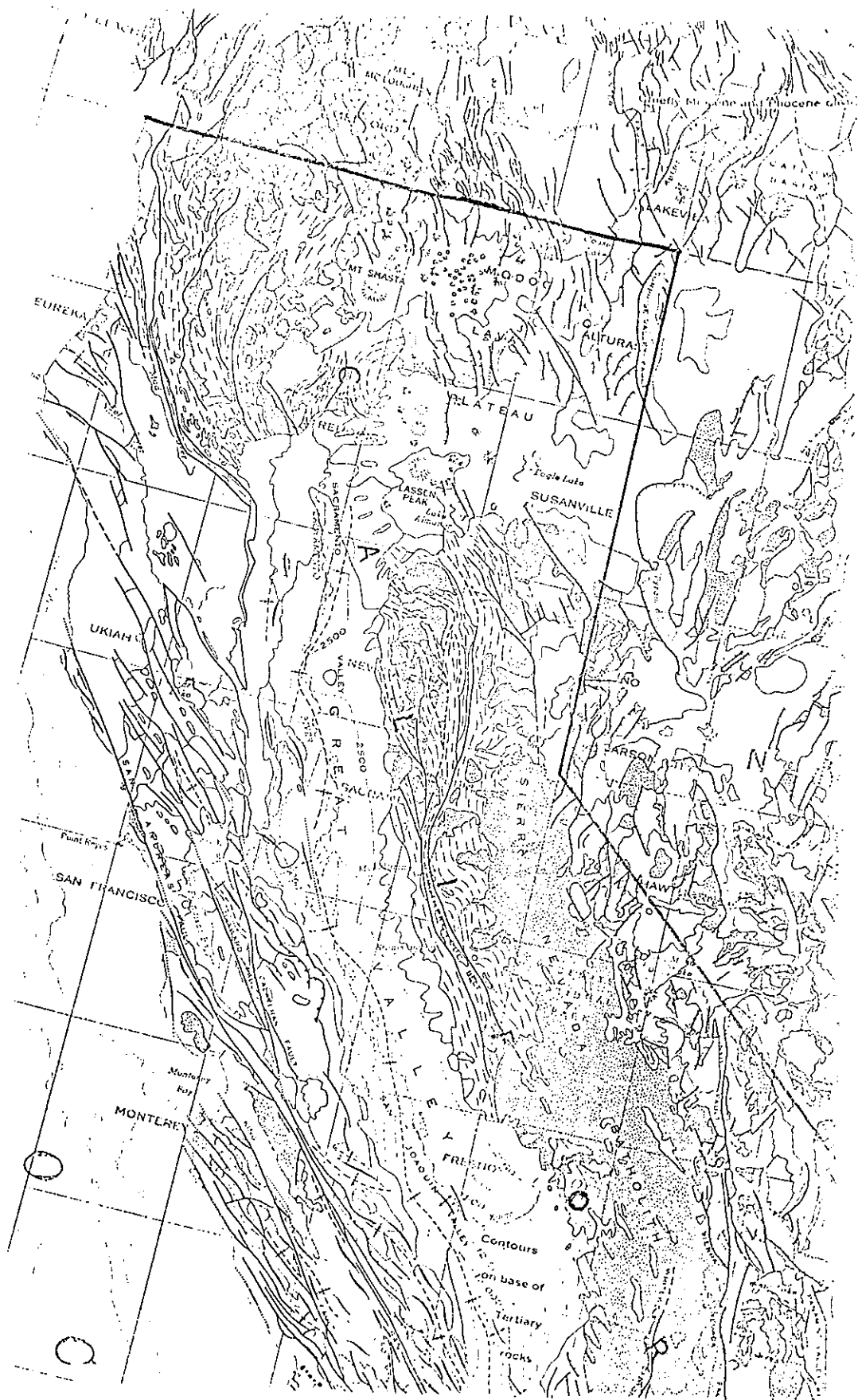
**Possible salt dome**



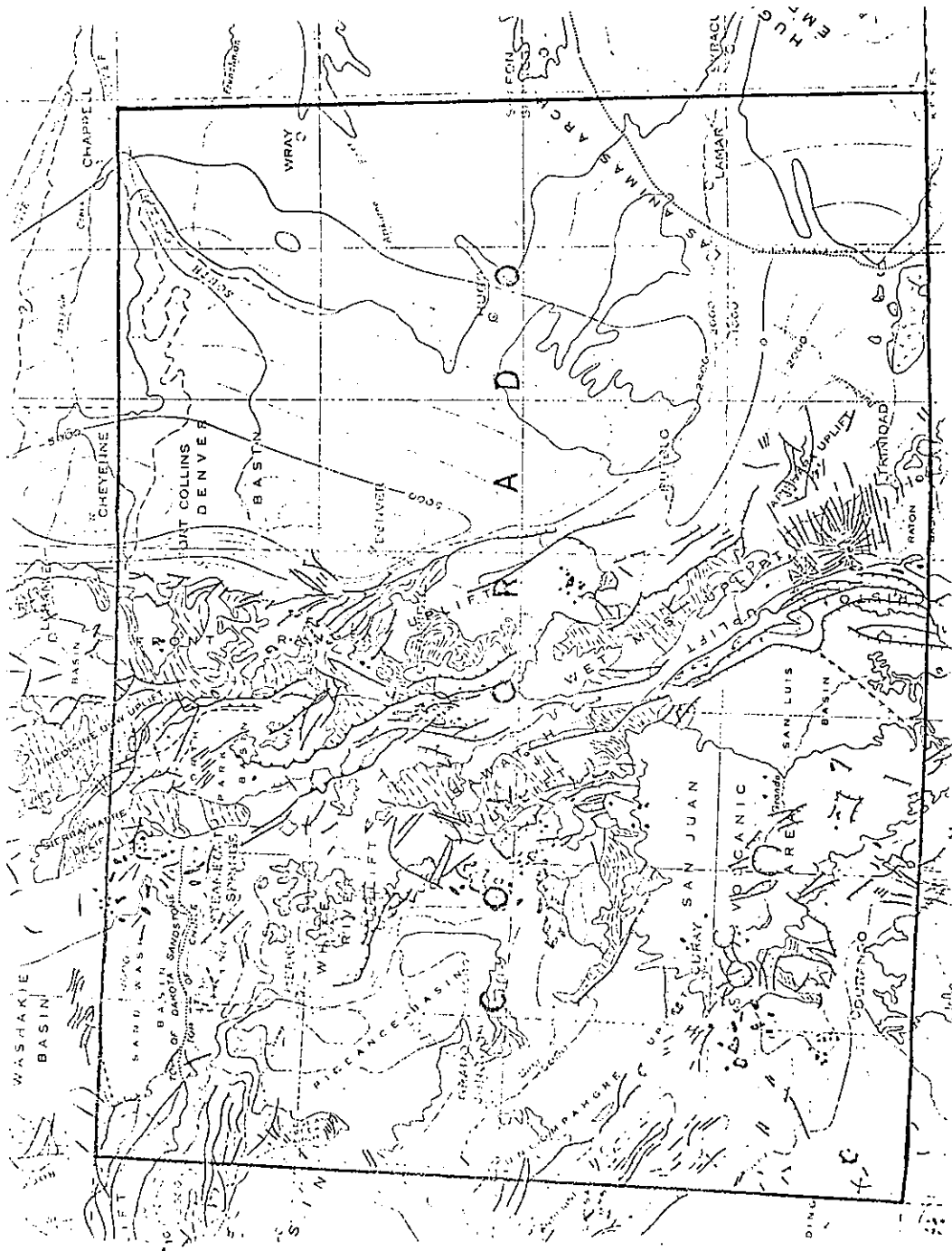
**Submarine contours**

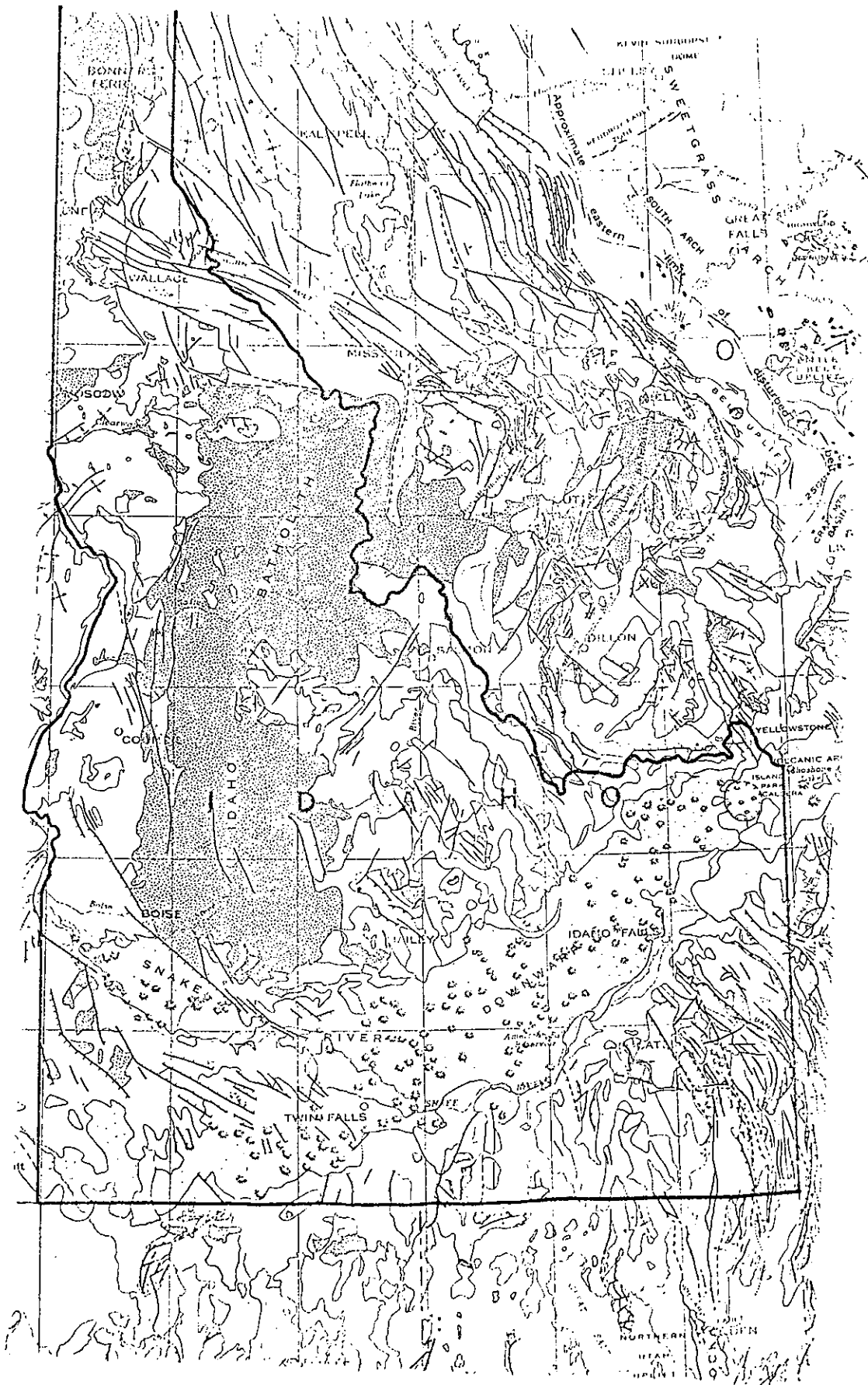
Interval 500 and 1000 feet; datum is sea level



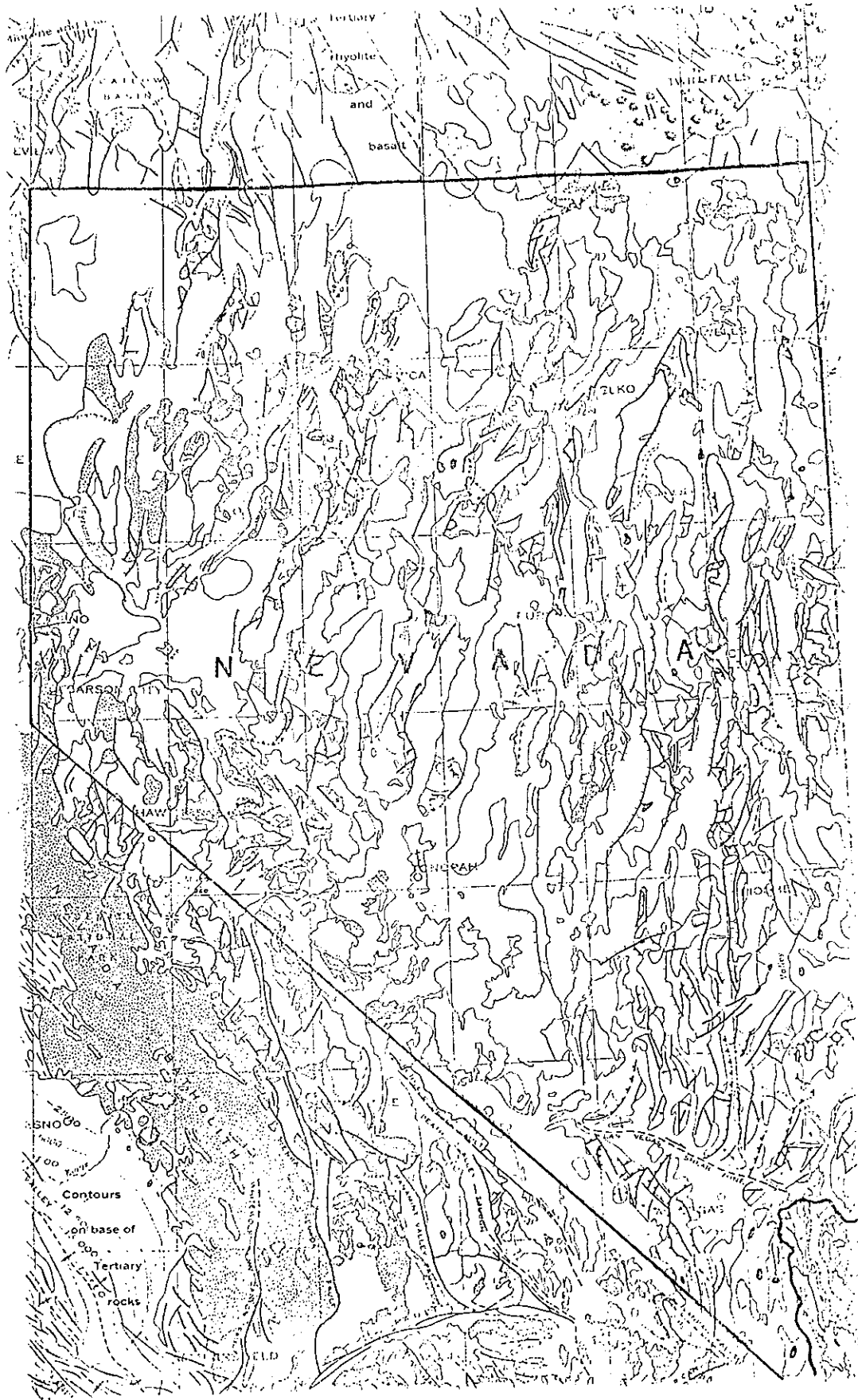




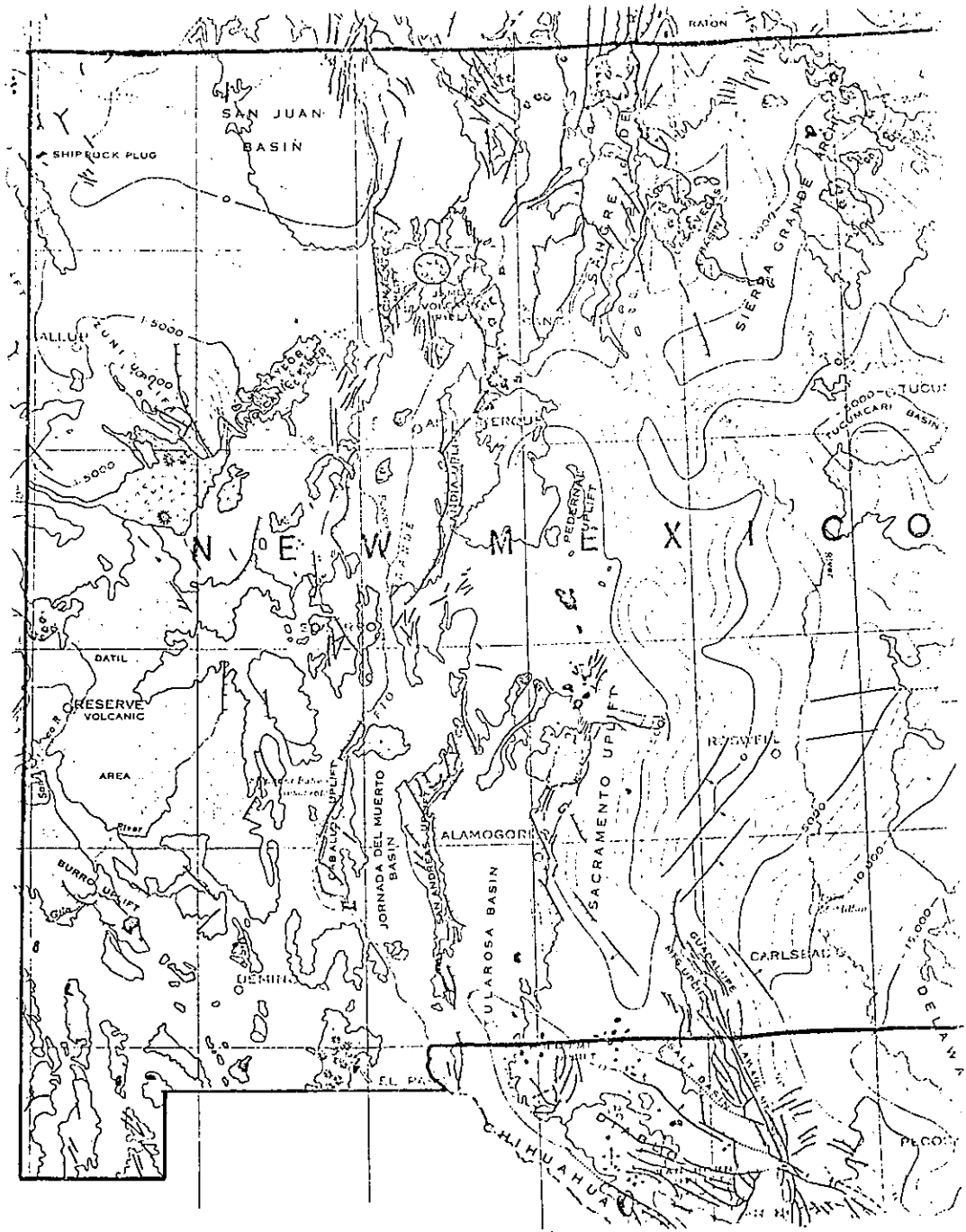


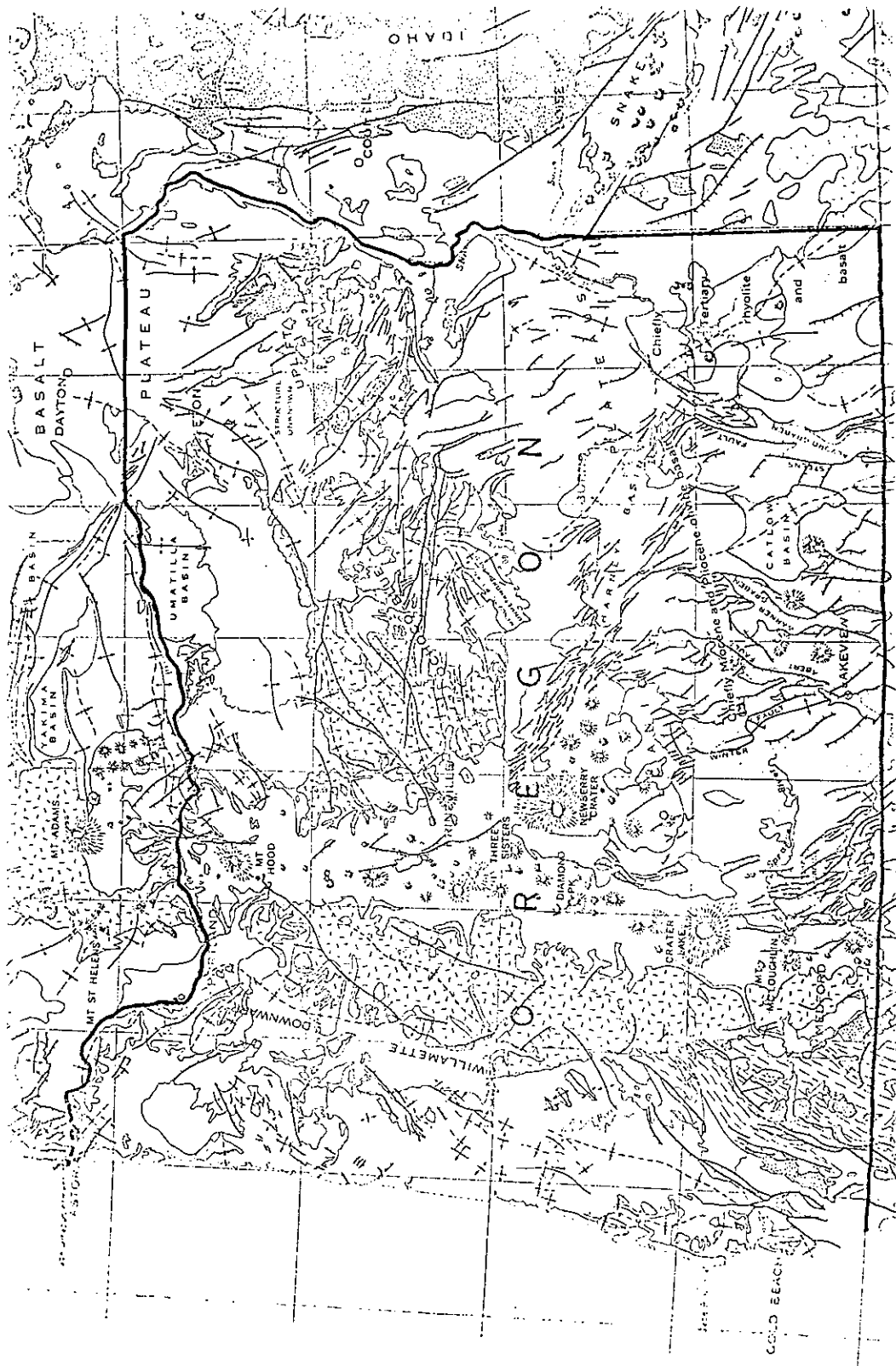


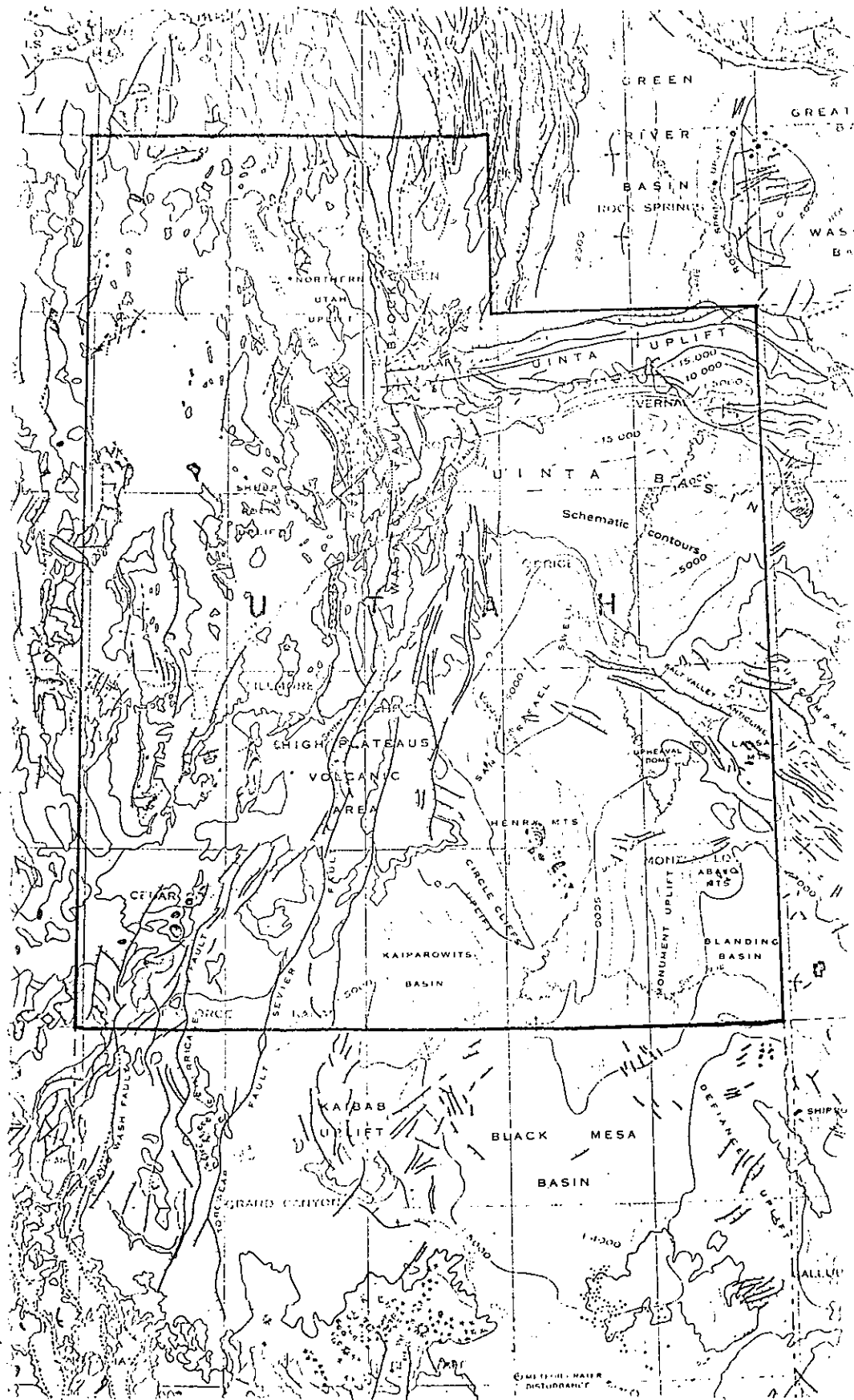


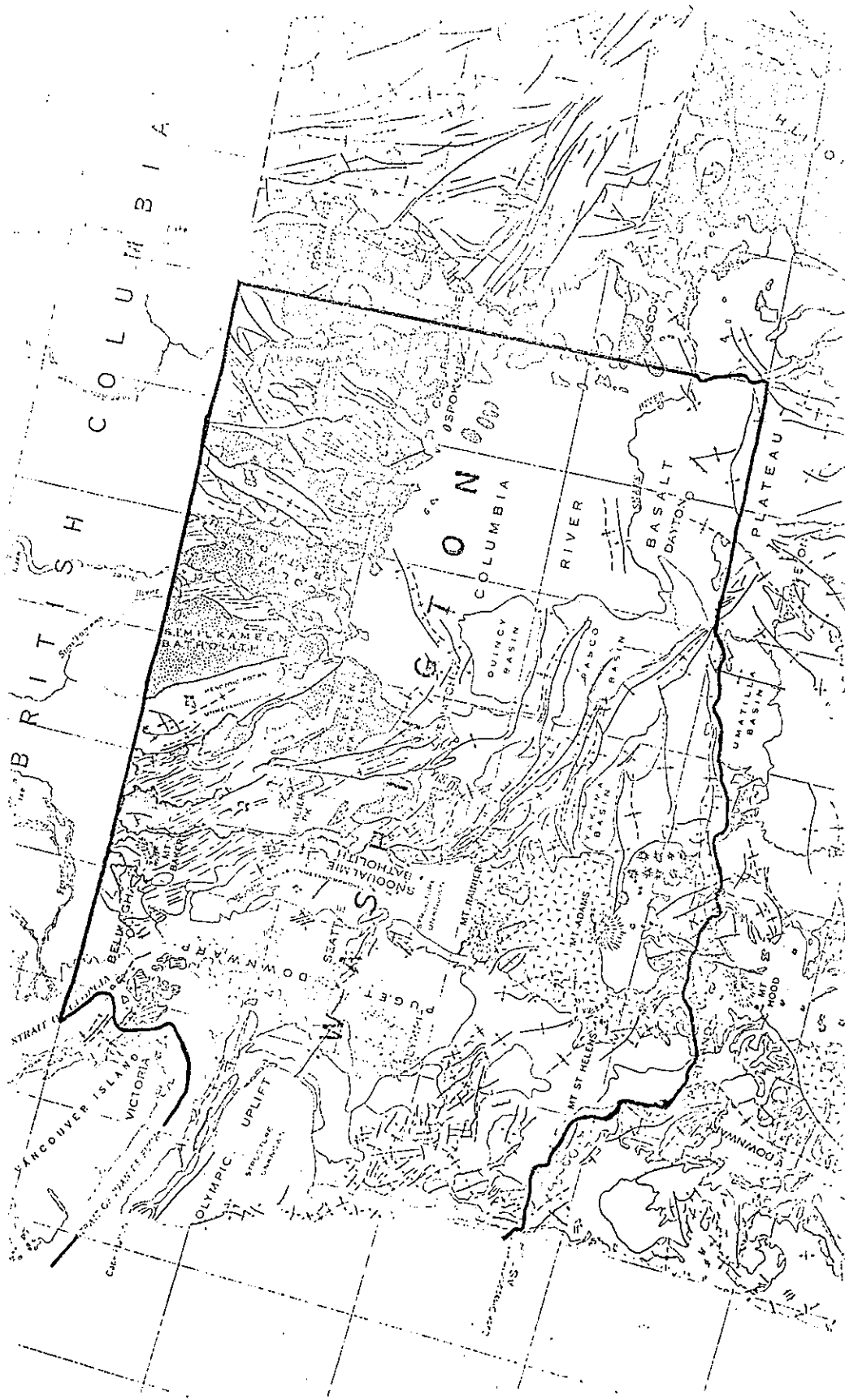




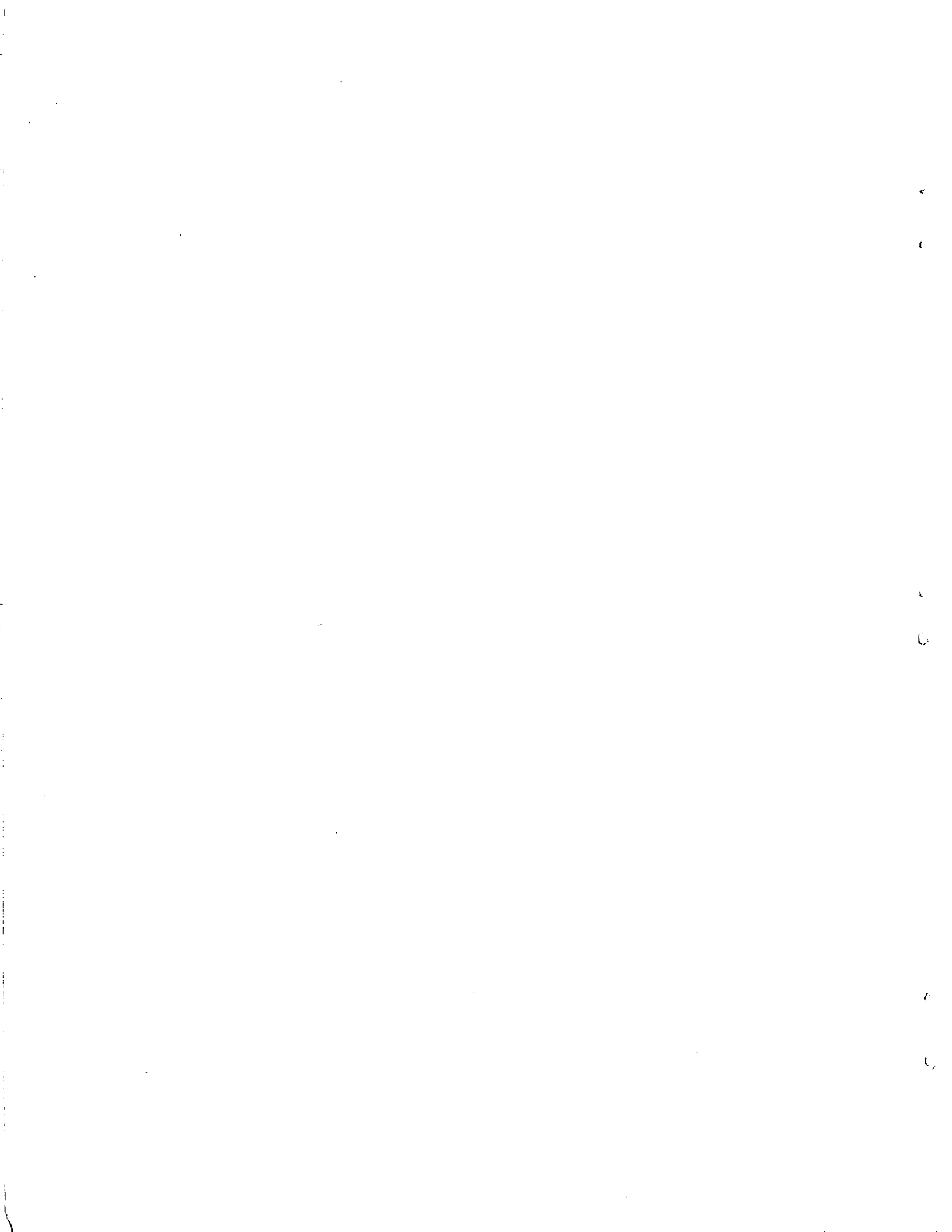












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